# Quinolone Antibacterials Containing the New 7-[3-(1-Aminoethyl)-1-pyrrolidinyl] Side Chain: The Effects of the 1-Aminoethyl Moiety and Its Stereochemical Configurations on Potency and in Vivo Efficacy

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A series of stereochemically pure 7-[3-(1-aminoethyl)-1-pyrrolidinyl]-1,4-dihydro-4-oxoquinoline and 1,8-naphthyridine-3-carboxylic acids, with varied substituents at the 1-, 5-, and 8-positions, were synthesized to study the effects of the 7-[3-(1-aminoethyl)-1-pyrrolidinyl] moiety on potency and in vivo efficacy relative to the known 7-[3-(aminomethyl)-1-pyrrolidinyl] derivatives. The antibacterial efficacies of the target compounds and their relevant reference agents were determined in vitro using an assortment of Gram-negative and Gram-positive organisms and in vivo using Escherichia coli and Streptococcus pyogenes mouse infection models. The effects of the 7-[3-(1-aminoethyl)-1-pyrrolidinyl] moiety were also examined at the level of the target enzyme by employing a DNA-gyrase supercoiling inhibition assay. Selected compounds were further evaluated for potential phototoxic and clastogenic liabilities using a phototoxicity mouse model and an in vitro mammalian cell cytotoxicity assay. It was found that the differences in in vitro antibacterial activity between the stereoisomers were significantly greater than previously reported for other optically pure 3-substituted pyrrolidinyl side chains. Relative to their 7-[3-(aminomethyl)-1pyrrolidinyl] analogs, the (3R,1S)-3-(1-aminoethyl)pyrrolidines generally conferred a 2-4-fold increase in Gram-positive in vitro activity and an average of 10-fold improvement in oral efficacy. The level of phototoxicity and cytotoxicity of the product quinolones was ultimately determined by the combined influence of the 7-[3-(1-aminoethyl)-1-pyrrolidinyl] side chains and the other quinolone substituents. From this study, several compounds were identified with outstanding antibacterial activity and low degrees of phototoxicity and mammalian cell cytotoxicity. One such agent, 34F-R,S (PD 140248), showed the best overall blend of safety and efficacy.

The fluoroquinolones 1 (Table I) have emerged as one of the premier classes of antibacterials and have been the subject of many recent reviews. Their broad-spectrum activity, 2,3,5,6 coupled with their oral and parenteral dosage options, has enabled them to successfully challenge all of the conventional drugs employed for hospital and serious community infections.

Structure-activity relationships for the quinolones have established that the 7-position is largely responsible for controlling the spectrum of antibacterial coverage 10-12 and the overall pharmacokinetics. 10 All of the fluoroquinolones currently approved for use in man contain a 1-piperazinyl moiety at the 7-position ( $R_7$  in 1), with ciprofloxacin (2) and of loxacin (3) generally considered to be the most active of these agents.<sup>4,5</sup> Both have demonstrated good clinical success in widespread use. 7,9,13 These piperazinylquinolones are particularly effective for the treatment of infections caused by most Gram-negative organisms, including Pseudomonas, and for some infections caused by Staphylococcus and mycobacteria.2-6 None of the current quinolones are considered front-line therapy for infections caused by the Gram-positive species Streptococcus or Enterococcus, or when anaerobes are among the causative pathogens.5,6,14-16

Gram-positive activity, especially against Streptococcus and Enterococcus, has been obtained for quinolones containing the 3-amino-substituted-1-pyrrolidines 4a and 4b (Figure 1) at the R<sub>7</sub>-position.<sup>3,10,11</sup> Several of these pyrrolidinylquinolones, such as 5-9 in Table I, also retain good Gram-negative potency and therefore represent truly broad-spectrum agents.<sup>3</sup> Unfortunately, the pyrrolidinyl

$$H_2N$$
 $A_3$ 
 $A_4$ 
 $A_4$ 
 $A_5$ 
 $A_4$ 
 $A_5$ 
 $A_5$ 
 $A_6$ 
 $A_7$ 
 $A_8$ 
 $A_8$ 
 $A_8$ 
 $A_8$ 
 $A_8$ 
 $A_9$ 
 $A_9$ 

3R,1S = RS; 3R,1R = RR; 3S,1S = SS; 3S,1R = SR

Figure 1. Substituted pyrrolidines.

side chains have less intrinsic oral absorption than the corresponding piperazines and require additional modifications of the quinolone structure to provide good oral efficacy. Such changes include a halogen at  $C_8$  (X = CF or CCl) as in 5, 6, and 8, 18 an amino acid prodrug as in 7, 19 or a difluorophenyl group at  $N_1$  as in  $9.^{12}$  The best in vivo efficacy was obtained with the 8-fluoro derivatives, but recently the fluorine at the 8-position has been shown to be associated with a number of undesirable effects such as increased phototoxicity and cytotoxicity  $^{21c-f}$  via inhibition of mammalian topoisomerase.  $^{21a-c}$  Other methods for improving in vivo efficacy include the alkylation of the distal basic nitrogen of the side chain  $^{10,11}$  or the alkylation of the heterocyclic ring. Alkylations of the aminopyr-

Table I. Fluoroquinolones

HN_N-	СН		<b>△</b>	Н
CH3N N-		COCH CCH >		
		-COCH <sub>2</sub> C(CH <sub>3</sub> )-		Н
H <sub>2</sub> N	CF		$\triangle$	Н
EINH N-	CF		$\triangle$	н
AlaNH N-	N		<b>△</b>	Н
H <sub>2</sub> N N-	CCl		$\Delta$	н
H <sub>2</sub> N	N		↓ F	н
	EtNH N- AlaNH N- H <sub>2</sub> N N-	EINH N-  EINH N-  AlanH N-  H <sub>2</sub> N N-  H <sub>2</sub> N N-  N	EINH ON-  EINH ON-  AlanH N  H <sub>2</sub> N CCl  H <sub>2</sub> N N	EINH $N$ CF  AlanH $N$ CCI $H_2N$ $N$ CCI $H_2N$ $N$ $N$

Table II. Quinolone and Naphthyridine Intermediates Used To Prepare Final Products

		uents on 1			
no.	$\overline{\mathbf{R}_7}$	X	$R_1$	$\mathbf{R}_{5}$	ref
17	Cl	N	c-C <sub>3</sub> H <sub>5</sub>	Н	27
18	F	CH	$c-C_3H_5$	H	18, 31
19	F	$\mathbf{CF}$	$c-C_3H_5$	$\mathbf{NH}_2$	28
20	F	CF	$c-C_3H_5$	H -	18
21	F	CH	c-C <sub>3</sub> H <sub>5</sub>	$NH_2$	28
22	F	CCF <sub>3</sub>	$c-C_3H_5$	H	20
23	F	N	$2,4-F_2Ph$	H	29
24	F	$COCH_3$	$c-C_3H_5$	H	30
25	F	COCH <sub>3</sub>	$C_2H_5$	H	30
26	F	CF	$C_2H_5$	H	31

rolidines 4a,b at positions  $2,^{23}$   $3,^{24,25}$  and  $4^{24}$  have all been reported to improve oral efficacy.

In this paper, we have studied the effects of adding a methyl group to the methylene spacer in 4b to produce the 3-(1-aminoethyl)pyrrolidines 10F-H (Figure 1). 26 In each case, all four possible stereoisomers have been prepared and coupled to a variety of quinolones and naphthyridines, which were tested for in vitro and in vivo activity. We wish to report that the 3-(1-aminoethyl)-1-pyrrolidinyl substituent and its stereochemical orientations profoundly influence both the in vitro and in vivo potency. Furthermore, this study has produced quinolones with enhanced Gram-positive activity and oral efficacy superior to those containing a halogen at the 8-position.

# Chemistry

The quinolone nuclei used for the preparation of final products were prepared according to literature procedures as indicated in Table II. The synthesis of the pure diastereomers of the 3-(1-aminoethyl)pyrrolidines 10 is outlined in Scheme I. Specific experimental details have been published elsewhere.<sup>32</sup> The key step is the separation of the diastereomeric oximes 12 by column chromatography ( $\geq 99\%$ ). Reduction of 12-3R produced 13-3R,1S and 13-3R,1R, and reduction of 12-3S gave 13-3S,1R and

Scheme I

<sup>a</sup> An asterisk denotes pure chiral centers.

13-3S,1S. The individual diastereomers of 13 were separated and purified by chromatography to ≥20:1 diastereomeric purity as determined by HPLC and NMR (see Experimental Section). In addition, the absolute configuration of each diastereomer of 13 was determined by X-ray crystallography.  $^{32}$  The pyrrolidinones 13 were then elab-

orated into the BOC-NH, BOC-N-methyl, and N-ethylpyrrolidines 14, 15, and 16, which were then coupled to the quinolone nuclei according to Scheme II. For the 7-haloquinolones 17-23 and 26, the aromatic nucleophilic displacement by the pyrrolidine followed standard procedures. 18,20,28,29 to provide final quinolones of the generic form 27 (Scheme II). For the 8-methoxyquinolones 24 and 25, the quinolone acid or ester was first converted to the borate ester 28, which was then coupled using the pyrrolidines and Hunig's base. The borate ester products were hydrolyzed back to the quinolone acids using triethylamine in ethanol. Deprotection as necessary provided the final products 27. All of the new and reference quinolones 30-39 prepared for this study are listed in Table III with physical constants and methods of preparation. In the numbering system chosen to identify compounds, the number refers to the quinolone nuclei and the letter, A-H, represents the side chain at the 7-position. These side chains are depicted in Figure 2. The R,S designations represent the stereochemistry at the 3- and 1-positions, respectively.

### Biological Assays

In Vitro Antibacterial. All of the 7-[3-(1-aminoethyl)-1-pyrrolidinyl]quinolones (30-39) in Table III and selected reference agents were tested against 10 representative Gram-positive and Gram-negative organisms using standard microtitration techniques.33 Their minimum inhibitory concentrations (MICs, µg/mL) were averaged from multiple experiments and recorded in Table IV. To determine the effects of the stereochemical orientations of the side chains at the target enzyme level, select compounds were tested for their inhibition of DNA gyrase (Table V), which was isolated and purified from Escherichia coli H560.34 The initial cleavage method was employed, which measures the lowest concentration of drug (µg/mL) that will produce a detectable level of cleavage from relaxed bacterial Col E1 plasmid DNA, as visualized by agarose gel electrophoresis and staining with ethidium bromide.34 The gyrase data is averaged from multiple experiments and is accurate to  $\pm 50\%$ .

In Vivo Antibacterial. As previously reported,  $^{35}$  the in vivo potency was determined in acute, lethal systemic infections in female Charles River CD-1 mice and expressed as the median protective dose (PD $_{50}$ , mg/kg, Table IV). Sixteen mice per method of administration were employed. Single doses of compound were administered at the time of the bacterial challenge. The 95% confidence limits are generally  $\pm 30\%$ .

Phototoxicity. Compounds with the best in vitro and in vivo efficacy were tested for oral phototoxicity in

Figure 2. Pyrrolidinyl side-chain substituents  $(\mathbf{R}_7)$  used in this study.

Figure 3. Stereochemical relationship between the most active stereoisomers of F-H vs D.

depilated female CD-1 mice.<sup>20</sup> Each day, five animals/dose and control were exposed to UV-A radiation (320–400 nm) for 3 h duration following oral administration. Dosing was continued until a definite positive response was elicited or for a maximum of 4 days. Any detectable redness or erythema (as observed 24 h postirradiation) relative to control animals was considered a positive photoresponse. The results are reported in Table VI as the highest dose producing no effect.

Mammalian Cell Cytotoxicity. Promising quinolones were also evaluated in a mammalian cell cytotoxicity assay<sup>21d</sup> to estimate their clastogenic potential. <sup>21c-f</sup> The clonogenic cytotoxicity was determined in Chinese hamster V-79 cells. The cells were grown overnight and treated with drug for 3 h at 37 °C, at which time the compound-containing media was replaced with fresh media. The cells were then incubated for 5 days and examined for colony formation. The concentration of drug-inhibiting colony formation by 50% (IC<sub>50</sub>,  $\mu$ g/mL) relative to control was determined and recorded in Table VI.

# Results and Discussion

For each parent quinolone 30-39, Table IV shows the effects of the reference side chains A-E (see Figure 2) vs the stereoisomers of the 3-(1-aminoethyl)pyrrolidines F-H on antibacterial activity. In order to discern the effects derived from the new pyrrolidines F-H, it is important to first summarize the structure-activity relationships of the reference pyrrolidines A-E, and those of the quinolone nuclei themselves.

Of the five reference pyrrolidines A-E, the data demonstrate that the 3-aminopyrrolidine (A) confers the best Gram-negative potency, including that against *Pseudomonas* (see 30A vs 30B-E, 31A vs 31B-E). This fact, coupled with the generally good Gram-positive activity provided by A, made the (3-aminopyrrolidinyl) quinolones the most balanced, broad-spectrum agents in the quinolone family<sup>18</sup> (see 30A, 32A, 34A, and 38A). Spacing the 3-amino group away from the ring by a single methylene to give B increased the Gram-positive potency of the quinolones by 2-8-fold, while decreasing the Gram-negative MICs by a similar margin (30A vs 30B, 31A vs 31B, 35A vs 35B). Alkylation of the distal nitrogen of B to give pyrrolidines

Table III. Physical Properties and Method of Preparation for Quinolone Antibacterials Used in This Study

no. f	method of prep.º or ref	method of purif of final prodb	% yield <sup>c</sup>	mp (°C)	analysis formula (elements analyzed) <sup>d</sup>
		F	ملہ	O <sub>2</sub> H	
		/A.W\^		30	
		(877)	Ä		
30A	ref 18				
30B 30C	ref 18 ref 18				
30D 30E	ref 18 A	isoelect prec	47	257-259	C <sub>18</sub> H <sub>21</sub> FN <sub>4</sub> O <sub>3</sub> ·H <sub>2</sub> O (C,H,N)
<b>30F</b> - <i>R</i> , <i>S</i>	A	trit 2-PrOH	82	>300	$C_{18}H_{21}FN_4O_{3}HCl-0.22H_2O$ (C,H,N,F,Cl)
<b>30F</b> - <i>R</i> , <i>R</i> <b>30F</b> - <i>S</i> , <i>S</i>	A A	trit 2-PrOH trit 2-PrOH	62 36	>300 >300	C <sub>18</sub> H <sub>21</sub> FN <sub>4</sub> O <sub>3</sub> ·HCl (C,H,N,F,Cl) C <sub>18</sub> H <sub>21</sub> FN <sub>4</sub> O <sub>3</sub> ·HCl (C,H,N,F,Cl)
<b>30F</b> -S,R <b>30G</b> -R,S	A A	trit 2-PrOH recryst MeOH/Et <sub>2</sub> O	65 90	>300 >300	C <sub>18</sub> H <sub>21</sub> FN <sub>4</sub> O <sub>3</sub> ·HCl (C,H,N,F,Cl) C <sub>19</sub> H <sub>23</sub> FN <sub>4</sub> O <sub>3</sub> ·1.1HCl-0.25H <sub>2</sub> O (C,H,N,Cl)
30G-R,R	A	trit 2-PrOH	76	>300	$C_{19}H_{23}FN_4O_3\cdot HCl\cdot 0.75H_2O\ (C,H,N,F,Cl)$
<b>30G</b> - <i>S</i> , <i>S</i> <b>30G</b> - <i>S</i> , <i>R</i>	A A	trit 2-PrOH recryst MeOH/Et <sub>2</sub> O	83 31	>300 >300	C <sub>19</sub> H <sub>23</sub> FN <sub>4</sub> O <sub>3</sub> ·HCl-0.5H <sub>2</sub> O (C,H,N,Cl) C <sub>19</sub> H <sub>23</sub> FN <sub>4</sub> O <sub>3</sub> ·HCl-0.6H <sub>2</sub> O (C,H,N,F,H <sub>2</sub> O)
30H-R,S 30H-R,R	B B	isoelect prec isoelect prec	83 75	194-195 208-210	C <sub>20</sub> H <sub>25</sub> FN <sub>4</sub> O <sub>3</sub> -0.5H <sub>2</sub> O (C,H,N,F) C <sub>20</sub> H <sub>25</sub> FN <sub>4</sub> O <sub>3</sub> -0.05HF (C,H,N,F)
30H-S,S	В	isoelect prec	77	208-210	$C_{20}H_{25}FN_4O_3-0.5H_2O$ (C,H,N,F)
<b>30H</b> -S,R	В	isoelect prec	88 O	198-199	C <sub>20</sub> H <sub>25</sub> FN <sub>4</sub> O <sub>3</sub> -0.45H <sub>2</sub> O (C,H,N,F)
		F		₂H 31	
		(A-H)	V N		
31A	ref 18		4		
31B 31C	ref 18 ref 18				
31D 31E	ref 18 A	isoelect prec	66	250-252	C <sub>19</sub> H <sub>22</sub> FN <sub>3</sub> O <sub>3</sub> -0.5H <sub>2</sub> O (C,H,N)
31 <b>F</b> -R,S	A	trit 2-PrOH	63	>300	$C_{19}H_{22}FN_3O_3$ ·HCl (C,H,N,Cl,F)
31 <b>F</b> -R,R 31 <b>F</b> -S,S	A A	trit 2-PrOH isoelect prec/dissolved	66 66	>300 >300	C <sub>19</sub> H <sub>22</sub> FN <sub>3</sub> O <sub>3</sub> ·HCl·2.6H <sub>2</sub> O (C,H,N,Cl,H <sub>2</sub> O) C <sub>19</sub> H <sub>22</sub> FN <sub>3</sub> O <sub>3</sub> ·1.6HCl·1.1H <sub>2</sub> O (C,H,N,F,Cl)
31F-S,R	A	HCl/conc/trit 2-PrOH isoelec prec/dissolved	75	>300	C <sub>19</sub> H <sub>22</sub> FN <sub>3</sub> O <sub>3</sub> ·1.4HCl·0.7H <sub>2</sub> O (C,H,N,F,Cl,H <sub>2</sub> O)
31G-R,S	A	HCl/conc/trit 2-PrOH trit 2-PrOH	32	229-230	C <sub>20</sub> H <sub>24</sub> FN <sub>3</sub> O <sub>3</sub> ·2HF-0.5H <sub>2</sub> O (C,H,N,F)
31G-R,R	A	trit 2-PrOH	82	>300	$C_{20}H_{24}FN_3O_3$ ·HCl-0.5 $H_2O$ (C,H,N)
31G-S,S 31G-S,R	A A	trit 2-PrOH recryst MeOH/Et <sub>2</sub> O	80 67	>300 >300	C <sub>20</sub> H <sub>24</sub> FN <sub>3</sub> O <sub>3</sub> ·HCl-0.4H <sub>2</sub> O (C,H,N,Cl) C <sub>20</sub> H <sub>24</sub> FN <sub>3</sub> O <sub>3</sub> ·HCl-0.8H <sub>2</sub> O (C,H,N,Cl)
31H-R,S 31H-R,R	B B	none	86 95	211-213 233-235	C <sub>21</sub> H <sub>26</sub> FN <sub>3</sub> O <sub>3</sub> -0.6HF (C,H,N,F) C <sub>21</sub> H <sub>26</sub> FN <sub>3</sub> O <sub>3</sub> -0.33HF (C,H,N,F)
31H-S,S	В	none none	94	233-234	$C_{21}H_{28}FN_3O_3$ -0.6HF (C,H,N,F)
31 <b>H</b> -S,R	В	none	86 NH 0	210–211	C <sub>21</sub> H <sub>26</sub> FN <sub>3</sub> O <sub>3</sub> -0.75HF (C,H,N,F)
		F	NH <sub>2</sub> O	O₂H	
		(A·H) ~		32	
224	600.00		Ė		
32A 32B	ref 28,36 ref 36				
32C 32D	ref 36 ref 28,36				
32E	A ·	isoelect prec	44	201-204	C <sub>19</sub> H <sub>22</sub> F <sub>2</sub> N <sub>4</sub> O <sub>3</sub> -0.5H <sub>2</sub> O (C,H,N)
32F-R,S 32F-R,R	A A	trit Et <sub>2</sub> O trit EtOAc	72 77	251-254 273-275	C <sub>19</sub> H <sub>22</sub> F <sub>2</sub> N <sub>4</sub> O <sub>3</sub> ·HCl-0.35H <sub>2</sub> O (C,H,N,Cl) C <sub>19</sub> H <sub>22</sub> F <sub>2</sub> N <sub>4</sub> O <sub>3</sub> ·1.2HCl-1.7H <sub>2</sub> O (C,H,N,Cl)
32F-S,S 32F-S,R	A A	trit EtOAc trit EtOAc	46 56	252-254 233-235	C <sub>19</sub> H <sub>22</sub> F <sub>2</sub> N <sub>4</sub> O <sub>3</sub> -1.1HCl·H <sub>2</sub> O (C,H,N,Cl) C <sub>19</sub> H <sub>22</sub> F <sub>2</sub> N <sub>4</sub> O <sub>3</sub> -1.1HCl·0.8H <sub>2</sub> O (C,H,N,Cl)
32G-R,S 32G-R,R	A A	none none	46 98	203-205 198-199	C <sub>20</sub> H <sub>24</sub> F <sub>2</sub> N <sub>4</sub> O <sub>3</sub> ·1.5HF·0.7H <sub>2</sub> O (C,H,N,F) C <sub>20</sub> H <sub>24</sub> F <sub>2</sub> N <sub>4</sub> O <sub>3</sub> ·1.5H <sub>2</sub> O (C,H,N)
32G-S,S	A	none	77	>250	$C_{20}H_{24}F_2N_4O_3$ -2.0HCl-0.6H <sub>2</sub> O (C,H,N)
32G-S,R 32H-R,S	A B	none isoelect prec	10 73	>250 148-150	C <sub>20</sub> H <sub>24</sub> F <sub>2</sub> N <sub>4</sub> O <sub>3</sub> -2HCl (C,H,N) C <sub>21</sub> H <sub>26</sub> F <sub>2</sub> N <sub>4</sub> O <sub>3</sub> -0.4H <sub>2</sub> O (C,H,N)
32H-R,R 32H-S,S	B B	isoelect prec isoelect prec	37 50	156-157 156-157	$C_{21}H_{28}F_2N_4O_{3'}^2.6H_2O$ (C,H,N,H <sub>2</sub> O) $C_{21}H_{28}F_2N_4O_{3'}^20.3H_2O$ (C,H,N)
32H-S,R	B	isoelect prec	50	136-138	$C_{21}H_{26}F_{2}N_{4}O_{3}\cdot 0.7H_{2}O$ (C,H,N)

Table III (Continued)

no./	method of prep.º or ref	method of purif of final prodb	% yield <sup>c</sup>	mp (°C)	analysis formula (elements analyzed) <sup>d</sup>
		F.	NH <sub>2</sub> O	D₃H	
		, , , , , , , , , , , , , , , , , , ,		33	
		(A-H) <sup>-</sup>	Ĭ		
33A	ref 28		4		
33D 33F-R.S	ref 28 A	trit 2-PrOH/Et <sub>2</sub> O	75	>300	C <sub>19</sub> H <sub>23</sub> FN <sub>4</sub> O <sub>3</sub> ·1.75HCl·2.5H <sub>2</sub> O (C,H,N,Cl)
33G-R,S	A	recryst EtOH/H <sub>2</sub> O	44	>280	C <sub>20</sub> H <sub>25</sub> FN <sub>4</sub> O <sub>3</sub> ·HCl·0.5H <sub>2</sub> O (C,H,N,Cl)
33H-R,S	A	dissolved HCl/conc/trit 2-PrOH	78 Q	>300	$C_{21}H_{27}FN_4O_3\cdot 1.8HCl\cdot 1.5H_2O$ (C,H,N,Cl)
		F		0₂H <b>34</b>	
		(A-H)	/ <sup>N</sup> Y <sup>N</sup> 加	34	
			<b>₽</b>		
34A (9*) 34B	ref 29 A	trit 2-PrOH/Et <sub>2</sub> O	∳ 74	272-275	C <sub>20</sub> H <sub>17</sub> F <sub>3</sub> N <sub>4</sub> O <sub>3</sub> ·HCl·2.3H <sub>2</sub> O (C,H,N,Cl)
34D	B	none	100	189-191	C <sub>22</sub> H <sub>21</sub> F <sub>3</sub> N <sub>4</sub> O <sub>3</sub> ·1.5H <sub>2</sub> O (C,H,N)
34F-R,S	A	none	86	263-265	$C_{21}H_{19}F_3N_4O_3\cdot 1.1HCl\cdot 0.9H_2O$ (C,H,N,Cl)
4 <b>F</b> -R,R 4 <b>G</b> -R,S	A A	none recryst EtOH/Et <sub>2</sub> O	95 82	265-268 278-280	C <sub>21</sub> H <sub>19</sub> F <sub>3</sub> N <sub>4</sub> O <sub>3</sub> ·1.1HCl·1.2H <sub>2</sub> O (C,H,N,Cl) C <sub>22</sub> H <sub>21</sub> F <sub>3</sub> N <sub>4</sub> O <sub>3</sub> ·HCl (C,H,N,Cl)
4G-R,S	В	isoelect prec	71	136-138	$C_{23}H_{23}F_3N_4O_3\cdot 2.0H_2O(C,H,N)$
34H-R,R	В	isoelect prec	77 0	203-205	$C_{23}H_{23}F_3N_4O_3\cdot 1.7H_2O (C,H,N)$
		F~		D <sub>2</sub> H	
		(A-H)		35	
			оме 🚶		
35A	c	isoelec prec/dissolved in	89	200-202	$C_{18}H_{20}FN_3O_4\cdot HCl\cdot 1.5H_2O\ (C,H,N)$
35B	С	HCl/conc/ trit 2-PrOH/ether	54	>300	C <sub>19</sub> H <sub>22</sub> FN <sub>3</sub> O <sub>4</sub> ·2.0HCl·3.0H <sub>2</sub> O (C,H,N,Cl,H <sub>2</sub> O)
35C	C	isoelect prec	42	207-208	$C_{21}H_{26}FN_3O_4\cdot 1.1H_2O(C,H,N)$
5 <b>E</b> 5 <b>F</b> - <i>R</i> ,S	C C	isoelect prec trit 2-PrOH	75 <b>89</b>	179-182 236-238	C <sub>20</sub> H <sub>24</sub> FN <sub>3</sub> O <sub>4</sub> ·0.5H <sub>2</sub> O C <sub>20</sub> H <sub>24</sub> FN <sub>3</sub> O <sub>4</sub> ·1.25HCl·2.5H <sub>2</sub> O (C,H,N,Cl)
5F-R,R	č	trit 2-PrOH	77	215-218	$C_{20}H_{24}FN_3O_4\cdot 1.1HCl\cdot 2.4H_2O (C,H,N,Cl)$
5F-S,R	C	recryst 2-PrOH	62	188-190	C <sub>20</sub> H <sub>24</sub> FN <sub>3</sub> O <sub>4</sub> ·HCl·0.2HF·2.2H <sub>2</sub> O (C,H,N,Cl)
5G-R,S 5G-R,R	c c c	trit EtOAc isoelect prec/dissolved HCl/conc	73 38	166-167 212-213	C <sub>21</sub> H <sub>26</sub> FN <sub>3</sub> O <sub>4</sub> ·HCl·1.4H <sub>2</sub> O (C,H,N,Cl) C <sub>21</sub> H <sub>26</sub> FN <sub>3</sub> O <sub>4</sub> ·HCl·2.0H <sub>2</sub> O (C,H,N,Cl)
5H-R,S	C	chromatographed	53		C <sub>22</sub> H <sub>28</sub> FN <sub>3</sub> O <sub>3</sub> ·1.22CH <sub>2</sub> Cl <sub>2</sub> (C,H,N)*
<b>35H</b> - <i>R</i> , <i>R</i>	С	chromatographed	66	97–99	$C_{22}H_{28}FN_3O_3$ (C,H,N)
		F_	ى الله	D₃H	
		Ĭ		36	
		(A-H)	CE. N		
36A	ref 20		o., 3 \square		
6 <b>D</b> 6 <b>F</b> - <i>R</i> ,S	ref 20 A	none	89	>300	C <sub>20</sub> H <sub>21</sub> F <sub>4</sub> N <sub>3</sub> O <sub>3</sub> ·1.8HCl·1.3H <sub>2</sub> O (C,H,N,Cl)
6F-R.R	Ā	trit 2-PrOH	47	>300	$C_{20}H_{21}F_{4}N_{3}O_{3}\cdot 1.6HCl\cdot 1.2HF\cdot 2.4H_{2}O$ (C,H,N,Cl)
6G-R,S	Ą	trit 2-PrOH	79	210-212	$C_{21}H_{23}F_4N_3O_3\cdot 1.1HCl\cdot 1.4H_2O(C,H,N,F,Cl)$
6 <b>G</b> -R,R 6 <b>H</b> -R,S	A B	trit 2-PrOH trit 2-PrOH	69 72	261-262 175-177	C <sub>21</sub> H <sub>23</sub> F <sub>4</sub> N <sub>3</sub> O <sub>3</sub> ·HCl·1.1H <sub>2</sub> O (C,H,N,F,Cl) C <sub>22</sub> H <sub>25</sub> F <sub>4</sub> N <sub>3</sub> O <sub>3</sub> ·0.2HF·1.1H <sub>2</sub> O (C,H,N,H <sub>2</sub> O)
36H-R,R	В	none	67	146-147	$C_{22}H_{25}F_4N_3O_3\cdot 0.5H_2O$ (C,H,N)
		r	<u> </u>	D₂H	
		· ``		37	
		(A-H)	ŶŅ"		
			ÓMe Et		
37 <b>A</b> 37 <b>F</b> - <i>R</i> ,S	C C	trit Et <sub>2</sub> O	73 72	229-231 >280	C <sub>17</sub> H <sub>20</sub> FN <sub>3</sub> O <sub>4</sub> ·1.1HCl·0.8H <sub>2</sub> O (C,H,N,Cl)
37F-R,S 37G-R,S	Ċ	trit 2-PrOH trit Et <sub>2</sub> O	72 70	>280 >280	C <sub>19</sub> H <sub>24</sub> FN <sub>3</sub> O <sub>4</sub> ·1.6HCl·1.8H <sub>2</sub> O (C,H,N,Cl) C <sub>20</sub> H <sub>26</sub> FN <sub>3</sub> O <sub>4</sub> ·1.1HCl-1.5H <sub>2</sub> O (C,H,N,Cl)
, <del>-</del>	_	-	-		

Table III (Continued)

no. f	method of prep. $^a$ or ref	method of purif of final prodb	% yield <sup>c</sup>	mp (°C)	analysis formula (elements analyzed)
		(A-H) F	° CO₂H	8	
38A (5) 38B (6) 38D 38F-R,S	ref 18 ref 18 ref 18 A	none	72	>300	C <sub>19</sub> H <sub>21</sub> F <sub>2</sub> N <sub>3</sub> O <sub>3</sub> -HCl-0.8H <sub>2</sub> O (C,H,N,Cl
		(A-H)	O CO₂H 39		
39A 39B	A ref 31	isoelect prec	Ét 48	240-242	$C_{16}H_{17}F_2N_3O_3\cdot 0.5H_2O$ (C,H,N)
39D-S 39F- <i>R,S</i> 39F- <i>R,R</i> 2 <sup>h</sup>	ref 37 A A ref 18	washed 2-PrOH washed 2-PrOH	72 68	289-290 285-286	$\begin{array}{l} C_{18}H_{21}F_2N_3O_3\cdot HCl\ (C,H,N,F,Cl) \\ C_{18}H_{21}F_2N_3O_3\cdot HCl\cdot H_2O\ (C,H,N,F,Cl) \end{array}$

a See Experimental Section for general methods. b Trituration (trit) refers to grinding the solids under solvent to produce a fine powder. Isoelectric precipitation (isoelec prec) refers to dissolving the compound in NaOH to pH 10.5−11.0 and then slowly adjusting to pH to 7.0−7.2 and filtering the resultant solids. Yields are those from coupling to final products and include the deprotection and hydrolysis steps. Symbols refer to those elements analyzed for. Analyses were ±0.4% of the theoretical values. Solvent was present in analytical sample and was observed in HNMR spectra. Chemical and optical purities were >98% by HPLC. See Figure 2 for structures of side chains A−H. Tosufloxacin. Ciprofloxacin.

C and D, or alkylation of the 3-position of the pyrrolidine ring to give E, produced quinolones with reduced Gramnegative and Gram-positive effectiveness (see any series of C, D, or E quinolones vs B).

In vivo, the (3-aminopyrrolidinyl)quinolones (A) displayed the best oral efficacy against systemic *E. coli* infections in mice, but were less effective vs the *Streptococcus pyogenes* infection model. Most of this variance was likely due to the 2-4-fold MIC differential between *E. coli* and *S. pyogenes*. The increased in vitro Grampositive activity conferred by pyrrolidine B did not translate into in vivo effectiveness (see 30B, 31B, 32B, and 38B). Alkylating the distal nitrogen or the pyrrolidine ring (C-E), even though in vitro activity was reduced, did improve in vivo efficacy vs *S. pyogenes* by 2-10-fold (32D vs 32A and 32B; 36D vs 36A; 38D vs 38A and 38B; and 30E vs 30B).

The quinolone nuclei can be ranked on the basis of combined in vitro and in vivo antibacterial effectiveness. This exercise, using all the data in Table IV, produces the following descending rank order: 6,8-difluoroquinolones  $38 > N_1$ -(difluorophenyl)naphthyridines  $34 \ge 5$ -amino-6,8-difluoroquinolones 32 > 8-methoxyquinolones 35 > 8-(trifluoromethyl)quinolones  $36 \ge 1$ -cyclopropylnaphthyridines  $30 \ge 5$ -aminoquinolones  $31 \ge 1$ -cyclopropylnaphthyridines  $30 \ge 5$ -aminoquinolones  $31 \ge 1$ -cyclopropylnaphthyridines  $31 \ge 1$ -cyclop

When analyzing the effects of adding a single methyl group to the methylene spacer of B, C, and D, giving the side chains F, G, and H, one notices that these side chains, like their precursors, confer much greater activity vs the Gram-positive organisms. It is also immediately clear that the stereochemistry of the 3- and 1-centers has a significant and consistent effect on in vitro and in vivo activity. For all the quinolones tested, the in vitro and in vivo potencies conferred by the side chains F-H tend to vary with stereochemical configuration in the descending rank order of  $3R,1S \geq 3R,1R \geq 3S,1R \geq 3S,1S$ . While the differences in activity between close pairs are small, usually 1-4-fold, the differences between the most active (3R,1S)-quinolones and the least active 3S,1S derivatives are consistently 2-8-

fold, with certain organisms approaching differences in potency of >30-fold (see S. pneumonia for 30G-R,S vs 30G-S,S and 32G-R,S vs 32G-S,S). These in vitro differences are reflected at the target enzyme level. The DNA gyrase inhibition data from Table V demonstrates that the R, S and R, R isomers of 30F and 31F are 3-15 times superior to the S,R and S,S isomers at inducing initial DNA cleavage. In vivo, the differences in oral efficacy between stereoisomers parallels the in vitro results with the R,S isomers, displaying an average of 6-10-fold superiority over the S,S and S,R derivatives. It appears that the center of asymmetry at the pyrrolidinyl 3-position has a greater overall influence on activity than the adjacent 1-position. The most active isomers (3R,1S and 3R,1R)bear the same configuration at C3 as the most favorable enantiomer of the 3-[(ethylamino)methyl]pyrrolidine D (Figure 3).37 In contrast to the large differences in quinolone activity conferred by the various stereoisomers of F-H, the quinolones containing the enantiomers of D showed only a small 1-3-fold variance. This is the first case where stereochemical configuration spatially distant from the quinolone nucleus has created such a large biological difference between stereoisomers. 12

If one compares the 3R.1S isomers of F-H to the reference side chains B-D, the effect of methylating the methylene spacer atom can be discerned. Examining all the quinolones tested shows that the methylated spacer has little effect on Gram-negative potency. (See 30F-R,S vs 30B, 30G-R,S vs 30C, 30H-R,S vs 30D, etc.) However, against Gram-positive organisms, the compounds containing pyrrolidines F-H confer an average 2-4-fold superiority relative to the quinolones with side chains B-D. (Compare 30-32G-R, S and 30-32H-R, S vs 30-32C and 30–32**D**.) Compared to the 3-aminopyrrolidine A, the R,Sand R.R isomers of F-H always provide quinolones with improved Gram-positive potency. The quinolones bearing the primary amino side chain F-R,S also approach the Gram-negative potency of the reference compounds containing side chain A.

Comparison of the oral efficacy for the quinolones containing side chains F-H vs the reference agents with side chains A-E immediately reveals that the 3-(1aminoethyl)pyrrolidines confer 4-10 times more activity in the S. pyogenes infection model with the average closer to 10 (see the R,S isomers of 30-32 vs the derivatives 30-32B-D). This in vivo oral superiority reaches 100-fold when comparing 39F-R, S to 39B. Against E. coli, quinolones with side chains F-R,S, G-R,S, or H-R,S show more modest advantages of 2-5-fold over comparable analogs of B, C, or D. They do not compete with the 3-aminopyrrolidine A for conferring oral efficacy vs E. coli. As with the reference agents, alkylation of the distal nitrogen further improves the in vivo performance (see corresponding stereoisomers of 31F relative to 31G and 31H, and 33F-R, S relative to 33H-R, S).

The in vivo enhancement of the 3-(1-aminoethyl) moiety is so great that even the quinolones with the S,R and R,Rside chains, which are the less active of the stereoisomers in vitro, are often more active orally than the reference quinolones, especially vs S. pyogenes (see 30F-R,R, 30F-S,R vs 30B, 31H-R,R vs 31D, and 39F-R,R vs 39B). The magnitude of the observed in vivo superiority of these quinolones can be partly explained by their generally better MICs for S. pyogenes vs the reference agents, and the fact that the reference quinolones are racemic mixtures, which has been shown to account for a 2-fold reduction in oral efficacy vs Gram-positive infections in mice.<sup>37</sup> Nevertheless, even with correction for these variables, the R,S isomers of the [3-(1-aminoethyl)-1-pyrrolidinyl]quinolones still have a baseline edge in oral efficacy of 3-fold over the reference quinolones. Other workers have shown that alkylation of the R<sub>7</sub> heterocyclic substituent of the quinolones tends to improve their water solubility<sup>23,40</sup> and increase lipophilicity.<sup>23</sup> Such physical chemical properties were associated with improved in vivo efficacy and this phenomena could help explain the in vivo enhancement described in this series.

The outstanding in vitro and in vivo Gram-positive potency conferred by the R,S isomers of F-H made it possible for several of the substituted quinolones to now compete with the 6,8-difluoroquinolones 38. Relative to the reference agents 5 (38A) and 6 (38D), several of the new quinolones meet or exceed these standards for combined in vitro and in vivo activity. These include 1-cyclopropylnaphthyridines 30F-R,S, 30G-R,S and 30G-R,R, and 30H-R,S and 30H-R,R; 5-amino-6,8-diffuoroquinolones 32F-R, S and 32F-R, R, 32G-R, S and 32G-R, R, 32H-R.S and 32H-R.R: 1-(2.4-diffuorophenyl)naphthyridine 34F-R,S; 8-methoxyquinolones 35F-R,S and 35F-R,R,35G-R,S, and 35H-R,S; 8-(trifluoromethyl)quinolones 36F-R,S and 36G-R,S; and 1-ethyl-6,8-difluoroquinolone **39F**-*R*,*S*.

As described above, the 8-fluoroguinolones have been reported to increase phototoxicity in animal models, 20 and also cytotoxicity in in vitro mammalian cell assays. 21a,b,d Therefore, the compounds with activity competitive with 38A and 38D were tested for both phototoxicity and cytotoxicity (Table VI).

We have previously reported that most of the quinolones in widespread clinical use display oral no-effect doses (NEDs) of ≥30 mg/kg in the mouse phototoxicity model.<sup>20,39</sup> The reference 6,8-difluoroguinolones 38A, 38D, and 39D and the agents 38F-R,S and 39F-R,S represent the high extremes of induced phototoxicity with NEDs ranging from 1.5 to 10 mg/kg. The [3-(1-ethylamino)-1-

pyrrolidinyl]naphthyridines 30 and the 5-amino-6,8difluoroquinolone analogs 32 show a fair degree of phototoxicity relative to their reference quinolones (see 30A vs 30F-R,S, 30G-R,S, and 30H-R,S). Clearly, the improved in vivo efficacy might account for the increased phototoxicity. However, several quinolones containing the F-H side chains show NEDs of ≥100 mg/kg.

Quinolone cytotoxicity in mammalian cells has become an important aspect of their overall safety profile. Several workers have previously demonstrated a strong correlation between mammalian topoisomerase II inhibition and cytotoxicity to mammalian cells. 21a,c,g,f Recently, these correlations were extended to include in vitro clastogenic endpoints.21c,e,f Over a wide variety of quinolones, the concentration causing mammalian cell cytotoxicity was well-correlated to that concentration inducing micronuclei formation, a clastogenic endpoint. 21,c,e From this and related work,21d it has been shown that most of the quinolones in clinical use display a 50% cytotoxicity concentration vs mammalian cells of ≥100 µg/mL.<sup>21c,d</sup> Of the new quinolones described in this study, only a small percentage have IC<sub>50</sub> values  $\geq$ 100  $\mu$ g/mL, and many, especially those with an 8-fluoro group, are cytotoxic at concentrations  $\leq 10 \,\mu\text{g/mL}$ . Only 34F-R,S, 35H-R,S, and 36G-R,S meet the phototoxicity and cytotoxicity criteria, while possessing overall antibacterial activity comparable to the reference standards 5 (38A) and 6 (38D).

In summary, we have shown that the 3-(1-aminoethyl)pyrrolidines confer excellent Gram-positive potency both in vitro and in vivo. Unlike their 3-(aminomethyl)pyrrolidinyl analogs, the stereoisomers of F-H, display significant differences in MICs and in oral efficacy ranging from 6- to 30-fold between the most active 3R,1S and the least active 3S,1S isomers. Relative to the reference quinolones containing the pyrrolidinyl side chains A-E. the R,S isomers of F-H confer a 2-4-fold improvement in in vitro Gram-positive activity and an average 10-fold improvement in oral efficacy. In particular, the quinolones 34F-R,S, 35H-R,S, and 36G-R,S display extremely high levels of antibacterial efficacy with low levels of phototoxicity and mammalian cell cytotoxicity. These agents are ideal candidates for further advancement with 34F-R,S (PD 140248) showing the best overall blend of safety and efficacy.

### **Experimental Section**

All melting points were determined on a Hoover capillary melting point apparatus and are uncorrected. Infrared (IR) spectra were determined in KBr on a Mattson Cygnus 100 or a Nicolet MX1 instrument. Proton magnetic resonance (NMR) were recorded on either a Varian XL-200 or a Bruker AM250 spectrometer; shifts are reported in  $\delta$  units relative to internal tetramethylsilane. All mass spectra were obtained on a Finnigan 4500 GCMS or a VG analytical 7070 E/F spectrometer. Elemental analyses were performed on a CEC Model 240 elemental analyzer and all compounds prepared had analytical results  $\pm 0.4\%$  of the theoretical values. Column chromatography was accomplished using E. Merck silica gel, 230-400 mesh, and concentrations were performed in vacuo at 10-30 mmHg. Final compounds were assayed for purity by using a Waters high-performance liquid chromatography (HPLC) system equipped with a 5-µM Alltech CN column and a mobile phase consisting of 20% THF/80% 0.5 M NH<sub>4</sub>H<sub>2</sub>PO<sub>4</sub> (adjusted to pH 3 with H<sub>3</sub>PO<sub>4</sub>); in all cases, the purity exceeded 97%.

General Method. Procedures for the Determination of Isomeric Purity of 13-R.S. 13-R.R. 13-S.R. and 13-S.S. The purity of pyrrolidinones 13 was determined via the two following methods which gave comparable results.

High-Pressure Liquid Chromatography. Small samples (<10 mg) of each of the chromatographed pyrrolidines 13 were

Table IV. In Vitro and in Vivo Antibacterial Activity of the [3-(1-Aminoethyl)-1-pyrrolidinyl]quinolones and the Reference Agents from Table III

	in vitro antibacterial activity (MICs $^a$ $\mu$ g/mL)					in vivo antibacterial activit mouse protection (PD <sub>50</sub> , mg/								
		Gram-negative organisms <sup>c</sup> Gram-positive organi			nisms <sup>c</sup>		E. c.		S. p	S. py.				
no.	$\overline{E. cl.}$	E. c.	K. pn.	P. r.	P. ae.	S. au. H	S. au. U	E. f.	S. pn.	S. py.	pod	8C <sup>e</sup>	pod	8C°
30A	0.025	0.013	0.025	0.05	0.1	0.2	0.013	0.2	0.2	0.2	2	0.6	32	14
30B 30C	0.1 0.2	0.05 0.1	0.1 0.2	0.4 0.4	0.2 0.8	0.1 0.2	0.003 0.025	0.025 0.2	0.1 0.025	0.025 0.05	17 3	1 1	39 21	4 7
30D	0.2	0.1	0.2	0.4	0.8	0.2	0.025	0.2	0.025	0.05	3 7	1	17	8
30E	0.1	0.2	0.1	0.4	0.8	0.025	0.006	0.05	0.025	0.05	10	2	6	2
30F-R,S	0.05	0.025	0.05	0.1	0.4	0.025	0.003	0.013	0.003	0.013	2	0.4	3	1
<b>30F</b> - <i>R</i> , <i>R</i> <b>30F</b> - <i>S</i> , <i>S</i>	0.05 0.05	0.05 0.05	$0.1 \\ 0.2$	0.4 0.4	0.8 1.6	$0.05 \\ 0.2$	0.00 <del>6</del> 0.025	0.05 0.1	0.006 0.05	0.013 0.05	6 8	1 1	14 28	2 10
30F-S,R	0.05	0.05	0.2	0.4	0.8	0.2	0.025	0.1	0.05	0.05	4	1	26 15	5
30G-R,S	0.1	0.1	0.2	0.4	1.6	0.1	0.013	0.1	0.013	0.025	1	0.5	2	2
30G-R,R	0.1	0.1	0.2	0.4	1.6	0.1	0.05	0.1	0.025	0.05	5	1	8	5
<b>30G</b> -S,S <b>30G</b> -S,R	0.4 0.1	0.2 0.1	0.4 0.2	0.8 0.8	3.1 1.6	0.8 0.2	0.2 0.05	0.8 0.2	0.4 0.1	0.4 0.1			23 11	15 5
30H-R,S	0.1	0.1	0.2	0.8	1.6	0.2	0.025	0.05	0.025	0.05	2	1	2	2
30H-R,R	0.4	0.1	0.2	1.6	1.6	0.05	0.025	0.05	0.025	0.05	7	6	6	4
30H-S,S	0.8	0.4	0.8	3.1	6.3	0.8	0.4	0.8	0.2	0.4			12	12
30H-S,R 31A	0.2 0.025	0.2 0.025	0.4 0.05	1.6 0.1	3.1 0.2	$\begin{array}{c} \textbf{0.2} \\ \textbf{0.2} \end{array}$	0.1 0.025	0.2 0.1	0.1 0.1	0.4 0.1	3 3	1 0.5	15 97	9 11
31B	0.025	0.023	0.03	0.1	0.8	0.05	0.025	0.05	0.013	0.013	>100	2	>100	4
31C	0.2	0.1	0.2	0.4	0.8	0.2	0.025	0.1	0.05	0.05	35	2	>50	5
31D	0.4	0.2	0.2	0.8	1.6	0.4	0.013	0.2	0.2	0.05	35	3	46	5
31E 31F- <i>R</i> , <i>S</i>	0.8 0.05	0.2 0.05	0.4 0.1	0.8 0.2	0.8 0.4	0.1 0.05	0.013 0.025	0.1 0.025	0.025	0.05	43 47	3 1	31	3
31F-R,R	0.03	0.05	0.1	0.2	1.6	0.03	0.025	0.025	0.013 0.025	0.003 0.025	72	2	11 >100	1 2
31F-S,S	0.2	0.2	0.4	0.8	1.6	0.8	0.1	0.4	0.2	0.1	100	4	>100	12
31F-S,R	0.2	0.2	0.4	0.8	1.6	0.4	0.05	0.2	0.1	0.1	35	2	>100	3
31G-R,S	0.2	0.2	0.4	0.8	1.6	0.2	0.05	0.1	0.05	0.05	7	1	7	1
31G-R,R 31G-S,S	0.2 0.4	0.1 0.4	0.4 0.8	0.8 3.1	1.6 6.3	0.2 1.6	0.05 0.2	0.2 0.8	0.05 0.4	0.05 0.4			36	3
31G-S,R	0.2	0.2	0.4	0.8	3.1	0.4	0.05	0.2	0.2	0.1			82	7
31H-R,S	0.2	0.2	0.4	0.8	6.3	0.1	0.025	0.1	0.05	0.05	13	2	8	1
31H-R,R	0.2	0.2	0.4 0.8	0.8 1.6	6.3 12.5	0.1 0.8	0.025	0.1	0.05	0.05			12	4
31H-S,S 31H-S,R	0.4 0.4	0.4 0.4	0.8	1.6	6.3	0.4	0.4 0.2	0.8 0.4	0.4 0.2	0.4 0.2				
32A	0.013	0.006	0.006	0.025	0.05	0.013	0.006	0.025	0.006	0.025	1	0.3	28	8
32B	0.025	0.013	0.025	0.05	0.1	0.006	0.003	0.013	0.003	0.003	12	1	11	0.6
32C	0.05	0.025	0.05	0.2	0.4	0.013	0.006	0.025	0.003	0.003	8	1		•
32D 32E	0.05 0.1	0.05 0.2	0.05 0.1	0.1 0.2	0.2 0.8	0.013 0.013	0.003 0.003	0.025 0.025	0.003 0.003	0.013 0.013	16 9	2 1	5 8	3 1
32F-R,S	0.013	0.013	0.013	0.025	0.4	0.003	0.003	0.003	0.003	0.003	2	0.3	0.6	0.1
32F-R,R	0.025	0.025	0.025	0.05	0.4	0.003	0.003	0.006	0.006	0.006	4	0.7	2	0.2
32F-S,S	0.05	0.05	0.1	0.4	0.4	0.025	0.013	0.05	0.025	0.05			>25	7
32F-S,R 32G-R,S	0.1 0.025	0.05 0.025	0.1 0.05	0.2 0.1	0.8 0.8	0.025 0.003	0.006 0.003	0.05 0.013	0.013 0.003	0.013 0.003	3	0.4	14 0.6	4 0.3
32G-R,R	0.05	0.05	0.1	0.2	0.8	0.006	0.003	0.013	0.006	0.006	3	2	2	0.7
<b>32G</b> -S,S	0.4	0.4	0.8	1.6	3.1	0.8	0.4	1.6	0.8	1.6				
32G-S,R	0.2	0.2	0.8	0.8	3.1	0.1	0.05	0.2	0.2	0.2			13	13
32H-R,S 32H-R,R	0.025 0.1	0.025 0.05	$0.05 \\ 0.1$	0.1 0.2	0.8 1.6	0.003 0.006	0.003 0.003	0.013 0.025	0.003 0.003	0.003	2	1	0.3 1	0.3 0.5
32H-S,S	0.4	0.2	0.4	1.6	3.1	0.1	0.05	0.025	0.05	0.003			13	9
<b>32H</b> -S,R	0.1	0.1	0.1	0.4	0.8	0.013	0.003	0.025	0.013	0.013	_		6	3
33A	0.013	0.013	0.025	0.1	0.4	0.1	0.025	0.1	0.025	0.05	8	1	110	6
33 <b>D</b> 33 <b>F</b> - <i>R</i> , <i>S</i>	0.1 0.025	0.1 0.025	0.4 0.05	0.8 0.1	1.6 0.4	0.2 0.05	0.025 0.003	0.05 0.013	0.05 0.003	0.1 0.003	90	3	>100 18	11 4
33G-R,S	0.020	0.020	0.2	0.4	1.6	0.05	0.006	0.025	0.013	0.025			8	1
<b>33H</b> - <i>R</i> , <i>S</i>	0.1	0.1	0.2	0.8	1.6	0.05	0.025	0.05	0.013	0.013		_	5	1
34A 34B	0.025 0.05	0.025 0.05	0.1 0.1	0.1 0.2	0.4	0.05 0.05	0.013	0.1 0.025	0.05	0.05	0.5	0.2	5 10	3
34D	0.05	0.05	0.1	0.2	0.4 1.6	0.05	0.013 0.025	0.025	$0.025 \\ 0.1$	$0.025 \\ 0.1$	6	4	10 8	1 3
<b>34F</b> - <i>R</i> , <i>S</i>	0.2	0.1	0.2	0.4	0.8	0.05	0.006	0.025	0.006	0.006	3	1	1	1
34F-R,R	0.1	0.1	0.2	0.4	1.6	0.1	0.025	0.1	0.1	0.05	•	•	9	3
34G- <i>R</i> , <i>S</i> 34H- <i>R</i> , <i>S</i>	0.8 0.8	0.4 0.4	0.8 0.8	1.6 1.6	3.1 3.1	0.1 0.05	0.025 0.013	0.2 0.2	$0.025 \\ 0.1$	0.05 0.1	6 25	2 13	4 2	1 2
34H-R,R	0.8	0.8	1.6	3.1	3.1	0.03	0.013	0.4	0.1	0.1	20	10	10	6
35A	0.025	0.025	0.05	0.1	0.2	0.05	0.013	0.05	0.025	0.025	3	0.4	10	3
35B	0.05	0.05	0.1	0.1	0.8	0.025	0.003	0.013	0.003	0.003			11	0.6
35D 35E	0.1 0.1	0.1 0.1	0.2 0.2	0.4 0.4	1.6 1.6	0.025 0.025	0.006 0.006	0.05 0.025	0.013 0.013	0.013 0.013	18	3	4 5	0.4 1
35F-R,S	0.05	0.025	0.05	0.05	0.4	0.025	0.003	0.006	0.003	0.003	4	1	1	0.3
35F-R,R	0.05	0.05	0.1	0.2	0.8	0.025	0.003	0.025	0.013	0.013			6	1
35F-S,R 35G-R,S	0.8 0.05	0.4 0.05	0.4 0.1	0.4 0.2	1.6 0.8	0.05 0.013	0.013 0.003	0.05 0.025	0.025 0.003	0.025 0.003	5	1	16 1	6 0.1
JJU-11,D	0.00	0.00	0.1	0.2	0.0	0.010	0.000	0.020	0.000	0.003	U		1	0.1

	in vitro antibacterial activity (MICs <sup>α</sup> μg/mL)											acterial act ion (PD50, n		
		Gram-n	egative or	ganisms	c		Gram-posi	tive orga	nisms		E. c.		S. py.	
no.	E. cl.	E. c.	K. pn.	P. r.	P. ae.	S. au. H	S. au. U	E. f.	S. pn.	S. py.	$po^d$	8C <sup>e</sup>	po <sup>d</sup>	sc°
35G-R.R	0.1	0.1	0.2	0.4	3.1	0.05	0.013	0.05	0.013	0.025	12	3	4	1
35H-R,S	0.1	0.1	0.2	0.4	3.1	0.013	0.003	0.025	0.006	0.006	5	2	0.5	0.2
35H-R.R	0.2	0.2	0.4	0.8	3.1	0.013	0.003	0.013	0.003	0.003	14	4	3	0.6
36A	0.1	0.1	0.2	0.4	1.6	0.2	0.1	0.4	0.4	0.4	3	1	32	22
36D	0.4	0.4	0.8	1.6	6.3	0.05	0.025	0.1	0.05	0.05	5	2	2	1
<b>36F</b> -R,S	0.2	0.2	0.2	0.4	1.6	0.013	0.006	0.025	0.006	0.006			0.7	0.3
36F-R.R	0.4	0.8	0.8	0.8	6.3	0.1	0.025	0.2	0.1	0.1			3	2
<b>36G</b> - <i>R</i> , <i>S</i>	0.4	0.4	0.8	0.8	3.1	0.025	0.013	0.1	0.013	0.013	4	2	0.8	0.3
36G-R,R	0.4	0.2	1.6	3.1	6.3	0.05	0.025	0.1	0.05	0.05			3	2
36H-R.S	0.4	0.8	1.6	1.6	6.3	0.05	0.013	0.1	0.013	0.025	5	3	0.7	0.5
36H-R.R	1.6	0.8	1.6	3.1	6.3	0.05	0.05	0.1	0.05	0.1		_	2	0.8
37A	0.2	0.2	0.2	0.4	0.8	0.2	0.05	0.4	0.4	0.2	13	7	74	25
37F-R,S	0.1	0.05	0.1	0.1	0.8	0.006	0.003	0.006	0.003	0.003	9	3	3	0.2
37G-R.S	0.8	0.2	0.2	0.4	1.6	0.025	0.003	0.025	0.013	0.013	22	4	ž	0.3
38A(5)	0.025	0.013	0.025	0.05	0.1	0.05	0.025	0.05	0.05	0.05	1	0.2	8	2
38B	0.013	0.025	0.05	0.1	0.2	0.013	0.003	0.013	0.006	0.003	6	1	4	0.4
38D(6)	0.1	0.05	0.1	0.2	0.4	0.025	0.006	0.025	0.025	0.025	4	1	i	0.8
38F-R,S	0.013	0.013	0.013	0.025	0.4	0.013	0.003	0.006	0.006	0.006	2	0.3	0.4	0.1
39A	0.1	0.05	0.1	0.1	0.2	0.2	0.05	0.4	0.4	0.2	3	1	21	7
39B	0.2	0.1	0.4	0.4	0.4	0.1	0.1	0.1	0.1	0.05	90	ī	100	5
39D-S	0.2	0.1	0.1	0.2	1.6	0.05	0.05	0.1	0.1	0.1	10	3	10	4
39F-R,S	0.05	0.05	0.1	0.2	0.8	0.025	0.006	0.013	0.003	0.003	3	ĭ	1	0.4
39F-R,R	0.1	0.1	0.2	0.4	1.6	0.1	0.025	0.1	0.05	0.05	15	2	19	2
2	0.05	0.05	0.05	0.1	0.4	3.1	0.2	0.8	1.6	0.8	1	0.3	180	19

<sup>&</sup>lt;sup>a</sup> Minimum inhibitory concentrations. <sup>b</sup> Dose required to prevent death in 50% of the animals. Dose administered at time of lethal infection. Definitions of organism abbreviations: E. cl. = Enterobacter cloacae MA 2646, E. c. = E. coli Vogel, K. pn. = Klebsiella pneumoniae MGH2, P. r. = Providencia rettgeri M1771, P. ae. = Pseudomonas aeruginosa UI-18, S. au. H = Staphylococcus aureus H 228, S. au. U = S. aureus UC 76, E. f. = Enterococcus faecalis MGH-2, S. pn. = Streptococcus pneumoniae SV-1, S. py. = S. pyogenes C 203. dOral administration. Subcutaneous administration.

Table V. DNA-Gyrase Inhibition<sup>a</sup> by Select Quinolones

no.	initial inhibn concn (µg/mL)	no.	initial inhibn concn (μg/mL		
30A	0.75	31 <b>E</b>	1.0		
30B	0.75	31F-R,S	0.50		
30E	5.00	31F-R.R	0.75		
30F-R.S	0.75	31 <b>F</b> -S,S	7.5		
30F-R,R	0.75	31F-S,R	7.5		
30F-S,S	2.5	<b>34F</b> - <i>R</i> , <i>S</i>	0.5		
30F-S,R	2.5	35G-R,S	<0.5		
31A	0.25	36H-R,S	0.88		
31B	0.50				

a Initial inhibition as measured by the concentration of drug that first induces cleavage of DNA-Gyrase complex. See ref 34.

dissolved in CH<sub>2</sub>Cl<sub>2</sub>, treated with 3 drops of Et<sub>3</sub>N and a small amount of di-tert-butyl dicarbonate, and stirred for 20 min. These BOC derivatives were then assayed on a Waters HPLC equipped with an Alltech silica gel column and monitored at a wavelength of 214 nm. The mobile phase consisted of 4:1 octane/2-propanol at a flow rate of 1.5 mL/min.

NMR Spectroscopy. The doublet in the NMR spectrum of 13 occurring at approximately  $\delta$  1.0 corresponds to the methyl group adjacent to the exocyclic amine. The exact position of this doublet was characteristic of the specific diastereomer, and the presence of extraneous doublets in the region was diagnostic of isomeric contamination. These diagnostic doublets appeared at following resonances:  $\delta$  0.82 for 13-R,R,  $\delta$  1.01 for 13-R,S,  $\delta$  1.09 for 13-S,R, and  $\delta$  1.19 for 13-S,S.

The purities obtained via method 1 correlated well with those obtained via method 2 and indicated isomeric purities of  $\geq 20:1$ .

General Method A. Preparation of (3R,1S)-7-[3-(1-Aminoethyl)-1-pyrrolidinyl]-1-cyclopropyl-6-fluoro-1,4-dihydro-4-oxo-1,8-naphthyridine-3-carboxylic Acid (30F-R,S). A mixture of 0.68 g (2.4 mmol) of 7-chloro-1-cyclopropyl-6-fluoro-1,4-dihydro-4-oxo-1,8-naphthyridine-3-carboxylic acid (17), 0.60 g (2.8 mmol) of 14, 0.74 (7.3 mmol) of  $Et_3N$ , and 25 mL of  $CH_3CN$ was refluxed for 2 h. The mixture was cooled to 5 °C, diluted with Et<sub>2</sub>O, and filtered. The solids were washed with H<sub>2</sub>O and Et<sub>2</sub>O and were dried to give 1.02 g (92%) of a white powder. This material was suspended in 20 mL of absolute EtOH, treated with

Table VI. Results of Select Quinolones in the Mouse Phototoxicity and Hamster V-79 Cell Cytotoxicity Assays

<del></del>					_ <u>-</u>
no.	photo- toxicity no-effect dose (mg/kg)	clonogenic cytotoxicity 50% inhibn (µg/mL)	no.	photo- toxicity no-effect dose (mg/kg)	clonogenic cytotoxicity 50% inhibn (µg/mL)
2	>100	>200	32H-R,S		42
5 (38A)	3	30	32H-R,R		15
6 (38 <b>D</b> )	10	11	33A	>300	12
38F-R,S	<3	<7.8	<b>33G</b> - <i>R</i> , <i>S</i>	>100	290
30A	>100	98	34A	>100	120
30F-R,S	30	33	34F-R,S	>100	280
30F-S,R	>100	84	34G-R,S	>100	310
30G-R,S	18	130	35A	>100	45
30G-R,R	30	88	35F-R,S	>100	<7.8
30H-R,S	<30	130	35F-R,R		41
30H-R,R		87	35G-R,S	>100	33
31G-R,S	100	180	35H-R,S	>100	110
32A	55	15	36A	>100	250
32C	>100	7.8	36F-R,S	>100	19
32F-R,S	<30	<7.8	36G-R,S	>100	130
32F-R,R		<7.8	37F-R,S		19
32G-R,S	<30	17	39D	1.5	190
32G-R,R	<30	24	<b>39F</b> - <i>R</i> , <i>S</i>	<10	9.8

5 mL of 1 N HCl, and refluxed for 2 h. The homogeneous solution was cooled to room temperature and stirred for 18 h. The solvent was evaporated; the residue was triturated with 2-PrOH, and the solids which formed were filtered, washed with 2-PrOH and Et<sub>2</sub>O, and dried to give 0.64 g (89%) of 30F-R, S: mp > 300 °C; IR 1703, 1631 cm<sup>-1</sup>; NMR (DMSO- $d_6$ )  $\delta$  1.05 (m, 2 H, cyclopropyl), 1.18-1.50 (m, 5 H, 2 cyclopropyl plus CH<sub>3</sub>), 1.65-1.95 (m, 1 H, pyrrolidine), 2.15 (m, 1 H, pyrrolidine), 2.48 (m, 1 H, pyrrolidine), 3.30 (m, 1 H, CHNH<sub>2</sub>), 3.47-3.87 (m, 3 H, pyrrolidine), 3.97 (m, 1 H, pyrrolidine), 4.10 (m, 1 H, cyclopropyl), 8.00 (d, J = 13 Hz, 1 H,  $C_5H$ ), 8.53 (s, 1 H,  $C_2H$ ).

Compounds 30F-R,R, 30F-S,R, and 30F-S,S were prepared in an identical fashion, giving rise to the following physical data.

**30F-R.R**: NMR (DMSO- $d_6$ )  $\delta$  1.00–1.47 (m, 7 H), 1.68–2.00 (m, 1 H), 2.11-2.35 (m, 1 H), 2.52 (m, 1 H), 3.17-3.87 (m, 4 H), 3.95 (m, 1 H), 4.17 (m, 1 H), 8.00 (d, J = 13 Hz, 1 H), 8.58 (s, 1)

**30F-S.R**: NMR (DMSO- $d_6$ )  $\delta$  0.93-1.17 (m, 2 H), 1.17-1.47 (m, 5 H), 1.63–1.95 (m, 1 H), 2.00–2.37 (m, 1 H), 2.47 (m, 1 H), 3.35 (m, 1 H), 3.50-3.87 (m, 3 H), 3.93 (m, 1 H), 4.25 (m, 1 H), 8.00 (d, J = 13 Hz, 1 H), 8.58 (s, 1 H).

**30F-S,S**: NMR (DMSO- $d_6$ )  $\delta$  1.00–1.50 (m, 7 H), 1.65–2.00 (m, 1 H), 2.22 (m, 1 H), 2.52 (m, 1 H), 3.15-3.85 (m, 4 H), 3.95 (m, 1 H), 4.17 (m, 1 H), 8.00 (d, J = 13 Hz, 1 H), 8.58 (s, 1 H).

Compounds 30G-S,S, 31F-R,S, 31F-R,R, 31G-R,S, 31G-R,R, 31G-S,S, 36G-R,S, 30E, 31E, 32E, 39F-R,S, and 39F-R,R were prepared in identical fashion to 30F-R,S. Compounds 31F-S,R and 31F-S,S were prepared similarly except that TFA in  $CH_2Cl_2$ was used to remove the tert-butoxycarbonyl protecting group instead of 1 N HCl in EtOH. Alternatively, the BOC-protected intermediate could be dissolved in 25 mL of CH<sub>2</sub>Cl<sub>2</sub>, cooled to 5 °C, and treated with gaseous HCl for 10 min. After warming to room temperature, the solution was concentrated and the residue triturated with 2-PrOH as before. This modified procedure was used to synthesize compounds 30G-S,R, 31G-S.R. 32F-R,S, 32F-S,R, 32F-R,R, 32G-S,R, 32G-S,S, 32G-R,R, 32G-R,S, 34F-R,R, 34G-R,S, 39F-R,R, 39F-R,S, 38F-R,S, 34B, and

At times, it was necessary to purify the tert-butoxycarbonyl intermediate prior to deprotection; chromatography on E. Merck silica gel, using 10:1 CH<sub>2</sub>Cl<sub>2</sub>/MeOH as eluent, proved to be the method of choice. Deprotection was then effected with gaseous HClin CH<sub>2</sub>Cl<sub>2</sub> as outlined previously. Compounds 36F-R,R, 36F-R,S, 30G-R,S, 34F-R,S, 33F-R,S, and 33G-R,S (eluting with 20:1 CH<sub>2</sub>Cl<sub>2</sub>/MeOH) were prepared in this manner, while compounds 30G-R,R and 36G-R,R (eluting with 40:1 CH<sub>2</sub>Cl<sub>2</sub>/MeOH) were deprotected with 1 N HCl in EtOH at reflux.

General Method B. Preparation of (3R,1S)-5-Amino-1cyclopropyl-7-[3-[1-(ethylamino)ethyl]-1-pyrrolidinyl]-6,8difluoro-1,4-dihydro-4-oxo-3-quinolinecarboxylic Acid (32H-R,S). A solution of 0.60 g (2.0 mmol) of 5-amino-1-cyclopropyl-6,7,8-trifluoro-1,4-dihydro-4-oxo-3-quinolinecarboxylic acid (19), 0.31 g (2.2 mmol) of pyrrolidine 16, 0.61 g (6.0 mmol) of Et<sub>3</sub>N, and 20 mL of CH<sub>3</sub>CN was refluxed for 5 h and then stirred at room temperature for 18 h. The solution was concentrated. The residue was dissolved in H<sub>2</sub>O and made basic (pH 11) with 10% NaOH; the solution was filtered through a fiberglass pad and neutralized to pH 7.5. The precipitate which formed was filtered, washed with H<sub>2</sub>O, and dried to give 0.61 g (73%) of the title compound: mp 148-150 °C; IR 1726, 1632 cm<sup>-1</sup>; NMR (DMSOd<sub>6</sub>) δ 1.04 (m, 10 H, 4 cyclopropyl plus NHCH<sub>2</sub>CH<sub>3</sub> plus CH<sub>3</sub> on pyrrolidine), 1.58 (m, 1 H, pyrrolidine), 2.00 (m, 2 H, pyrrolidine), 2.48 (m, 2 H, CHNHCH<sub>2</sub>CH<sub>3</sub>), 2.67 (m, 1 H, CHNHEt), 3.33 (m, 1 H, pyrrolidine), 3.66 (m, 3 H, pyrrolidine), 3.95 (m, 1 H, cyclopropyl), 7.10 (bs, 2 H,  $NH_2$  at C-5), 8.41 (s, 1 H,  $C_2H$ ).

The three remaining stereoisomers—32H-R,R, 32H-S,R, and 32H-S,S—were prepared in an identical fashion, giving rise to the following physical data.

**32H-R,R**: NMR (DMSO- $d_6$ )  $\delta$  1.03 (m, 10 H), 1.60 (m, 1 H), 2.11 (m, 2 H), 2.65 (m, 2 H), 3.48 (m, 4 H), 3.80 (m, 1 H), 4.05 (m, 1 H), 7.13 (bs, 2 H), 8.44 (s, 1 H).

**32H-S,R**: NMR (DMSO- $d_6$ )  $\delta$  1.05 (m, 10 H), 1.59 (m, 1 H), 1.97 (m, 1 H), 2.07 (m, 1 H), 2.51 (m, 2 H), 2.69 (m, 1 H), 3.61 (m, 4 H), 3.96 (m, 1 H), 7.10 (bs, 2 H), 8.41 (s, 1 H).

**32H-S,S**: NMR (DMSO- $d_6$ )  $\delta$  1.03 (m, 10 H), 1.63 (m, 1 H), 2.10 (m, 2 H), 2.51-2.67 (m, 3 H), 3.55 (m, 3 H), 3.77 (m, 1 H), 3.97 (m, 1 H), 7.12 (bs, 2 H), 8.43 (s, 1 H).

Compounds 30H-R,R, 30H-R,S, 30H-S,R, 30H-S,S, 34H-R,R, and 34H-R,S were synthesized in identical fashion to 32H-R,S. For compound 32H-R,S, the product obtained from the isoelectric precipitation was dissolved in concentrated HCl; the solution was concentrated and the residue triturated with 2-PrOH to give the desired product as the HCl salt. In many cases, the target compounds precipitated from the cooled CH<sub>3</sub>CN solution in excellent purity, and for these compounds (31H-R,R, 31H-R,S, 31H-S,R, 31H-S,S, 34D, and 36H-R,R) no further purification was necessary. For the quinolone 36H-R,S the reaction mixture was concentrated to a paste which was triturated with H<sub>2</sub>O to give the desired product.

General Method C. Preparation of (3R,1S)-7-[3-(1-Aminoethyl)-1-pyrrolidinyl]-1-cyclopropyl-6-fluoro-1,4-dihydro-8-methoxy-4-oxo-3-quinolinecarboxylic Acid (35F-R,S). A solution of 0.53 g (1.6 mmol) of 1-cyclopropyl-6,7-difluoro-1,4dihydro-8-methoxy-4-oxo-3-quinolinecarboxylic acid-boron difluoride complex (28,  $R_1 = c - C_3 H_5$ ), 0.40 g (1.9 mmol) of pyrrolidine 14, 0.60 g (4.7 mmol) of diisopropylethylamine, and 25 mL of CH<sub>3</sub>CN was stirred at room temperature for 18 h. The solution was concentrated to a paste which was dissolved in 20 mL of EtOH and 5 mL of Et<sub>3</sub>N. The solution was refluxed for 5 h and stirred at room temperature for 18 h. The solvent was evaporated, and the residue was chromatographed on silica gel, eluting with 90:10 CH<sub>2</sub>Cl<sub>2</sub>/MeOH, to give a yellow foam, mp 161-163 °C.

The penultimate intermediate was dissolved in 25 mL of CHCl<sub>3</sub>, cooled to 5 °C, and treated with gaseous HCl for 10 min. The mixture was allowed to warm to room temperature and was concentrated to a paste. The residue was triturated with EtOAc, and the solids were filtered, washed with ether, and dried to give 0.57 g (89% overall) of the title compound as the hydrochloride salt: mp 236-238 °C; IR 1718, 1621 cm<sup>-1</sup>; NMR (TFA) δ 1.21 (m, 1 H, cyclopropyl), 1.33 (m, 1 H, cyclopropyl), 1.50 (m, 1 H, cyclopropyl), 1.61 (m, 1 H, cyclopropyl), 1.69 (d, J = 5.5 Hz, 3 H,  $CH_3$  on pyrrolidine), 2.21 (m, 1 H, pyrrolidine), 2.57 (m, 1 H, pyrrolidine), 3.09 (m, 1 H, pyrrolidine), 3.93 (s, 4 H, OCH<sub>3</sub> plus CHNH<sub>2</sub>), 4.14 (m, 2 H, pyrrolidine), 4.32 (m, 2 H, pyrrolidine), 4.55 (m, 1 H, cyclopropyl), 8.18 (d, 1 H,  $C_5H$ ), 9.41 (s, 1 H,  $C_2H$ ).

The remaining stereoisomers 35F-R.R and 35F-S.R were prepared in an identical fashion, giving rise to the following physical data.

35F-R,R: NMR (DMSO- $d_6$ )  $\delta$  0.98-1.17 (m, 4 H), 1.26 (d, J = 6.0 Hz, 3 H, 1.78 (m, 1 H), 2.21 (m, 1 H), 2.45 (m, 1 H), 3.26(m, 1 H), 3.56 (bs, 6 H), 3.77 (m, 1 H), 4.14 (m, 1 H), 7.66 (d, J = 14 Hz, 1 H), 8.66 (s, 1 H); IR 1728,  $1622 \text{ cm}^{-1}$ .

35F-S,R: NMR (DMSO- $d_6$ )  $\delta$  0.93 (m, 1 H), 1.05–1.25 (m, 3 H), 1.31 (d, J = 6.7 Hz, 3 H), 1.74 (m, 1 H), 2.10 (m, 1 H), 2.50(m, 1 H), 3.30 (m, 1 H), 3.56 (bs, 5 H), 3.70 (m, 2 H), 4.14 (m, 1 H), 7.67 (d, J = 14 Hz, 1 H), 8.66 (s, 1 H).

Compounds 35G-R,S, 37F-R,S, 37G-R,S, 35B, and 37A were prepared in the same manner as was 35F-R,S. For compounds 35G-R.R and 35A, the crude HCl salt was dissolved in water, filtered, and neutralized to pH 7.2. The solids that formed were filtered, redissolved in 6 N HCl, and concentrated. Trituration of this residue with 2-PrOH gave the desired products in pure

Pyrrolidine 16 does not contain a tert-butoxycarbonyl protecting group and therefore did not undergo acid hydrolysis. Rather, the crude product obtained in the initial coupling was simply refluxed in 20 mL of EtOH and 5 mL of Et<sub>3</sub>N and then cooled to room temperature. The reaction mixtures containing 35H-R,R and 35H-R,S were concentrated, and the desired products were chromatographed on silica gel, eluting with a solution of CH<sub>2</sub>Cl<sub>2</sub>/2.6% NH<sub>3</sub> in CH<sub>3</sub>OH/H<sub>2</sub>O (1900 mL:95 mL:5

Preparation of 7-Chloro-1-cyclopropyl-6-fluoro-1,4-dihydro-4-oxo-1,8-naphthyridine-3-carboxylic Acid<sup>27</sup> (17). A solution of 9.8 g (35 mmol) of ethyl (2,6-dichloro-5-fluoronicotinyl)acetate,29 triethyl orthoformate (7.8 g, 53 mmol), and Ac2O (50 mL) was refluxed for 3 h. The solution was cooled to room temperature and concentrated. The residue was dissolved in 150 mL of Et<sub>2</sub>O, cooled to 5 °C, and treated dropwise with 2.2 g (38 mmol) of cyclopropylamine. The suspension was stirred at 5-10 °C for 1 h and then at room temperature for 2 h. Concentration gave a tan solid which was dissolved in 30 mL of dry DMSO. This solution was treated with 14 g (135 mmol) of Et<sub>3</sub>N, stirred at room temperature overnight, and poured into H<sub>2</sub>O (200 mL). The suspension was extracted with CH<sub>2</sub>Cl<sub>2</sub>, and the extract was washed (H2O), dried, and concentrated. This residue was suspended in 100 mL of 6 N HCl, refluxed for 5 h, cooled to room temperature, and filtered. The solids were washed with H<sub>2</sub>O and dried to give 5.1 g of the title compound: mp 210-212 °C; NMR (DMSO- $d_6$ )  $\delta$  1.18-1.26 (m, 4 H, cyclopropyl), 3.83 (m, 1 H, cyclopropyl), 8.70 (d, 1 H,  $C_5H$ ), 8.84 (s, 1 H,  $C_2H$ ); IR (KBr) 1735, 1613 cm<sup>-1</sup>.

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