

# First total synthesis of carbazomycin C and D<sup>1</sup>

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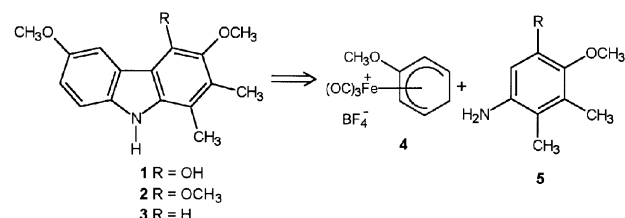
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**The first total synthesis of the antibiotic carbazomycins C and D using a convergent iron-mediated construction of the carbazole framework is described.**

The carbazomycins isolated from *Streptovercillium ehimense* were the first antibiotics to contain a carbazole framework.<sup>2</sup> Moreover, the carbazomycins B and C were shown to inhibit 5-lipoxygenase.<sup>2h</sup> The biological activity and unusual structure of the carbazomycins stimulated the development of diverse strategies directed towards their total synthesis.<sup>3</sup> We reported a novel methodology, *via* iron-mediated consecutive C–C and C–N bond formation, for the coupling of cyclohexa-1,3-diene and the corresponding arylamine, which was applied to the total synthesis of carbazomycin A, B and E.<sup>4</sup> Here we describe an extension of our method which is used for the first total synthesis of carbazomycin C **1**,<sup>2f</sup> carbazomycin D **2**<sup>2f</sup> and the non-natural 4-deoxycarbazomycin C **3**. Retrosynthetic analysis of the carbazomycins **1–3** based on the iron-mediated construction of the carbazole framework leads to tricarbonyl-[3-methoxy-(1,2,3,4,5-η)-cyclohexadienyl]iron tetrafluoroborate **4** and the arylamines **5** (Scheme 1). Complex **4** is readily

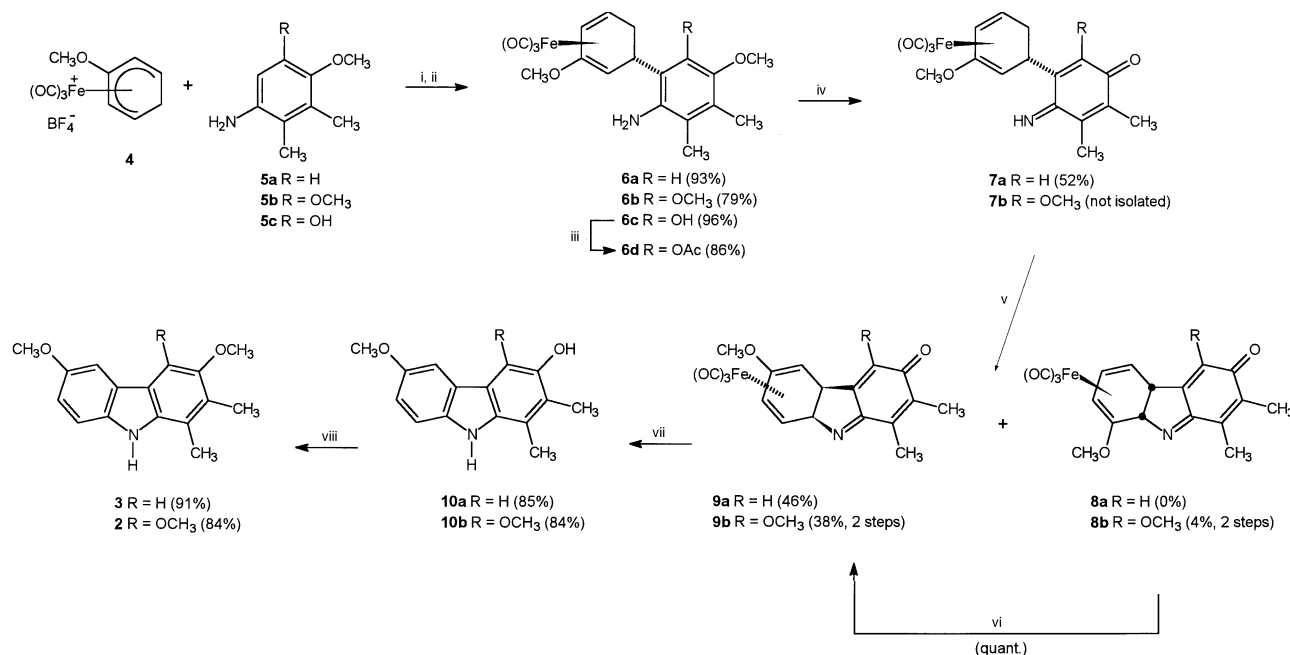
available in 3 steps from 1,3-dimethoxybenzene.<sup>5</sup> The arylamines **5** have been described by us in our previous studies.<sup>6,7</sup>

The synthesis of the alkaloids **2** and **3** was achieved by the iron-mediated quinone imine cyclization (Scheme 2). Electrophilic substitution of the arylamines **5a** and **5b** with **4** provided the complexes **6a** and **6b**. Chemoselective oxidation of the aromatic nucleus to afford **7a** and **7b** was achieved with commercial manganese dioxide.† Oxidative cyclization of the quinone imines using very active manganese dioxide<sup>8</sup> provided the stable tricarbonyliron-complexed 4b,8a-dihydrocarbazol-3-ones **8** and **9**. In the case of **7a** the desired 6-methoxy substituted regioisomer **9a** was exclusively isolated (46% yield), while **7b** gave minor quantities (4%) of the 8-methoxy isomer **8b** along with 38% of the 6-methoxy isomer **9b**. The regioselectivity of these cyclizations can be rationalized by the results previously obtained in our deuterium labelling studies.<sup>9</sup> Cyclizations with two-electron oxidants, such as manganese dioxide, initially lead to the product resulting from exclusive attack at C-4 of the cyclohexadiene ligand, which is represented by isomer **8** in the present case. However, a subsequent proton-catalysed isomerization of the kinetic product may occur. The isomerization of **8** to **9** is overriding due to the well-established regio-directing effect<sup>10</sup> of the 2-methoxy substituent of the intermediate cyclohexadienyl cation **11**, which forces the nucleophile to attack at the 5-position (Scheme 3). Consequently, the proton-catalysed isomerization of **8b** results in smooth conversion to

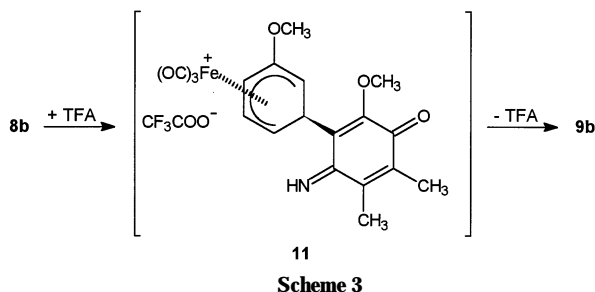


Scheme 1

† Manganese dioxide (precipitated active) from Merck-Schuchardt (art. 805958).

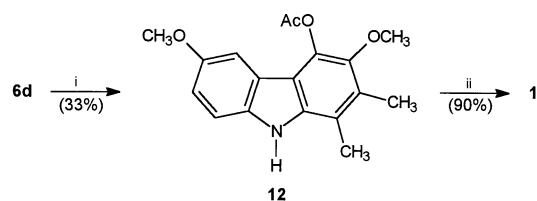


**Scheme 2** Reagents and conditions: i, MeCN, 25 °C; ii, dil. aq. NaOH; iii, Ac<sub>2</sub>O, CH<sub>2</sub>Cl<sub>2</sub>, pyridine, DMAP, 25 °C; iv, MnO<sub>2</sub>, CH<sub>2</sub>Cl<sub>2</sub>, 25 °C; v, very active MnO<sub>2</sub>, CH<sub>2</sub>Cl<sub>2</sub>, 25 °C; vi, cat. CF<sub>3</sub>CO<sub>2</sub>H, CH<sub>2</sub>Cl<sub>2</sub>, 25 °C; vii, Me<sub>3</sub>NO·2 H<sub>2</sub>O, Me<sub>2</sub>CO, 25–40 °C; viii, MeI, K<sub>2</sub>CO<sub>3</sub>, Me<sub>2</sub>CO, 56 °C



the desired 6-methoxy isomer **9b**. The tricarbonyliron-complexed **4b**, 8a-dihydrocarbazol-3-ones are useful synthetic precursors for 3-hydroxy-9*H*-carbazole alkaloids.<sup>11</sup> Thus, demetallation of the complexes **9a** and **9b** using trimethylamine *N*-oxide<sup>12</sup> afforded the 3-hydroxycarbazole derivatives **10a** and **10b**, which after *O*-methylation gave 4-deoxycarbazomycin C **3** and carbazomycin D **2**.<sup>‡</sup>

Carbazomycin C **1** was obtained *via* the iron-mediated arylamine cyclization. Electrophilic substitution of the aminophenol **5c** by **4** afforded the complex **6c** which was transformed into the acetate **6d** (Scheme 2). Oxidative cyclization of **6d** using very active manganese dioxide<sup>8</sup> to give the carbazole **12** followed by saponification of the ester provided carbazomycin C **1** (Scheme 4).§



**Scheme 4** Reagents and conditions: i, very active  $\text{MnO}_2$ ,  $\text{CH}_2\text{Cl}_2$ , 25 °C; ii,  $\text{NaOH}$ ,  $\text{H}_2\text{O}$ , reflux

‡ This synthesis affords carbazomycin D **2** in 5 steps and 23% overall yield based on **4** as pale yellow needles, mp 125 °C (from cyclohexane) (lit.,<sup>2f</sup> mp 129.5–130 °C, colourless needles from hexane–dichloromethane). All spectral data (UV, IR,  $^1\text{H}$  and  $^{13}\text{C}$  NMR, MS) are in full agreement with those reported for the natural product.

§ Carbazomycin C **1** was obtained in 4 steps and 25% overall yield based on **4** as colourless crystals, mp 190–191 °C (from hexane–ethyl acetate) (lit.,<sup>2f</sup> mp 198–198.5 °C, pale yellow prisms from hexane–ethyl acetate). All spectral data (UV, IR,  $^1\text{H}$  and  $^{13}\text{C}$  NMR, MS) are in full agreement with those reported for the natural product.

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## References

- Part 34 of transition metal complexes in organic synthesis. Part 33: H.-J. Knölker and C. Hofmann, *Tetrahedron Lett.*, 1996, **37**, 7947.
- (a) K. Sakano, K. Ishimaru and S. Nakamura, *J. Antibiot.*, 1980, **33**, 683; (b) K. Sakano and S. Nakamura, *J. Antibiot.*, 1980, **33**, 961; (c) M. Kaneda, K. Sakano, S. Nakamura, Y. Kushi and Y. Iitaka, *Heterocycles*, 1981, **15**, 993; (d) K. Yamasaki, M. Kaneda, K. Watanabe, Y. Ueki, K. Ishimaru, S. Nakamura, R. Nomi, N. Yoshida and T. Nakajima, *J. Antibiot.*, 1983, **36**, 522; (e) S. Kondo, M. Katayama and S. Marumo, *J. Antibiot.*, 1986, **39**, 727; (f) T. Naid, T. Kitahara, M. Kaneda and S. Nakamura, *J. Antibiot.*, 1987, **40**, 157; (g) M. Kaneda, T. Naid, T. Kitahara and S. Nakamura, *J. Antibiot.*, 1988, **41**, 602; (h) D. J. Hook, J. J. Yacobucci, S. O'Connor, M. Lee, E. Kerns, B. Krishnan, J. Matson and G. Hesler, *J. Antibiot.*, 1990, **43**, 1347.
- U. Pindur, *Chimia*, 1990, **44**, 406; J. Bergman and B. Pelcman, *Pure Appl. Chem.*, 1990, **62**, 1967; D. P. Chakraborty, in *The Alkaloids*, ed. A. Brossi, Academic Press, New York, 1993, vol. 44, p. 257; C. J. Moody, *Synlett*, 1994, 681.
- H.-J. Knölker, in *Organic Synthesis via Organometallics*, ed. K. H. Dötz and R. W. Hoffmann, Vieweg, Braunschweig, 1991, p. 119; H.-J. Knölker, *Synlett*, 1992, 371; H.-J. Knölker, in *Advances in Nitrogen Heterocycles*, ed. C. J. Moody, JAI Press, Greenwich, CT, 1995, vol. 1, p. 173.
- A. J. Birch, L. F. Kelly and D. J. Thompson, *J. Chem. Soc., Perkin Trans. 1*, 1981, 1006.
- H.-J. Knölker, M. Bauermeister, J.-B. Pannek, D. Bläser and R. Boese, *Tetrahedron*, 1993, **49**, 841.
- H.-J. Knölker and M. Bauermeister, *Helv. Chim. Acta*, 1993, **76**, 2500.
- A. J. Fatiadi, *Synthesis*, 1976, 65.
- H.-J. Knölker, F. Budei, J.-B. Pannek and G. Schlechtingen, *Synlett*, 1996, 587.
- A. J. Birch, K. B. Chamberlain, M. A. Haas and D. J. Thompson, *J. Chem. Soc., Perkin Trans. 1*, 1973, 1882; A. J. Pearson, *Acc. Chem. Res.*, 1980, **13**, 463; A. J. Pearson, *Metallo-organic Chemistry*, Wiley, Chichester, 1985, ch. 8.
- H.-J. Knölker, M. Bauermeister, J.-B. Pannek and M. Wolpert, *Synthesis*, 1995, 397.
- Y. Shvo and E. Hazum, *J. Chem. Soc., Chem. Commun.*, 1974, 336; H.-J. Knölker, *J. Prakt. Chem.*, 1996, **338**, 190.

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