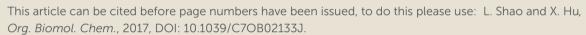
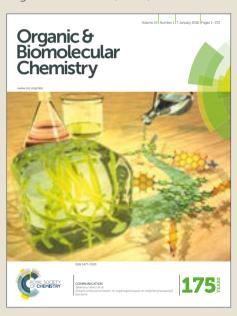
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Cu-catalyzed asymmetric Friedel-Crafts propargylic alkylation of phenol derivatives

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Long Shao, a,b and Xiang-Ping Hu*a

A copper-catalyzed asymmetric Friedel-Crafts propargylic alkylation of electron-rich phenol derivatives with a variety of propargylic esters has been described. With Cu(OTf)₂ decorated with a chiral tridentate ketimine P,N,N-ligand as the catalyst, asymmetric Friedel-Crafts propargylic alkylation of 3,5-dialkoxyphenol derivatives proceeded smoothly in high yields and with good to excellent enantioselectivities. The present study suggested that the presence of an electron-rich substituent on the meta-position of phenol is essential for the promotion of Friedel-Crafts propargylic alkylation, and the substrate bearing two electron-rich groups on both of 3,5-positions of phenol tends to give the satisfactory performance.

Introduction

Since van Maarseveen¹ and Nishibayashi² respectively reported the first Cu-catalyzed asymmetric propargylic amination in 2008, copper-catalyzed asymmetric propargylic transformation,3 featuring Cu-allenylidene complexes as the key intermediates, has attracted increasing attention due to its high potentials in the enantioselective formation of C-C and C-heteroatom bonds⁴ and the stereoselective construction of complex cyclic frameworks.5 In the past decade, many C-, N-, and Onucleophiles have proved to be suitable reaction partners for this important transformation. However, Cu-catalyzed asymmetric Friedel-Crafts propargylic alkylation with electronrich aromatic compounds as C-nucleophiles remains less successful although catalytic asymmetric Friedel-Crafts alkylation has made significant achievements in the past decades.⁶ To our knowledge, only one example has been reported by van Maarseveen recently, in which a Cu-catalyzed asymmetric Friedel-Crafts propargylic alkylation of indole has been described.7 However, Cu-catalyzed asymmetric Friedel-Crafts propargylic alkylation of phenol derivatives is still unexplored. There is therefore an urgent need for the development of Cu-catalyzed asymmetric Friedel-Crafts propargylic alkylation of phenol derivatives.

The challenge for the realization of Cu-catalyzed asymmetric Friedel-Crafts propargylic alkylation of phenols is obvious since phenols readily underwent the propargylic *O*-alkylation with propargylic esters in the presence of a copper catalyst as reported by Nishibayashi's group and us (Scheme 1a).8 Very recently, we disclosed a copper-catalyzed sequential Friedel-Crafts alkylation/intramolecular hydroalkoxylation process

b) Our work on the propargylic cycloaddition of electron-rich phenols:

$$R^{2}$$
 R^{2}
 R^{2

c) This work on the Friedel-Crafts propargylic alkylation of phenols:

Scheme 1 Cu-catalyzed asymmetric transformation between phenols and propargylic esters.

between electron-rich phenols and propargylic esters.⁹ This reaction suggested that the development of a copper-catalyzed Friedel-Crafts propargylic alkylation should be possible if the last hydroalkoxylation process can be efficiently interrupted (Scheme 1b). Undoubtedly, if a Friedel-Crafts propargylic alkylation is controlled to take place at the para-position of phenol, the intramolecular hydroalkoxylation would be completely inhibited. Indeed, this strategy has been successfully employed by us in the Cu-catalyzed asymmetric propargylic dearomatization of phenols very recently.¹⁰ As part of our ongoing interest in the development of Cu-catalyzed asymmetric propargylic transformation, we herein wished to

a) Nishibayashi's and our works on the propargylic etherification of phenols:

^a Dalian Institute of Chemical Physics, Chinese Academy of Sciences, 457 Zhongshan Road, Dalian 116023, China. E-mail: xiangping@dicp.ac.cn.

b. University of Chinese Academy of Sciences, Beijing 100049, China.

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report our detail studies on the copper-catalyzed asymmetric Friedel-Crafts propargylic alkylation of phenols.

Results and discussion

We started our investigation by screening a series of chiral ligands for the model reaction of 1-phenyl-2-propynyl acetate 1a with 3,5-dimethoxyphenol 2a in methanol at room temperature in the presence of Cu(OAc)2·H2O and Pr2NEt, and the results are summarized in Table 1. 3,5-Dimethoxyphenol 2a was selected as the standard substrate to examine the possibility of the Friedel-Crafts propargylic alkylation since we have recently disclosed that electron-rich phenols preferentially took the Friedel-Crafts-type reaction and the dearomatization reaction rather than the propargylic O-alkylation. Pleasingly, the ligand screening disclosed that the Friedel-Crafts propargylic alkylation took place smoothly by the use of chiral tridentate P,N,N-ligands developed within our group, in which the ketimine P,N,N-ligand (S)-L4 was identified as the most promising ligand in terms of yield and enantioselectivity (entries

Table 1 Optimization of the reaction conditions

(S,S)-Me-pybox (L_1)

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(S)-BINAP (L₂)

(S)-L₃: R = H (S)-L₄: R = Ph

Entry	[Cu]	L	Base	Т	Yield ^b	ee^c	
				(°C)	(%)	(%)	
1	Cu(OAc) ₂ ·H ₂ O	L ₁	[/] Pr ₂ NEt	rt	-	-	
2	Cu(OAc) ₂ ·H ₂ O	L_2	'Pr ₂ NEt	rt	16	12	
3	Cu(OAc) ₂ ·H ₂ O	L_3	$^{\prime}Pr_{2}NEt$	rt	51	62	
4	Cu(OAc) ₂ ·H ₂ O	L_4	'Pr ₂ NEt	rt	82	87	
5	Cu(OAc) ₂ ·H ₂ O	L_4	DBU	rt	63	86	
6	Cu(OAc) ₂ ·H ₂ O	L_4	NEt ₃	rt	55	87	
7	Cu(OAc) ₂ ·H ₂ O	L_4	K_2CO_3	rt	93	87	
8	Cu(OAc) ₂ ·H ₂ O	L_4	Cs_2CO_3	rt	66	84	
9	Cu(OAc) ₂ ·H ₂ O	L_4	-	rt	<10	-	
10	Cu(MeCN) ₄ BF ₄	L_4	K_2CO_3	rt	82	85	
11	CuCl	L_4	K_2CO_3	rt	87	87	
12	Cu(OTf)₂	L_4	K_2CO_3	rt	84	88	
13	Cu(OAc) ₂ ·H ₂ O	L_4	K_2CO_3	0	87	91	
14	CuCl	L_4	K_2CO_3	0	82	90	
15	Cu(OTf) ₂	L_4	K ₂ CO ₃	0	93	91	
16	Cu(OTf)₂	L ₄	K_2CO_3	-20	96	93	

^a Reaction condition: 1a (0.3 mmol), 2a (0.36 mmol), [Cu] (0.015 mmol, 5 mol%), L* (0.0165 mmol, 5.5 mol%), and base (0.36 mmol, 1.2 equiv) in 3 mL of MeOH at indicated reaction temperature for 12 h. b Yield of isolated product. ^c Determined by HPLC using a chiral stationary phase.

1-4). A brief base-additive screening revealed that K2CO3 was the best base-additive for this transformation, affording 3aa in 93% yield and with 87% ee (entries 4-8). The result also indicated that the base-additive was necessary for this transformation since very low conversion was observed in its absence (entry 9). Cu salts showed less influence on the reactivity and enantioselectivity. All of Cu salts tested gave the similar performance to that with Cu(OAc)2·H2O. Lowering the reaction temperature could further improve the reaction performance, in which Cu(OTf)2 displayed the best catalytic activity and enantioselectivity. In particular, when the reaction was performed at -20 °C, Friedel-Crafts propargylic alkylation product 3aa could be obtained in 96% yield and with 93% ee (entry 16).

With the optimized reaction conditions in hands, we proceeded to investigate the applicability of propargylic acetates 1 in the Friedel-Crafts propargylic alkylation of 3,5dimethoxyphenol 2a, and the results are summarized in Table 2. In all cases, only Friedel-Crafts propargylic alkylation products were observed. The resulted indicated that the reaction was sensitive to the substitution pattern on the phenyl ring. Thus, the reactions with 3-Cl or 4-Cl substituted propargylic acetates (1c and 1d) proceeded smoothly to give the desired products (3ca and 3da) in high yields and with excellent enantioselectivity (95% and 92% ee, respectively) (entries 2 and 3), while the substrate 1b with a 2-Cl substituent led to an obvious decrease in the enantioselectivity to 76% ee although good yield (86%) was maintained (entry 1). The electronic properties of the substituent at the para-position of the phenyl ring showed little influence on the reactivity and enantioselectivity, and all substrates 1d-i gave rise to the corresponding Friedel-Crafts

Table 2 Scope with respect to propargylic acetates

Entry	Substrate (R1)	Product (3)	Yield ^b (%)	ee ^c (%)
1	1b : $R^1 = 2 - CIC_6H_4$	3ba	86	76
2	1c: $R^1 = 3-CIC_6H_4$	3ca	89	95
3	1d : $R^1 = 4 - CIC_6H_4$	3da	88	92
4	1e : $R^1 = 4 - FC_6H_4$	3ea	90	90
5	1f : $R^1 = 4 - BrC_6H_4$	3fa	86	92
6	1g : $R^1 = 4$ -MeC ₆ H ₄	3ga	93	93
7	1h : $R^1 = 4$ -MeOC ₆ H ₄	3ha	86	87
8	1i: $R^1 = 4 - CF_3C_6H_4$	3ia	87	92
9	1j: R1 = 2-naphthyl	3ja	84	93
10	1k : R ¹ = 2-thienyl	3ka	84	91
11^d	1l : R ¹ = Bn	3la	39	72

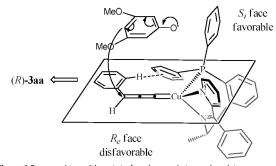
^a Reaction condition: 1 (0.3 mmol), 2a (0.36 mmol), Cu(OTf)₂ (0.015 mmol, 5 mol%), (S)-L4 (0.0165 mmol, 5.5 mol%), and K2CO3 (0.36 mmol, 1.2 equiv) in 3 mL of MeOH at -20 °C for 12 h. b Yield of isolated product, c Determined by HPLC using a chiral stationary phase. d The corresponding pentafluorobenzoate was used instead of acetate.

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Table 3 Scope with respect to phenol derivatives ^a

 a Reaction condition: **1a** (0.3 mmol), **2** (0.36 mmol), Cu(OTf)₂ (0.015 mmol, 5 mol%), (*S*)-L₄ (0.0165 mmol, 5.5 mol%), and K₂CO₃ (0.36 mmol, 1.2 equiv) in 3 mL of MeOH at -20 $^{\circ}$ C for 12 h. b Yield of isolated product. c Determined by HPLC using a chiral stationary phase.

Scheme 2 The substituent effect in the Cu-catalyzed asymmetric transformation between phenols and propargylic esters.



 $\textbf{Figure 1} \ \textbf{Proposed transition state for observed stereochemistry}.$

propargylic alkylation products **3da-ia** in good yields (86-93% yield) and with good enantioselectivities (87-93% ee) (entries 3-8). 2-Naphthyl-substituted substrate **1j** served as a suitable reaction partner, giving **3ja** in 84% yield and with 93% ee (entry 9). 2-Thienyl substituted heterocyclic substrate **1k** also worked well for the reaction, producing **3ka** in 83% yield and with 91% ee (entry 10). However, aliphatic substrate **1l** proved to be less suitable to the reaction, with which only low yield and moderate enantioselectivity were obtained (entry 11).

Table 3 shows the scope and limitation with regard to phenol derivatives that underwent the Friedel-Crafts propargylic alkylation. In general, the presence of the alkoxy group on both of 3,5-positions of phenols tended to give satisfactory yield and enantioselectivity, in which only Friedel-Crafts propargylic alkylation products were observed. However, the increased steric hindrance of alkoxy group significantly decreased the reactivity and enantioselectivity of Friedel-Crafts propargylic alkylation. Thus, 3,5-diisopropoxyphenol 2d led to the corresponding Friedel-Crafts propargylic alkylation product in 59% yield and with 68% ee. Different alkoxy groups on the 3,5-positions of phenols were well tolerated, and the corresponding Friedel-Crafts propargylic alkylation products (3ae-ag) were obtained in good yields (77-84%) and with good enantioselectivity (81-88% ee). However, when 3.5dimethoxyaniline was used as the substrate, no Friedel-Crafts propargylic alkylation product was detected.

To investigate the role of the substitutent on phenols, some control experiments were performed as shown in Scheme 2. With phenol as the substrate, only O-alkylation product 4 was observed. The introduction of a methoxy group into meta-position of phenol significantly inhibited the propargylic O-alkylation, promoting the Friedel-Crafts propargylic alkylation at the para-position and Friedel-Crafts alkylation/intramolecular hydroalkoxylation at the ortho-position. Further introduction of an electron-rich group into 5-position of 2i led to the Friedel-Crafts propargylic alkylation as the only observed reaction. Thus, 3-methoxy-5-methylphenol 2j gave the Friedel-Crafts propargylic alkylation product 3aj in 57% yield and with 70% ee. These results suggested that the presence of electron-rich substituents on the 3,5-positions of phenol should be necessary to efficiently promote the Friedel-Crafts propargylic alkylation of phenol. The absolute configuration of 3ai was determined as R-

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configuration by the derivatization and comparison to the known compound.11

Based on the experimental results and an edge-to-face aromatic interaction between a phenyl group of the substrate and a phenyl group of the ligand, a transition state of Cuacetylide complex with chiral P,N,N-ligand (S)-L4, is proposed to explain the observed stereochemistry as shown in Figure 1. Due to the steric hindrance of the ligand, the attack of the γ -carbon atom happened favourably from the Si face to form (R)-3aa while the Re face was hampered.

Conclusions

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In conclusion, we have realized an enantioselective coppercatalyzed Friedel-Crafts propargylic alkylation of phenol derivatives with propargylic esters. The research indicated that the presence of an electron-rich substituent on the meta-position of phenol is essential for the realization of Friedel-Crafts propargylic alkylation of phenol, and phenol derivatives bearing two electronrich groups on both of 3,5-positions tend to give the satisfactory performance. With Cu(OTf)₂ in combination with a structurally rigid chiral tridentate ketimine P,N,N-ligand as the catalyst, the Friedel-Crafts propargylic alkylation of 3,5-dialkoxyphenols proceeded smoothly, therefore giving rise to a variety of Friedel-Crafts propargylic alkylation products in good to high yields with a high to excellent enantioselectivity (up to 95% ee). To our knowledge, the present research represents the first successful example of Cu-catalyzed asymmetric Friedel-Crafts propargylic alkylation of phenol derivatives.

Experimental

General methods

Commercially available compounds were used without further purification. Solvents were purified by standard procedure before use. Flash chromatography was performed on silica gel 60 (40-63 μm, 60 Å). Thin layer chromatography (TLC) was performed on glass plates coated with silica gel 60 with F254 indicator. Proton nuclear magnetic resonance (1H NMR) spectra were recorded on a Bruker 400 MHz spectrometer. Chemical shifts for protons are reported in parts per million downfield from tetramethylsilane and are referenced to residual protium in the NMR solvent (CHCl₃ = δ 7.26 or DMSO = δ 2.50). Carbon nuclear magnetic resonance (13C NMR) spectra were recorded on a Bruker 100 MHz spectrometer. Chemical shifts for carbon reported in parts per million downfield from tetramethylsilane and are referenced to the carbon resonances of the solvent (CDCl₃ = δ 77.23 or DMSO = δ 39.60). Data are represented as follows: chemical shift, multiplicity (br = broad, s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet), coupling constants in Hertz (Hz), integration. Only the most important and relevant frequencies are reported. Enantiomeric ratios were determined by chiral HPLC with n-hexane and i-PrOH as solvents. IR was recorded on a Nicolet-is50 infrared

spectrometer. Optical rotations were recorded on a JASCO P-1020 polarimeter.

General procedure for Cu-catalyzed asymmetric Friedel-Crafts propargylic alkylation reaction

A solution of Cu(OTf)₂ (5.4 mg, 0.015 mmol) and (S)-L₄ (7.8 mg, 0.0165 mmol) in 1 mL of anhydrous methanol placed in an ovendried Schlenk flask was stirred at room temperature under a nitrogen atmosphere for 1 h. A solution of propargylic esters 1 (0.3 mmol), phenol derivatives 2 (0.36 mmol) and K₂CO₃ (49.8mg, 0.36 mmol) in 2 mL of anhydrous methanol was added. The mixture was stirred at -20 °C for 12 h. The reaction mixture was then concentrated under vaccum, and the residue was purified by silica gel chromatography to afford the desired products 3.

3,5-dimethoxy-4-[(R)-1-phenylprop-2-yn-1-yl]phenol Employing the general procedure afforded compound 3aa as a pale-yellow oil (77.2 mg, 96% yield). HPLC (Daicel Chiralpak AD-H, n-hexane/i-propanol = 80/20, flow rate 0.8 mL min⁻¹, detection at 230 nm): t_R = 8.7 min (major enantiomer), t_R = 10.0 min (minor enantiomer); 93% ee. $[\alpha]^{21}_D$ = +97.4 (c = 1.00, CH_2CI_2). ¹H NMR (400 MHz, DMSO-d₆) δ 9.53 (s, 1H), 7.32 (d, J =7.6 Hz, 2H), 7.24 (t, J = 7.3 Hz, 2H), 7.14 (t, J = 7.1 Hz, 1H), 6.09 (s, 2H), 5.48 (s, 1H), 3.65 (s, 6H), 2.99 (s, 1H); ¹³C NMR (101 MHz, DMSO-d₆) δ 141.4, 128.2, 127.4, 126.3, 108.3, 93.0, 85.0, 71.8, 56.0, 30.5. IR (KBr): 3428, 3280, 2937, 2845, 2112, 1597, 1473, 1116, 996, 814, 730, 633 cm⁻¹; HRMS (ESI): m/z calcd. for C17H17O3 [M + H] 269.1178, found 269.1174.

4-(1-(2-chlorophenyl)prop-2-yn-1-yl)-3,5-dimethoxyphenol

(3ba). Employing the general procedure afforded compound **3ba** as a pale-yellow solid (77.8 mg, 86% yield). M.p. 42-44 °C. HPLC (Daicel Chiralpak AD-H, n-hexane/i-propanol = 80/20, flow rate 0.8 mL min⁻¹, detection at 230 nm): t_R = 8.7 min (major enantiomer), t_R = 10.3 min (minor enantiomer); 76% ee. [α]²²_D = +99.3 (c = 1.00, CH₂Cl₂). ¹H NMR (400 MHz, CDCl₃) δ 7.95 (d, J= 7.9 Hz, 1H), 7.37-7.20 (m, 2H), 7.12 (t, J = 7.5 Hz, 1H), 6.01 (s, 2H), 5.80 (d, J = 2.4 Hz, 1H), 5.44 (s, 1H), 3.63 (s, 6H), 2.28 (d, J =2.3 Hz, 1H); 13 C NMR (101 MHz, CDCl₃) δ 158.8, 156.5, 137.8, 133.0, 131.5, 129.2, 127.6, 125.7, 108.8, 93.0, 84.3, 69.7, 55.8, 29.8. IR (KBr): 3421, 3302, 2930, 2926, 2113, 1600, 1472, 1149, 1120, 995, 800, 757, 640 cm⁻¹; HRMS (ESI): m/z calcd. for C17H16O3Cl [M + H] 303.0788, found 303.0780.

4-(1-(3-chlorophenyl)prop-2-yn-1-yl)-3,5-dimethoxyphenol

(3ca). Employing the general procedure afforded compound 3ca as a pale-yellow solid (81.1 mg, 89% yield). M.p. 52-54 °C. HPLC (Daicel Chiralpak AD-H, n-hexane/i-propanol = 80/20, flow rate 0.8 mL min⁻¹, detection at 230 nm): t_R = 7.6 min (major enantiomer), t_R = 10.5 min (minor enantiomer); 95% ee. [α]²²_D = +109.7 (c = 1.00, CH₂Cl₂). ¹H NMR (400 MHz, CDCl₃) δ 7.41 (s, 1H), 7.28 (d, J = 6.9 Hz, 1H), 7.23-6.94 (m, 2H), 6.02 (s, 2H), 5.62(d, J = 2.3 Hz, 1H), 5.57 (s, 1H), 3.64 (s, 6H), 2.29 (d, J = 2.7 Hz,1H); 13 C NMR (101 MHz, CDCl₃) δ 158.5, 156.6, 143.0, 133.6, 129.1, 127.4, 126.2, 125.5, 109.8, 93.0, 84.1, 69.8, 55.9, 30.3. IR (KBr): 3518, 3288, 2972, 2112, 1594, 1470, 1227, 1097, 720, 664, 620 cm⁻¹; HRMS (ESI): m/z calcd. for C17H16O3Cl [M + H] 303.0788, found 303.0779.

4-(1-(4-chlorophenyl)prop-2-yn-1-yl)-3,5-dimethoxyphenol (3da). Employing the general procedure afforded compound

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3da as a colourless oil (79.9 mg, 88% yield). HPLC (Daicel Chiralpak AD-H, n-hexane/i-propanol = 85/15, flow rate 0.8 mL min^{-1} , detection at 230 nm): $t_R = 13.8 min$ (major enantiomer), $t_{\rm R}$ = 15.2 min (minor enantiomer); 92% ee. [α]²⁴_D = +115.6 (c = 1.00, CH₂Cl₂). ¹H NMR (400 MHz, CDCl₃) δ 7.35 (d, J = 8.2 Hz, 2H), 7.19 (d, J = 8.5 Hz, 2H), 6.04 (s, 2H), 5.62 (d, J = 2.5 Hz, 1H), 5.32 (s, 1H), 3.68 (s, 6H), 2.27 (d, J = 2.7 Hz, 1H); ¹³C NMR (101 MHz, $CDCl_3$) δ 158.5, 156.5, 139.5, 131.6, 128.6, 127.8, 110.1, 92.8, 84.3, 69.4, 55.9, 30.1. IR (KBr): 3412, 3294, 2931, 2115, 1602, 1475, 1431, 1216, 1122, 995, 780, 630 cm⁻¹; HRMS (ESI): m/z calcd. for C17H16O3Cl [M + H] 303.0788, found 303.0783.

4-(1-(4-fluorophenyl)prop-2-yn-1-yl)-3,5-dimethoxyphenol

(3ea). Employing the general procedure afforded compound **3ea** as a pale-yellow solid (77.5 mg, 90% yield). M.p. 76-78 °C. HPLC (Daicel Chiralpak AD-H, n-hexane/i-propanol = 90/10, flow rate 0.8 mL min⁻¹, detection at 230 nm): t_R = 20.1 min (major enantiomer), t_R = 21.8 min (minor enantiomer); 90% ee. $[\alpha]^{24}$ _D = +120.9 (c = 1.00, CH₂Cl₂). ¹H NMR (400 MHz, DMSO-d₆) δ 9.54 (s, 1H), 7.34–7.31 (m, 2H), 7.09–7.04 (m, 2H), 6.09 (s, 2H), 5.46 (s, 1H), 3.66 (s, 6H), 3.02-3.01 (m, 1H); ¹³C NMR (101 MHz, DMSO-d₆) δ 161.0 (d, J = 241.7 Hz), 158.8, 158.4, 137.5 (d, J = 2.9 Hz), 129.1 (d, J = 8.1 Hz), 115.0, 114.8, 108.1, 93.0, 84.8, 72.1, 56.0, 29.9. IR (MeOH): 3360, 3280, 2950, 1600, 1510, 1470, 1220, 1120, 1020, 778, 630, 562 cm⁻¹; HRMS (ESI): m/z calcd. for C17H16O3F [M + H] 287.1083, found 287.1079.

4-(1-(4-bromophenyl)prop-2-ynyl)-3,5-dimethoxyphenol (3fa). Employing the general procedure afforded compound 3fa as a pale-yellow oil (89.5 mg, 86% yield). HPLC (Daicel Chiralpak AD-H, n-hexane/i-propanol = 85/15, flow rate 0.8 mL min⁻¹, detection at 230 nm): $t_R = 14.7$ min (major enantiomer), $t_R = 16.6$ min (minor enantiomer); 92% ee. $[\alpha]^{22}D$ = +80.7 (c = 1.00, CH_2Cl_2). ¹H NMR (400 MHz, CDCl₃) δ 7.35 (d, J = 8.5 Hz, 2H), 7.29 (d, J = 8.4 Hz, 2H), 6.04 (s, 2H), 5.60 (d, J = 2.5 Hz, 1H), 5.30 (s, 2H)1H), 3.67 (s, 6H), 2.27 (d, J = 2.7 Hz, 1H); ¹³C NMR (101 MHz, CDCl₃) δ 158.5, 156.6, 140.0, 130.8, 129.1, 119.7, 110.0, 92.8, 84.2, 69.5, 55.9, 30.2. IR (KBr): 3419, 2966, 2950, 2936, 2113, 1600, 1486, 1430, 1344, 777, 633 cm⁻¹; HRMS (ESI): m/z calcd. for C17H16O3Br [M + H] 347.0283, found 347.0273.

3,5-dimethoxy-4-(1-(p-tolyl)prop-2-yn-1-yl)phenol Employing the general procedure afforded compound 3ga as a pale-yellow oil (78.9 mg, 93% yield). HPLC (Daicel Chiralpak AD-H, n-hexane/i-propanol = 90/10, flow rate 0.8 mL min⁻¹, detection at 230 nm): t_R = 17.4 min (major enantiomer), t_R = 18.9 min (minor enantiomer); 93% ee. $[\alpha]^{24}D = +93.7$ (c = 1.00, CH_2Cl_2). ¹H NMR (400 MHz, CDCl₃) δ 7.30 (d, J = 7.9 Hz, 2H), 7.04 (d, J = 7.9 Hz, 2H), 6.00 (s, 2H), 5.63 (s, 1H), 5.47 (s, 1H), 3.64 (s, 2H), 5.63 (s, 2H), 5.63 (s, 2H), 5.47 (s, 2H), 3.64 (s, 2H), 5.63 (s, 2H), 5.63 (s, 2H), 5.47 (s, 2H), 3.64 (s, 2H), 5.63 (s, 2H), 5.63 (s, 2H), 5.47 (s, 2H), 3.64 (s, 2H), 5.63 (s, 2H), 5.63 (s, 2H), 5.47 (s, 2H), 3.64 (s, 2H), 5.63 (s, 2H), 5.66H), 2.28 (s, 3H), 2.24 (d, J = 2.2 Hz, 1H); ¹³C NMR (101 MHz, $CDCl_3$) δ 158.6, 156.3, 137.8, 135.5, 128.6, 127.1, 110.7, 93.0, 85.2, 68.9, 56.0, 30.2, 21.0. IR (MeOH): 3275, 3029, 2836, 2116, 1595, 1505, 1452, 1292, 1194, 1033, 952, 831, 699 cm⁻¹; HRMS (ESI): m/z calcd. for C18H19O3 [M + H] 283.1134, found 283.1132.

3,5-dimethoxy-4-(1-(4-methoxyphenyl)prop-2-yn-1-yl)phenol (3ha). Employing the general procedure afforded compound **3ha** as a pale-yellow solid (77.4 mg, 86% yield). M.p. 124–126 °C. HPLC (Daicel Chiralpak AD-H, n-hexane/i-propanol = 80/20, flow rate 0.8 mL min⁻¹, detection at 230 nm): t_R = 20.7 min

(minor enantiomer), $t_R = 26.3$ min (major enantiomer); 87% ee. $[\alpha]^{24}$ _D = +104.7 (c = 1.00, CH₂Cl₂). ¹H NMR (400 MHz, DMSO-d₆) δ 9.50 (s, 1H), 7.24 (d, J = 8.3 Hz, 2H), 6.81 (d, J = 8.5 Hz, 2H), 6.09 (s, 2H), 5.43 (s, 1H), 3.68 (d, J = 12.1 Hz, 9H), 2.94 (d, J = 1.7)Hz, 1H); 13 C NMR (101 MHz, DMSO-d₆) δ 158.6, 158.4, 157.9, 133.3, 128.4, 113.6, 108.6, 93.0, 85.4, 71.5, 56.0, 55.4, 29.7. IR (KBr): 3354, 3281, 2932, 2116, 1596, 1511, 1475, 1239, 1118, 1000, 630 cm⁻¹; HRMS (ESI): m/z calcd. for C18H19O4 [M + H] 299.1283, found 299.1280.

3,5-dimethoxy-4-(1-(4-(trifluoromethyl)phenyl)prop-2-yn-1yl)phenol (3ia). Employing the general procedure afforded compound 3ia as a colourless oil (88.2 mg, 87% yield). HPLC (Chiralcel OD-H, n-hexane/i-propanol = 90/10, flow rate 0.8 mL min^{-1} , detection at 230 nm): $t_R = 12.9 min$ (major enantiomer), t_R = 14.2 min (minor enantiomer); 92% ee. [α]²⁴_D = +91.3 (c = 1.00, CH_2CI_2). ¹H NMR (400 MHz, DMSO-d₆) δ 9.59 (s, 1H), 7.62 (d, J = 8.3 Hz, 2H), 7.52 (d, J = 8.2 Hz, 2H), 6.10 (s, 2H), 5.56 (s, 2H)1H), 3.66 (s, 6H), 3.10 (d, J = 2.6 Hz, 1H); ¹³C NMR (101 MHz, DMSO-d₆) δ 158.8(d, J = 70.2 Hz), 146.2, 128.1, 127.1 (d, J = 31.6 Hz), 126.2, 125.2 (d, J = 3.9 Hz), 123.5, 107.5, 93.0, 84.0, 72.7, 56.0, 30.5. IR (MeOH): 3360, 3280, 2830, 1600, 1470, 1320, 1220, 1110, 1020, 630 cm⁻¹; HRMS (ESI): m/z calcd. for C18H16O3F3 [M + H] 347.0283, found 347.0273.

3,5-dimethoxy-4-(1-(naphthalen-2-yl)prop-2-yn-1-yl)phenol (3ja). Employing the general procedure afforded compound 3ja as a colourless oil (80.1 mg, 84% yield). HPLC (Daicel Chiralpak AD-H, n-hexane/i-propanol = 80/20, flow rate 0.8 mL min⁻¹, detection at 230 nm): t_R = 10.3 min (major enantiomer), t_R = 13.5 min (minor enantiomer); 93% ee. $[\alpha]^{21}_D$ = +187.9 (c = 1.00, CH_2Cl_2). ¹H NMR (400 MHz, CDCl₃) δ 7.94 (s, 1H), 7.87–7.59 (m, 3H), 7.51-7.26 (m, 3H), 5.96 (s, 2H), 5.81 (d, J = 1.5 Hz, 1H), 5.59(s, 1H), 3.56 (d, J = 3.8 Hz, 6H), 2.34 (d, J = 2.6 Hz, 1H); ¹³C NMR (101 MHz, CDCl₃) δ 158.7, 156.5, 138.3, 133.3, 132.1, 127.9, 127.5, 127.3, 126.0, 125.7, 125.4, 125.2, 110.4, 92.9, 84.8, 69.4, 56.0, 30.7. IR (MeOH): 3340, 3290, 2950, 2940, 1600, 1480, 1220, 1110, 1020, 998, 816, 633 cm⁻¹; HRMS (ESI): m/z calcd. for C21H19O3 [M + H] 319.1334, found 319.1311.

3,5-dimethoxy-4-(1-(thiophen-2-yl)prop-2-ynyl)phenol (3ka). Employing the general procedure afforded compound 3ka as a colourless oil (69.1 mg, 84% yield). HPLC (Chiralcel OJ-H, nhexane/i-propanol = 80/20, flow rate 0.8 mL min⁻¹, detection at 230 nm): t_R = 9.7 min (minor enantiomer), t_R = 12.3 min (major enantiomer); 91% ee. [α]²¹D = +52.8 (c = 1.00, CH₂Cl₂). ¹H NMR (400 MHz, CDCl₃) δ 6.98 (d, J = 4.6 Hz, 1H), 6.87 (d, J = 3.4 Hz, 1H), 6.77 (dd, J = 5.0, 3.6 Hz, 1H), 5.93 (s, 2H), 5.74 (d, J = 1.7 Hz, 1H), 5.47 (s, 1H), 3.59 (s, 6H), 2.17 (d, J = 2.6 Hz, 1H); 13 C NMR (101 MHz, CDCl₃) δ 158.6, 156.6, 145.2, 126.4, 124.4, 123.5, 109.8, 93.0, 84.7, 68.5, 56.0, 26.7. IR (MeOH): 3390, 3290, 2940, 2840, 1600, 1470, 1430, 1210, 1110, 995, 815, 701, 630 cm⁻¹; HRMS (ESI): m/z calcd for C15H15O3S [M + H] 275.0742, found 275.0741.

3,5-dimethoxy-4-(1-phenylbut-3-yn-2-yl)phenol (3la). Employing the general procedure afforded compound 3la as a colourless oil (32.8 mg, 39% yield). HPLC (Daicel Chiralpak AD-H, n-hexane/i-propanol = 90/10, flow rate 0.8 mL min⁻¹, detection at 230 nm): t_R = 12.5 min (major enantiomer), t_R = 13.8 min (minor enantiomer); 72% ee. $[\alpha]^{24}_D = + (c = 1.00, CH_2Cl_2)$. ¹H

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NMR (400 MHz, CDCl₃) δ 7.24–7.13 (m, 5H), 5.99 (s, 2H), 5.04 (s, 1H), 4.49-4.46 (m, 1H), 3.71 (d, J = 22.9 Hz, 6H), 3.25 (dd, J =13.0, 8.0 Hz, 1H), 3.02 (dd, J = 12.8, 7.7 Hz, 1H), 2.02–2.01 (m, 1H); 13 C NMR (101 MHz, CDCl₃) δ 158.8, 156.0, 140.2, 129.2, 127.9, 126.0, 109.4, 94.3, 92.7, 86.9, 67.3, 55.9, 55.4, 40.0, 27.9. IR (MeOH): 3310, 2940, 2830, 1410, 1120, 1020, 691, 614 cm⁻¹; HRMS (ESI): m/z calcd for C18H19O3 [M + H] 283.1334, found 283.1329

3,5-diethoxy-4-(1-phenylprop-2-yn-1-yl)phenol (3ab).

Employing the general procedure afforded compound 3ab as a pale-yellow solid (69.2 mg, 78% yield). M.p. 82-84 °C. HPLC (Chiralcel OJ-H, n-hexane/i-propanol = 80/20, flow rate 0.8 mL min⁻¹, detection at 230 nm): t_R = 8.3 min (major enantiomer), t_R = 13.0 min (minor enantiomer); 82% ee. $[\alpha]^{20}$ D = +122.3 (c = 1.00, CH_2Cl_2). ¹H NMR (400 MHz, CDCl₃) δ 7.46 (d, J = 7.7 Hz, 2H), 7.23 (dd, J = 8.7, 6.3 Hz, 2H), 7.14 (t, J = 7.3 Hz, 1H), 5.95 (d, J = 4.0)Hz, 2H), 5.71 (d, J = 2.3 Hz, 1H), 5.16 (s, 1H), 3.91 (dt, J = 14.1, 7.1 Hz, 2H), 3.81 (dd, J = 14.9, 7.9 Hz, 2H), 2.25 (d, J = 2.7 Hz, 1H), 1.27 (t, J = 7.0 Hz, 6H); ¹³C NMR (101 MHz, CDCl₃) δ 157.8, 156.1, 141.1, 127.7, 127.4, 125.8, 110.9, 93.3, 84.9, 69.3, 64.3, 30.6, 14.6. IR (KBr): 3380, 3306, 2982, 2926, 2115, 1597, 1458, 1229, 1150, 1120, 696 cm⁻¹; HRMS (ESI): m/z calcd for C19H21O3 [M + H] 297.1491, found 297.1489.

3,5-bis(benzyloxy)-4-(1-phenylprop-2-ynyl)phenol (3ac). Employing the general procedure afforded compound 3ac as a pale-yellow solid (95.2 mg, 76% yield). HPLC (Daicel Chiralpak AD-H, n-hexane/i-propanol = 80/20, flow rate 0.8 mL min⁻¹, detection at 230 nm): t_R = 11.8 min (major enantiomer), t_R = 13.3 min (minor enantiomer); 85% ee. [α]²⁰_D = + (c = 1.00, CH₂Cl₂). ¹H NMR (400 MHz, CDCl₃) δ 7.40 (d, J = 7.5 Hz, 2H), 7.35–7.00 (m, 13H), 6.02 (s, 2H), 5.83 (s, 1H), 5.63 (s, 1H), 4.92 (d, J = 12.0 Hz, 2H), 4.82 (d, J = 12.0 Hz, 2H), 2.22 (dd, J = 2.6, 0.8 Hz, 1H); ¹³C NMR (101 MHz, CDCl₃) δ 157.6, 156.3, 140.8, 136.8, 128.5, 127.9, 127.8, 127.5, 127.4, 126.0, 111.0, 94.1, 84.9, 70.4, 69.8, 30.8. IR (KBr): 3360, 3280, 2930, 1600, 1450, 1230, 1110, 1020, 730, 695, 631 cm⁻¹; HRMS (ESI): m/z calcd for C29H25O3 [M + H] 421.1804, found 421.1801.

3,5-diisopropoxy-4-(1-phenylprop-2-ynyl)phenol (3ad). Employing the general procedure afforded compound 3ad as a colorless oil (57.2 mg, 59% yield). HPLC (Daicel Chiralpak AD-H,

n-hexane/i-propanol = 85/15, flow rate 0.8 mL min⁻¹, detection at 230 nm): t_R = 7.5 min (major enantiomer), t_R = 8.3 min (minor enantiomer); 68% ee. $[\alpha]^{31}_D$ = +140.6 (c = 1.00, CH₂Cl₂). ¹H NMR (400 MHz, CDCl₃) δ 7.44 (d, J = 7.9 Hz, 2H), 7.21 (dd, J = 12.2, 4.9 Hz, 2H), 7.13 (d, J = 7.3 Hz, 1H), 5.98 (s, 2H), 5.68 (d, J = 2.2 Hz, 1H), 5.40 (s, 1H), 4.38 (dt, J = 12.1, 6.0 Hz, 2H), 2.23 (d, J = 2.7Hz, 1H), 1.32-1.26 (m, 6H), 1.06 (d, J = 6.0 Hz, 6H); 13 C NMR (101 MHz, CDCl₃) δ 156.8, 156.0, 141.4, 127.5, 127.5, 125.7, 112.5, 94.0, 84.8, 70.4, 69.4, 30.8, 22.0, 21.7. IR (MeOH): 3260, 2980, 1600, 1460, 1370, 1100, 1060, 1020, 720, 654 cm⁻¹; HRMS (ESI): m/z calcd for C21H25O3 [M + H] 325.1804, found 325.1804.

3-ethoxy-5-methoxy-4-(1-phenylprop-2-ynyl)phenol Employing the general procedure afforded compound 3ae as a colourless oil (70.7 mg, 84% yield). HPLC (Daicel Chiralpak AD-H, n-hexane/i-propanol = 80/20, flow rate 0.8 mL min⁻¹, detection at 230 nm): $t_R = 7.2$ min (major enantiomer), $t_R = 8.1$ min (minor enantiomer); 87% ee. $[\alpha]^{20}$ _D = +123.4 (c = 1.00, CH₂Cl₂). ¹H NMR (400 MHz, CDCl₃) δ 7.36 (d, J = 7.5 Hz, 2H), 7.15 (t, J = 7.6 Hz, 2H), 7.07 (d, J = 7.2 Hz, 1H), 5.92 (dd, J = 10.0, 2.2)Hz, 2H), 5.61 (d, J = 1.9 Hz, 1H), 4.64 (s, 1H), 3.85–3.78 (m, 1H), 3.67 (d, J = 16.9 Hz, 1H), 3.57 (s, 3H), 2.31-2.05 (m, 1H), 1.16 (t, 3.67 (d, 3.67J = 7.0 Hz, 3H); ¹³C NMR (101 MHz, CDCl₃) δ 158.5, 157.9, 156.4, 141.0, 127.8, 127.3, 125.9, 110.6, 93.6, 92.7, 84.9, 69.2, 64.3, 55.9, 30.6, 14.6. IR (MeOH): 3350, 3280, 2940, 2930, 1600, 1470, 1120, 1022, 699, 636 cm⁻¹; HRMS (ESI): m/z calcd for C18H19O3 [M + H] 283.1334, found 283.1359.

3-(benzyloxy)-5-methoxy-4-(1-phenylprop-2-ynyl)phenol (3af). Employing the general procedure afforded compound 3af as a colorless oil (79.2 mg, 77% yield). HPLC (Daicel Chiralpak AD-H, n-hexane/i-propanol = 80/20, flow rate 0.8 mL min⁻¹, detection at 230 nm): t_R = 13.6 min (major enantiomer), t_R = 16.7 min (minor enantiomer); 88% ee. $[\alpha]^{30}_D$ = +97.3 (c = 1.00, CH_2CI_2). 1H NMR (400 MHz, CDCl₃) δ 7.40 (d, J = 7.5 Hz, 2H), 7.34–7.00 (m, 8H), 6.10-5.93 (m, 2H), 5.87 (s, 1H), 5.74 (s, 1H), 4.89 (d, J = 12.0Hz, 1H), 4.79 (d, J = 12.0 Hz, 1H), 3.57 (s, 3H), 2.23 (d, J = 2.7 Hz, 1H); ¹³C NMR (101 MHz, CDCl₃) δ 158.6, 157.6, 156.4, 140.8, 136.8, 128.4, 127.4, 126.1, 110.8, 94.0, 93.2, 85.0, 70.4, 69.7, 56.0, 30.7. IR (MeOH): 3300, 2940, 2830, 1600, 1450, 1110, 1020, 698, 630 cm⁻¹; HRMS (ESI): m/z calcd for C23H21O3 [M + H] 345.1491, found 345.1484.

3-isopropoxy-5-methoxy-4-(1-phenylprop-2-ynyl)phenol (3ag). Employing the general procedure afforded compound 3ag as a colorless oil (69.3 mg, 79% yield). HPLC (Daicel Chiralpak AD-H, n-hexane/i-propanol = 80/20, flow rate 0.8 mL min⁻¹, detection at 230 nm): t_R = 6.0 min (major enantiomer), t_R = 6.8 min (minor enantiomer); 81% ee. [α]³⁰_D = +139.0 (c = 1.00, CH₂Cl₂). ¹H NMR (400 MHz, CDCl₃) δ 7.55–7.37 (m, 2H), 7.22 (t, J = 7.6 Hz, 2H), 7.13 (t, J = 7.3 Hz, 1H), 5.98 (dd, J = 11.1, 1.9 Hz, 2H), 5.75 (s, 1H), 5.67 (d, J = 2.2 Hz, 1H), 4.50-4.18 (m, 1H), 3.62 (s, 3H), 2.24 (d, 3H)J = 2.7 Hz, 1H), 1.25 (d, J = 6.0 Hz, 3H), 1.02 (d, J = 6.0 Hz, 3H); ¹³C NMR (101 MHz, CDCl₃) δ 158.7, 156.6, 156.2, 141.1, 127.7, 127.4, 125.8, 111.6, 94.5, 92.6, 84.9, 70.5, 69.4, 55.9, 30.68 (s), 22.0, 21.6. IR (MeOH): 3360, 3280, 2980, 2940, 1600, 1470, 1110, 1020, 697, 632 cm⁻¹; HRMS (ESI): m/z calcd for C19H21O3 [M + H] 297.1491, found 297.1490.

3-methoxy-4-(1-phenylprop-2-ynyl)phenol (3ai). Employing the general procedure afforded compound 3ai as a white solid (9.3 mg, 13% yield). M.p. 96-97 °C. HPLC (Daicel Chiralpak AD-H, n-hexane/i-propanol = 80/20, flow rate 0.8 mL min⁻¹, detection at 230 nm): t_R = 8.2 min (minor enantiomer), t_R = 9.8 min (major enantiomer); 65% ee. [α]³¹_D = +55.4 (c = 1.00, CH₂Cl₂). ¹H NMR (400 MHz, CDCl₃) δ 7.37 (d, J = 7.8 Hz, 2H), 7.31 (d, J = 8.2 Hz, 1H), 7.26 (t, J = 7.6 Hz, 2H), 7.17 (s, 1H), 6.34 (dd, J = 12.4, 5.3 Hz, 2H), 5.53 (s, 1H), 5.36 (d, J = 1.9 Hz, 1H), 3.67 (s, 3H), 2.36 (dd, J = 1.8, 0.7 Hz, 1H); ¹³C NMR (101 MHz, CDCl₃) δ 157.2, 155.8, 141.4, 129.6, 128.4, 127.7, 126.7, 122.2, 107.3, 99.2, 85.5, 71.5, 55.6, 35.4. IR (KBr): 3407, 3274, 3025, 2838, 2114, 1596, 1506, 1455, 1203, 1037, 952, 699 cm⁻¹; HRMS (ESI): m/z calcd for C16H15O2 [M + H] 239.1072, found 239.1076.

3-methoxy-5-methyl-4-(1-phenylprop-2-ynyl)phenol (3aj). Employing the general procedure afforded compound 3aj as a colorless oil (42.8 mg, 57% yield). HPLC (Daicel Chiralpak AD-H, n-hexane/i-propanol = 90/10, flow rate 0.8 mL min⁻¹, detection at 230 nm): t_R = 16.0 min (major enantiomer), t_R = 19.8 min

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(minor enantiomer); 70% ee. $[\alpha]^{31}_D$ = +64.1 (c = 1.00, CH₂Cl₂). ¹H NMR (400 MHz, CDCl₃): δ 7.35 (d, J = 7.5 Hz, 2H), 7.25 (t, J = 7.6 Hz, 2H), 7.17 (s, 1H), 6.32 (d, J = 2.3 Hz, 1H), 6.22 (d, J = 2.2 Hz, 1H), 5.87 (s, 1H), 5.46 (s, 1H), 3.73 (s, 3H), 2.36 (d, J = 2.7 Hz, 1H), 2.09 (s, 3H); ¹³C NMR (101 MHz, CDCl₃) δ 158.0, 155.3, 140.1, 140.1, 128.2, 127.0, 126.2, 120.2, 110.5, 97.1, 84.0, 71.2, 56.0, 31.9, 20.3. IR (MeOH): 3290, 2940, 1610, 1460, 1340, 1090, 1020, 725, 630 cm⁻¹; HRMS (ESI): m/z calcd for C17H17O2 [M + H] 253.1229, found 253.1255.

(*S*)-2,3-dihydro-5-methoxy-2-methylene-1-phenyl-1H-indene (*S*). Employing the general procedure afforded compound **5** as a white solid (8.7 mg, 12% yield). M.p. 68–70 °C. HPLC (Chiralcel OJ-H, n-hexane/i-propanol = 90/10, flow rate 0.8 mL min⁻¹, detection at 230 nm): t_R = 11.3 min (major enantiomer), t_R = 16.4 min (minor enantiomer); 68% ee. [α]³¹_D = -102.4 (c = 0.20, CH₂Cl₂). ¹H NMR (400 MHz, DMSO-d₆) δ 7.35 (t, J = 7.4 Hz, 2H), 7.27 (d, J = 7.2 Hz, 1H), 7.22 (t, J = 10.3 Hz, 2H), 6.92 (d, J = 8.2 Hz, 1H), 6.69 (d, J = 2.2 Hz, 1H), 6.54 (dd, J = 8.3, 2.2 Hz, 1H), 5.24 (s, 1H), 4.73 (d, J = 2.4 Hz, 1H), 4.14 (t, J = 2.3 Hz, 1H), 3.75 (s, 3H); ¹³C NMR (101 MHz, DMSO-d₆) δ 166.5, 160.7, 158.4, 143.2, 129.2, 128.2, 127.4, 125.7, 122.0, 108.5, 96.4, 87.2, 55.9, 49.4. IR (MeOH): 1690, 1590, 1490, 1440, 1290, 1090, 950, 820, 693, 565, 439 cm⁻¹; HRMS (ESI): m/z calcd for C16H15O2 [M + H] 239.1072, found 239.1082.

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Cu-catalyzed asymmetric Friedel-Crafts propargylic alkylation of phenol

derivatives

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Long Shao and Xiang-Ping Hu

A copper-catalyzed asymmetric Friedel-Crafts propargylic alkylation of electron-rich phenol derivatives with a variety of propargylic esters has been described. With Cu(OTf)₂ decorated with a chiral tridentate ketimine P,N,N-ligand as the catalyst, asymmetric Friedel-Crafts propargylic alkylation of 3,5-dialkoxyphenol derivatives proceeded smoothly in high yields and with good to excellent enantioselectivities (up to 95% ee).