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Synthesis and Class III Type Antiarrhythmic Activity of 4-Aroyl (and Aryl)-1-aralkylpiperazines

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Abstract—The synthesis and in vitro Class III antiarrhythmic activity of several 4-aroyl (and aryl)-1-aralkylpiperazine and piperidine derivatives are described. Among several potent compounds identified in the series, RWJ-28810 (3), with its EC_{20} of 3 nM, ranks as one of the most potent (in vitro) compounds reported. © 2000 Elsevier Science Ltd. All rights reserved.

Sudden cardiac death (SCD), a primary cause of mortality in patients with cardiovascular diseases, is caused by the loss of regular cardiac rhythm resulting from electrical instability of the cardiac muscle. These malignant ventricular arrhythmias leading to the lethal event are recognized as ventricular tachycardia (VT) and ventricular fibrillation (VF).¹ Although therapeutic interventions are available for VF, they are unsatisfactory from the view point of safety and general efficacy² Since the first observation of antiarrhythmic activity of a cinchona alkaloid preparation in a malaria patient in 1914, the alkaloid quinidine, subsequently identified as the most potent ingredient of such natural product preparation, remains an important antiarrhythmic therapeutic agent.³ However, quinidine and other conventional⁴ Class I antiarrhythmic agents, such as encainide and flecainide, with ionic mechanism of that action include interference with sodium channels in the cell membrane, are effective against simple ventricular, and in some cases, supraventricular arrhythmias, but are ineffective for the prevention of sudden cardiac death by VF. Moreover, the recent results of the Cardiac Arrhythmia Suppression Trial (CAST)⁵ have raised concerns about the safety of the Class I type agents focusing attention on Class III agents. These provide antiarrhythmic protection by selectively prolonging the cardiac action potential and thereby increasing the effective refractory period (ERP).⁶ d-Sotalol (35),⁷ Dofetilide (36),⁸ E-4031 (37),⁹ Amiodarone,¹⁰ and Azimilide¹¹ are a few representative examples of Class III antiarrhythmic agents.

Herein we describe the synthesis and potent in vitro ERP activity of a series of 4-aroyl-1-aralkylpiperazines (Table 1), 4-aryl-1-aralkylpiperazines (Table 2), and a few piperidine derivatives (Table 3).

Chemistry¹²

Various 4-aroyl-1-aralkylpiperazines I (3–11, Table 1) were prepared according to Scheme 1. Acylation of piperazine with 4-nitrobenzoyl chloride afforded monoacylated product 2^{13} separated from the diacylated product 1 by chromatography. 4-Nitroaralkylation of 2 with the various 4-nitroaralkyl halides (or sulphonate IIe) in refluxing MeCN in the presence of anhydrous K₂CO₃ and NaI gave 3 (70%), 6 (94%), 9 (80%), 10 (63%), and 11 (70%). Reduction of NO₂ to NH_2 followed by N-mesylation gave, respectively, 4 and 5 from 3 and a mixture of 7 and 8 from 6. Compounds 12-21 (V, Table 1) were obtained according to Scheme 2 by first aralkylating mono-t-Boc protected piperazine to give IIIa (52%) and IIIb (33%), the latter converted sequentially to NH₂ (IIIc) and NHSO₂Me (IIId) followed by deprotection to IV and then acylation to give V (12–21). 4-Aryl-1-aralkylpiperazines 22–30 (VII, Table 2) were similarly prepared according to Scheme 3 by aralkylating commercially available 4-arylpiperazines VI to VIIa (27, 50%), VIIb [22, (80%), 23 (70%), and 26 (80%)] and VIId,e, [28 (48%) and 31(88%), respectively], reduction of NO₂ to NH₂ (VIIc, 24, 25; VIId, 29), and then N-mesylation of 29 to 30. Three piperidine derivatives 32-34 (IX, Table 3) were similarly obtained according to Scheme 4.

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 Table 1.
 4-Aroyl-1-aralkylpiperazines



#	Ar ₁ -X	Y	Ar ₂	Emperical formula	Mol. Wt.	Mp °C	Antiarrhythmic activity	
							$EC_{20}\;\mu M$	Tension% Δ
1	4-O ₂ N-PhCO-	СО	-Ph-4-NO ₂	C ₁₈ H ₁₆ N ₄ O ₆	384.34	>300	7	-6
2	4-O ₂ N–PhCO–	Н		$C_{11}H_{13}N_3O_3$	235.24	149-50	20	-33
3	4-O ₂ N-PhCO-	$(CH_{2})_{2}$	-Ph-4-NO ₂	C ₁₉ H ₂₀ N ₄ O ₅ /0.25H ₂ O	384.38/388.88	123-25	0.003	-3
4	4-O ₂ N–PhCO–	$(CH_{2})_{2}$	-Ph-4-NH ₂	C ₁₉ H ₂₄ N ₄ O/0.25 H ₂ O	324.41/333.41	154-56	1.40	-14
5	4-MeO ₂ SNH–PhCO–	$(CH_{2})_{2}$	-Ph-4-NHSO ₂ Me	C ₂₁ H ₂₈ N ₄ O ₅ S ₃ /0.25H ₂ O	480.59/485.09	130-135	2.0	+1
6	4-O ₂ N–PhCO–	$(CH_2)_2O$	-Ph-4-NO ₂	C ₁₉ H ₂₀ N ₄ O ₆ /0.5H ₂ O	400.38/404.88	160-61	0.060	0
7	4-MeO ₂ SNH–PhCO–	$(CH_2)_2O$	-Ph-4-NHSO ₂ Me	$C_{21}H_{28}N_4O_6S_2/0.5H_2O$	496.50/505.58	107-09	15	-13
8	4-MeO ₂ SNH–PhCO–	$(CH_2)_2O$	-Ph-4-N(SO ₂ Me) ₂	$C_{22}H_{30}N_4O_8S_3$	674.67	212-13	25	-14
9	4-O ₂ N–PhCO–	CH ₂ CONH	-Ph-4-NO ₂	$C_{19}H_{19}N_5O_6$	413.38	192–93	0.035	+5
10	4-O ₂ N–PhCO–	CH_2	-Ph-4-NO ₂	$C_{18}H_{18}N_4O_5/0.25H_2O$	370.36/374.86	211-12	3.5	-44
11	4-O ₂ N–PhCO–	$(CH_{2})_{2}$	-Th ^a -5-NO ₂	$C_{17}H_{18}N_4O_5S$	390.41	137-38	0.01	-17
12	4-O ₂ N–PhCO–	$(CH_{2})_{2}$	–Ph	$C_{19}H_{21}N_3O_3$	339.38	101-02	0.16	-8
13	PhCO-	$(CH_{2})_{2}$	-Ph-4-NO ₂	$C_{19}H_{21}N_3O_3$	339.38	138-39	0.14	-7
14	4-O ₂ N–PhCO–	$(CH_{2})_{2}$	-Ph-4-NHSO ₂ Me	C ₂₀ H ₂₄ N ₄ O ₅ S/0.25H ₂ O	432.48/436.96	186-87	0.16	+13
15	4-MeO ₂ SNH–PhCO–	$(CH_{2})_{2}$	-Ph-4-NO ₂	$C_{20}H_{24}N_4O_5S$	432.48	170-71	0.06	1
16	4-MeO ₂ SNH–PhCO–	$(CH_{2})_{2}$	–Ph	$C_{20}H_{25}N_3O_3S$	387.48	155-56	0.68	5
17	5-O ₂ N-Th ^a -	$(CH_{2})_{2}$	-Ph-4-NO ₂	$C_{17}H_{18}N_4O_5S$	390.42	101-02	0.012	0
18	5-O ₂ N-Fu ^b -	$(CH_{2})_{2}$	-Ph-4-NO ₂	$C_{17}H_{18}N_4O_6$	374.35	91–92	0.14	-2
19	4-PyrCO-	$(CH_2)_2$	-Ph-4-NO ₂	$C_{18}H_{20}N_4O_3/0.25H_2O$	340.37/344.87	145-46	1	+19
20	3-O ₂ N-PhCO-	$(CH_2)_2$	-Ph-4-NO ₂	$C_{19}H_{20}N_4O_5/0.25H_2O$	384.38/393.38	145-46	0.27	-7
21	$4-O_2N-PhSO_2-$	$(CH_{2})_{2}$	$-Ph-4-NO_2$	$C_{18}H_{20}N_4O_6S$	420.43	154–55	1.0	-24

 ${}^{a}Th = Thiophene.$ ${}^{b}Fu = Furan.$

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#	Ar ₁	Y	Ar ₂	Emperical formula	Mol. Wt.	Mp °C	Antiarrhythmic activity	
							$EC_{20}\;\mu M$	Tension% Δ
22	4-O ₂ N-Ph-	(CH ₂) ₂	-Ph-4-NO ₂	C ₁₈ H ₂₀ N ₄ O ₄	356.37	148-49	0.03	+ 5
23	$4-O_2N-Ph-$	(CH ₂) ₂ O	$-Ph-4-NO_2$	$C_{18}H_{20}N_4O_5$	372.37	150-51	0.35	-27
24	$4-H_2N-Ph-$	$(CH_2)_2$	$-Ph-4-NH_2$	C ₁₈ H ₂₄ N ₄ /0.8H ₂ O	296.40/310.82	158-59	>100	-53^{*}
25	$4-H_2N-Ph-$	$(CH_2)_2O$	$-Ph-4-NH_2$	C ₁₈ H ₂₄ N ₄ O/0.75H ₂ O	312.40/325.91	105-06	>100	-43^{*}
26	$4-O_2N-Ph-$	$(CH_2)_2$	–Ph	C ₁₈ H ₂₁ N ₃ O ₂	311.39	163-64	0.5	-6
27	Ph–	$(CH_2)_2$	-Ph-4-NO ₂	C ₁₈ H ₂₁ N ₃ O ₂ /0.125H ₂ O	311.39/313.64	140-41	0.8	+10
28	4-Pyr–	$(CH_2)_2$	$-Ph-4-NO_2$	C ₁₇ H ₂₀ N ₄ O ₂ /0.5H ₂ O	312.36/321.36	146-48	0.03	+ 8
29	4-Pyr-	$(CH_2)_2$	-Ph-4-NH ₂	C17H22N4/0.5H2O	282.38/291.38	184-85	15	+9
30	4-Pyr-	$(CH_2)_2$	-Ph-4-NHSO ₂ Me	C ₁₈ H ₂₄ N ₄ O ₂ S/0.5H ₂ O	360.46/369.46	202-03	0.22	+2
31	5-O ₂ N-2-Pyr-	$(CH_2)_2$	-Ph-4-NO ₂	$C_{17}H_{19}N_5O_4/0.5H_2O$	357.36361.86	137–38	1.3	-21

Table 3. 4-Aryl(or aroyl)-1-aralkylpiperidines



#	Ar ₁ X	Y	Ar ₂	Emperical formula	Mol. Wt.	Mp °C	Antiarrhy	thmic activity
							$EC_{20}\;\mu M$	Tension% Δ
32 33 34	$\begin{array}{l} \text{4-O}_2\text{N-Ph-}\\ \text{4-H}_2\text{N-Ph-}\\ \text{4-Cl-PhCO-} \end{array}$	(CH ₂) ₂ (CH ₂) ₂ (CH ₂) ₂	$\begin{array}{c} -Ph\text{-}4\text{-}NO_2\\ -Ph\text{-}4\text{-}NH_2\\ -Ph\text{-}4\text{-}NO_2\end{array}$	$\begin{array}{c} C_{19}H_{21}N_3O_4/0.25H_2O\\ C_{19}H_{25}N_3/0.25H_2O\\ C_{20}H_{21}ClN_2O_3 \end{array}$	355.38/359.88 295.43/299.91 372.84	133–34 140–42 1501	0.3 2.0 0.3	$0\\+15\\-4$

Biological Materials and Methods

Measurement of effective refractory period (ERP)

Papillary muscles were dissected from the right ventricle of anesthetized male ferrets, secured at one end and attached to a force transducer. Muscles were placed in a tissue bath containing Tyrode's solution of the following composition (mM): NaCl, 130; KCl, 4; MgCl₂, 1.0; NaHCO₃, 25; KH₂PO₄, 1.2; CaCl₂, 2; dextrose, 11; pH 7.3, gassed with 5% CO₂. 95% O₂ at 37 °C. Timolol (0.1 mM) was added to prevent variations in tension produced by spontaneous or stimulation-induced release of catecholamines. Isometric force was recorded during stimulation with constant current pulses (3 msec; $1.3 \times$ threshold) using platinum bipolar electrodes. The ERP was determined using the extra stimulus technique by applying trains of 20 basic stimuli (S₁) at cycle length of 300 msec (1 Hz), followed by application of an extra



Scheme 1. Reagents: (a) 4-O₂NPhCOCl, acetone, HCl; (b) 4-O₂NPh(CH₂)₂-Br (Cl), K₂CO₃, NaI, MeCN, Δ ; (c) Ra-Ni, H₂, EtOH; (d) MeSO₂Cl, Pyr, CH₂Cl₂; (e) 4-O₂NPhO(CH₂)₂Cl, K₂CO₃, NaI, MeCH, Δ ; (f) 4-O₂NPhNHCOCH₂Cl, K₂CO₃, MeCN, Δ ; (g) O₂NPhCH₂Cl, K₂CO₃, NaI, MeCN, Δ ; (h) (Ac)₂O, H₂SO₄; (i) HNO₃, (Ac)₂O, -10 °C \rightarrow 0 °C; (j) 10% H₂SO₄/H₂O, Δ ; (k) MeSO₂Cl, Et₃N, CH₂Cl₂, 0°C; (l) IIe, EtOH, Δ .



Scheme 2. Reagents: (a) Ar_2-Y-Br , K_2CO_3 , NaI, MeCN, Δ ; (b) Ra-Ni, H₂, EtOH; (c) MeSO₂Cl, Pyr, CH₂Cl₂; (d) CF₃CO₂H; (e) Ar_1-X-C , Et₃N, CH₂Cl₂.



Scheme 3. Reagents: (a) Ar_2 -Y-Cl (or Br), K_2CO_3 , NaI, MeCN, Δ ; (b) Ra-Ni, H₂, EtOH; (c) MeSO₂Cl, Pyr.



Scheme 4. Reagents: (a) HNO₃, H₂SO₄; (b) Ar_2 -(CH₂)₂-Br, K₂CO₃, NaI, MeCN, Δ ; (c) Pd/C, H₂, EtOH.

#			Antiarrhythmic activity	
			$EC_{20} \ \mu M$	Tension% Δ
35	d-Sotalol	MeSO ₂ NH-CH(OH)CH ₂ NHCH(Me) ₂	3.5	-13
36	Dofetilide	MeSO ₂ NH-(CH ₂) ₂ N(CH ₂) ₂ O-(NHSO ₂ Me	0.015	+ 5
37	E-4031	$MeSO_2NH \longrightarrow N-(CH_2)_2 \longrightarrow Ne$	0.02	+9

Table 4. Class III antiarrhythmic standards

stimulus (S₂) at decreasing coupling intervals (S₁–S₂) in decrements of 2 ms until muscle activation failed. ERP was defined as the shortest S₁–S₂ interval producing muscle activation. Test compounds were dissolved in dimethylsulfoxide (DMSO) at a stock concentration of 1×10^{-2} M and added in appropriate amounts directly to Tyrode's solution to achieve the final concentrations. Following duplicate control measurements, compound concentrations were taken at steady state, usually 15 min after addition of each concentration. The mean dose that produced a 20% increase in ERP is defined as EC₂₀ and is reported in μ M (1×10⁻⁶ M), as a measure of Class III antiarrhythmic activity.

Measurement of tension

Tension was measured as peak isometric tension during the basic stimulus trains (S_1) during control, and compound induced changes in tension were expressed as a percent of control.

Results and Discussion

SAR highlights

In in vitro tests, the most potent compound of the series, RWJ-28810 (3), increased ERP ~5 times more potently than Dofetilide (36),⁸ the most potent reference standard tested (Table 4). In the piperazine series, the general potency order for the $4,4^{-1}$ -substituents on the two terminal aromatic rings, keeping NO₂ constant on one end while varying the other is, for Ar_1CO (I and V): NO_2 (3) >NHSO₂Me (15) > H (13); for Ar₂: NO₂ (3) > $H(12) > MeSO_2NH(14) > NH_2(4)$ [3-NO₂ on Ar₁(20) is much weaker than $4-NO_2$ on Ar_1 (3)]; for any any logazines (VII), on Ar_1 : NO₂ (22) > H (27); on Ar_2 : NO₂ $(22) > H(26) > NH_2$ (24). Of the two possible regioisomeric mono-NO₂ analogues 12 and 13 of the di-NO₂ lead 3, the 4-NO₂-aroyl analogue 12, although 5 times less potent than 3, is 10 times more potent than the 4-NO₂-aralkyl isomer 13. With mixed functionality, present on different aryl moieties, the potencies are reversed in favor of MeSO₂NH-aroyl/NO₂-aralkyl (15) versus NO_2 -aroyl/MeSO₂NH-aralkyl (14). For the aromatic ring Ar_1 potency order is: benzoyl (3) > thienoyl (17) >> furancyl (18) > 4-pyridcyl (19); for Ar₂ potency order is: phenyl > thienyl. The non-basic amido nitrogen of aroyl-piperazine core (as in I) is replaceable by a

CH group (e.g. piperidine derivative 34) as also is the anilino nitrogen of VII, but with significant loss of potency (compare 22 with 32). For the spacer group X, whereas CO is better than SO₂ (21), it is altogether dispensable (e.g. 22). For the spacer group Y, $(CH_2)_2$ (3) is better than $(CH_2)_2O$ (6) or CH_2 (10). There were only two compounds showing significant decrease in tension (24 and 25, possibly 10) and these were at high μ M concentrations where vehicle (DMSO) is known to decrease tension. Others had little or no change in tension indicating selectivity of Class III action, since selective block of K channels is associated with no change or an actual increase in isometric tension in the absence of an effect on Na or Ca channels.

Summary

We have synthesized and identified a series of 4-aroyl (andaryl)-1-aralkylpiperazine derivatives as potent Class III Type antiarrhythmics as evidenced by in vitro ERP prolongation.

The lead compound RWJ-28810 (3) with its EC_{20} of **3nM** represents one of the most potent (in vitro) compounds reported.

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