imidoethane Iodide (7). To 7.3 g (1.98 mmoles) of 5 in 150 ml of abs EtOH was added gradually MeI (3.76 g, 26.5 mmoles) and the mixt was refluxed for 1 hr. On standing, 7 pptd; some more 7 pptd on addn of dry Et₂O. The light-sensitive salt was filtered, washed (Et₂O), and dried at 0°, yield 9.2 g (91.1%), mp 186-187°. Anal. $(C_{22}H_{27}IN_2O_4)$ C, H, N; m/e 368 (M^{*} - CH₃I).

The homologous salt (8) prepd from 6 was obtd in 99% yield, mp $152-153^{\circ}$. Anal. ($C_{23}H_{29}IN_2O_4$) C, H, N. Hydrazinolysis of Methiodides 7 and 8. To a suspension of 8 or

Hydrazinolysis of Methiodides 7 and 8. To a suspension of 8 or 7 (2.4-2.55 g, 4.5-5 mmoles) in 25 ml of 95% EtOH was added 0.7-0.75 g (ca. 14.5 mmoles) of 64% hydrazine hydrate. The mixt turned pale yellow and the solid went into soln on heating. After 2 hr of reflux and cooling overnight, a white ppt had formed. HCl (37%) was added dropwise to Congo Red, and the yellowish ppt was filtered off and washed with EtOH and then H₂O, and the filtrate was evapd. Recrystn of the pale yellow residue from EtOH and from hexane gave 1.28 g (61.5%) of 3-methoxy-4-trimethylammoniumethoxyphenethylamine-HCl·T (9), mp 181-182° dec [Anal. ($C_{14}H_{26}CIIN_2O_2$) C, H], and 1-(3-methoxy-4-trimethylammoniumethoxyphenyl)-2-aminopropane-HCl·T (10) (1.4 g, 71%), mp 177-179° dec [Anal. ($C_{15}H_{25}CIIN_2O_2$) C, H, N], respectively. Hydrazinolysis of 5 and 6. To an iced aqueous soln of salts 5

Hydrazinolysis of 5 and 6. To an iced aqueous soln of salts 5 (or 6) (0.01 mole) was added NaOH (0.01 mole) and the liberated amine was extd with Et₂O (3×50 ml). The solvent was evapd, and the residue dissolved in 25 ml of EtOH and a 10% excess of 64% hydrazine hydrate. The warm mixt was refluxed for 45 min, then 6 N HCl (2 ml) was added dropwise. Phthalhydrazide was filtered off and washed (EtOH, H₂O), the filtrate was concd, and a white ppt was filtered off. The yellow residue from the evapd filtrate was recrystd from abs EtOH: 4-dimethylaminoethoxy-3-methoxyphenethylamine-2HCl (11), mp 198-200° dec, yield 36% [A nal. (C₁₃H₂₄Cl₂N₂O₂·H₂O) C, H, N]; 1-(4-dimethylaminoethoxy-3-methoxyphenyl)-2-aminopropane-2HCl (12), mp 209-211°, yield 54% [A nal. (C₁₄H₂₆Cl₂N₂O₂) C, H, N; m/e 252 (M⁺)].

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1,4-Disubstituted Piperazines. 3. Piperazinylbenzothiazoles[†]

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In the course of synthesis of 1,4-disubstituted piperazines,¹ 4-(2-benzothiazolyl)-*N*,*N*-diethyl-1-piperazinecarboxamide (2) was prepared and appeared to exhibit activity against coccidiosis in chickens. Since 2-(1-piperazinyl)benzothiazole was new to the literature, the synthesis and biological studies of this parent and its derivatives I, reported in this paper, were undertaken. Retesting of 2, however, failed to sustain coccidiostatic activity; nor was this activity shown by any of the type I compounds. Of interest, though, was the antifungal action and CNS effects shown by some of these piperazinylbenzothiazoles.

Biological Data.[‡] Compds 1, 5, and 6 were marginal psychomotor stimulants in mice at 300 mg/kg po. Compd 1 also showed antihypertensive activity in rats (approx $ED_{50} = 6 \text{ mg/kg sc}$). Compds 8 and 14 produced decreased locomotor activity in mice at 16 mg/kg po, and 64 mg/kg po,

respectively. Mice were hyperactive at 256 mg/kg po with 20, 21, 22, and 23. They showed ataxia at 256 mg/kg po with 24. In spot tests against *Trichophton mentagrophytes*, *Asperigillus niger*, and *Candida albicans*, 21, 22, 23, and 24 were active. Compd 9 showed marginal in vitro activity against *Staphylococcus aureus*, *Pseudomonas aeruginosa*, *Escherichia coli*, and *Proteus vulgaris*, but was inactive against staphylococcal infections in mice. Many of these compounds were also tested for possible anthelmintic, amebi-



asis, antimalarial, schistosomiasis, antiviral, and antiinflammatory activities.

Experimental Section §

All microanalyses were performed at the Sterling-Winthrop Research Institute. The chemicals were either purchased from Eastman, K and K, or Aldrich, including 1-methylpiperazine and 2-(1piperazinyl)ethanol. The following monosubstituted piperazines were prepd by literature methods: 1-(N,N-diethylcarbamoyl)piperazine,² 1-formylpiperazine,³ 1-diphenylmethylpiperazine,⁴ 1-(2-N,N-dimethylaminoethyl)piperazine,⁵ 1-benzylpiperazine,⁶ and 1-pchlorophenylpiperazine.⁷

Also prepd as intermediates were: 2-bromo-4-chlorobenzothiazole,⁶ 2-chloro-6-methoxybenzothiazole,⁹ 2-chloro-6-nitrobenzothiazole,¹⁰ and 1-chloroacetyl-3-methylurea.¹¹ 2-Chloro-6-ethoxybenzothiazole was prepd according to the procedure used in making 2-chloro-6-methoxybenzothiazole, but was not analyzed: mp 63-67° from hexane.

1,4-Disubstituted Piperazines (Table I). Procedure A. One equiv of the alkyl halide was slowly added to a vigorously agitated mixt of 2 equiv of piperazine in 80% alcohol plus excess NaHCO₃ heated at reflux. After filtration, the solvent was removed under reduced pressure. Addn of concd HCl resulted in pptn of the corresponding salt for 1. For 15, in addn to many other compds here described, dil HCl dissolved the residue, which was then washed with Et_2O or EtOAc and repptd by addn of NaOH. The products were recrystd from the solvents shown in Table I.

Procedure B. Two equiv of the appropriate monosubstituted piperazine was used to 1 equiv of the alkyl halide in PhH.

Procedure C. A 10% molar excess of the appropriate alkyl halide plus 2-(1-piperazinyl)benzothiazole HCl in an excess of NaHCO₃ and EtOH-H₂O represented the reaction mixt.

Procedure D. Two equiv of the appropriate monosubstituted piperazine to 1 equiv of the alkyl halide in EtOH was used.

Procedure E. A 10% molar excess of the alkyl halide to the appropriate monosubstituted piperazine in excess NaHCO₃ plus EtOH-H₂O was used.

Procedure F. Here the mixt consisted of an EtOH soln of a 20% excess of the corresponding C=O compd plus 2-(4-amino-1-piperazinyl)benzothiazole.[#]



 $Where analyses are indicated only by symbols of the elements or functions, analytical results obtained were within <math>\pm 0.4\%$ of the theoretical values.

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[‡]The author wishes to extend his thanks to the Biological Division of Sterling-Winthrop Research Institute for conducting these studies.

[#]This compd was prepd impure from 9 according to Conroy's¹² method of reduction of 1-(p-chlorophenyl)-4-nitrosopiperazine; mp 78-85° from petr ether (90-120°), followed by washing with ether.

Table I. 1,4-Disubstituted Piperazines (II)

No.	R	X	Yield, %	Mp, ^a °C	Procedure ^b	Reaction ^c time, hr	Recrystn ^d solvent	Formula	Analy- ses
1	н	H(HCl)	75	290 dec	A	8	A	C ₁₁ H ₁₃ N ₃ S·HCl	Cl, S
2	Н	$CON(C_2H_5)$	54	86-87	В	0.5 (60°)	В	C ₁₆ H ₂₂ N ₄ OS	N, S
3	Н	CH ₃	58	89-9 0	В	48 (room temp)	В	$C_{12}H_{15}N_{3}S$	N, S
4	Н	CHÔ	83	142-143	В	1 (50°)	S + C	C ₁ , H ₁ , N ₂ OS	N, S
5	Н	CH ₂ C ₄ H ₄	92	134-135	С	4	B + C	C ₁ ^s H ₁ ^s N ₃ S	N, S
6	Н	$CH_{2}C(Br) = CH_{2}$	89	140-141	С	3	В	$C_{14}H_{16}BrN_{3}S$	Br, N, S
7		CH ₂ -Cl	47	136-137	С	3	E (first), N (second)	C ₁₈ H ₁₇ Cl ₂ N ₃ S	CI, N, S
8	Н	(CH ₂) ₄ CH ₂	60	89-89.5	С	1.8	F	C, H, N,S	N, S
9	Н	NO	68	129-130	е		G	C ₁₁ H ₁₂ N ₄ OS	N, S
10	Н	$CH(C_6H_5)_2$	57	166.5-167.5	С	1.5	н	C ₂₄ H ₂₃ N ₃ S	N, S
11	Η	CH,COOH	54	263 dec	С	1.8 (room temp)	I	C13H16N3O2S	N, S
12	O ₂ N	H(HCl)	26	250 dec	С	2 (room temp)	J	C ₁₁ H ₁₂ N ₄ O ₂ S · HCl	CI, S
13	O ₂ N	CH ₂ CH ₂ N(C ₂ H ₄), HCl	54	248-250 dec	Е	5 (room temp)	К	C ₁₇ H ₂ N ₂ O ₂ S·HCl	Cl, N
14	CĤ₄O	CH,	60	102.5-103.5	D	9	L	C ₁ ,H ₁ ,N ₃ OS	N, S
15	CH O	н	71	152-153	Α	2.5	N + C	C ₁₂ H ₁ N ₃ OS	N, S
16	Н	N=C(CH ₃) ₂	20	122-124	F	1	F (first),	$C_{14}H_{18}N_4S$	N, S
17	CHO	сн с н	60	111-112	F	10	I (second)	CHNOS	NS
10	сп ₃ 0 ц	CH CONNCONNCH	83	205_207	č	1	Ď	C H N O S	N S
10	11		05	203-207	C	1	0	C151119145025	11, 0
19	Н	<	70	208	Ε	1.5	E + T (1:3)	C ₁₇ H ₁₆ ClN ₃ S	N, S
20	Н	N=CH-	65	165-166	F	1	N + H (1:1)	$C_{17}H_{17}N_{5}S$	N, S
21	н	CH.CH.N(C.H.).	58	54.5-55	Е	1.8	Р	C. H. N.S	N. S
22 ^f	H	CH ₃	82	80-81	Đ	9	L (first), R (second)	C ₁₂ H ₁₄ CIN ₃ S	N, S
22f	н	СН СН ОН	69	107 5-108 5	F	85	S + N(1.4)	C H CIN OS	NS
2.3		CH_2CH_2OH	49	52-52 5	F	0.0	P	C H N OS	NS
24	002115		72	52-52.5	L).2	x	C1911301400	11, 5
25	н		83	210-213	С	2 (room temp)	Т	$C_{17}H_{15}N_{5}O_{4}S$	N, S
		NO ₂							
26	Н	N=CH-O-NO ₂ g	71	201-202	F	1.3	U	C ₁₆ H ₁₅ N ₅ O ₃ S	N, S
2 7	Н	COOCH ₂ CH ₂ CH ₃	84	88-89	С	h	N	$C_{15}H_{19}N_{3}O_{2}S$	N, S

^aMelting points (uncorr) were taken on a Fisher-Johns block. ^bThese procedures are described in the Experimental Section. ^cUnless otherwise specified, the reactions were carried out at reflux temps. ^dA, 92% O in water; B, petr ether (65-110°); C, CCl₄; E, acetone; F, O plus H₂O; G, O first, then triturated and washed with Et₂O; H, EtOAc; I, AcOH-MeOH; J, first O, second H₂O; K, abs O; L, petr ether (60-90°); N, petr ether (90-120°); O, EtOH; P, hexane; R, petr ether (30-60°); S, benzene; T, dioxane; U, not recrystd: washed with O then ether. ^ePrepd according to a lit. method for making 1-(*p*-chlorophenyl)-4-nitrosopiperazine.¹² fAlso bears Cl at the 4 position of the benzothiazole ring. ^gPrepd by acid treatment of 5-nitrofurfural diacetate followed by ether extn of the liberated aldehyde. ^hHeated to 60° for 5 min and set to stand overnight at room temp.

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Synthesis of 1,2,4-Triazoles as Potential Hypoglycemic Agents

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Impetus for a study of triazoles was provided by a report that certain 4-alkyl-5-aryl-4H-1,2,4-triazole-3-thiols produced hypoglycemia in normal and alloxan-diabetic rats. The potency of these compounds was comparable to that of N^1 -(p-tolylsulfonamido)- N^3 -(n-butyl)urea (tolbutamide) while their duration of action was greater.^{1,2} Consequently,