

**Irreversible Enzyme Inhibitors. CXXXIX.<sup>1,2</sup> *p*-(4,6-Diamino-1,2-dihydro-2,2-dimethyl-s-triazin-1-yl)phenylpropionylsulfanyl Fluoride, an Active-Site-Directed Irreversible Inhibitor of Dihydrofolic Reductase. VI.<sup>3</sup> Further Studies on Effects of Substitution on the Propionamide Bridge on Isozyme Specificity**

B. R. BAKER AND GERHARDUS J. LOURENS<sup>4</sup>

*Department of Chemistry, University of California at Santa Barbara, Santa Barbara, California 93106*

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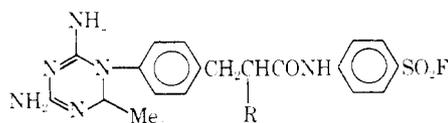
Thirteen candidate irreversible inhibitors of dihydrofolic reductase that have an alkyl, aryl, or aralkyl group on the propionamide bridge of the title compound (**1**) have been investigated with this enzyme from Walker 256 rat tumor, rat liver, three strains of L1210 mouse leukemia, and mouse liver. Substitution of an  $\alpha$ -methyl (**6**),  $\alpha$ -phenethyl (**7**), or N-methyl (**9**) eradicated the irreversible inhibition seen with **1**. Irreversible inhibition could be maintained with an  $\alpha$ -aryl substituent such as phenyl, tolyl, anisyl, or naphthyl, but such a substitution was detrimental to the good reversible binding of **1** needed for irreversible inhibition at low concentration. The best compound emerging from the study was the  $\beta$ -methyl derivative (**8**), which showed specificity by inactivation of L1210 enzyme but not liver; however, **8** did not give complete irreversible inhibition of the L1210 enzyme due to the competition of enzyme-catalyzed hydrolysis of its SO<sub>2</sub>F group.

In an earlier study on substitution<sup>5</sup> on the propionamide bridge of **1**,<sup>6</sup> it was observed that an  $\alpha$ -phenyl substituent (**2**) gave an irreversible inhibitor that was more effective on rat liver dihydrofolic reductase than the enzyme from Walker 256 rat tumor. Furthermore, an *o*-tolyl (**3**) or *m*-tolyl (**4**) substituent gave an irreversible inhibitor more effective on the rat tumor enzyme than the rat liver enzyme; however, the separation of this irreversible inhibition on the enzymes from the two sources was considered insufficient to be chemotherapeutically useful. With the *p*-tolyl substituent, irreversible inhibition was abolished. Com-

tion<sup>5</sup> of tumor *vs.* liver enzyme could be achieved. The results are the subject of this paper.

**Enzyme Results.**—The previous data<sup>5,6</sup> with **1** and **2** on the dihydrofolic reductase from Walker 256, rat liver, L1210/FR8, and mouse liver are summarized in Table I for comparison purposes. The  $\alpha$ -phenyl derivative (**2**) was near equally effective as an irreversible inhibitor of the dihydrofolic reductase from the three L1210 strains; however, **2** was not a good enough irreversible inhibitor<sup>9</sup> of the L1210 enzymes to warrant assay on the mouse liver enzyme. Introduction of the *o*-methyl group (**3**) on **2** enhanced irreversible inhibition of the enzyme from the L1210 strains; irreversible inhibition of the liver enzyme was appreciably less. Similar results were observed with the *m*-tolyl derivative (**4**); furthermore, even though **4** was not a good irreversible inhibitor<sup>9</sup> of the enzyme from L1210/DF8 and L1210/FR8, little irreversible inhibition of the mouse liver enzyme was observed with **4**. The *p*-tolyl derivative (**5**) showed no appreciable irreversible inhibition of the L1210 enzymes.

The alkyl and aralkyl derivatives (**6–9**) were then investigated as irreversible inhibitors of the dihydrofolic reductase from the various strains of L1210; the  $\alpha$ -methyl (**6**),  $\alpha$ -phenethyl (**7**), and N-methyl (**9**) derivatives failed to inactivate the enzyme. The  $\beta$ -methyl derivative (**8**) was previously reported<sup>6</sup> to be a fair irreversible inhibitor of the enzyme from Walker 256 and L1210/FR8; **8** has now been observed to be an irreversible inhibitor of the L1210/DF8 and L1210/0 enzymes. Furthermore, **8** showed no irreversible



- |                                      |  |
|--------------------------------------|--|
| 1, R = H                             | 3, R = <i>o</i> -C <sub>6</sub> H <sub>4</sub> CH <sub>3</sub> |
| 2, R = C <sub>6</sub> H <sub>5</sub> | 4, R = <i>m</i> -C <sub>6</sub> H <sub>4</sub> CH <sub>3</sub> |
|                                      | 5, R = <i>p</i> -C <sub>6</sub> H <sub>4</sub> CH <sub>3</sub> |

pounds **2–5** were only poor irreversible inhibitors of L1210/FR8 mouse leukemia enzyme. In more recent studies,<sup>7</sup> the dihydrofolic reductase from the parent L1210/0 strain and a different mutant, L1210/DF8, have been investigated. Therefore, compounds **2–5**, as well as the compounds with other substituents on the propionamide bridge described earlier,<sup>5</sup> have now been assayed on the enzyme from L1210/0 and L1210/DF8; furthermore, some additional  $\alpha$ -aryl analogs have now been synthesized to determine if a greater separation of active-site-directed irreversible inhibi-

(8) B. R. Baker, "Design of Active-Site-Directed Irreversible Enzyme Inhibitors. The Organic Chemistry of the Enzymic Active-Site," John Wiley and Sons, Inc., New York, N. Y., 1967.

(9) A good irreversible inhibitor is defined as one that at an  $I_{50} > 6K_1$  concentration gives greater than 70% inactivation of an enzyme; an excellent irreversible inhibitor gives greater than 70% inactivation at  $I_{50}/6 = K_1$ ; a fair irreversible inhibitor shows >70% inactivation at  $30K_1 \approx 5I_{50}$ . The dependence of the extent of irreversible inhibition on concentration of SO<sub>2</sub>F-type irreversible inhibitors is due to the competition between enzyme inactivation and enzyme-catalyzed hydrolysis of the SO<sub>2</sub>F group within the enzyme-inhibitor complex;<sup>10</sup> the ratio of these two rates is sensitive to structural change.<sup>3,5,11</sup>

(10) (a) B. R. Baker and J. A. Hurlbut, *J. Med. Chem.*, **11**, 233 (1968), paper CXIII of this series; (b) B. R. Baker and E. H. Erickson, *ibid.*, **11**, 245 (1968), paper CXV of this series.

(11) B. R. Baker and G. J. Lourens, *ibid.*, **11**, 666 (1968), paper CXXVII of this series.

(1) This work was generously supported by Grant CA-08695 from the National Cancer Institute, U. S. Public Health Service.

(2) For the previous paper in this series see B. R. Baker and N. M. J. Vermeulen, *J. Med. Chem.*, **12**, 89 (1969).

(3) For the previous paper in this subseries see B. R. Baker and G. J. Lourens, *ibid.*, **11**, 677 (1968), paper CXXIX of this series.

(4) G. J. L. wishes to thank the Council for Scientific and Industrial Research, Republic of South Africa, for a tuition fellowship.

(5) B. R. Baker and G. J. Lourens, *J. Med. Chem.*, **11**, 672 (1968), paper CXXVIII of this series.

(6) B. R. Baker and G. J. Lourens, *ibid.*, **10**, 1113 (1967), paper CV of this series.

(7) B. R. Baker, G. J. Lourens, R. B. Meyer, Jr., and N. M. J. Vermeulen, *ibid.*, **12**, 67 (1969), paper CXXXIII of this series.

TABLE I  
 INHIBITION<sup>a</sup> OF DIHYDROFOLIC REDUCTASE BY

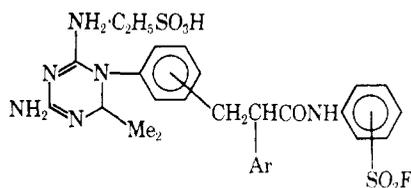
No.	R	Enzyme source	Reversible <sup>b</sup>		Irreversible <sup>c</sup>			
			I <sub>50</sub> , <sup>d</sup> μM	K <sub>i</sub> , <sup>e</sup> μM	Inhib, μM	% EI <sup>f</sup>	Time, min	% inactv <sup>g</sup>
1	H	Walker 256 <sup>g</sup>	0.020	0.003	0.020	87	1, 3	50, 90 <sup>h</sup>
		Rat liver <sup>g</sup>	0.006	0.001	0.020	95	8 <sup>i</sup>	50 <sup>h</sup>
		L1210/FR8 <sup>g</sup>	0.080	0.01	0.050	98	<2, 60	70, 70 <sup>h</sup>
		L1210/DF8 <sup>j</sup>	0.025	0.004	0.070	84	<2, 10	84, 84 <sup>h</sup>
		L1210/0 <sup>j</sup>	0.012	0.002	0.070	94	60	73 <sup>h</sup>
		Mouse liver <sup>g</sup>			0.070	98	60	53 <sup>h</sup>
					0.070		2, 60	38, 38 <sup>h</sup>
2	α-C <sub>6</sub> H <sub>5</sub>	Walker 256 <sup>g</sup>	0.074	0.01	0.074	87	4, 9, 60	50, 71, 71 <sup>h</sup>
		Rat liver <sup>g</sup>	0.18	0.03	0.074	84	2, 8, 30	50, 88, 94 <sup>h</sup>
		L1210/FR8	0.20	0.03	1.2	96	<2, 60	65, 65 <sup>h</sup>
					0.16	86	60	27
		L1210/DF8	0.045	0.007	0.48	99	60	34 <sup>h</sup>
3	α-( <i>o</i> -C <sub>6</sub> H <sub>4</sub> CH <sub>3</sub> )	L1210/0			0.48		60	47
		L1210/DF8	0.35 <sup>n</sup>		0.70		60	68 <sup>h</sup>
		L1210/0			0.70		60	73 <sup>h</sup>
4	α-( <i>m</i> -C <sub>6</sub> H <sub>4</sub> CH <sub>3</sub> )	Mouse liver			0.70		60	19 <sup>h</sup>
		L1210/DF8			0.42		60	65 <sup>h</sup>
		L1210/0			0.42		60	52 <sup>h</sup>
5	α-( <i>p</i> -C <sub>6</sub> H <sub>4</sub> CH <sub>3</sub> )	Mouse liver	0.20	0.03	0.42	92	60	20 <sup>h</sup>
		L1210/FR8	0.63	0.1	3.1	97	60	13
6	α-Me	L1210/0			1.2		60	0
		L1210/0	0.050 <sup>n</sup>		0.10		60	0
		L1210/DF8			0.10		60	0
7	α-(CH <sub>2</sub> ) <sub>2</sub> C <sub>6</sub> H <sub>5</sub>	L1210/0	1.9 <sup>n</sup>		3.8		60	0
		L1210/DF8			3.8		60	0
8	β-Me	L1210/DF8 <sup>j</sup>			0.16		60	40 <sup>h</sup>
		L1210/0 <sup>j</sup>	0.044	0.007	0.16	97	60	25 <sup>h</sup>
		Mouse liver <sup>j</sup>	0.080	0.01	0.40	97	60	0 <sup>h</sup>
		Mouse spleen <sup>j</sup>			0.16		60	38 <sup>h</sup>
9	N-Me	L1210/0	0.041 <sup>n</sup>		0.082		60	0
10	α-C <sub>6</sub> H <sub>5</sub> <sup>i</sup>	Walker 256	0.54	0.09	2.7	97	60	0
		L1210/FR8	0.48	0.06	2.4	97	60	0
		L1210/0			0.96		60	0
11	α-C <sub>6</sub> H <sub>5</sub> <sup>m</sup>	Walker 256			0.65		60	10 <sup>h</sup>
		L1210/DF8	0.13	0.02	0.65	97	60	0
		L1210/0			0.65		60	10 <sup>h</sup>
12	H <sup>m</sup>	Rat liver	0.012	0.002	0.060	96	2, 20	72, 71
		L1210/0			0.10		60	53 <sup>h</sup>
		L1210/DF8	0.018	0.003	0.10	98	60	49 <sup>h</sup>
		Mouse liver			0.20		60	26 <sup>h</sup>
13	α-( <i>o</i> -C <sub>6</sub> H <sub>4</sub> OCH <sub>3</sub> )	Walker 256			1.0		60	83
		Rat liver			1.0		60	75
		L1210/DF8	0.47		2.4	97	60	76 <sup>h</sup>
		L1210/0			2.4		60	57 <sup>h</sup>
14	α-( <i>m</i> -C <sub>6</sub> H <sub>4</sub> OCH <sub>3</sub> )	Walker 256			0.70		60	73
		Rat liver			0.70		60	57
		L1210/DF8	0.35		1.8	97	60	71 <sup>h</sup>
15	α-(α-Naphthyl)	Walker 256			1.1		60	78
		Rat liver			1.1		60	54
		L1210/DF8	0.57		2.8		60	61 <sup>h</sup>

<sup>a</sup> The technical assistance of Sharon Lafler, Diane Shea, and Carolyn Wade with these assays is acknowledged. <sup>b</sup> Assayed with 6 μM dihydrofolate and 30 μM TPNH in pH 7.4 Tris buffer containing 0.15 M KCl as previously described.<sup>7</sup> <sup>c</sup> Incubated at 37° in pH 7.4 Tris buffer in the presence of 60 μM TPNH as previously described.<sup>7</sup> <sup>d</sup> I<sub>50</sub> = concentration for 50% inhibition. <sup>e</sup> Estimated from  $K_i = K_m[I_{50}]/[S]$  which is valid since  $[S] = 6K_m = 6\mu\text{M}$  dihydrofolate; see ref 8, p 202. <sup>f</sup> Calculated from  $[EI] = [E_t]/(1 + K_i/[I])$  where  $[EI]$  is the amount of the total enzyme ( $E_t$ ) reversibly complexed; see ref 8, Chapter 8. <sup>g</sup> Data from ref 5. <sup>h</sup> From a time-study plot; see ref 6. <sup>i</sup> Half-time of reaction. <sup>j</sup> Data from ref 7. <sup>k</sup> Zero point obtained by adding inhibitor to assay cuvette prior to addition of enzyme aliquot.<sup>6,7</sup> <sup>l</sup> Propionamide bridge connected *meta* to 1-phenyl-*s*-triazine moiety. <sup>m</sup> SO<sub>2</sub>F connected *meta* to propionamide bridge. <sup>n</sup> I<sub>50</sub> on L1210/FR8 enzyme.<sup>5</sup>

inhibition of the mouse liver enzyme at a concentration of 5I<sub>50</sub>.

From this evaluation of 2-9 on the dihydrofolic reductases from L1210/0 emerged two studies worthy

of pursuit: (1) further structural variations of the β-methyl derivative (8) should be pursued, and (2) further α-aryl derivatives related to 2-5 should be synthesized to see if better and more selective ir-

TABLE II  
 PHYSICAL PROPERTIES OF


No. <sup>a</sup>	Bridge position	SO <sub>2</sub> F position	Ar	% yield	Mp, °C dec	Formula <sup>d</sup>
10	<i>meta</i>	<i>para</i>	C <sub>6</sub> H <sub>5</sub>	32 <sup>b</sup>	235-236	C <sub>26</sub> H <sub>27</sub> FN <sub>6</sub> O <sub>3</sub> S · C <sub>2</sub> H <sub>5</sub> SO <sub>3</sub> H
11	<i>para</i>	<i>meta</i>	C <sub>6</sub> H <sub>5</sub>	47 <sup>c</sup>	215-216	C <sub>26</sub> H <sub>27</sub> FN <sub>6</sub> O <sub>3</sub> S · C <sub>2</sub> H <sub>5</sub> SO <sub>3</sub> H
13	<i>para</i>	<i>para</i>	<i>o</i> -C <sub>6</sub> H <sub>4</sub> OMe	60 <sup>b</sup>	208-209	C <sub>27</sub> H <sub>29</sub> FN <sub>6</sub> O <sub>4</sub> S · C <sub>2</sub> H <sub>5</sub> SO <sub>3</sub> H
14	<i>para</i>	<i>para</i>	<i>m</i> -C <sub>6</sub> H <sub>4</sub> OMe	43 <sup>b</sup>	230-231	C <sub>27</sub> H <sub>29</sub> FN <sub>6</sub> O <sub>4</sub> S · C <sub>2</sub> H <sub>5</sub> SO <sub>3</sub> H
15	<i>para</i>	<i>para</i>	$\alpha$ -Naphthyl	45 <sup>c</sup>	229-231	C <sub>30</sub> H <sub>29</sub> FN <sub>6</sub> O <sub>3</sub> S · C <sub>2</sub> H <sub>5</sub> SO <sub>3</sub> H

<sup>a</sup> Compounds were prepared by method B previously described;<sup>5</sup> each had an ir band at 1395-1405 cm<sup>-1</sup> characteristic of SO<sub>2</sub>F. <sup>b</sup> Recrystallized from *i*-PrOH-H<sub>2</sub>O. <sup>c</sup> Recrystallized from EtOH-H<sub>2</sub>O. <sup>d</sup> All compounds gave correct analyses for C, H, F.

reversible inhibition of tumor enzymes could be achieved; such a study is reported below.

When the propionamide side chain of **2** was moved to the *meta* position of the 1-phenyl group as in **10**, irreversible inhibition of the enzyme from Walker 256, L1210/FR8, or L1210/0 was lost. When the sulfonyl fluoride group of **1** was moved to the *meta* position, the resultant **12** was a fair irreversible inhibitor of the enzyme from Walker 256;<sup>12</sup> **12** has now been observed to be a fair irreversible inhibitor<sup>9</sup> of the rat liver enzyme and a fair to poor irreversible inhibitor of the dihydrofolate reductase from L1210/0, L1210/DFS, and mouse liver, being least effective on the latter enzyme. Insertion of the  $\alpha$ -phenyl group (**11**) on **12** eradicated the irreversible inhibition of the L1210 and Walker 256 enzymes.

Three additional  $\alpha$ -aryl derivatives (**13-15**) were then synthesized for evaluation. The *o*-anisyl derivative (**13**) was a fair irreversible inhibitor<sup>9</sup> of both the Walker 256 rat tumor and rat liver enzymes, but a fair to poor irreversible inhibitor of the L1210/DFS and L1210/0 enzymes; the *m*-anisyl (**14**) and  $\alpha$ -naphthyl (**15**) derivatives gave similar results with the two rat enzymes and the L1210 enzyme.

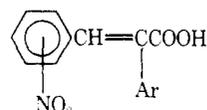
From this study it can be concluded that the  $\alpha$ -aryl derivatives of **1** are not good candidates for animal studies; irreversible inhibition is poor to fair and relatively nonselective and secondly, this type of substitution is detrimental to reversible inhibition. Whether or not more effective compounds derived from the  $\beta$ -methyl derivative (**8**) can be achieved remains to be determined.

### Experimental Section<sup>13</sup>

The synthesis of a number of the compounds in Table I have been previously described as indicated. The remaining compounds (**10**, **11**, **13-15**) were synthesized by proper modification

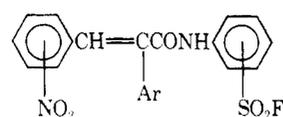
(12) B. R. Baker and G. J. Lourens, *J. Med. Chem.*, **11**, 39 (1968), paper CXII of this series.

of the routes described earlier.<sup>5,8,12</sup> The physical properties of **10-24** are given in Tables II-IV.

 TABLE III  
 PHYSICAL PROPERTIES OF


No. <sup>a</sup>	NO <sub>2</sub> position	Ar	% yield	Mp, °C dec	Formula <sup>d</sup>
16	<i>meta</i>	C <sub>6</sub> H <sub>5</sub>	37	178-181 <sup>b</sup>	
17	<i>para</i>	<i>o</i> -C <sub>6</sub> H <sub>4</sub> OMe	47	200-202	C <sub>16</sub> H <sub>13</sub> NO <sub>5</sub>
18	<i>para</i>	<i>m</i> -C <sub>6</sub> H <sub>4</sub> OMe	35	220-222	C <sub>16</sub> H <sub>13</sub> NO <sub>5</sub>
19	<i>para</i>	$\alpha$ -Naphthyl	54	>215 <sup>c</sup>	C <sub>19</sub> H <sub>13</sub> NO <sub>4</sub>

<sup>a</sup> Compounds were prepared by the method previously described for *p*-nitro- $\alpha$ -(*p*-tolyl)cinnamic acid<sup>5</sup> and recrystallized from EtOH. <sup>b</sup> T. R. Lewis, M. G. Pratt, E. D. Homiller, B. F. Tullar, and S. Archer, *J. Am. Chem. Soc.*, **71**, 3749 (1949), have recorded mp 183-185°. <sup>c</sup> Gradually decomposes over this temperature. <sup>d</sup> Compounds **17-19** gave correct analyses for C, H, N.

 TABLE IV  
 PHYSICAL PROPERTIES OF


No. <sup>a</sup>	NO <sub>2</sub> position	SO <sub>2</sub> F position	Ar	% yield	Mp, °C	Formula <sup>d</sup>
20	<i>meta</i>	<i>para</i>	C <sub>6</sub> H <sub>5</sub>	56 <sup>b</sup>	157-158	C <sub>31</sub> H <sub>13</sub> FN <sub>2</sub> O <sub>5</sub> S
21	<i>para</i>	<i>meta</i>	C <sub>6</sub> H <sub>5</sub>	58 <sup>b</sup>	141-142	C <sub>31</sub> H <sub>13</sub> FN <sub>2</sub> O <sub>5</sub> S
22	<i>para</i>	<i>para</i>	<i>o</i> -C <sub>6</sub> H <sub>4</sub> OMe	62 <sup>c</sup>	203-204	C <sub>32</sub> H <sub>17</sub> FN <sub>2</sub> O <sub>6</sub> S
23	<i>para</i>	<i>para</i>	<i>m</i> -C <sub>6</sub> H <sub>4</sub> OMe	66 <sup>b</sup>	204-205 dec	C <sub>32</sub> H <sub>17</sub> FN <sub>2</sub> O <sub>6</sub> S
24	<i>para</i>	<i>para</i>	$\alpha$ -Naphthyl	53 <sup>b</sup>	240-242	C <sub>35</sub> H <sub>17</sub> FN <sub>2</sub> O <sub>6</sub> S

<sup>a</sup> All compounds were prepared by method A previously described.<sup>5</sup> <sup>b</sup> Recrystallized from MeOEtOH-H<sub>2</sub>O. <sup>c</sup> Recrystallized from MeOEtOH. <sup>d</sup> All compounds gave correct analyses for C, H, N.

(13) Melting points were taken in capillary tubes on a Mel-Temp block and are uncorrected. Each analytical sample had ir and uv spectra compatible with their assigned structure; each gave combustion values for C, H, and N or F within 0.4% of theoretical.