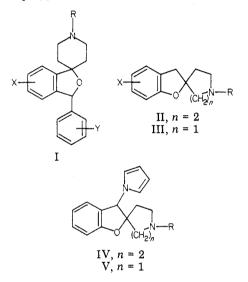
2,3-Dihydro-3-(1-pyrryl)spiro[benzofuran-2,4'-piperidine]s and 2,3-Dihydro-3-(1-pyrryl)spiro[benzofuran-2,3'-pyrrolidine]s: Novel Antihypertensive Agents

Larry Davis,*,[†] Marc N. Agnew,[†] Richard C. Effland,[†] Joseph T. Klein,[†] Jan M. Kitzen,[‡] and Mary A. Schwenkler[‡]

Chemical Research Department and Department of Biological Sciences, Hoechst-Roussel Pharmaceuticals Inc., Somerville, New Jersey 08876. Received April 11, 1983

A series of 2,3-dihydro-3-(1-pyrryl)spiro[benzofuran-2,4'-piperidine]s (IV) and 2,3-dihydro-3-(1-pyrryl)spiro[benzofuran-2,3'-pyrrolidine]s (V) was synthesized and evaluated for cardiovascular activity. The majority of the compounds displayed good antihypertensive activity in the spontaneous hypertensive rat model at 50 mg/kg po. Compounds 5 (2,3-dihydro-1'-methyl-3-(1-pyrryl)spiro[benzofuran-2,4'-piperidine]) and 12a (2,3-dihydro-1'-ethyl-3-(1-pyrryl)-spiro[benzofuran-2,4'-piperidine]) and 12a (2,3-dihydro-1'-ethyl-3-(1-pyrryl)-spiro[benzofuran-2,4'-piperidine]) were selected for a more detailed cardiovascular evaluation in the renal hypertensive rat and for standard cardiovascular challenges in anesthetized dogs and the sinoaortic-deafferented dog.

The 3-aryl-1,3-dihydrospiro[isobenzofuran-1(3H),4'piperidine]s (I) have demonstrated marked activity in

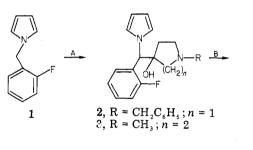


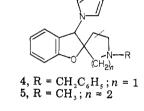
pharmacological tests, indicating antidepressant,¹ neuroleptic,² diuretic,³ and antihypertensive⁴ utility. This, coupled with our successful synthesis of a related series of 2,3-dihydrospiro[benzofuran-2,4'-piperidine]s (II) and 2,3-dihydrospiro[benzofuran-2,3'-pyrrolidine]s (III) via a nucleophilic aromatic fluorine displacement,⁵ as well as our interest in pyrrole chemistry, prompted us to explore further this chemistry and prepare a series of 2,3-dihydro-3-(1-pyrryl)spiro[benzofuran-2,4'-piperidine]s (IV) and 2,3-dihydro-3-(1-pyrryl)spiro[benzofuran-2,3'pyrrolidine]s (V) (Table I). As part of our continuing program to find new antihypertensive agents with nonclassical structures that are devoid of the side effects that commonly plague existing blood pressure lowering agents, these compounds were evaluated for antihypertensive activity in the spontaneously hypertensive rat assay, resulting in the selection of 5 and 12a for further study.

Chemistry. The synthesis of the initial spiro compounds 4 and 5 is shown in Scheme I. 1-(2-Fluorobenzyl)pyrrole (1), prepared by condensation of 2-fluorobenzylamine and 2,5-dimethoxytetrahydrofuran, was metalated with *n*-butyllithium in tetrahydrofuran at -80 °C and then reacted with 1-methyl-4-piperidone or 1-benzyl-3-pyrrolidone (method A) to give the alcohols 2 and 3 in good yield. The alcohols 2 and 3 were cyclized to the spiro compounds 4 and 5 through a base-induced nucleophilic displacement of fluoride with sodium hydride in a

[†]Chemical Research Department.

Scheme I





10% solution of dimethylformamide in benzene at 70 °C (method B).

Aromatic fluoride displacements of this type generally proceed in high yields; the lower yields (30-40%) encountered in this cyclization step are accounted for by a competing reaction in which the alcohol 2 or 3, in a retro fashion, eliminates the stabilized carbanion of 1 and the corresponding piperidone or pyrrolidone.

A number of attempts were made to minimize this retro elimination and to improve the yield of this cyclization step. Using sodium hydride, but increasing the ratio of dimethylformamide-benzene, or using dimethylformamide alone at various temperatures (30-80 °C), gave lower yields of 4 or 5 with increased retro elimination of 1. Using tetrahydrofuran or tetrahydrofuran-dimethylformamide mixtures gave similar results. Sodium hydride and benzene alone gave no reaction. Warming the lithium alkoxide of alcohol 3, generated in the synthesis of 3 (procedure A), at 30-65 °C gave no further reaction. Addition of dimethylformamide to the tetrahydrofuran solution of the lithium alkoxide resulted only in retro elim-

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[‡]Department of Biological Sciences.

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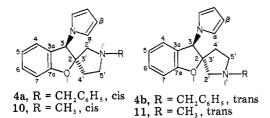
Table I. 2,3-Dihydro-3-(1-pyrryl)spiro[benzofuran-2,4'-piperidine]s and -spiro[benzofuran-2,3'-pyrrolidine]s^a

no.	\mathbf{R}^{i}	n	method ^b	mp, °C	yield, ^c %	recrystn solvent ^d	formula	anal.
4	CH ₂ C ₆ H ₅ ^e	1	В	136-138	35	A-B	$C_{22}H_{22}N_2O\cdot C_2H_2O_4^{f}$	C, H, N
5	CH ₃	2	B	217 - 219	20	Α	$C_{17}H_{20}N_{2}OHCl$	C, N, H^g
8	H ^e	1	D	161-163	. 30	A-B	$\mathbf{C}_{15}\mathbf{H}_{16}\mathbf{N}_{2}\mathbf{O}\cdot\mathbf{C}_{4}\mathbf{H}_{4}\mathbf{O}_{4}\mathbf{h}_{4}$	C, H, N
9	Н	2	D	125 - 126	75	A-B	$\mathbf{C}_{16}^{\dagger}\mathbf{H}_{18}^{\dagger}\mathbf{N}_{2}^{2}\mathbf{O}\cdot\mathbf{C}_{2}^{\dagger}\mathbf{H}_{2}^{\dagger}\mathbf{O}_{4}^{\dagger}\mathbf{f}$	C, H, N
10	CH_3 (cis)	1	F	82-84	80	С	$C_{16}H_{18}N_2O$	С, Н
11	CH_3 (trans)	1	F	149-150	55	A-B	$\mathbf{C}_{16}\mathbf{H}_{18}\mathbf{N}_{2}\mathbf{O}\cdot\mathbf{C}_{2}\mathbf{H}_{2}\mathbf{O}_{4}^{f}$	С, Н
12a	CH, CH,	2	\mathbf{F}	115-118	50	A-B	$\mathbf{C}_{18}^{\circ}\mathbf{H}_{22}^{\circ}\mathbf{N}_{2}^{\circ}\mathbf{Q}\cdot\mathbf{C}_{2}^{\circ}\mathbf{H}_{2}^{\circ}\mathbf{O}_{4}^{\circ}f$	C, H, N
12b	CH,CH,OH	2	\mathbf{F}	i	50	j	$C_{18}H_{22}N_2O_2$	С, Н
12c	CH ₂ CH ₂ CH ₃	2	F	130-132	51	A	$C_{19}H_{24}N_{2}OC_{2}H_{2}O_{4}f$	C, H, N
12d	$CH_{2}CH_{2}N(CH_{2}CH_{3})_{2}$	2	Е	115-117	83	Α	$C_{22}H_{31}N_{3}O 2C_{2}H_{2}O_{4}f$	C, H, N
12e	CH ₂ CH ₂ C ₆ H ₅	2	F	112 - 115	56	Α	$C_{14}H_{16}N_{1}O\cdot C_{2}H_{2}O_{4}\cdot 0.5H_{2}O^{T}$	C, H, N
12f	$CH_{2}CH_{2}CH_{2}CH_{2}CO(4-FC_{6}H_{4})$	2	\mathbf{E}	120-123	78	Α	$C_{26}^{24}H_{27}^{40}FN_{2}O_{2}C_{2}H_{2}O_{4}$	C, H, N
13	C(=NOĤ)NĤ ₂	2		160-162	71	C-D	$C_{17}^{20}H_{20}^{20}N_{4}O_{2}$	C, H, N

^a All compounds exhibited IR and ¹H NMR spectra consistent with the structure. ^b See Experimental Section. ^c Isolated yield; no efforts were made to optimize these yields. ^d A = ethyl acetate; B = methanol; C = hexane; D = ethyl ether. ^e Mixture of cis and trans isomers. ^f Acid oxalate salt. ^g H: calcd, 6.95; found, 7.42. ^h Acid maleate salt. ⁱ Isolated as a heavy oil. ^j Purified by Kugelrohr distillation, 200-210 °C (0.4 mmHg).

ination and the isolation of 1.

Further transformations of the spiro compounds 4 and



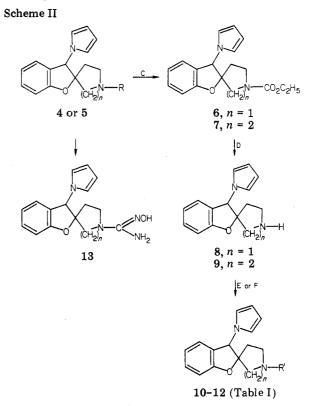
5 are shown in Scheme II. Dealkylation of the tertiary amines to the secondary amines was effected by the reaction of the N-benzyl or N-methyl tertiary amines 4 or 5 with ethyl chloroformate and potassium carbonate in refluxing benzene to give the corresponding carbamates 6 and 7 (method C). The carbamates 6 and 7 were hydrolyzed by refluxing with 10% aqueous potassium hydroxide solution in 1-propanol to afford the secondary amines 8 and 9 (method D). Reduction of the *cis*- and *trans*-carbamates 6 with lithium aluminum hydride in tetrahydrofuran gave the *cis*- and *trans*-N-methylpyrrolidines 10 and 11.

The secondary amines 8 and 9 were either alkylated directly with an alkyl halide in dimethylformamide with potassium carbonate at 70 °C (method E) or acylated with an acid chloride, followed by reduction of the resulting amide with lithium aluminum hydride in tetrahydrofuran (method F), to yield the N-substituted spiro compounds 12a-f.

Reaction of the N-methyl compound 5 with cyanogen bromide and sodium carbonate in refluxing chloroform gave the cyanamide, which was then reacted with hydroxylamine hydrochloride and sodium carbonate in dimethylformamide to give the N-amino(hydroxyimino)methyl 13 in an overall yield of 60%.

The mixture of spiro pyrrolidine diastereoisomers 4 was separated by preparative HPLC. The stereochemical assignments of each diastereoisomer were made based on carbon-13 NMR and proton NMR.

Carbon-13 NMR is a technique that is remarkably sensitive to the orientation of the carbon in a molecule and has been used previously to aid in the assignment of dia-



stereoisomers.⁶ When the chemical shifts of the model compounds *N*-methylpyrrolidine,⁷ 2,3-dihydro-2,2-dimethylbenzofuran,⁸ and benzylamine,⁹ in conjunction with the techniques of quaternary enhancement,¹⁰ continuous-wave off-resonance decoupling,¹¹ and second-order

- (6) F. Wehrli and T. Wirthlin, "Interpretation of Carbon-13 Nuclear Magnetic Resonance Spectra", Heyden, London, 1976, p 188.
- (7) "Sadtler Standard ¹⁸C NMR Spectra", Vol. 46, Sadtler Research Laboratories, Philadelphia, 1980, p. 9049.
- (8) Reference 7, Vol. 18, 1978, p 3414.
- (9) Reference 7, Vol. 1, 1976, p 140.
- (10) I. H. Sadtler, J. Chem. Soc., Chem. Commun., 809 (1973).
- (11) J. B. Stothers, "Carbon-13 NMR Spectroscopy", Academic Press, New York and London, 1972, p 38.

effects in the latter,¹² were used, all aliphatic and most aromatic carbons were unambiguously assigned (Table II).

The chemical shifts of the ring carbons among isomers 4a,b, 10, and 11 remain essentially unchanged, with the exception of C-2' and C-4', the only carbons influenced significantly by their proximity to the pyrrole ring. This proximity and the resultant steric crowding produce a steric compression shift to higher field of the pyrrolidine ring carbon oriented cis to the 3-pyrryl group.¹³ For the N-benzyl compounds, the chemical shift of C-2' in 4a is 6.63 ppm to higher field than in 4b, implying that C-2' is sterically compressed in 4a and is, therefore, cis to the 3-pyrryl substitutent. The chemical shift of C-4' in 4b is 9.51 ppm to higher field than in 4a, indicating that C-4' is sterically compressed in 4b, andd, therefore, C-2' is trans to the pyrrole ring.

Similar results are obtained for the N-methyl compounds, where in 10, C-2' is 7.57 ppm to higher field than in 11, and 11, C-4' is 9.39 ppm to higher field than in 10.

In the 60-MHz proton NMR spectra of isomers 4a and 10, the methylene protons at the 2'-position appeared as an AB quartet ($J \approx 11$ Hz, $\Delta \delta \approx 0.55$). This magnetic nonequivalence of the 2'-methylene protons is presumably due to the effect of the anisotropic shielding cone generated by the cis orientation of the pyrrole ring.¹⁴

In contrast, the 2'-protons of isomer 4b appear as the predicted singlet at the appropriate chemical shift. However, the 2'-protons in 11 appear not as a singlet but as a multiplet centered at about 2.75 ppm. For both compounds 4b and 11, however, the 4'-protons appear as a characteristic pair of shielded multiplets ($J \approx 11$ Hz, $\Delta \delta$ ≈ 0.55), again due, apparently, to their proximity to the anisotropic field of the pyrrole ring. The 5'-protons in all of the compounds appear to be unaffected and always appear as a multiplet at about 2.7 ppm (Table III).

Pharmacology and Discussion of Results

All compounds were evaluated for oral hypotensive activity in spontaneously hypertensive rats (SHR) by using the indirect tail-cuff method (Table IV), and all were significantly active at 50 mg/kg po, except compounds 11, 12c, and 13, which were only marginally active at this dose. In compounds where n = 2, the length and size of the N-alkyl chain appeared to be important for maintaining hypotensive activity at lower doses. Optimal activity was found in compounds 5 ($R^1 = CH_3$) and 12a ($R^1 = CH_2CH_3$). Compounds 12b-f, which possess longer and bulkier side chains, were less active than compounds 5 and 12a. The pyrrolidine secondary amine 8, as a mixture of diastereoisomers, and the piperidine 9 were essentially equally potent. Of the two N-methylpyrrolidine diastereoisomers, the cis isomer 10 was significantly more active than the trans isomer 11, and at the screening dose of 50 mg/kg po, it was equal in potency to the N-methylpiperidine 5.

Compounds 5 and 12a were selected for further study because of their potency, good dose-response, and temporal activity (no decrease in hypotensive response over the duration of treatment) in the SHR. As indicated in Table V, both of these compounds were very active at lowering the blood pressure in renal hypertensive rats [one

Journal of Medicinal Chemistry, 1983, Vol. 26, No. 10 1507

compd	C-2 (C-3′)	C-3	C-3a	C.₄	C-5	0-6 C	C-7	C-7a	C-2′	C-4'	C-5′	C-a	C-6	E E	
4a (cis)	97.39	68.17	125.46	130.76	121.37	126.34	110.71	159.04	58.72	39.78	52.70	120.04	108.66 ($CH_{2}C_{6}H_{3}:CH_{2}:CH_{2}(C_{1})$	2, (60.05) 7, (138.44) 7, (128.17) ^a
4b (trans) 97.01	97.01	67.67	125.29	130.54	121.21	126.34	110.38	159.38	65.35	30.27	52.87	120.10	108.66 (CH ₂ C ₆ H ₅ : CH ₂ CH ₁ C ₆ H ₅ : CH ₂ C-1 ³	$\begin{array}{c} (126.95) \\ (126.95) \\ (59.94) \\ (138.50) \\ (138.50) \\ (138.50) \end{array}$
10 (cis)		67.84	125.46	130.76	121.37	126.18	110.76		60.77				108.72	C-2 C-3" C4": CH.: CH,	$\begin{array}{c} (128.66)^{\alpha} \\ (128.66)^{\alpha} \\ (127.06) \\ (122.10) \\ (42.10) \end{array}$
11 (trans)	97.83	67.01	125.46	130.60	121.21	126.23	110.54	159.26	68.34	31.05	55.13	120.04	108.72	CH ₃ : CH ₃	_
a Assignment of the C-2" and C-3" carbons is tentative and only based upon the relative positions of the equivalent carbons in benzylamine.	t of the C-2	" and C	3'' carbon	s is tentativ	re and only	based upo	n the relati	ive position	s of the e	equivalent	t carbons	in benzylaı	nine.		
Table III. Proton NMR Chemical Shifts	oton NMR (Chemical	Shifts												i
compd	R	H-3		H-4, H-5, H-6, H-7	H	H-2′		H-4'		H-5′	-π-Η	β-H	CH3	CH2C6Hs	$CH_2C_6H_5$ $CH_2C_6H_5$
4a (cis)	CH ₂ C ₆ H ₅		_	6.6-7.5 (m)	2.63 (d), 2.10	l), 2.10 (d)	2.15 (1	u)		2 (m)	6.45 (t)	6.09 (t)		3.53 (s)	7.25 (s)
4b (trans)			_	6.6-7.5 (m)	2.76 (s		1.88 (1	1.88 (m), 1.33 (m)		2.68 (m)	6.49(t)			3.64 (s)	7.30 (s)
10 (cis)		ł	_	6.8-7.6 (m)	2.61 (d	l), 2.00 (d)	2.17 (1	n)		5 (m)	6.56 (t)	6.20 (t)			
11 (trans)		5.43 (s)	_	6.8-7.6 (m)	~2.75 (n	(u	1.92 (1	1.92 (m), 1.35 (m)		~2.75 (m)	6.57 (t)	6.20 (t)			

 Table II.
 Carbon-13 NMR Chemical Shifts

⁽¹²⁾ F. Wehrli and T. Wirthlin, "Interpretation of Carbon-13 Nuclear Magnetic Resonance Spectra", Heyden, London, 1976, p

⁽¹³⁾ G. Levy and G. Nelson, "Carbon-13 Nuclear Magnetic Resonance for Organic Chemists", Wiley-Interscience, New York, 1972, p 24.

⁽¹⁴⁾ L. M. Jackman and S. Sternhell, "Applications of Nuclear Magnetic Resonance Spectroscopy in Organic Chemistry", 2nd ed., Vol. 5, Pergaman Press, Oxford, 1969, p 94.

Table IV. Hypotensive Activity of 2,3-Dihydro-3-(1-pyrryl)spiro[benzofuran-2,4'-piperidine]s and -spiro[benzofuran-2,3'-pyrrolidine]s in Spontaneously Hypertensive Rats (SHR)

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$\begin{array}{c ccccccccccccccccccccccccccccccccccc$					
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$		of	mg/kg	BP,	ΔBP^{a}
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	compd	rats	ро	mmHg	
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	4	4	50	211	42 ± 15.9
$\begin{array}{cccccccccccccccccccccccccccccccccccc$		8	25	185	18 ± 5.4
$\begin{array}{cccccccccccccccccccccccccccccccccccc$		8	10	171	$10 \pm 8.1^{b,c}$
$\begin{array}{cccccccccccccccccccccccccccccccccccc$		8	5	198	27 ± 8.9^{c}
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	5	4	50	202	39 ± 6.3
$\begin{array}{cccccccccccccccccccccccccccccccccccc$		8	25	196	58 ± 8.4
$\begin{array}{cccccccccccccccccccccccccccccccccccc$		8		190	50 ± 5.6
$\begin{array}{cccccccccccccccccccccccccccccccccccc$		8			
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	8	2		223	
$\begin{array}{cccccccccccccccccccccccccccccccccccc$		8			
$\begin{array}{cccccccccccccccccccccccccccccccccccc$		7			36 ± 14.2^{c}
$\begin{array}{cccccccccccccccccccccccccccccccccccc$		8			
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	9	4			
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$					
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$ \begin{array}{cccccccccccccccccccccccccccccccccccc$					
$\begin{array}{cccccccccccccccccccccccccccccccccccc$		4			
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	12a	4			
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$		7			
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$ \begin{array}{cccccccccccccccccccccccccccccccccccc$		8			
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$		3			39 ± 18.8
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$		4			
12f 4 50 199 37 ± 11.2 13 4 50 209 24 ± 15.6 15.6		4			
12f 4 50 199 37 ± 11.2 13 4 50 209 24 ± 15.6 15.6	12e	4			
12f 4 50 199 37 ± 11.2 13 4 50 209 24 ± 15.6 15.6		8			
12f 4 50 199 37 ± 11.2 13 4 50 209 24 ± 15.6 15.6		7			
13 4 50 209 24 ± 15.6 cloni- 0.25 38		8			
cloni- 0.25 38		4			37 ± 11.2
		4		209	
dine			0.25		38
	dine				

^a Peak decrease in systolic blood pressure on 3rd day, 2 h postadministration of compound, ±SEM. ^b Increase in systolic blood pressure on 3rd day, 2 h postadministration of compound. ^c Fifth day, 2 h postadministration of compound.

kidney, one wrap (Grollman)]. Compound 5 was particularly active on day 3 of the test, with a decrease of 109 mmHg. The reason for the greater hypotensive activity of compound 5 in RHR vs. SHR is not clear at this time.

In anesthetized neurogenic hypertensive dogs (Table VI), both compounds 5 and 12a exerted marked hypotensive effects within 5 min of dosage. Both caused moderate bradycardia but had little effect on $dP dT^{-1}P_{max}^{-1}$, a parameter of cardiac contractility. Compound 12a did, however, lower the cardiac output somewhat, and this may have contributed to its hypotensive activity.

In anesthetized normotensive dogs in standard cardiovascular challenge experiments, neither of these compounds produced any striking effects on responses to standard challenge drugs, suggesting little interaction with autonomic receptors. Compound 12a did have a slight tendency to potentiate responses to norepinephrine, epinephrine, and tyramine, but these effects were minimal (Table VII). Additional hemodynamic studies performed in anesthetized, ganglion-blocked dogs suggested that the hypotensive effects were not mediated via the central nervous system, since activity was still present under these conditions.

Experimental Section

The structures of all compounds are supported by their IR (Perkin-Elmer 457) and ¹H NMR (JEOL C6OHL) spectra. Melting points were obtained on a Thomas-Hoover capillary melting point apparatus and are uncorrected. Where analyses

Table V. Hypotensive Activity of Compounds 5 and 12a in Renal Hypertensive Rats $(RHR)^a$

compd	no. of rats	dose, mg/ kg po	initial BP,	postdrug BP, mmHg, (day/h)	∆BP, mmHg
5	4	10	206 ± 13	$196 \pm 8 (1/2) \\ 184 \pm 11 (2/0) \\ 170 \pm 12b (2/2) \\ 183 \pm 6b (3/0) \\ 97 \pm 12b (3/2) $	37 22 36 23 109
12a	4	50	162 ± 8	$\begin{array}{c} 131 \pm 10 \ (1/2) \\ 110 \pm 4 \ (2/0) \\ 113 \pm 5^{b} \ (2/2) \\ 116 \pm 5^{b} \ (3/0) \\ 129 \pm 23^{b} \ (3/2) \end{array}$	$31 \\ 52 \\ 49 \\ 46 \\ 33$
capto- pril	4	10	192 ± 15	$\begin{array}{c} 181 \pm 16 \ (1/2) \\ 185 \pm 9 \ (2/0) \\ 169 \pm 4 \ (2/2) \\ 163 \pm 12 \ (3/0) \\ 152 \pm 12 \ (3/2) \end{array}$	11 7 23 29 40

 a Grollman model of hypertension (one kidney, one wrap). b Data for three rats.

are indicated only by symbols of the elements, the analytical results obtained for those elements (performed by Micro-Tech Laboratories, Skokie, IL) were within 0.4% of theoretical values.

Preparative HPLC separation of 4a,b was performed on a Waters Associates Prep 500 system using a Prep Pak silica cartridge. The mobile phase was ether/hexane (35:65), and detection was by refractive index.

Carbon-13 NMR spectra were recorded as deuteriochloroform solutions at 15.03 MHz on a JEOL FX-60 Fourier-transform spectrometer equipped with an internal deuterium lock. Operating parameters were as follows: pulse width, 3μ s; tip angle, 30°; pulse repetition, 1.5 s; spectral width, 3.4 kHz; data size, 8K. All proton and carbon chemical shifts are reported relative to internal tetramethylsilane.

1-(2-Fluorobenzyl)pyrrole (1). To 120 mL of glacial acetic acid was added 2-fluorobenzylamine (30.0 g, 0.24 mol) with cooling, followed by 2,5-dimethoxytetrahydrofuran (32.0 g, 0.24 mol). After stirring at reflux for 2 h, the cooled solution was concentrated in vacuo to a dark oil, which was added to 100 mL of a 10% solution of sodium carbonate and then extracted with ethyl ether (3 × 100 mL). The dried (MgSO₄) etheral solution was concentrated in vacuo to a dark oil, which was distilled to give 27 g (64%), bp 73 °C (0.5 mm), of a water-white liquid. Anal. (C₁₁H₁₀FN) C, H, N.

4-[(2-Fluorophenyl)(1-pyrryl)methyl]-1-methyl-4piperidinol (3). Method A. To 35 mL of THF was added 1 (10.0 g, 0.057 mol), and, after this solution was cooled to -80 °C, a solution of n-butyllithium (25 mL of a 2.4 M solution, 0.06 mol) was added dropwise with stirring over a period of 10 min. After the thick solution was stirred at -80 °C for 1 h, a solution of 1-methyl-4-piperidone (6.4 g, 0.055 mol) in 15 mL of THF was added in 5 min. After stirring at ambient temperature for 1 h, the mixture was poured into 250 mL of water, stirred for 10 min, and then extracted with ethyl ether $(2 \times 50 \text{ mL})$. The organic extracts were washed with water (2 times) then dried (MgSO₄) and concentrated under vacuum to a yellow oil, which upon trituration with *n*-hexanes gave 3 as a tan solid: yield 10 g (62%); mp 128-130 °C, 135 °C after crystallization from ethyl acetate/methanol (10:1). Anal. (C₁₇H₂₁FN₂O·HCl) C, H, N. Compound 2 was prepared in a similar manner in 58% yield, mp 87-89 °C. Anal. $(C_{22}H_{23}FN_2O)$ C, H, N.

2,3-Dihydro-1'-methyl-3-(1-pyrryl)spiro[benzofuran-2,4'-piperidine] Hydrochloride (5). Method B. To a suspension of sodium hydride (0.9 g, 0.036 mol) in 50 mL of benzene was added a solution of 3 (9.0 g, 0.03 mol) in 100 mL of benzene. After the solution was heated to reflux, 40 mL of DMF was added slowly, and reflux was continued for 3 h. After cooling, the mixture was poured into 500 mL of water, stirred for 5 min, and then extracted with ethyl ether (2 × 100 mL). The combined organic layer was washed with water (2 times), dried (MgSO₄), and then concentrated in vacuo to a brown oil, which was converted to a

Table VI.	Hemodynamic	Effects in	the Sino	aortic-Deaffe	erented l	Hypertensive 1	Dog^a
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compd	no. of dogs	dose, mg/kg iv	time, min	mean arterial BP, mmHg	heart rate, beats/min	$\frac{\mathrm{d}P\mathrm{d}t^{-1}}{P_{\mathrm{max}^{-1}},\mathrm{s}^{-1}}$	cardiac output, mL/min
5	3	5	0	175 ± 15	194 ± 26	24.2 ± 5.8^{b}	
			5	123 ± 24^{c}	154 ± 32^{c}	26.0 ± 0.7^{b}	
			30	129 ± 16^{c}	174 ± 16^{d}	26.2 ± 2.2^{b}	
			60	129 ± 19^{c}	160 ± 21^{c}	23.4 ± 0.2^{b}	
12a	3	10	0	165 ± 8.5	170 ± 10	27.2 ± 5.2	2120 ± 350
			5	121 ± 17^{c}	134 ± 7^{c}	20.3 ± 4.9^{d}	1600 ± 260^d
			30	98 ± 29 ^c	126 ± 16^{c}	21.2 ± 8.2^d	1530 ± 610^{d}
			60	103 ± 20^{c}	132 ± 16^{c}	25.6 ± 5.3^d	1600 ± 560^{d}

^a Data indicated as mean plus or minus SEM. ^b Data from two dogs, no statistics. ^c p < 0.05 vs. time 0, analysis of variance (ANOVA) with repeated measures on time factor, Duncan's test. ^d Data not statistically different from time 0, ANOVA with repeated measures, Duncan's test.

 Table VII.
 Effects on Standard Cardiovascular Challenges

 in Anesthetized Normotensive Dogs

		0	
challenge drug		5 ^a	12a ^a
epinephrine	pre	+ 53.7 ^b	+44.8
	post	+67.0	+59.7
	$\overline{\Delta}$	+13.3	+14.9
norepinephrine	pre	+57	+56.7
	\mathbf{post}	+58	+66.0
	Δ	+1	+ 9.3
isoproterenol	pre	-38	-34.3
	post	-30	-39.3
	Δ	-8	+5.0
carbachol	pre	-49.2	-52
	post	-29.7	-37
	Δ	-19.5	-15
tyramine	pre	+52.5	+64.3
	post	+48.0	+84.3
	Δ	-4.5	+20.0
mean arterial BP <i>c</i>	pre	117	132.7
	post	82.3	106.3
	Δ	-34.7	-26.4
mean heart rate ^d	pre	147	167
	post	113	126
	Δ	-34	-41

^a Administered 10 mg/kg iv. ^b mmHg from base line. ^c Fifteen minutes postdose, mmHg. ^d Beats per minute.

crystalline hydrochloride, 5, in ether. Properties of 5 and 4, prepared in a similar manner, are included in Table I.

2,3-Dihydro-1'-(ethoxycarbonyl)-3-(1-pyrryl)spiro[benzofuran-2,4'-piperidine] (7). Method C. To 200 mL of benzene were added 5 (15.5 g, 0.058 mol) and potassium carbonate (25 g, 0.18 mol), followed by a solution of ethyl chloroformate (8.1 g, 0.75 mol) in 50 mL of benzene. After stirring at reflux for 20 h, the mixture was cooled and filtered, and the filtrate was washed with water and 3 N hydrochloric acid, dried (MgSO₄), and then concentrated in vacuo to a brown oil, which upon trituration with hexanes gave 7: yield 18.6 g (80%); mp 132-133 °C. Anal. (C₁₉H₂₂N₂O₃) C, H, N. Compound 6 (both isomers) was prepared in a similar manner in 70% yield.

2,3-Dihydro-3-(1-pyrryl)spiro[benzofuran-2,4'-piperidine] Oxalate (9). Method D. To a solution of potassium hydroxide (20.0 g, 0.36 mol) in 6 mL of water and 100 mL of 1-propanol was added a solution of 7 (13.3 g, 0.41 mol) in 150 mL of 1-propanol. After stirring at reflux for 20 h, the mixture was concentrated in vacuo to a dark oil, which was dissolved in CHCl₃, washed with water (2 times), and then dried (MgSO₄). After concentrating in vacuo, the resultant brown oil was converted to a crystalline oxalate (9) in ether. Properties of 9 and 8, prepared in a similar manner, are included in Table I.

cis-2,3-Dihydro-1'-methyl-3-(1-pyrryl)spiro[benzofuran-2,3'-pyrrolidine] (10). To a suspension of lithium aluminum hydride (1.1 g, 0.029 mol) in 75 mL of THF was added a solution of 6 (cis isomer; 4.5 g, 0.0138 mol) in 25 mL of THF. After stirring at ambient temperature for 3 h, the mixture was diluted with 100 mL of ethyl ether and then quenched with 5 mL of saturated ammonium chloride solution. After the solution was filtered, the organic layer was washed with water (2 times), dried (MgSO₄), and then concentrated in vacuo to give a light yellow oil, which crystallized from hexanes. Properties of 10 and 11 (trans isomer), prepared in a similar manner, are included in Table I.

1'-[2-(Diethylamino)ethyl]-2,3-dihydro-3-(1-pyrryl)spiro-[benzofuran-2,4'-piperidine] Dioxalate (12d). Method E. To 100 mL of 1-butanol were added 9 (4.5 g, 0.018 mol), 2-(diethylamino)ethyl chloride (2.7 g, 0.02 mol), K_2CO_3 (20 g, 0.145 mol), and KI (0.01 g). After stirring at reflux for 48 h, the mixture was concentrated in vacuo, dissolved in ethyl ether, washed with water, and then dried (MgSO₄). After concentration in vacuo, the oil was converted to a crystalline dioxalate, 12d, in ether. Properties of 12d,f, prepared in a similar manner, are included in Table I.

2,3-Dihydro-1'-ethyl-3-(1-pyrryl)spiro[benzofuran-2,4'piperidine] Oxalate (12a). Method F. To 100 mL of CH_2Cl_2 were added 9 (5.5 g, 0.022 mol) and triethylamine (3.8 mL, 0.025 mol), followed by the addition of a solution of acetyl chloride (1.8 mL, 0.025 mol) in 30 mL of CH_2Cl_2 with cooling. After stirring at ambient temperature for 20 h, the mixture was concentrated in vacuo, dissolved in ethyl acetate, washed with water and 2 N hydrochloric acid, and then dried (MgSO₄). Concentration in vacuo resulted in a brown semisolid: yield 4.8 g (84%); mp 45 °C. No further purification was attempted. The amide was reduced to the amine as in 10, to yield a brown oil, which was converted to a crystalline oxalate, 12a, in ether. Properties of 12a,c,e, prepared in a similar manner, are included in Table I.

2,3-Dihydro-1'-(2-hydroxyethyl)-3-(1-pyrryl)spiro[benzofuran-2,4'-piperidine] (12b). To 50 mL of DMF were added 9 (6.0 g, 0.024 mol), ethyl bromoacetate (4.6 g, 0.028 mol), NaHCO₃ (20 g, 0.24 mol), and KI (0.01 g). After stirring at 70 °C for 1 h, the mixture was concentrated in vacuo, dissolved in ethyl acetate, washed with water, dried (MgSO₄), and then concentrated in vacuo to give 5 g (62%) of a dark oil. No further purification was attempted. This ester was reduced to the alcohol 12b as in 10, to yield a yellow oil. Properties of 12b are included in Table I.

l'-[Amino(hydroxyimino)methyl]-2,3-dihydro-3-(1-pyrryl)spiro[benzofuran-2,4'-piperidine] (13). To a refluxing solution of cyanogen bromide (3.7 g, 0.035 mol) and K_2CO_3 (20.0 g, 0.145 mol) in 100 mL CHCl₃ was added a solution of 5 (8.0 g, 0.03 mol) in 75 mL of CHCl₃. After refluxing for 5 h, the mixture was filtered, and the filtrate was concentrated in vacuo to a brown oil, which solidified to a tan solid upon trituration with petroleum ether to give the cyanamide: yield 6.9 g (82%); mp 165–168 °C. Anal. ($C_{17}H_{17}N_3O$) C, H, N.

To 20 mL of DMF were added the cyanamide (6.4 g, 0.023 mol), hydroxylamine hydrochloride (2.6 g, 0.04 mol), and Na_2CO_3 (7.7 g, 0.07 mol). After stirring at 105 °C for 1.5 h, the mixture was poured into 500 mL of iced water, and the resultant tan precipitate was collected and dried to yield 5.1 g of 13. Properties of 13 are included in Table I.

Hypotensive Activity in Spontaneously Hypertensive Rats (SHR) and Renal Hypertensive Rats (RHR). All compounds were screened for hypotensive activity in spontaneously hypertensive rats of the Okamoto-Aoki strain. The screening dose was 50 mg/kg po. Systolic blood pressures were determined at the following times by tail-cuff plethysmography: day 1, predose and 2 h postdose; day 3, predose and 2 h postdose; day 5, predose and 2 and 4 h postdose. Details of the method are described in Buggy et al.¹⁵ The test compounds were suspended in distilled water

⁽¹⁵⁾ J. Buggy, G. D. Fink, A. K. Johnsen, and M. J. Brody, Circ. Res., Suppl., 40(5), 1977, I-110.

with Tween 80 and administered at either 50, 30, 10, or 3 mg/kg po, unless otherwise indicated. Animals were dosed every day. Four animals were used in the preliminary screen, while four to eight animals per dose were used in the dose-response screen.

Renal hypertension was produced by a modification of the method of Grollman.^{15,16} Briefly, male Sprague–Dawley rats were anesthetized with pentobarbital sodium (50 mg/kg ip), and the right kidney was removed through a right subcostal incision. The left kidney was exposed through an incision in the left flank. Silk sutures (size 00) were passed around both poles of the kidney and tightened to produce a visible constriction of the renal parenchyma. Animals were returned to their cages and given food and water ad libitum. After 3 weeks of stabilization, systolic blood pressures were determined as above, and rats with blood pressures >150 mmHg were selected for use in evaluating drugs

Standard Cardiovascular Challenges in Anesthetized Dogs. Beagle dogs of either sex were anesthetized with barbital sodium (200 mg/kg iv), thiopental sodium (15 mg/kg iv), and pentobarbital sodium (60 mg iv). The trachea was intubated, and the animals were artificially respired with a Harvard apparatus respirator. A femoral artery and vein were cannulated with PE tubing for measurement of blood pressure and intravenous administration of drugs, respectively. Bilateral vagotomies were performed, and blood pressure and heart rate were continuously displayed on a Beckman R 511 recorder. The effects of various challenge drugs on the blood pressure and heart rate were determined; test drug was administered, and responses to the challenge drugs were again obtained. Each challenge drug was intravenously administered twice predrug to obtain matched responses and once postdrug. Challenge drugs and doses were as follows: epinephrine and norepinephrine, 0.5 to 1.0 μ g/kg; isoproterenol and carbachol, 0.25 μ g/kg; tyramine, 100 μ g/kg. Adjustments were made in test doses to compensate for animal to animal variation. The effect of the test drug on the responses to the challenge drugs was noted, and results were expressed as change from predrug control responses.

Hypotensive Activity in the Sinoaortic-Deafferented Dogs. Adult dogs of either sex were anesthetized with barbital sodium (200 mg/kg iv), thiopental sodium (15 mg/kg iv), and pentobarbital sodium (60 mg iv). A femoral vein and artery were cannulated with PE tubing for intravenous administration of drugs

(16) A. Groolman, Proc. Soc. Exp. Biol. Med., 57, 102 (1944).

and to record blood pressure and heart rate, respectively. Left ventricular pressure was recorded from a catheter inserted into the left ventricle via the left common carotid artery (postdeafferentation), and the first derivative, dP/dt, was derived from it. Cardiac output was determined by thermodilution technique using ice-cold 5% dextrose in water injected into a Swan-Ganz catheter inserted into the right side of the heart with the tip placed in the pulmonary artery.

Deafferentation was accomplished by clearing both of the carotid arteries up to the internal and external carotid artery bifurcation. The carotid sinus nerves were isolated, ligated, and sectioned, and a bilateral vagotomy was performed to produce neurogenic hypertension (mean arterial pressure >150 mmHg). Dogs were allowed to stabilize for approximately 30 min, and then a bolus intravenous injection of the test compound was administered. Heart rate, arterial pressure, left ventricular pressure, dP/dt, and cardiac output were monitored for 90 min postdose.

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Registry No. 1, 63880-21-7; 2, 79227-66-0; 3-HCl, 79227-46-6; cis-4, 82830-95-3; cis-4 oxalate, 82830-96-4; trans-4, 82831-05-8; trans-4 oxalate, 82831-06-9; 5, 79227-49-9; 5-HCl, 79227-48-8; cis-6, 82831-04-7; trans-6, 82831-03-6; 7, 79227-53-5; cis-8, 86632-59-9; cis-8 maleate, 86632-60-2; trans-8, 86632-61-3; trans-8 maleate, 86632-62-4; 9, 79227-51-3; 9 oxalate, 79227-52-4; 10, 82830-99-7; 11, 82830-97-5; 11 oxalate, 82830-98-6; 12a, 79227-71-7; 12a oxalate, 79227-72-8; 12a ($R^1 = COCH_3$), 79227-73-9; 12b, 79227-75-1; 12b $(R^1 = CH_2CO_2Et)$, 79227-76-2; 12c, 79227-63-7; 12c oxalate, 79227-64-8; 12c (R¹ = COCH₂CH₃), 79227-65-9; 12d, 79227-58-0; 12d 2-oxalate, 79227-59-1; 12e, 79227-61-5; 12e oxalate, 79227-62-6; 12e ($R^1 = COCH_2C_6H_5$), 79227-60-4; 12f, 79227-56-8; 12f oxalate, 79227-57-9; 13, 79227-55-7; 13 ($\mathbb{R}^1 = \mathbb{CN}$), 79227-54-6; 2-fluorobenzylamine, 89-99-6; 2,5-dimethoxytetrahydrofuran, 696-59-3; 1-methyl-4-piperidone, 1445-73-4; 1-benzyl-3-pyrrolidinone, 775-16-6; ethyl chloroformate, 541-41-3; 2-(diethylamino)ethyl chloride, 100-35-6; ethyl bromoacetate, 105-36-2; cyanogen bromide, 506-68-3; hydroxylamine, 7803-49-8.

9-Acridinyl and 2-Methoxy-6-chloro-9-acridinyl Derivatives of Aliphatic Di-, Tri-, and Tetraamines. Chemistry, Cytostatic Activity, and Schistosomicidal Activity

John B. Hansen,*,[†] Eyvind Langvad,[‡] Flemming Frandsen,[§] and Ole Buchardt*,[†]

Chemical Laboratory II, The H. C. Ørsted Institute, University of Copenhagen, Universitetsparken 5, DK-2100 Copenhagen Ø, The Fibiger Laboratory, Nordre Frihaunsgade 50, DK-2100 Copenhagen Ø, and Danish Bilharziasis Laboratory, Jaegersborg Alle 1D, DK-2920 Charlottenlund, Denmark. Received December 30, 1982

9-acridinyl derivatives of 1,6-hexanediamine, 1,8-octanediamine, bis(3-aminopropyl)amine, N,N'-bis(3-aminopropyl)piperazine, and N-ethyl-1,6-hexanediamine in the form of their hydrochlorides were prepared in high yields and converted into potential hetero bis DNA intercalating diacridines. The corresponding potential homo bis DNA intercalating reagents were prepared by heating the above amines with 9-chloroacridines. The chemical stability of the acridines was examined. Their cytostatic activity against Cloudman melanoma cells, in vitro, has been determined. The strongest cytostatic activity was observed for the acridine derivatives of the tri- and tetraamines. The schistosomicidal activity of selected acridine and diacridine derivatives against Schistosoma mansoni in mice was found to be insignificant. The S. mansoni egg development was apparently suppressed by this treatment.

In the design of biomolecular tools and drugs, it is important to obtain high target specificity. The target could either be an entire organ, a type of tissue, or it could be molecular. Thus, compounds that could be directed not only toward DNA but even toward particular nucleotide

sequences in DNA might exhibit high specific biological activity. Such compounds would be potential drugs against tumors,¹ virus,^{1,2} bacteria,³ and parasites.⁴

(3) A. C. R. Dean, in ref 1, p 789.

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[†]The H. C. Ørsted Institute.

[‡]The Fibiger Laboratory.

[§] Danish Bilharziasis Laboratory.

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