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From Aryliodonio-quinoline-8-olates to Arylquinolinium-8olates by a Two Step Aryl Migration

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Abstract: The reaction of 4-nitro-8-hydroxyquinoline with (diacetoxyiodo)benzene affords aryliodonio-quinoline-8-olates. Thermal migration of the aryl group from iodine to oxygen and photochemical migration from oxygen to nitrogen leads to the formation of intesively coloured arylquinolinium-8-olates.

Aryliodonio phenolates 1 belong to the major class of zwitterionic iodonium compounds, the chemistry of which, as well as of other classes of hypervalent iodine compounds, has been reviewed by Varvoglis.¹ The term aryliodonio phenolates refers to 1,4 dipoles with the negative charge on the oxygen atom of a phenolic group. The presence of electron withdrawing groups, mostly nitro groups, is necessary for the stabilization of these compounds.



Aryliodonio phenolates react with nucleophiles and electrophiles² and afford cyclization products with various dipolophiles.³ Recently we reported the formation of aryliodonio phenolates from 1,3-dihydroxy-benzene derivatives.⁴

Continuing the exploration of the chemistry of zwitterionic iodonium compounds,⁵ we considered the possibility of forming aryliodonio phenolates from 8-hydroxy-quinoline (oxine) and its derivatives. Oxine, upon treatment with (diacetoxyiodo)arenes 3, gave complicated mixtures of oxidation products, but its 5-nitro derivative 2 afforded 5-nitro-6-aryliodonio-quinoline-8-olates 4a-c, or their acetoxy salts 5a-c, depending on the nucleophilicity of the solvent used for the reaction (Scheme 1).





Both 4 and 5 were isolated by filtration. They are fairly stable and elucidation of their structure was based on spectroscopic data and elemental analyses. Quinoline-8-olates 4 are converted to the corresponding iodo ethers 6a-c in refluxing acetonitrile (Scheme 2). This aryl migration has been observed in other aryliodonium phenolates and involves an intermediary spiro-Meisenheimer complex.^{2a,6} Ethers 6 are new compounds and were fully characterized. Substituent R is found at *para* position, confirming the *ipso* migration of aryl group.



Scheme 2

Solutions of ethers $\mathbf{6}$ in various solvents have the tendency to obtain a deep purple colour, when they are exposed even to diffused sunlight. Chromatographic separation showed that this intense colour is a result of the formation of aryl-quinolinium-8-olates, **8**. Best results (yields of **8** up to 40-60%) are obtained when benzene solutions of ethers **6** are irradiated with a 250-Watt low pressure Hg lamp for 2 hrs or exposed to direct sunlight for 8-10 hrs. The reaction takes place even in the solid state, as it is indicated by the change of colour, when crystals of **6** are exposed to light. The only known compound analogous to **8** is methyl-quinolinium-8-olate, prepared from the corresponding methyl iodide salt of oxine.⁷ Compounds **8** are crystallized in deep purple needles and their spectroscopic data and elemental analyses are consistent with their structure.⁸

This photochemical aryl migration is most unusual since Kappe and co-workers⁹ showed that the photochemical reaction in analogous (but without ring nitrogen) iodo-aryl ethers takes a completely different path: Loss of HI gives rise to dibenzofurans.

The migration of aryl group from 6 to 8 takes place again at the carbon *ipso* to oxygen (R is found at the *para* position in the cases of 8b and 8c), indicating the possible intermediacy of a spiro-Meisenheimer complex of type 7. This assumption is strongly strengthened by the fact that alkyl ethers 9, prepared by alkylation of the silver salt of 5-nitro-8-hydroxy-quinoline, remained unchanged under irradiation and no alkyl migration to 10 is observed (Scheme 3).



It must be noted that some attempts to prepare aryl ethers analogous to 9 (with R = Ar) by other arylation methods (Williamson, diaryliodonium salts) have failed. The two step migration of aryl group described provides a facile access to aryl-8-quinolinates, these interesting 1,4 dipoles, the properties of which will be investigated.

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- 8. Selected spectroscopic data for **8c**. UV-Vis (EtOH): λ_{max} (lg ε) 313 (3.67), 385 (3.60), 495 (3.32) nm; ¹H NMR (300 MHz, CDCl₃): δ 3.88 (s, 3H), 6.95 (d, 2H, J=9 Hz), 7.17 (d, 2H, J=9 Hz), 7.88 (dd, 1H, J=9 Hz, 7 Hz), 8.37 (d, 1H, J=7 Hz), 9.42 (s, 1H), 10.43 (d,1H, J=9 Hz); MS m/z (rel intensity) 422 (M⁺, 40), 392 (10), 249 (41), 206 (56), 178 (100), 151 (39).
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