View Article Online View Journal



Organic & Biomolecular Chemistry

Accepted Manuscript

This article can be cited before page numbers have been issued, to do this please use: F. Secci, A. Frongia, G. Sarais, D. J. Aitken, A. Luridiana and R. Guillot, *Org. Biomol. Chem.*, 2016, DOI: 10.1039/C6OB00160B.



This is an *Accepted Manuscript*, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. We will replace this Accepted Manuscript with the edited and formatted Advance Article as soon as it is available.

You can find more information about *Accepted Manuscripts* in the **Information for Authors**.

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard <u>Terms & Conditions</u> and the <u>Ethical guidelines</u> still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this *Accepted Manuscript* or any consequences arising from the use of any information it contains.



www.rsc.org/obc

Published on 26 February 2016. Downloaded by KUNGL TEKNISKA HOGSKOLAN on 29/02/2016 09:26:46.

COYAL SOCIETY

Journal Name

ARTICLE

Received 00th January 20xx, Accepted 00th January 20xx

DOI: 10.1039/x0xx00000x

www.rsc.org/

Deracemizing Organocatalyzed Michael Addition Reactions of 2-(Arylthio)cyclobutanones with β -Nitrostyrenes

Alberto Luridiana,^a Angelo Frongia,^a David J. Aitken,^b Regis Guillot,^b Giorgia Sarais,^c Francesco Secci^a*

Organocatalyzed Michael addition reactions of 2-(arylthio)cyclobutanones with *trans*-β-nitrostyrenes have been carried out using a bifunctional thiourea-primary amine catalyst, providing diastereoisomerically and enantiomerically enriched 2-alkyl-2-(arylthio)cyclobutanones having two contiguous stereocenters of which one is a chiral quaternary center. The absolute configuration of these novel adducts was assigned by X-ray diffraction analysis and a transition-state model is proposed to explain the observed stereoselectivities.

Dedicated to Dr Jean Ollivier, a mentor and a friend, on the occasion of his retirement

Introduction

Cyclobutanones are a family of strained small-ring carbocyclic molecules which are stable yet chemically reactive and are recognised as powerful intermediates for the stereo- and enantioselective synthesis of natural and/or biologically-active products.¹ The interest on this class of compounds arises from both the α -proton acidity and the marked carbonyl group electrophilicity.² These intrinsic characteristics have been used advantageously in synthesis, for the construction of highly functionalized building blocks containing quaternary stereogenic centres.³ Over the years, cyclobutanone derivatives have been exploited in various synthetic applications such as ring-opening,⁴ ring-expansion,⁵ and ringcontraction reactions.⁶ while useful intramolecular cyclization reactions⁷ and oxidation protocols⁸ have been established. More recently, metal-catalyzed and organocatalytic procedures have been developed in order to facilitate controlled selective transformations of cyclobutanones, including $\alpha\text{-alkylation,}^9$ aldolization 10 and Mannich reactions 11 of cvclobutanone itself; aldolization,¹² nitrogen insertion,¹³ and α -amination¹⁴ of 2-hydroxycyclobutanone; and the aldoldesymmetrization of prochiral 3-substituted cyclobutanones.¹¹

Concerning organocatalyzed Michael addition reactions, Ley and co-workers were the first to consider (unsubstituted) cyclobutanone in reactions with β -nitrostyrenes, employing Lproline derivatives as catalysts.¹⁶ Since then, however, little development of this type of reaction has been reported. In one significant result in the field, reported by Rodriguez and coworkers, hydrogen bond-forming 2-carboxamide derivatives of cyclobutanone were engaged successfully in straightforward enantio- and diastereoselective Michael addition reactions with β -nitrostyrenes using bifunctional organocatalysts.¹⁷ Our own recent work in this area revealed that 3-substituted cyclobutanones can be desymmetrized in Michael addition reactions with nitroalkenes in mild organocatalyzed conditions, yielding the corresponding α -functionalized adducts with high selectivity.¹⁸ However, reaction times were long and many cyclobutanone derivatives are moisture and air sensitive,¹⁹ so we considered that further exploration of Michael reactions of cyclobutanones was warranted.

Our rationale for the developments described in this paper is based on the added synthetic value of using cyclobutanones bearing an α -sulfanyl group as substrates. With such compounds, we anticipated an enhanced reactivity of the cyclobutanone α -proton and an improved stability of the intermediates during the enolization processes. Indeed, the literature concerning β -ketosulfides shows that the enolate of α -(phenylthio)cyclobutanone has a pronounced pronucleophilic character, comparable with ketoamides and 1,3dicarbonyl compounds (p K_a (DMSO) = 16–19).²⁰ It is also known that α -(phenylthio)cyclobutanones show a strong tendency to enolize in the presence of triethylamine/Ac₂O, generating the corresponding stable enolthioether.²¹ These β ketosulfides, as well as their enolthioethers derivatives, have been described in a number of synthetic applications²² but to the best of our knowledge they have never been used in

^{a.} Dipartimento di Scienze Chimiche e Geologiche, Università degli Studi di Cagliari, Complesso Universitario di Monserrato, Monserrato (Ca) ITALY, Phone : (+39)-0706754384, Fax: (+39)-0706754456, e-mail: fsecci@unica.it

^{b.} CP3A Organic Synthesis Group & Services Communs, ICMMO (UMR 8182), CNRS, Université Paris Sud, Univ. Paris Saclay, 15 rue Georges Clemenceau, 91405 Orsay cedex, FRANCE

^c Dipartimento di Scienze della Vita e dell'Ambiente, Università degli Studi di Cagliari, Palazzo delle Scienze, via Ospedale 82, 09124 Cagliari, ITALY

Electronic Supplementary Information (ESI) including 1 H and 13 C NMR spectra, HPLC traces, CIF data for compound **5a** and additional experimental procedures available: DOI: 10.1039/x0xx00000x

Published on 26 February 2016. Downloaded by KUNGL TEKNISKA HOGSKOLAN on 29/02/2016 09:26:46.

DOI: 10.1039/C6OB00160B Journal Name

enantioselective procedures. The chemical versatility of sulfurcontaining building blocks should also facilitate the use of such compounds in a wide number of subsequent transformations. For all of these reasons, we undertook a study of 2-(arylthio)cyclobutanone derivatives as nucleophiles in new organocatalyzed Michael addition reactions with β nitrostyrenes.

Results and discussion

To begin, a catalyst screening was conducted. The catalysts examined are illustrated in Figure 1 and fall into four categories: L-Proline derivatives I-V, cinchona alkaloid derivatives VI-VII, thiourea–amine derivatives VIII-XII, and two squaramide derivatives XIII-XIV. The model Michael addition reaction between (±)-2-(phenylthio)cyclobutanone $1a^{23}$ and *trans*- β -nitrostyrene 2a was selected to examine using each of these catalysts in appropriate solvents at room temperature for 2 days.



The first series of experiments was conducted in the presence of 20 mol% of a proline derivative (Table 1). The use of L-proline I itself in DMSO (entry 1) led to the formation of two Michael adducts in a 10:1 ratio and a combined yield of 44%. The major product was isolated and ¹H NMR and NOESY analysis revealed that its structure was iso-3a, obtained with a notable 10:1 diastereoselectivity. The quaternary regioisomeric adduct 3a was identified as the minor product (configuration not determined). Formation of this product is rationalized by the hypothesis that the reaction proceeds preferentially via the formation of the less-hindered enamine (implicating the C4 center) due to the steric bulk of the PhS group.²⁴ Improved results were achieved using catalyst II in DMSO (entry 2), which provided a higher conversion of 1a, an excellent product selectivity in favour of iso-3a, and a very high diastereoselectivity for this compound. In toluene (entry 3) the high selectivities observed using catalyst **II** were maintained but the conversion was diminished; the reaction failed to proceed in dichloromethane (entry 4). The use of catalysts **III-V** in DMSO (entries 5-7) progressed with lower conversions and reduced selectivities. While several aspects of this first-round study were encouraging, unfortunately all of the isolated samples of the main product *iso*-**3a** were close to racemic. Likewise, HPLC analysis of the recovered unreacted cyclobutanone **1a** showed this material to be racemic indicating no kinetic or dynamic resolution was operating.



Table 1. Michael addition reaction of cyclobutanone 1a with $\textit{trans-}\beta\text{-nitrostyrene}~2a$ using catalysts I-V. a

Entry	Cat.	Solvent	Y(3a + <i>i</i> - 3a)(%) ^b	3a: <i>i</i> -3a [°]	<i>i-</i> 3a <i>d.r.</i> ^c	<i>i-</i> 3a <i>e.r.</i> ^d
1	1	DMSO	44	17:83	9:91	57:43
2	П	DMSO	62	2:98	2:98	55:45
3	П	toluene	25	4:96	4:96	53:47
4	П	CH_2CI_2	_e			
5	Ш	DMSO	35	26:74	21:79	52:48
6	IV	DMSO	28	23:77	16:84	59:41
7	v	DMSO	30	19:81	13:87	43:57

^a Reactions were carried out in a 5 mL vial with **1a** (0.56 mmol), **2a** (0.56 mmol), cat. **I-V** (20 mol %) in the reported solvents (1.5 mL) at room temperature (48 h). ^b Yields were determined by weight after chromatography. ^c **3a***iso***-3a** ratios and *iso***-3a** *d.r.* values were determined by ¹H NMR analysis (CDCl₃) of the crude reaction mixtures. ^d *iso***-3a** *e.r.* values were determined by chiral HPLC. ^e No reaction.

Unsatisfied with these results, we turned our attention to the model reaction of 1a with 2a in the presence of 20 mol % of the other catalysts VI-XIV; toluene was a more convenient solvent for these reactions (Table 2). The cinchona-derived catalysts VI and VII induced the formation of the guaternary Michael adduct 3a (implicating reaction at the C2 center) in low yields and diastereoselectivities (entries 1 and 2). When the reaction was carried out in the presence of thioureatertiary amines VIII or X (entries 3 and 5) the yields of 3a were improved although diastereoselectivities remained low. The related thiourea-primary amine IX performed less well, providing 3a in a lower yield and with a comparable diastereoselectivity; however, in contrast to its tertiary amine congeners, IX provided sample of the major diastereomer of 3a with a more encouraging enantiomeric ratio, better than 4:1 (entry 4). Further improvements were obtained using catalyst XI which gave 3a in 68% yield and a better than 7:1 diastereoselectivity; the diastereoselection in this case was in favour of the syn-3a isomer (entry 6).²⁵ The best result in this series of test experiments was achieved using the thioureaprimary amine catalyst XII, which afforded anti-3a, in 84% yield, almost exclusively as a single diastereomer which had an enantiomeric ratio of 99:1 (entry 7). The catalyst screening was

completed by examining the squaramide derivatives **XIII** and **XIV** which gave disappointing results in terms of both product yields and selectivities (entries 8 and 9). The *anti* configuration observed for the major diastereoisomer in all these experiments (with the exception of entry 6) is attributed here on the basis of the x-ray diffraction analysis described later in this manuscript. *Anti-* and *syn*-stereoisomers showed quite similar ¹H NMR spectra, although one of the four cyclobutanone ring protons showed a configuration-dependent chemical shift, appearing at 2.31 ppm in *anti-***3a** and at 1.95 ppm for *syn-***3a**.



Table 2. Michael addition reaction of cyclobutanone 1a with $\textit{trans-}\beta\text{-nitro}$ styrene 2a using catalysts VI-XIV

Entry	Cat.	Yield 3a (%) ^b	syn:anti 3a °	anti- 3a e.r. ^d
1	VI	16	42:58	23:77
2	VII	12	24:76	_e
3	VIII	64	31:69	53:47
4	IX	21	22:78	19:81
5	Х	50	27:73	52:48
6	XI	68	88:12	90:10 ^f
7	XII	84	1:99	1:99
8	XIII	49	14:86	8:92
9	XIV	26	20:80	18:82

^a Reactions were carried out in a 5 mL vial with **1a** (0.56 mmol), **2a** (0.56 mmol), cat. **VI-XIII** (20 mol %) in toluene (1.5 mL) at room temperature (48 h). ^b Yields were determined by weight after chromatography. ^c **3a** *d.r.* values were determined by ¹H NMR analysis (CDCl₃) of the reaction crude mixtures. ^d anti-**3a** *e.r.* values were determined by chiral HPLC. ^e Not determined. ^fThe value is given for the major *syn*-**3a** adduct.

The pleasing results obtained using the thiourea–amine **XII** inspired us to pursue studies with this catalyst and explore different reaction conditions in which the solvent, the reagent stoichiometry, the reaction time and the catalyst loading were varied, with a view to optimization (Table 3).



Table 3. Michael addition reaction of cyclobutanone 1a with $\textit{trans-}\beta\text{-nitro}$ styrene 2a using catalyst XII in different conditions.



1	<i>n</i> -hexane	20	1:1	168	27	1:99	10:90
2	EtOAc	20	1:1	48	20	1:99	10:90
3	<i>m</i> -xylene	20	1:1	48	19	7:93	_e
4	CH_2CI_2	20	1:1	36	13	13:87	13:87
5	CHCl₃	20	1:1	72	29	10:90	9:91
6	dioxane	20	1:1	48	22	28:72	_e
7	THF	20	1:1	48	20	21:79	22:78
8	toluene	20	1:1	48	80	1:99	1:99
9	toluene	10	1:1	120	48	4:96	3:97
10	toluene	15	1:1	48	72	1:99	2:98
11	toluene	30	1:1	48	71	2:98	2:98
12	toluene	15	1:3	48	81	1:99	2:98
13	toluene	15	3:1	48	49	4:96	4:96
14	toluene	15	1:6	48	63	2:98	3:97
15	toluene	15	1:10	48	28	8:92	7:93

DOI: 10.1039/C6OB00160B

ARTICLE

^a Reactions were carried out in a 5 mL vial with **1a** (0.56 mmol), **2a** (0.56 mmol), cat. **XII** (15-30 mol %) in the appropriate solvent (1.5 mL) at room temperature. ^b Yields were determined by weight after chromatography. ^c **3a** *d.r.* values were determined by ¹H NMR analysis (CDCl₃) of the reaction crude mixtures. ^d anti-**3a** *e.r.* values determined by chiral HPLC. ^e Not determined.

Firstly, the effect of the solvent was assessed in reactions using 20 mol% catalyst and 1 equivalent each of 1a and 2a. In nhexane the diastereoselectivity was high but the reaction was very slow and inefficient; after seven days, compound 3a was obtained in only 27% yield (entry 1). Reactions remained sluggish in several other solvents (entries 2-7) and in some cases selectivity was compromised. We therefore returned to toluene as the solvent of choice for the evaluation of the reagent stoichiometry and catalyst loading parameters. Using 20 mol% catalyst and 1 equivalent each of 1a and 2a the adduct 3a was obtained in 80% yield after 48 hours with excellent diastereo- and enantioselectivity (entry 8). When the catalyst loading was reduced to 10 mol%, the reaction was considerably slower, giving 3a in 48% yield only after 5 days (entry 9). On the other hand, an increased catalyst loading of 30 mol% did not provide any benefit in terms of yield or selectivity (entry 11). A 72% yield of 3a with very high selectivity was achieved in 2 days, when 15 mol% of catalyst XII was employed (entry 10). The use of a 3-fold excess of the cyclobutane substrate 1a reduced the yield of the adduct 3a somewhat (entry 13), whereas the use of an 3-fold excess of trans- β -nitrostyrene **2a** resulted in an increase of the yield of 3a to 81%, still characterized by high diastereo- and enantioselectivities (entry 12) and considered by us the best reaction condition for this transformation. Further increases in the excess of the nitrostyrene did not improve on this result (entries 14 and 15). The optimized protocol was therefore concluded as being the use of 15 mol% of catalyst XII in toluene at room temperature for 2 days with a 1:3 1a/2a ratio. We then turned our attention to the XII-catalyzed Michael addition reactions involving other 2-(arylthio)cyclobutanones in order to establish the reaction scope and to evaluate the electronic effects, if any, of substituents present on the sulfanyl aromatic moiety. For these purposes, a panel of pderivatives racemic substituted of 2-(phenylthio)cyclobutanone **1b-g** was prepared²³ and each was submitted to a Michael addition reaction with trans- β -

nitrostyrene **2a** to afford the corresponding C2 quaternary cyclobutanone **3b-g** (Table 4). In each case, the Michael adduct **3b-g** was obtained in good yield, with very high *anti* diastereoselectivity, and with high enantiomeric enrichment. The major *anti* configuration was suggested by the uniformly low-field chemical shift values (>2.2 ppm) observed for the cyclobutanone proton discussed above for **3a**. Only marginal differences in the terms of yield and selectivity were observed across the substrate panel, suggesting no significance of the electronic effects present in the arylthio moiety for either the formation of the intermediate enamine or the progress of the reaction.



Table 4. Michael addition reactions of cyclobutanones **1b-g** with *trans*- β -nitrostyrene **2a** using catalyst **XII** in optimized conditions.^{a-d}



^a Reactions were carried out in a 5 mL vial with **1b-g** (0.56 mmol), **2a** (1.68 mmol), cat. **XII** (15 mol %) in toluene (1.5 mL), room temperature (48 h). ^b *d.r.* values were determined by ¹H NMR analysis (CDCl₃). ^c Yields were determined by weight after chromatography. ^d *e.r.* values determined by chiral HPLC.

In the next series of experiments, we evaluated the scope of the reaction and the significance, if any, of the electronic and steric effects prevalent in the *trans*-β-nitrostyrene component. For this purpose, a panel of differently substituted β nitrostyrene derivatives 2b-I was selected and each was submitted to a Michael addition reaction with racemic 2-(phenylthio)cyclobutanone 1a, using the optimized reaction conditions established above, to afford the corresponding C2 quaternary cyclobutanones 4b-I (Table 5). Compound 1a reacted smoothly with the *p*-substituted trans-β-nitrostyrene derivatives 2b-g, bearing either an electron-donating or an electron-withdrawing group, affording the corresponding anti derivatives 4b-f (showing low-field chemical shift values for the relevant cyclobutanone proton in ¹H NMR spectra) with good yields (57-78%) and with excellent diastereoselectivities and *e.r.* values. The reaction with the *m,p*-methylenedioxy derivative 2h to give 4h worked equally well, and the reaction also proceeded with the *o*,*p*-disubstituted and 0.0disubstituted *trans*- β -nitrostyrenes, **2i** and **2j**, with only marginally lower yields and unaffected high stereoselectivities. These observations gratifyingly illustrated a general tolerance of electronic and steric effects on the aryl moiety of the *trans*- β -nitrostyrene component.

To probe the limits of the reaction scope, cyclobutanone **1a** was placed with nitroalkenes **2k** and **2l** in the standard Michael addition conditions. The reaction failed to proceed with these substrates and almost complete recovery of the unchanged starting materials was observed. This result is in line with comparable observations, reported previously by Wennemers and co-workers,²⁶ of the low reactivity of α - or β -substituted β -nitrostyrenes with sterically hindered nucleophiles in C–C bond-forming reactions.









^a Reactions were carried out in a 5 mL vial with **1a** (0.56 mmol), nitrostyrene **2b-I** (1.68 mmol), cat. **XII** (15 mol %) in toluene (1.5 mL) at room temperature (48 h). ^b *d.r.* values of compounds **4b-j** were determined by ¹H NMR analysis (CDCl₃) of the reaction crude mixtures. ^c Yields were determined by weight after chromatography. ^d *e.r.* values determined by chiral HPLC.

DOI: 10.1039/C6OB00160B ARTICLE Scheme 2. In this representation, cyclobutanone **1a** interacts

Satisfied with the results obtained thus far in our investigation, we extended the methodology by applying it to other related ketone substrates (Scheme 1). The α -sulfonyl cyclobutanone (±)-5 reacted only very slowly with *trans*- β -nitrostyrene 2a in the presence of catalyst XII, yielding the quaternary Michael adduct *anti*-8 in low yield as a single diastereoisomer. The α -sulfanyl cyclopentanone and cyclohexanone substrates, (±)-6 and (±)-7, both reacted with *trans*- β -nitrostyrene 2a in the standard conditions to provide the single diastereoisomeric adducts *anti*-9 and *anti*-10, in 57 % (*e.r.* 4:96) and 52 % (*e.r.* 1:99) yields, respectively.



Scheme 1. Reaction of other cyclic ketone substrates 5-7 with *trans*- β -nitrostyrene 2a using catalyst XII in optimized conditions.

In order to establish the stereochemical profile of the Michael addition reactions described in this work, the major diastereoisomer of 3a (described in Table 2) was oxidized using m-CPBA in dichloromethane, to give sulfone 8, in 88% yield (Scheme 2). This compound was spectroscopically identical (superimposable ¹H and ¹³C NMR spectra, same sign for specific rotation) with the sample of 8 obtained in Scheme 1. X-Ray diffraction analysis of a single crystal of the new sample of 8 revealed the anti diastereoisomeric structure with the R configuration at the quaternary cyclobutanone C2 center and the S configuration at the new exocyclic stereogenic center (Scheme 2).²⁷ This observation is in good agreement with our previous work using catalyst XII for Michael addition reactions of 3-substituted cyclobutanones with *trans*-β-nitrostyrenes, which exhibited the same stereochemical selectivities at the two above-mentioned sereocenters.¹⁸ On the basis of the similarities in the ¹H and ¹³C-NMR spectroscopic signatures it is proposed by extension that the same configuration pattern prevails for the major stereoisomer throughout the series of compounds 3a-g, 4b-j, 8-10 prepared in this work.

A plausible transition state model, which explains the formation of the major *anti*-cyclobutanone adducts **3a-g** and **4a-i** with the attributed stereochemistry, is illustrated in



with catalyst XII leading to the formation of an enamine

intermediate able to interact with the β -nitrostyrene **2a** bound

to the thiourea moiety of the catalyst via hydrogen-bonding

interactions.^{17,18} The geometrical alignment of the two reactive

Scheme 2. a) Plausible transition state (TS) model for the formation of cyclobutanone **3a**. b) Oxidation of adduct **3a** to sulfone **8** and ORTEP plot of the single crystal X-ray structure of this compound (see the Supporting Information for details).

88% vield

Conclusions

In summary, we have developed a new, deracemizing organocatalyzed Michael addition reaction between 2-(arylthio)cyclobutanones and *trans*-β-nitrostyrenes which provides access to a selection of new C2 quaternary cyclobutanone derivatives in good yields and with excellent regio-, diastereo- and enantio-selectivities. The extent of substrate tolerance in both the electrophile and the nucleophile, the successful preliminary extensions to other cyclic ketone substrates, the reactivity potential embedded in the highly-functionalized adducts, and the general simplicity and feasibility of this C–C bond-forming reaction make it a useful addition to the existing inventory of organocatalyzed reactions currently available as tools for fine organic synthesis.

Acknowledgements

The authors gratefully acknowledge R.A.S. "Sardinia Regional Government" for financial support (P.O.R. Sardegna F.S.E. Operational Programme of the Autonomous Region of Sardinia, European Social Fund 2007-2013-Axis IV Human Resources, Objective I.3, Line of Activity I.3.1) and C.I.N.M.P.I.S. (Consorzio Interuniversitario Nazionale di Ricerca

DOI: 10.1039/C6OB00160B

in Metodologie e Processi Innovativi di Sintesi). Funding by the Université Paris-Sud, the CHARM3AT LabEx, and the Ile-de-France Region (SESAME program 2012 N°12018501) for Instrument resources is acknowledged.

Experimental section

General information

¹H NMR spectra were recorded on 400 and 500 MHz Varian spectrometers at 27 °C using CDCl₃, DMF-d₇, acetone-d₆ or DMSO-d₆ as solvent. ¹³C NMR spectra were recorded at 100 and 125 MHz at 27 °C using CDCl₃, DMF-d₇ or DMSO-d₆ as solvent. Chemical shifts (\delta) are given in ppm. Coupling constants (J) are reported in Hz. Infrared spectra were recorded on a FT-IR Bruker spectrophotometer and are reported in wavenumbers. High resolution mass spectra (HRMS) were obtained using positive mode Electrospray ionization (ESI). Optical rotation values were recorded with a PolAAr 32 instrument. Analytical thin layer chromatography was performed using 0.25 mm silica gel 60-F plates. Flash chromatography was performed using 70-200 mesh silica gel. Yields refer to chromatographically and spectroscopically pure materials. Enantiomeric ratios (e.r.) of Michael adduct cvclobutanones reported in the text were determined by analytical HPLC using a Diacel Chiralpak AD-H, a Chiracel OJ, a Chiracel AS-H or a Phenomenex Lux Cellulose-1 analytical column, with i-PrOH/Hexane as the moble phase. Authentic racemic samples were used for reference comparisons. All 2-(arylthio)cycloalkanones 1a-g, 6 and 7 were prepared using the recent literature protocol.²³ β-Nitrostyrene and its derivatives 2a-1 were purchased from Sigma-Aldrich or Alfa-Aesar and used without further purification. Catalyst XII was purchased from Sigma-Aldrich or prepared according to the literature procedure.²⁸

Synthesis of 2-(phenylsulfonyl)cyclobutanone 5.

To a stirred solution of cyclobutanone 1a (300 mg, 1.68 mmol) in CHCl₃ (20 mL) cooled to -20 °C, commercial m-CPBA (70% pure; 752 mg, 3.36 mmol), was added portion-wise over 20 min. The reaction mixture was monitored by TLC and stirred until the reaction was complete (5 h). The resulting suspension was filtered through a pad of celite, which was washed through with CHCl₃. The organic layer was washed successively with saturated aqueous NaHCO₃ solution, water, then brine. It was then dried over Na₂SO₄ and concentrated under reduced pressure. The crude product was purified by flash chromatography (hexanes:diethyl ether, gradient 90:10 to 50:50) affording the pure product 5 as a pale yellow solid, in 78% yield. Mp = 68-70 °C; ¹H NMR (400 MHz, CDCl₃) δ : 7.85-7.80 (m, 2H), 7.64-7.58 (m, 1H), 7.55-7.47 (m, 2H), 4.72-4.61 (m, 1H), 3.23 (dddd, J = 18.3, 11.1, 7.2, 2.8 Hz, 1H), 3.14-3.02 (m, 1H), 2.57 (ddt, J = 12.8, 11.1, 6.4 Hz, 1H), 2.35 (dtd, J = 12.7, 9.8, 7.2 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ: 195.0, 137.9, 134.2, 129.3, 128.5, 78.8, 47.6, 13.6. This compound was prepared previously using a different method.²⁹

General procedure for the organocatalyzed Michael reactions of ketones 1a-g and 5-7 with nitrostyrenes.

In a 5 mL vial, a solution of the appropriate ketone (0.56 mmol), the appropriate nitrostyrene (1.68 mmol) and catalyst **XII** (15 mol %), in dry toluene (1.5 mL) was stirred for 36 h at room temperature. The reaction mixture was directly purified by flash chromatography (hexanes:diethyl ether, gradient 95:5 to 50:50) to affording the corresponding pure product.

(2R, 1'S)-2-(2-nitro-1-phenylethyl)-2-(phenylthio)cyclobutanone anti-3a. Yellow oil. FTIR (neat) v 2952, 2342, 1790, 1566, 1051 cm⁻¹; $[\alpha]^{25}_{D} = +19.2$ (c. 0.36, CHCl₃); ¹H NMR (500 MHz, CDCl₃) δ: 7.51-7.48 (m, 2H), 7.39-7.34 (m, 1H), 7.32 (t, J = 7.2 Hz, 2H), 7.23 (dt, J = 6.9, 6.2 Hz, 3H), 7.19-7.15 (m, 2H), 5.25 (dd, J = 13.4, 3.8 Hz, 1H), 4.88 (dd, J = 13.3, 11.9 Hz, 1H), 3.89 (dd, J = 11.8, 3.7 Hz, 1H), 2.88 (ddd, J = 17.9, 10.6, 7.3 Hz, 1H), 2.44 (ddd, J = 17.9, 10.3, 5.6 Hz, 1H), 2.31 (ddd, J = 12.3, 10.4, 7.4 Hz, 1H), 1.76 $(ddd, J = 12.4, 10.7, 5.6 Hz, 1H); {}^{13}C NMR (125 MHz, CDCl_3)$ δ: 203.2, 136.2, 133.9, 130.1, 129.5, 129.2, 129.2, 129.0, 128.5, 75.0, 72.6, 45.2, 43.6, 21.2; HRMS (ESI): calcd for C₁₈H₁₇NNaO₃S: 350.0827 (M+Na⁺), found: 350.0829. HPLC analysis (Phenomenex Lux-1 column, i-PrOH/Hexane 5:95, 1.0 mL/min. $\lambda = 254$ nm, t_R (major) = 20.78 min, t_R (minor) = 18.44 min.

2-(2-nitro-1-phenylethyl)-2-(phenylthio)cyclobutanone syn-**3a** Colourless oil. FTIR (neat) v: 2994, 2344, 1791, 1542, 1048 cm⁻¹; [α]²⁵_D = +19.2 (c. 0.35, CHCl₃); ¹H NMR (500 MHz, CDCl₃) δ: 7.50-7.47 (m, 2H), 7.40-7.32 (m, 3H), 7.27-7.19 (m, 3H), 7.08-7.04 (m, 2H), 5.27 (dd, *J* = 13.3, 3.8 Hz, 1H), 5.06 (dd, *J* = 13.2, 11.5 Hz, 1H), 3.79 (dd, *J* = 11.4, 3.8 Hz, 1H), 2.85 (ddd, *J* = 18.2, 10.5, 7.6 Hz, 1H), 2.39 (ddd, *J* = 18.3, 10.2, 5.8 Hz, 1H), 1.95 (ddd, *J* = 16.5, 10.7, 5.8 Hz, 1H), 1.87 (ddd, *J* = 12.4, 10.3, 7.6 Hz, 1H); ¹³C NMR (125 MHz, CDCl₃) δ: 205.8, 136.1, 135.7, 131.9, 129.9, 129.5, 129.1, 128.6, 128.5, 76.7, 73.7, 48.9, 43.7, 22.0; HRMS (ESI): calcd for C₁₈H₁₇NNaO₃S: 350.0827 (M+Na⁺), found: 350.0831. HPLC analysis (chiral pack AD-H column, *i*-PrOH/Hexane 25:75, 1.0 mL/min. λ = 254 nm, t_R (major) = 24.18 min, t_R (minor) = 22.62 min.

(2R,1'S)-2-(4-fluorophenylthio)-2-(2-nitro-1-phenylethyl)

cyclobutanone anti-**3b.** Yellow solid. Mp = 135-137 °C; FTIR (KBr) v: 2993, 2357, 1789, 1555, 1053 cm⁻¹; [α]²⁵_D = +69.2 (*c*. 0.87, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ: 7.56 (dd, *J* = 8.5, 5.3 Hz, 2H), 7.36-7.25 (m, 3H), 7.23 (d, *J* = 7.7 Hz, 2H), 7.09 (t, *J* = 8.5 Hz, 2H), 5.31 (dd, *J* = 13.3, 3.7 Hz, 1H), 5.01-4.89 (m, 1H), 3.91 (dd, *J* = 11.7, 3.7 Hz, 1H), 2.97 (ddd, *J* = 17.7, 10.6, 7.2 Hz, 1H), 2.56-2.47 (m, 1H), 2.44-2.35 (m, 1H), 1.87-1.78 (m, 1H); ¹³C NMR (100 MHz, CDCl₃) δ: 202.8, 165.2, 138.4 (d, *J* = 8.5 Hz), 133.4 (d, *J* = 29.0 Hz), 129.0, 128.9, 128.4, 124.3, 116.5 (d, *J* = 21.9 Hz), 74.7, 72.4, 45.0, 43.5, 20.7; HRMS (ESI): calcd for C₁₈H₁₆FNNaO₃S: 368.0727 (M+Na⁺), found: 368.0714. HPLC analysis (Phenomenex Lux-1 column, *i*-PrOH/Hexane 5:95, 1.0 mL/min. λ = 254 nm, t_R (major) = 21.51 min, t_R (minor) = 23.97 min.

(2R, 1'S)-2-(4-chlorophenylthio)-2-(2-nitro-1-

phenylethyl)cyclobutanone anti-3c. Yellow solid. Mp = 91-94

Organic & Biomolecular Chemistry Accepted Manuscript

6 | J. Name., 2012, 00, 1-3

°C; FTIR (KBr) v: 2984, 2344, 1789, 1557, 1060 cm⁻¹; $[\alpha]^{22}_{D}$ = +107.4 (*c*. 0.48, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ : 7.50 (d, *J* = 8.4 Hz, 2H), 7.42-7.35 (m, 2H), 7.34-7.27 (m, 3H), 7.24 (t, *J* = 7.5 Hz, 2H), 5.29 (dd, *J* = 13.4, 3.7 Hz, 1H), 4.95 (t, *J* = 12.5 Hz, 1H), 3.93 (dd, *J* = 11.7, 3.6 Hz, 1H), 3.00 (ddd, *J* = 17.7, 10.5, 7.1 Hz, 1H), 2.60-2.46 (m, 1H), 2.45-2.34 (m, 1H), 1.90-1.75 (m, 1H); ¹³C NMR (100 MHz, CDCl₃) δ : 202.7, 161.8, 137.3, 133.5, 129.6, 129.0, 128.9, 128.5, 127.6, 74.7, 72.4, 45.0, 43.5, 20.8; HRMS (ESI): calcd for C₁₈H₁₆ClNNaO₃S: 384.0432 (M+Na⁺), found: 384.0415. HPLC analysis (Phenomenex Lux-1 column, *i*-PrOH/Hexane 5:95, 1.0 mL/min. λ = 254 nm, t_R (major) = 27.01 min, t_R (minor) = 24.15 min.

(2R,1'S)-2-(4-bromophenylthio)-2-(2-nitro-1-

phenylethyl)*cyclobutanone anti*-**3d.** White solid. Mp = 132-133; FTIR (KBr) v: 2949, 2350, 1790, 1563, 1064 cm⁻¹; $[\alpha]^{20}_{D}$ = +39.8 (*c*. 0.10, CHCl₃); ¹H NMR (500 MHz, CDCl₃) δ: 7.45 (d, *J* = 8.5 Hz, 2H), 7.35 (d, *J* = 8.5 Hz, 2H), 7.27-7.19 (m, 3H), 7.17-7.14 (m, 2H), 5.21 (dd, *J* = 13.4, 3.8 Hz, 1H), 4.87 (dd, *J* = 13.3, 11.8 Hz, 1H), 3.86 (dd, *J* = 11.7, 3.7 Hz, 1H), 2.92 (ddd, *J* = 17.9, 10.6, 7.2 Hz, 1H), 2.45 (ddd, *J* = 18.0, 10.4, 5.7 Hz, 1H), 2.33 (ddd, *J* = 12.4, 10.4, 7.2 Hz, 1H), 1.75 (ddd, *J* = 12.5, 10.6, 5.7 Hz, 1H); ¹³C NMR (125 MHz, CDCl₃) δ: 202.8, 137.6, 133.7, 132.7, 129.1, 129.0, 128.6, 125.0, 74.6, 72.5, 45.1, 43.7, 21.0; HRMS (ESI): calcd for C₁₈H₁₆BrNNaO₃S: 427.9926 (M+Na⁺), found: 427.9907. HPLC analysis (Phenomenex Lux-1 column, *i*-PrOH/Hexane 5:95, 1.0 mL/min. λ = 254 nm, t_R (major) = 32.83 min, t_R (minor) = 28.65 min.

(2R, 1'S)-2-(4-nitrophenylthio)-2-(2-nitro-1-

phenylethyl)cyclobutanone anti-**3e.** Yellow oil. FTIR (neat) v: 2997, 2340, 1790, 1576, 1478, 1242 cm⁻¹; $[\alpha]^{20}{}_{\rm D}$ = +31.7 (c. 0.26, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ: 8.24 (d, *J* = 8.5 Hz, 2H), 7.61 (d, *J* = 8.5 Hz, 2H), 7.35 (d, *J* = 4.5 Hz, 3H), 7.14–7.11 (m, 2H), 5.19 (dd, *J* = 13.2, 4.2 Hz, 1H), 5.11-5.01 (m, 1H), 3.97 (dd, *J* = 10.7, 4.2 Hz, 1H), 3.18 (ddd, *J* = 18.2, 10.4, 7.6 Hz, 1H), 2.73-2.61 (m, 1H), 2.29-2.15 (m, 1H), 2.14-2.01 (m, 1H); ¹³C NMR (100 MHz, CDCl₃) δ: 202.1, 139.0, 134.6, 133.7, 133.3, 129.3, 129.1, 128.7, 124.3, 74.8, 44.9, 43.7, 22.7, 20.9; HRMS (ESI): calcd for C₁₈H₁₆N₂NaO₅S: 395.0678 (M+Na⁺), found: 395.0683. HPLC analysis (Phenomenex Lux-1 column, *i*-PrOH/Hexane 5:95, 1.0 mL/min. λ = 254 nm, t_R (major) = 35.90 min, t_R (minor) = 41.57 min.

(2R,1'S)-2-(4-methoxyphenylthio)-2-(2-nitro-1-phenylethyl)

cyclobutanone anti-**3f.** Colourless solid. Mp = 123-124 °C; FTIR (ATR) v: 2934, 2345, 1790, 1556, 1052 cm⁻¹; $[\alpha]^{20}_{D}$ = +68.5 (*c*. 1.342, CHCl₃); ¹H NMR (400 MHz, CDCl₃) & 7.22 (ddt, *J* = 18.5, 11.7, 4.2 Hz, 6H), 7.10-7.04 (m, 2H), 6.94-6.85 (m, 1H), 5.24 (dd, *J* = 13.4, 3.8 Hz, 1H), 4.87 (dd, *J* = 13.2, 12.0 Hz, 1H), 3.92 (dd, *J* = 11.8, 3.8 Hz, 1H), 3.75 (s, 3H), 2.91 (ddd, *J* = 17.7, 10.6, 7.2 Hz, 1H), 2.42 (ddd, *J* = 17.8, 10.3, 5.5 Hz, 1H), 2.30 (ddd, *J* = 12.3, 10.4, 7.2 Hz, 1H), 1.75 (ddd, *J* = 12.3, 10.8, 5.5 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) & 203.2, 130.2, 130.0, 129.0, 128.8, 128.4, 127.9, 120.4, 116.3, 74.8, 72.3, 55.4, 44.9, 43.5, 20.9; HRMS (ESI): calcd for C₁₉H₁₉NNaO₄S: 380.0927 (M+Na⁺), found: 380.0915. HPLC analysis (Phenomenex Lux-1 column, *i*-PrOH/Hexane 5:95, 1.0 mL/min. λ = 254 nm, t_R (major) = 27.82 min, t_R (minor) = 25.75 min.

DOI: 10.1039/C6OB00160B

ARTICLE

(2*R*, *I*'S)-2-(*p*-tolylthio)-2-(2-nitro-1-phenylethyl)cyclobutanone anti-**3g.** White solid. Mp = 83-85 °C; FTIR (ATR) v: 2989, 2354, 1790, 1585, 1440 cm⁻¹; $[\alpha]^{21}_{D}$ = +71.0 (*c*. 1.07, CHCl₃); ¹H NMR (500 MHz, CDCl₃) δ: 7.37 (d, *J* = 8.0 Hz, 2H), 7.26-7.20 (m, 3H), 7.20-7.14 (m, 2H), 7.12 (d, *J* = 8.0 Hz, 2H), 5.26 (dd, *J* = 13.4, 3.8 Hz, 1H), 4.88 (dd, *J* = 13.3, 12.0 Hz, 1H), 3.87 (dd, *J* = 11.8, 3.7 Hz, 1H), 2.89 (ddd, *J* = 17.9, 10.5, 7.3 Hz, 1H), 2.43 (ddd, *J* = 17.9, 10.3, 5.6 Hz, 1H), 2.31 (s, 1H), 2.30-2.24 (m, 1H), 1.75 (ddd, *J* = 12.4, 10.7, 5.6 Hz, 1H); ¹³C NMR (125 MHz, CDCl₃) δ: 203.0, 140.4, 136.2, 133.9, 130.1, 129.0, 128.8, 128.3, 125.5, 74.9, 72.5, 45.0, 43.5, 21.3, 20.9; HRMS (ESI): calcd for C₁₉H₁₉NNaO₃S: 364.0978 (M+Na⁺), found: 364.0965. HPLC analysis (Phenomenex Lux-1 column, *i*-PrOH/Hexane 5:95, 1.0 mL/min. λ = 254 nm, t_R (major) = 37.25 min, t_R (minor) = 32.88 min.

(2R, 1'S)-2-(1-(4-fluorophenyl)-2-nitroethyl)-2-(phenylthio)

cyclobutanone anti-4**b**. Colourless oil. FTIR (neat) cm⁻¹ v: 2889, 2350, 1789, 1562, 1436, 1257 cm⁻¹; $[α]^{20}{}_D = +61.2$ (*c*. 0.20, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ: 7.49 (d, J = 6.9 Hz, 1H), 7.41-7.30 (m, 4H), 7.20-7.13 (m, 2H), 6.94 (dd, J = 11.9, 5.2 Hz, 2H), 5.25 (dd, J = 13.4, 3.7 Hz, 1H), 4.87-4.78 (m, 1H), 3.87 (dd, J = 11.9, 3.6 Hz, 1H), 2.99-2.88 (m, 1H), 2.53-2.43 (m, 1H), 2.31-2.19 (m, 1H), 1.78 (ddd, J = 12.5, 10.7, 5.6 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ: 202.9, 162.5 (d, J = 248.3 Hz), 136.1, 130.7 (d, J = 8.1 Hz), 130.1, 129.5, 129.4, 128.9, 115.9 (d, J = 21.6 Hz), 74.9, 72.3, 44.5, 43.5, 20.9; HRMS (ESI): calcd for C₁₈H₁₆FNNaO₃S: 368.0727 (M+Na⁺), found: 368.0716. HPLC analysis (Phenomenex Lux-1 column, *i*-PrOH/Hexane 5:95, 1.0 mL/min. $\lambda = 254$ nm, t_R (major) = 17.63 min, t_R (minor) = 18.64 min.

(2R,1'S)-2-(1-(4-chlorophenyl)-2-nitroethyl)-2-(phenylthio)

cyclobutanone anti-4c. Yellow oil. FTIR (neat) v: 2967, 2340, 1789, 1558, 1442, 1378 cm⁻¹; $[\alpha]^{25}_{D} = +53.4$ (c. 0.22, CHCl₃); ¹H NMR (500 MHz, CDCl₃) δ: 7.48-7.45 (m, 2H), 7.40-7.35 (m, 5H), 6.95 (d, J = 8.4 Hz, 2H), 5.23 (dd, J =13.4, 3.9 Hz, 1H), 5.00 (dd, J = 13.3, 11.6 Hz, 1H), 3.76 (dd, J = 11.5, 3.7 Hz, 1H), 2.92 (ddd, J = 18.3, 10.5, 7.7 Hz, 1H), 2.45 (ddd, J = 18.3, 10.1, 5.6 Hz, 1H), 1.96 (ddd, J = 12.4, 10.7, 5.6Hz, 1H), 1.86 (ddd, J = 12.5, 10.2, 7.8 Hz, 1H); ¹³C NMR (125 MHz, CDCl₃) δ: 205.2, 136.1, 134.7, 132.3, 130.3, 130.0, 129.5, 129.3, 122.6, 76.4, 73.3, 48.1, 43.7, 21.9; HRMS (ESI): calcd for $C_{18}H_{16}CINNaO_3S$: 384.0432 (M+Na⁺), found: 384.384.0426. HPLC analysis (Phenomenex Lux-1 column, i-PrOH/Hexane 5:95, 1.0 mL/min. $\lambda = 254$ nm, t_R (major) = 25.41 min, t_R (minor) = 21.935 min.

(2R, 1'S)-2-(1-(4-bromophenyl)-2-nitroethyl)-2-(phenylthio)

cyclobutanone anti-4d. Yellow solid, Mp = 121-124 °C; FTIR (ATR) v: 2923, 2356, 1790, 1460, 1045 cm⁻¹; $[\alpha]^{26}{}_{\rm D}$ = +63.4 (*c*. 0.43, CHCl₃); ¹H NMR (500 MHz, CDCl₃) &: 7.46 (m, 2H), 7.39-7.34 (m, 5H), 6.95 (d, *J* = 5.0 Hz, 2H), 5.23 (dd, 1H, *J* = 5.0, 10.0 Hz), 5.00 (dd, *J* = 10.0, 15.0 Hz, 1H), 3.76 (dd, *J* = 5.0, 15.0 Hz, 1H), 2.95-2.89 (m, 1H), 2.47-2.45 (m, 1H), 1.99-1.93 (m, 1H), 1.91-1.85 (m, 1H); ¹³C NMR (125 MHz, CDCl₃) &: 205.2, 136.1, 134.7, 132.3, 130.3, 130.0, 129.5, 129.3, 122.6,

Published on 26 February 2016. Downloaded by KUNGL TEKNISKA HOGSKOLAN on 29/02/2016 09:26:46.

ARTICLE

41.76 min.

(2R,1'S)-2-(1-(4-methoxyphenyl)-2-nitroethyl)-2-(phenylthio) cyclobutanone anti-**4e.** Yellow oil. FTIR (neat) v: 2996, 2354, 1790, 1554, 1033 cm⁻¹; $[\alpha]^{21}_{D} = +74.3$ (c. 1.22, CHCl₃); ¹H NMR (500 MHz, CDCl₃) δ: 7.52-7.48 (m, 2H), 7.38-7.29 (m, 3H), 7.09 (d, J = 8.7 Hz, 2H), 6.77-6.73 (m, 2H), 5.22 (dd, J = 13.2, 3.9 Hz, 1H), 4.82 (dd, J = 13.2, 12.0 Hz, 1H), 3.83 (dd, J = 11.9, 3.8 Hz, 1H), 3.69 (s, 2H), 2.88 (ddd, J= 18.0, 10.6, 7.3 Hz, 1H), 2.44 (ddd, J = 18.0, 10.3, 5.7 Hz, 1H), 2.29 (ddd, J = 12.3, 10.3, 7.3 Hz, 1H), 1.75 (ddd, J = 12.4, 10.6, 5.7 Hz, 1H); ¹³C NMR (125 MHz, CDCl₃) δ: 203.5, 159.6, 136.2, 130.3, 130.0, 129.4, 125.7, 114.4, 107.4, 75.2, 72.7, 55.3, 44.5, 43.6, 21.0; HRMS (ESI): calcd for C₁₉H₁₉NNaO₄S: 380.0927 (M+Na⁺), found: 380.0925. HPLC analysis (Phenomenex Lux-1 column, i-PrOH/Hexane 2:98, 1.0 mL/min. λ = 254 nm, t_R (major) = 43.07 min, t_R (minor) = 39.23 min.

76.4, 73.3, 48.1, 43.7, 21.9; HRMS (ESI): calcd for $C_{18}H_{16}BrNNaO_3S$: 427.9926 (M+Na⁺), found: 427.9913. HPLC

analysis (Chiralpack AD-H column, i-PrOH/Hexane 10:90, 1.0

(2*R*, *I*'S)-2-(2-nitro-1-p-tolylethyl)-2-(phenylthio)cyclobutanone anti-**4f.** Yellow solid, Mp = 92-95 °C; FTIR (ATR) v: 2899, 2348, 1790, 1380, 1061 cm⁻¹; $[α]^{27}_{D}$ = +26.5 (*c*. 0.62, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ: 7.59-7.48 (m, 1H), 7.38 (ddd, *J* = 10.6, 6.4, 2.2 Hz, 2H), 7.08 (s, 2H), 5.26 (dd, *J* = 13.3, 3.8 Hz, 1H), 4.89 (dd, *J* = 13.3, 11.9 Hz, 1H), 3.88 (dd, *J* = 11.8, 3.8 Hz, 1H), 2.91 (ddd, *J* = 17.7, 10.6, 7.2 Hz, 1H), 2.47 (ddd, *J* = 17.8, 10.3, 5.6 Hz, 1H), 2.39-2.30 (m, 1H), 2.26 (s, 2H), 1.79 (ddd, *J* = 12.4, 10.6, 5.6 Hz, 1H), 0.86 (ddt, *J* = 10.3, 6.5, 5.0 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ: 203.4, 138.3, 136.2, 130.8, 130.0, 129.7, 129.4, 129.0, 75.1, 72.6, 44.8, 43.6, 21.2, 21.1; HRMS (ESI): calcd for C₁₉H₁₉NNaO₃S: 364.0983 (M+Na⁺), found: 364.1035. HPLC analysis (Phenomenex Lux-1 column, *i*-PrOH/Hexane 5:95, 1.0 mL/min. λ = 254 nm, t_R (major) = 23.38 min, t_R (minor) = 20.80 min.

(2R,1'S)-2-(1-(4-(benzyloxy)phenyl)-2-nitroethyl)-2-

(*phenylthio*) *cyclobutanone anti*-**4g.** White solid. Mp = 62-63 °C; FTIR (ATR) v: 2964, 1791, 1586, 1253 cm⁻¹; $[\alpha]^{27}_{D}$ = +31.2 (*c*. 0.71, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ: 7.47 (d, *J* = 6.4 Hz, 2H), 7.38-7.24 (m, 8H), 6.98 (d, *J* = 8.6 Hz, 2H), 6.84 (d, *J* = 8.6 Hz, 2H), 5.25 (dd, *J* = 13.2, 3.8 Hz, 1H), 5.01 (dd, *J* = 22.2, 9.2 Hz, 1H), 4.93 (s, 2H), 3.74 (dd, *J* = 11.5, 3.8 Hz, 1H), 2.82 (ddd, *J* = 18.2, 10.2, 7.9 Hz, 1H), 2.39 (ddd, *J* = 18.3, 9.9, 6.2 Hz, 1H), 2.00-1.84 (m, 2H); ¹³C NMR (100 MHz, CDCl₃) δ: 206.3, 158.9, 136.7, 136.2, 130.0, 129.9, 129.7, 129.6, 128.7, 128.2, 127.8, 127.6, 115.4, 76.8, 73.9, 70.2, 48.3, 43.8, 22.1; HRMS (ESI): calcd for C₂₅H₂₃NNaO₄S: 456.1240 (M+Na⁺), found: 456.1286. HPLC analysis (chiralpack-OJ column, *i*-PrOH/Hexane 25:75, 1.0 mL/min. λ = 254 nm, t_R (major) = 31.18 min, t_R (minor) = 46.53 min.

(2R,1'S)-2-(1-(benzo[d][1,3]dioxol-6-yl)-2-nitroethyl)-2-

(phenylthio) cyclobutanone anti-**4h.** Yellow oil. FTIR (neat) v: 2994, 1791, 1378, 1256 cm⁻¹; $[\alpha]^{24}_{D}$ = +78.6 (c. 0.54, CHCl₃); ¹H NMR (500 MHz, CDCl₃) δ : 7.52 (dd, J = 7.5, 6.0 Hz, 1H), 7.40-7.31 (m, 2H), 7.04 (dd, J = 8.0, 1.5 Hz, 1H), 6.96 (d, J = 1.6 Hz, 1H), 6.83 (d, J = 8.0 Hz, 1H), 6.71-6.65 (m, 2H), 5.90 (s, 2H), 5.23 (dd, J = 13.3, 3.8 Hz, 1H), 4.81 (dd, J = 13.2, 12.1 Hz, 1H), 3.83 (dd, J = 11.9, 3.8 Hz, 1H), 3.00-2.86 (m, 1H), 2.62-2.49 (m, 1H), 2.33 (ddd, J = 12.4, 10.3, 7.5 Hz, 1H), 1.81 (ddd, J = 12.4, 10.7, 5.7 Hz, 1H); ¹³C NMR (125 MHz, CDCl₃) δ : 203.2, 147.9, 147.6, 136.0, 129.9, 129.3, 129.1, 127.3, 122.5, 109.3, 108.4, 101.3, 75.1, 72.4, 44.9, 43.5, 21.1; HRMS (ESI): calcd for C₁₉H₁₇NNaO₅S: 394.0720 (M+Na⁺),

 $16.99 \text{ min}, t_{R} \text{ (minor)} = 26.84 \text{ min}.$ (2R, 1'S)-2-(1-(2, 4-dichlorophenyl)-2-nitroethyl)-2-(phenylthio) cyclobutanone anti-4i. Yellow viscous oil. FTIR (neat) v: 2996, 2360, 1788, 1436, 1058 cm⁻¹; $[\alpha]^{26}_{D} = +47.4$ (c. 0.33, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ: 7.64-7.55 (m, 2H), 7.49-7.40 (m, 4H), 7.27-7.22 (m, 1H), 7.18 (d, *J* = 8.5 Hz, 1H), 5.44 (dd, J = 13.8, 3.9 Hz, 1H), 5.21 (dd, J = 13.8, 11.4 Hz, 1H), 4.62 (dd, J = 11.4, 3.9 Hz, 1H), 3.05 (ddd, J = 18.4, 10.7, 7.7 Hz, 1H), 2.51 (ddd, J = 18.2, 10.2, 5.6 Hz, 1H), 2.16-2.08 (m, 1H), 1.79 (ddd, J = 12.5, 10.2, 7.9 Hz, 1H); ¹³C NMR (125 MHz, CDCl₃) δ: 205.0, 136.8, 136.3, 135.0, 132.4, 130.4, 130.3, 129.7, 129.3, 128.8, 128.1, 107.5, 75.6, 73.5, 44.1, 43.9, 22.1; HRMS (ESI): calcd for C₁₈H₁₅Cl₂NNaO₃S: 418.0042 (M+Na⁺), found: 418.0025. HPLC analysis (Phenomenex Lux-1 column, *i*-PrOH/Hexane 3:97, 1.0 mL/min. λ = 254 nm, t_R $(major) = 30.23 \text{ min}, t_R (minor) = 40.38 \text{ min}.$

found: 394.0701. HPLC analysis (Phenomenex Lux-1 column,

i-PrOH/Hexane 5:95, 1.0 mL/min. $\lambda = 254$ nm, t_R (major) =

(2R, 1'S)-2-(1-(2-chloro-6-fluorophenyl)-2-nitroethyl)-2-

(phenylthio) cyclobutanone anti-4j. Yellow oil. FTIR (neat) v: 2999, 2355, 1789, 1484, 1258 cm⁻¹; $[\alpha]^{24}_{D} = +38.2$ (c. 0.22, CHCl₃); ¹H NMR (500 MHz, CDCl₃) δ : 7.62 (d, J = 7.8 Hz, 2H), 7.45 (dd, J = 10.7, 3.8 Hz, 1H), 7.40 (t, J = 7.5 Hz, 2H), 7.17 (dd, J = 11.2, 4.9 Hz, 1H), 7.13 (t, J = 7.8 Hz, 1H), 6.97 (dd, J = 11.4, 8.2 Hz, 1H), 5.27 (dd, J = 13.2, 4.3 Hz, 1H), 5.19 (t, J = 11.5 Hz, 1H), 4.65 (dd, J = 10.8, 4.3 Hz, 1H), 3.34-3.23 (m, 1H), 2.80 (dd, J = 10.2, 5.4 Hz, 1H), 2.76-2.67 (m, 1H), 1.93-1.82 (m, 1H); ¹³C NMR (100 MHz, CDCl₃) δ: 200.5, 163.0, 160.5, 136.7, 130.2, 130.1, 129.2, 126.6, 120.6 (d, J = 16.1 Hz), 115.1 (d, J = 24.6 Hz), 73.5 (d, J = 8.6 Hz), 72.4, 43.5, 39.8, 21.1 (d, J = 9.4 Hz); HRMS (ESI): calcd for C₁₈H₁₅ClFNNaO₃S: 402.0337 (M+Na⁺), found: 402.0324. HPLC analysis (Chiralpack OJ column, i-PrOH/Hexane 5:95, 1.0 mL/min. $\lambda = 254$ nm, t_R (major) = 20.78 min, t_R (minor) = 18.44 min.

(2R,1'S)-2-(2-nitro-1-phenylethyl)-2-

(phenylsulfonyl)cyclobutanone anti-**8**. White solid, Mp = 160-162 °C; FTIR (ATR) v: 3000, 1790, 1480, 1440 cm⁻¹; [α]²³_D = +51.4 (c. 0.51, CHCl₃); ¹H NMR (500 MHz, CDCl₃) δ: 8.00 (d, *J* = 7.5 Hz, 2H), 7.82 (t, *J* = 7.5 Hz, 1H), 7.70 (t, *J* = 7.8 Hz, 2H), 7.28 (d, *J* = 1.5 Hz, 1H), 7.27 (d, *J* = 1.7 Hz, 2H), 7.09 (dd, *J* = 6.5, 2.8 Hz, 2H), 5.42 (dd, *J* = 13.7, 3.7 Hz, 1H), 4.99 (dd, *J* = 13.6, 12.0 Hz, 1H), 4.07 (dd, *J* = 11.9, 3.7 Hz, 1H), 3.08-2.99 (m, 1H), 2.99-2.90 (m, 1H), 2.61-2.52 (m, 1H), 2.48-2.38 (m, 1H); ¹³C NMR (126 MHz, CDCl₃) δ: 197.4, 135.1, 134.7, 132.3, 130.3, 129.5, 129.2, 129.0, 128.7, 88.9, 74.3, 46.8, 42.9, 15.3; HRMS (ESI): calcd for C₁₈H₁₇NNaO₅S: 382.0725 (M+Na⁺), found: 382.0718; HPLC analysis (Chiralpack AD-H column, *i*-PrOH/Hexane 5:95, 1.0 mL/min. $\lambda = 254$ nm, t_R (major) = 22.35 min, t_R (minor) = 19.57 min.

DOI: 10.1039/C6OB00160B

Journal Name

(2R,1'S)-2-(2-nitro-1-phenylethyl)-2-(phenylthio)

anti-9. Yellow cyclopentanone oil. FTIR (neat) v: 2998, 1747, 1485, 1254 cm⁻¹; $[\alpha]^{27}_{D} = +42.3$ (c. 0.12, CHCl₃); ¹H NMR (500 MHz, CDCl₃) δ: 7.51-7.46 (m, 2H), 7.42 (d, J = 7.5 Hz, 2H), 7.26-7.15 (m, 5H), 5.71 (dd, J = 13.2, 3.4 Hz, 1H), 4.99 (t, J = 12.8 Hz, 1H), 4.04 (dd, J = 12.4, 3.3 Hz, 1H), 2.44 (dd, J = 18.3, 8.3 Hz, 1H), 2.37-2.29 (m, 1H), 2.17-2.05 (m, 1H), 1.86 (dd, J = 13.6, 6.7 Hz, 1H), 1.79 (dd, J =13.6, 8.6 Hz, 1H), 1.72 (dd, J = 18.9, 9.2 Hz, 1H); ¹³C NMR (125 MHz, CDCl₃) δ: 204.9, 135.0, 131.8, 128.3, 128.0, 127.1, 126.3, 125.9, 109.9, 74.0, 60.8, 43.0, 34.0, 28.7, 16.0; HRMS (ESI): calcd for C₁₉H₁₉NNaO₃S: 364.0978 (M+Na⁺), found: 364.0969. HPLC analysis (Chiralpack AD-H column, i-PrOH/Hexane 5:95, 1.0 mL/min. $\lambda = 254$ nm, t_R (major) = $30.50 \text{ min}, t_{R} \text{ (minor)} = 24.96 \text{ min}.$

(2R,1'S)-2-(2-nitro-1-phenylethyl)-2-(phenylthio)

cyclohexanone anti-**10.** White solid. Mp = 92-94 °C; FTIR (ATR) v: 2964, 1732, 1556, 1261 cm⁻¹; [α]²⁷_D = +37.8 (*c*. 0.30, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ: 7.41-7.29 (m, 4H), 7.25-7.06 (m, 6H), 5.49 (dd, *J* = 13.0, 3.2 Hz, 1H), 4.84 (t, *J* = 12.7 Hz, 1H), 4.16 (dd, *J* = 12.4, 3.1 Hz, 1H), 2.99 (ddd, *J* = 15.8, 13.9, 6.3 Hz, 1H), 2.31-2.19 (m, 1H), 2.18-2.10 (m, 1H), 2.06-1.93 (m, 1H), 1.87 (ddd, *J* = 25.0, 11.0, 2.7 Hz, 1H), 1.70-1.61 (m, 2H), 1.39-1.25 (m, 1H); ¹³C NMR (100 MHz, CDCl₃) δ: 201.5, 135.8, 134.8, 131.8, 130.9, 130.2, 129.4, 128.1, 127.7, 64.6, 56.4, 46.3, 37.8, 32.5, 25.0, 20.7; HRMS (ESI): calcd for $C_{20}H_{21}NNaO_3S$: 378.1134 (M+Na⁺), found: 378.1172. HPLC analysis (Chiralpack AD-H column, *i*-PrOH/Hexane 5:95, 1.0 mL/min. λ = 254 nm, t_R (major) = 20.61 min, t_R (minor) = 19.18 min.

Notes and references

- (a) J. C. Namyslo and D. E. Kaufmann, *Chem. Rev.*, 2003, **103**, 1485; (b) E., Lee-Ruff and G. Mladenova, *Chem. Rev.*, 2003, **103**, 1449.
- 2 *The Chemistry of Cyclobutanes*, ed. Z. Rappoport and J. F. Liebman, John Wiley & Sons, Chichester, 2005.
- 3 (a) F. Secci, A. Frongia and P. P. Piras, *Molecules*, 2013, 18, 15541; (b) A. M. Bernard, A. Frongia, F. Secci and P. P. Piras, *Chem. Commun.*, 2005, 3853; (c) F. Secci, A. Frongia, J. Ollivier and P. P. Piras, *Synthesis*, 2007, 999; (d) A. Frongia, C. Girard, J. Ollivier, P. P. Piras and F. Secci, *Synlett*, 2008, 2823; (e) F. Secci, A. Frongia, M. G. Rubanu, M. L. Sechi, G. Sarais, M. Arca and P. P. Piras, *Eur. J. Org. Chem.*, 2014, 6659.
- 4 (a) R. D. Miller and D. R. McKean, *Tetrahedron. Lett.*, 20, 1979, 1003; (b) X. Zhang and Z. W. Li, *Synth. Commun.*, 2006, 36, 249.
- 5 (a) M. Murakami, K. Takahashi, H. Amii and Y. Ito, J. Am. Chem. Soc., 1997, 119, 9307; (b) Z. Chai and T. J. Rainey, J. Am. Chem. Soc., 2012, 134, 3615; (c) E. Zhang, C. A. Fan, Y. Q. Tu, F. M. Zhang and Y. L. Song, J. Am. Chem. Soc., 2009, 131, 14626; (d) D. C. Moebius and J. S. Kingsbury, J. Am. Chem. Soc., 2009, 131, 878; (e) J. A. Dabrowski, D. C. Moebius, A. J. Wommack, A. F. Kornahrens and J. S. Kingsbury, Org. Lett., 2010, 12, 3598; (f) M. Yoshida, H. Nemoto and M. Ihara, Tetrahedron Lett., 1999, 40, 8583.
- 6 (a) J. M. Conia and M. J. Robson, Angew. Chem., Int. Ed., 1975, 14, 473; (b) Cleavage of Carbon-Carbon Single Bonds by Transition Metals, M. Murakami and N. Chatani, Wiley-VCH, Weinheim, 2015.

- 7 (a) G. Alberti, A. M. Bernard, A. Frongia, P. P. Piras, F. Secci and M. Spiga, *Synlett*, 2006, 2241; (b) A. M. Bernard, E. Cadoni, A. Frongia, P. P. Piras and F. Secci, *Org. Lett.*, 2002, 4, 2565; (c) A. M Bernard, C. Floris, A. Frongia, P. P Piras and F. Secci, *Tetrahedron*, 2004, 60, 449.
- 8 (a) N. Sasakura, K. Nakano, Y. Ichikawa and H. Kotsuki, RSC Adv., 2012, 2, 6135; (b) Y. Imada, H. Iida, S.-I. Murahashi and T. Naota, Angew. Chem., Int. Ed., 2005, 44, 1704; (c) S.-I. Murahashi, S. Ono and Y. Imada, Angew. Chem., Int. Ed., 2002, 41, 2366; (d) S. Xu, Z. Wang, X. Zhang, X. Zhang and K. Ding, Angew. Chem., Int. Ed., 2008, 47, 2840.
- 9 (a) A. Mastracchio, A. A. Warkentin, A. M. Walji and D. W. C. MacMillan, *Proc. Nat. Acad. Sci. USA*, 2010, **107**, 20648; (b) L. Zhang, L. Cui, X. Li, J. Li, S. Luo and J.-P. Cheng, *Chem. Eur. J.*, 2010, **16**, 2045; (b) C. M. Reeves, C. Eidamshaus, J. Kim and B. M. Stoltz, *Angew. Chem., Int. Ed.*, 2013, **52**, 6718.
- (a) X. Ma, C.-S. Da, L. Yi, Y.-N. Jia, Q.-P. Guo, L.-P. Che, F.-C. Wu, J.-R. Wang and W.-P. Li, *Tetrahedron: Asymmetry*, 2009, 20, 1419; (b) P. Kotrusz, I. Kmenttová, B. Gotov, Š. Toma and E. Solčániová, *Chem. Commun.*, 2002, 2510; (c) B. Alcaide, P. Almendros and A. Luna, *Tetrahedron*, 2007, 63, 3102.
- (a) E. Veverková, J. Štrasserová, R. Šebesta and Š. Toma, *Tetrahedron: Asymmetry*, 2010, **21**, 58; (b) E. Alza, C. Rodríguez-Escrich, S. Sayalero, A. Bastero and M. A. Pericàs, *Chem. Eur. J.*, 2009, **15**, 10167; (c) A. J. A. Cobb, D. M. Shaw and S. V. Ley, *Synlett*, 2004, 558.
- 12 (a) D. J. Aitken, F. Capitta, A. Frongia, J. Ollivier, P. P. Piras and F. Secci, *Synlett*, 2012, 727; (b) D. J. Aitken, F. Capitta, A. Frongia, D. Gori, R. Guillot, J. Ollivier, P. P. Piras, F. Secci and M. Spiga, *Synlett*, 2011, 712.
- 13 F. Capitta, A. Frongia, J. Ollivier, P. P. Piras and F. Secci, Synlett, 2011, 89.
- (a) D. J. Aitken, P. Caboni, H. Eijsberg, A. Frongia, R. Guillot, J. Ollivier, P. P. Piras and F. Secci, Adv. Synth. Catal., 2014, 356, 941; (b) N. Melis, L. Ghisu, R. Guillot, P. Caboni, F. Secci, D. J. Aitken and A. Frongia, Eur. J. Org. Chem., 2015, 4358; (c) N. Melis, F. Secci, T. Boddaert, D. J. Aitken and A. Frongia, Chem. Commun., 2015, 51, 15272.
- 15 D. J. Aitken, A. M. Bernard, F. Capitta, A. Frongia, R. Guillot, J. Ollivier, P. P. Piras, F. Secci and M. Spiga, *Org. Biomol. Chem.*, 2012, **10**, 5045.
- 16 A. J. A. Cobb, D. M. Shaw, D. A. Longbottom, J. B. Gold and S. V. Ley, Org. Biomol. Chem., 2005, 3, 84.
- 17 D. Mailhol, M. del Mar Sanchez Duque, W. Raimondi, D. Bonne, T. Constantieux, Y. Coquerel and J. Rodriguez, Adv. Synth. Catal., 2012, 354, 3523.
- 18 F. Capitta, A. Frongia, J. Ollivier, D. J. Aitken, F. Secci, P. P. Piras and R. Guillot, Synlett, 2015, 26, 123.
- Mechanisms of Atmospheric Oxidation of the Oxygenates, J. G. Calvert, A. Mellouki, J. J. Orlando, M. J. Pilling and T. J. Wallington, Oxford University Press, New York, 2011.
- 20 (a) The Proton in Chemistry, R. P. Bell, Cornell University Press, Ithica, 1959; (b) F. G. Bordwell, X. Zhang and M. S. Alnajjar, J. Am. Chem. Soc., 1992, **114**, 7623; (c) F. G. Bordwell, H. E. Fried, D. L. Hughes, T. Y. Lynch, A. V. Satish and Y. E. Whang, J. Org. Chem., 1990, **55**, 3330.
- 21 B. M. Trost, W. C. Vladuchick and A. J. Bridges, J. Am. Chem. Soc., 1980, **102**, 3548.
- 22 (a) P. Leiverend and C. Huard, *Sulfur Lett.*, 1990, **11**, 219; (b)
 A. Bury, H. A. Earl and C. J. M. Stirling, *J. Chem. Soc. Perkin Trans II*, 1987, **9**, 1281; (c) T. Wakanatsu, N. Miyachi, F. Ozuki and Y. Ban, *Heterocycles*, 1987, **26**, 1445; (d) M. D. Lawlor, T. W. Lee and R. L. Danheiser, *J. Org. Chem.*, 2000, **65**, 4375.
- 23 A. F. Vaquer, A. Frongia, F. Secci, E. Tuveri, *RSC Adv.*, 2015, **5**, 96695.
- 24 A. M. Bernard, A. Frongia, P. P. Piras, F. Secci and M. Spiga, *Tetrahedron Lett.*, 2008, **49**, 3037.

This journal is C The Royal Society of Chemistry 20xx

25 The predominant formation of the *syn* adduct of **3a** using the tertiary amine catalyst **XI** is an indication that a different catalytic mechanism is probably operating, as would be expected since formation of an enamine intermediate (as suggested in Scheme 2 for catalyst **XII**) is precluded.



Scheme 3. Hypothetic transition states evoked for the formation of *anti*-**3a** (Ts2) and *syn*-**3a** (Ts3) using catalyst **XI.**

- 26 J. Duschmal and H. Wennemers, *Chem. Eur. J.*, 2012, **18**, 1111.
- 27 Details of the x-ray diffraction study of 8 are given in the SI document. CCDC 1442520 contains the crystallographic data for compound 8; these data can be obtained free of charge from the Cambridge Crystallographic Data Centre via http://www.ccdc.cam.ac.uk/Community/Requestastructure.
- 28 A. G. Wenzel and E. N. Jacobsen, J. Am. Chem. Soc., 2002, 124, 12964.
- 29 J. J. Eisch, J. E. Galle and L. E. Hallenbeck, J. Org. Chem., 1982, 47, 1608.

Page 10 of 10