

GENISTEIN 7-(2"-p-COUMAROYLGLUCOSIDE) FROM *TRIFOLIUM REPENS*

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Key Word Index—*Trifolium repens*; Leguminosac; genistein 7-O-(2"-p-coumaroyl- β -D-glucopyranoside); isoflavone.

Abstract—A new acetylated isoflavone, genistein 7-(2"-p-coumaroylglucoside), has been characterized from *Trifolium repens*.

Earlier reports of the occurrence of genistein and biochanin-A [1] from *Trifolium repens* [2] encouraged us to investigate this plant further.

A new acetylated isoflavone (1) was isolated from a 95% ethanolic extract of the dried plant by extraction with ethyl acetate and column chromatography of the concentrated ethyl acetate extract to give a single spot on TLC. It gave positive tests for an isoflavone glycoside [3], $C_{30}H_{26}O_{12}$, $[M]^+$ at m/z 578, mp 250–252°, and analysed for C = 62.28%, H = 4.49%, calculated; found: C = 62.24%, H = 4.53%.

UV spectral analysis indicated the presence of free hydroxyl groups at C-5 and C-4', and acid hydrolysis yielded genistein, p-coumaric acid and D-glucose (by co-PC and co-TLC). On treatment with methanolic NaOMe it gave genistein 7-O- β -D-glucopyranoside, mp 263–265° and methyl coumarate (R_f 0.73), which were identified by UV, IR, 1H NMR and MS analysis.

Permethylation [4] of 1 gave a hexa-O-methyl ether, 2, $C_{36}H_{38}O_{12}$, $[M]^+$ at m/z 662, which on acid hydrolysis yielded genistein 5,4'-dimethyl ether indicating that the acyl residue was attached to the sugar moiety. On acetylation with acetic anhydride-pyridine it gave a hexaacetate, 3, $C_{42}H_{38}O_{18}$, $[M]^+$ at m/z 830. 1H NMR analysis of 3 showed the presence of one p-coumaroyl residue, and the chemical shifts for the protons H-1" (5.52 ppm) and H-2" (5.06 ppm) indicated acylation at position C-2" [5], thus confirming 1 to be genistein 7-O-(2"-p-coumaroyl- β -D-glucopyranoside), which was further confirmed by its mass spectral analysis.

EXPERIMENTAL

Plants of *Trifolium repens* (L.) were supplied by M/s Himalaya range drug field, Simla, and authenticated by the Botany Department of this university.

Spectral data for 1. UV λ_{max}^{MeOH} nm: 265, 332; + NaOMe 275, 284, 360; + $AlCl_3$ 274, 304, 378; + $AlCl_3$ -HCl 243, 380; + NaOAc 278, 323; + H_3BO_3 264, 330. IR ν_{max}^{KBr} cm^{-1} : 3348, 1774, 1705, 1560–1610, 1282, 1216, 1150, 825.

Acid hydrolysis of 1. A methanolic soln of 1 (350 mg) was refluxed with 10% HCl (50 ml) and run on TLC (C_6H_6 - Me_2CO - H_2O , 4:3:1) giving genistein (R_f 0.36), p-coumaric acid (R_f 0.47) and D-glucose. Genistein, $C_{15}H_{10}O_5$, mp 296–297°; UV λ_{max}^{MeOH} nm: 260, 331; + NaOMe 271, 326; + $AlCl_3$ 274, 304, 370; + $AlCl_3$ -HCl 280, 312, 380; + NaOAc 280, 324; + H_3BO_3 268, 330.

Alkaline methylation of 1. 1 in absolute MeOH was treated with 2% NaOMe overnight, neutralized, evaporated to a syrup and extracted with dry Et_2O . The Et_2O -soluble fraction on TLC ($EtOAc$ - $HOAc$ - H_2O , 3:2:3) gave methyl p-coumarate (R_f 0.72) and genistein 7-glucoside (R_f 0.38). **Methyl p-coumarate:** 1H NMR ($CDCl_3$, TMS as internal standard) δ 7.70 (d, J = 8 Hz, H- β), 7.56 (d, J = 4.6 Hz, H-2, H-6), 6.88 (d, J = 4.5 Hz, H-3, H-5), 6.28 (d, J = 7.8 Hz, H- α), 2.46 (s, OAc), 3.83 (s, OMe). IR ν_{max}^{KBr} cm^{-1} : 3400, 2970, 1720, 1680, 1610, 1560, 1330, 820. MS m/z : 178 $[M]^+$ 147, 119, 91, 69, 55. **Genistein 7-glucoside:** UV λ_{max}^{MeOH} nm: 265, 328; + NaOMe 275, 320; + $AlCl_3$ 270, 308, 372; + $AlCl_3$ -HCl 282, 308, 382; + NaOAc 278, 320; + H_3BO_3 272, 334.

Genistein 7-O- β -D-glucopyranoside permethyl ether was prepared as described by Brimacombe [6] and worked up in the usual manner.

Genistein 5,4'-di-O-methyl ether. The permethyl ether of genistein 7-glucoside on acid hydrolysis gave genistein 5,4'-dimethyl ether.

1 Hexa-O-acetate (3). 1 was heated with Ac_2O -pyridine for 4 hr and the acetate worked up in the usual manner. 3 crystallized from MeOH, $CHCl_3$, mp 240–242°. 1H NMR ($CDCl_3$, TMS as internal standard): δ 8.04 (s, H-2), 6.75 (d, J = 2.5 Hz, C₆-H) 2.51 (s, 3H of 5-OAc), 7.42 (d, J = 2.4 Hz, C₈-H), 2.28 (s, C₄-OAc), 7.46 (d, J = 9 Hz, C₂-H, C₆-H) 7.12 (d, J = 9 Hz, C₃-C₅-H), 4.43 (d, J = 7.0 Hz, 1" anomeric proton) 2.12 (s, C-3" OAc), 2.02 (s, C-4" OAc), 3.93 (s, C-6" OAc) 5.46 (m-6 proton of glucose unit). MS m/z : 578 $[M]^+$, 459, 431, 270, 269, 165, 153, 152, 135, 124, 118.

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TWO 12a-HYDROXYROTENOIDS FROM *BOERHAAVIA COCCINEA*

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Abstract—Two new 12a-hydroxyrotenoids, (–)-4,11,12a-trihydroxy-9-methoxyrotenoid and (–)-4,9,11,12a-tetrahydroxyrotenoid, were isolated from the roots of *Boerhaavia coccinea*. Their structures were established by spectroscopic methods.

INTRODUCTION

Boerhaavia coccinea is a Brazilian plant known by the trivial name of pega-pinto. The roots are used in traditional medicine for the treatment of liver, loins and urinary diseases [1]. Only two species of *Boerhaavia* have been extensively studied, *B. repens* [2, 3] and *B. diffusa* [4], from which alkaloids and polysaccharides have been isolated. In this paper we report the isolation of (–)-4,11,12a-trihydroxy-9-methoxyrotenoid (1) and (–)-4,9,11,12a-tetrahydroxyrotenoid (2) from a methanolic extract of roots of *B. coccinea*.

RESULTS AND DISCUSSION

Compound 1, $\text{C}_{17}\text{H}_{14}\text{O}_7$, $[\text{M}]^+$ at m/z 330, $[\alpha]_{\text{D}}^{20}$ –339°, was obtained as an amorphous powder. It showed UV maxima (MeOH) at 340 (sh), 293 and 216 nm, and IR bands (CHCl_3) at 3550 (br, OH) and 1630 cm^{-1} (chelated C=O). The ^1H NMR spectrum of 1 exhibited the signals of five aromatic protons, two of which were *meta*-coupled (δ 6.08 and 6.10, 2H, *dd*, $J = 2\text{ Hz}$), a methoxyl group (δ 3.86, 3H, *s*), a chelated hydroxyl (δ 11.85, 1H, *s*, exchangeable with D_2O) and an ABC system (δ 4.44, 1H, *dd*, $J = 5.5$ and 10 Hz; δ 4.48, 1H, *t*, $J = 10\text{ Hz}$; δ 4.77, 1H, *dd*, $J = 5.5$ and 10 Hz), attributed to a O–CH₂–CH–O sequence. In agreement with the latter assignment, the ^{13}C NMR spectrum of 1 (see Experimental) showed

resonances at 77.1 and 62.4 ppm (doublet and triplet, respectively, in the off-resonance decoupled ^{13}C NMR spectrum), together with a signal at 67.0 ppm (singlet).

The aforementioned data suggested for 1 the structure of a 12a-hydroxyrotenoid. After acetylation, 1 was transformed into the corresponding derivative 3, $\text{C}_{23}\text{H}_{20}\text{O}_{10}$, which showed in its ^1H NMR spectrum the signals of an aliphatic acetyl group (δ 1.88), assigned to the 12a position, and two aromatic acetyl groups (2.29 and 2.37).

In the mass spectrum of 1 the base peak at m/z 167 ($\text{C}_8\text{H}_7\text{O}_4$) originated from the typical retro-Diels–Alder fragmentation of 6a,12a-saturated rotenoids [5], thus confirming the suggested structure and the assignment of the methoxyl and the chelated hydroxyl group to the D-ring (Scheme 1). On account of the presence in the ^1H NMR spectrum of 1 of a low-field shifted signal (δ 7.78, 1H, *dd*, $J = 3$ and 7 Hz), the third hydroxyl group could be located on C-4 and, therefore, the structure of 4,11,12a-trihydroxy-9-methoxyrotenoid was unambiguously assigned to 1. The *trans*-B/C ring junction was assigned to 1 on the basis of the H-1 chemical shift value [6], whereas it was impossible to correlate the Cotton effect curve to the absolute configuration on account of the absence of proper models. Compound 2, $\text{C}_{16}\text{H}_{12}\text{O}_7$, $[\text{M}]^+$ at m/z 316, crystals from CHCl_3 –MeOH, mp 237–240°, $[\alpha]_{\text{D}}^{20}$ –440°, exhibited UV maxima (MeOH) at 335 (sh), 293 and 231 nm. The structure of a 9-demethyl derivative of 1 could be assigned to 2 by comparison of the ^1H NMR and mass spectra of the two compounds (see Experimental). Moreover, 1 and 2 when treated with diazomethane gave the same methyl derivative by TLC comparison.

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