SYNTHESIS OF 1-[ω-[(ARYLAMINO)CARBONYL]ALKYL]-4-(BENZOCYCLOALKYL)PIPERAZINES

Youssef El-Ahmad, Philippe Maillet, Elisabeth Laurent, Akram Talab, Gilles Tran, and Roland Ollivier*

Centre de Recherche, Coopération Pharmaceutique Française, 13-15 rue Benjamin Franklin, 77000 La Rochette, France

Abstract- A series of $1-[\omega-[(arylamino)carbonyl]alkyl]-4-(benzocycloalkyl)$ $piperazines (1a-v) was prepared either by reacting the precursor 4-[<math>\omega$ -[(arylamino)carbonyl]alkyl]piperazine (2a-j) with 1-chlorobenzocycloalkanes (3ac) (Procedure A) or by reacting the *N*-aryl- ω -chloroalkanamides (5a-j) with the 4-(benzocycloalkyl)piperazines (10a-c) (Procedure B). The best yields were obtained using procedure A.

Aminotetralins exhibit various pharmacological activities, for example antidepressant¹, anxiolytic² or antipsychotic.³ In connection with our ongoing work on the synthesis of potential central-nervous-system active compounds,³ we were interested in the preparation of a series of $1-[\omega-[(arylamino)carbonyl]alkyl]-4-(benzocycloalkyl)piperazines (1a-v).$

Derivatives (1a-v) were prepared by condensing the compounds (2a-j) with 1-chlorobenzocycloalkanes (3a-c) as described in Scheme 1 (Procedure A).

Scheme 1



1-Chlorobenzocycloalkanes (**3a-c**) were synthesised by the method described by Bogeso⁴ consisting in reducing the corresponding ketones (**11a-c**) with sodium borohydride, then by the treatment of the formed alcohols (**12a-c**) with thionyl chloride. (See Scheme 2).

Scheme 2



4-[(Arylaminocarbonyl)alkyl]piperazines (2a-j) were obtained according to the route illustrated in Scheme 3. The reaction of ω -halo-*N*-alkanoyl chlorides (6) with the appropriate arylamines (7) in the presence of Et₃N gave the ω -halo-*N*-arylalkanoylamides (5a-j),⁵ which after condensation with two equivalents of 1-benzylpiperazine (one equivalent was used as the basic agent for the neutralization of the halogen hydride formed in the reaction) yielded the 1-[(arylaminocarbonyl)alkyl]-4-benzylpiperazines (4a-j),⁵ which upon debenzylation under palladium on active carbon gave the desired 2a-j.⁶

Scheme 3



As described in Scheme 4, the condensation of the *N*-aryl- ω -haloalkanamides (**5a-j**) with the benzocycloalkylpiperazines (**10a-c**) (procedure B) obtained by decarboxylation of 1-(benzocycloalkyl)-4-ethoxycarbonylpiperazines (**9a-c**) under basic conditions with potassium hydroxide, also permitted to obtain the benzocycloalkylpiperazines (**1a-v**). Compounds (**9a-c**) were obtained by the condensation of



(3a-c) with ethyl N-piperazinecarboxylate (8) (See Scheme 4).

Scheme 4

EXPERIMENTAL

Melting points were measured on a büchi 535 apparatus and are uncorrected. IR spectra were recorded on a Perkin Elmer spectrophotometer; ¹H NMR spectra were recorded on a Hitachi 1500 FT spectrometer (60 MHz) with TMS as internal standard. Chemical shifts are given in ppm; s, d, t, q, m designated respectively singlet, doublet, triplet, quartet and multiplet. Thin layer chromatography was performed on Merck silica gel 60 plates with fluorescent indicator. The plates were visualised with UV light (254 nM). Flash chromatography was conducted on Merck Kieselgel 60 (0.040-0.063). Elemental analyses were carried out by microanalysis laboratory in the faculty of pharmacy in Chatenay-Malabry, France.

1-Hydroxybenzocycloalkanes (12a-c)

Sodium borohydride (2.32 g, 61 mmol) was added in portions with stirring at 12 °C to a solution of 11 (183 mmol) in methanol (500 mL). The mixture was stirred for 2 h and then evaporated. The resulting oil was treated with water and ether, and the organic phase was separated, washed with water and 0.1 N HCl, dried over MgSO₄ and evaporated to dryness to give compounds (**12 a-c**). Compound **12a** (m = 1): yield 88%, ¹H NMR (CDCl₃): δ = 1.8-2.5 (m, 3H), 2.9 (t, 2H, J = 7 Hz), 5.2 (t, 1H, J = 7 Hz), 7.25 (m, 4H). Anal. Calcd for C₉H₁₀O: C, 80.57; H, 7.51. Found: C, 80.28; H, 7.45. Compound **12b** (m = 2): yield 97%, ¹H NMR (CDCl₃): δ = 1.6 -2.3 (m, 5H), 2.8 (t, 2H, J = 7 Hz), 4.95 (t, 1H, J = 7 Hz), 7.0-7.6 (m, 4H). Anal. Calcd for C₁₀H₁₂O: C, 81.06; H, 8.16. Found: C, 80.97; H, 8.22. Compound **12c** (m = 3): mp 102 °C (from 2-propanol), yield 85%, ¹H NMR (CDCl₃): δ = 1.4 -2.2 (m, 7H), 2.8 (t, 2H, J = 7 Hz), 4.75 (t, 1H, J = 7 Hz), 7.0-7.4 (m, 4H). Anal. Calcd for C₁₁H₁₄O: C, 81.44; H, 8.70. Found: C, 81.53; H,

8.62.

1-Chlorobenzocycloalkanes (3a-c)

Thionyl chloride (26 mL, 0.35 mol) was added with stirring and cooling at 15°C to a solution of **12** (0.238 mol) in toluene (340 mL). The mixture was stirred at room temperature for 30 min and then heated to 55°C for 1 h. The mixture was cooled, washed twice with ice-water, dried over MgSO₄ and evaporated to give a quantitative yield of **3** as an oil, which was used without any further purification in the following step. Compound **3a** (m = 1): ¹H NMR (CDCl₃): $\delta = 2.35$ (q, 2H, J = 7 Hz), 2.95 (t, 2H, J = 7 Hz), 5.4 (t, 1H, J = 7 Hz), 7.1-7.5 (m, 4H). Anal. Calcd for C₉H₉Cl: C, 70.81; H, 5.94. Found: C, 70.53; H, 5.71. Compound **3b** (m = 2): ¹H NMR (CDCl₃): $\delta = 1.7-2.5$ (m, 4H), 2.8 (t, 2H, J = 7 Hz), 5.3 (t, 1H, J = 7 Hz), 7.0-7.3 (m, 4H). Anal. Calcd for C₁₀H₁₁Cl: C, 72.06; H, 6.65. Found: C, 71.81; H, 6.42. Compound **3c** (m = 3): ¹H NMR (CDCl₃): $\delta = 1.4$ -2.6 (m, 6H), 2.8 (t, 2H, J = 7 Hz), 5.2 (t, 1H, J = 7 Hz), 7.2 (m, 4H). Anal. Calcd for C₁₁H₁₃Cl: C, 73.11; H, 7.25. Found: C, 72.82; H, 6.96.

3-Bromo-N-(4-fluorophenyl)propionamide (5a)

A mixture of 4-fluoroaniline (22.2 g, 0.2 mol), 20 mL of triethylamine and 50 mL of benzene was cooled to 5°C. 3-Bromopropionyl chloride (34.3 g, 0.2 mol) dissolved in 150 mL of benzene was then added dropwise. When the addition was complete, the reaction medium was left to stand for 4 h. The benzene was evaporated off under vacuum. The solid was taken up with dichloromethane and then washed with ice water. The organic phase was dried over MgSO4 and then concentrated under vacuum to give 39 g of **5a** (79%). Compounds (**5b-j**) were synthesised similarly. Compound **5a** (n = 2): mp 108 °C (from toluene): vield 79%; ¹H NMR (CDCl₃): δ = 2.9 (t, 2H, J = 7 Hz), 3.6 (t, 2H, J = 7 Hz), 6.8-7.65 (m, 4H), 8 (s, NH). Anal. Calcd for C₉H₉NOBrF: C, 43.93; H, 3.68; N, 5.69. Found: C, 43.71; H, 3.58; N, 5.56. Compound **5b** (n = 2): mp 106 °C (from 2-propanol); yield 86%; ¹H NMR (CDCl₃): δ = 2.9 (t, 2H, J = 7 Hz), 3.7 (t, 2H, J = 7 Hz), 7.6 (m, 4H), 7.95 (s, NH). Anal. Calcd for $C_{10}H_0NOBrF_3$: C, 40.56; H, 3.06; N, 4.73. Found: C, 40.32; H, 3.12; N, 4.56. Compound 5c (n = 2): mp 137 °C (fom 2-propanol); yield 70%; ¹H NMR (CDCl₃): δ = 2.3 (s, 3H, CH₃), 2.9 (t, 2H, J = 7 Hz), 3.7 (t, 2H, J = 7 Hz), 7.1 (d, 2H, J = 9 Hz), 7.4 (d, 2H, J = 9 Hz), 7.6 (s, NH). Anal. Calcd for $C_{10}H_{12}NOBr$: C, 49.60; H, 4.99; N, 5.78. Found: C, 49.47; H, 4.81; N, 5.58. Compound **5d** (n = 2): mp 160 °C (from toluene); yield 78%; ¹H NMR (CDCl₁): $\delta = 2.95$ (t, 2H, J = 7 Hz), 3.65 (t, 2H, J = 7 Hz), 7.3-7.7 (m, 9H), 7.9 (s, NH). Anal. Calcd for C15H14NOBr: C, 59.22; H, 4.60; N, 4.60. Found: C, 59.02; H, 4.43; N, 4.56. Compound 5e (n = 2); mp 112 °C (from toluene); yield 95%; ¹H NMR (CDCl₃); $\delta = 2.85$ (t, 2H, J = 7 Hz), 3.65 (t, 2H,

J = 7 Hz), 3.8 (s, 3H, OCH₃), 6.8 (d, 2H, J = 9 Hz), 7.4 (d, 2H, J = 9 Hz), 8.1 (s, NH), Anal. Calcd for C₁₀H₁₂NO₂Br: C, 46.52; H, 4.69; N, 5.43. Found: C, 46.41; H, 4.73; N, 5.56. Compound **5f** (n = 3): mp 98 °C (from toluene); yield 85%; ¹H NMR (CDCl₃): $\delta = 2.2$ (m, 2H), 2.5 (t, 2H, J = 7 Hz), 3.6 (t, 2H, J = 7 Hz), 6.8-7.6 (m, 4H), 7.8 (s, NH). Anal. Calcd for C₁₀H₁₁NOClF: C, 55.70; H, 5.14; N, 6.49. Found: C, 55.56; H, 5.12; N, 6.56. Compound 5g (n = 3): mp 118 °C (from pentane); yield 98%; ¹H NMR $(CDCl_3): \delta = 2.2 \text{ (m, 2H)}, 2.5 \text{ (t, 2H, } J = 7 \text{ Hz}), 3.65 \text{ (t, 2H, } J = 7 \text{ Hz}), 7.6 \text{ (m, 4H)}, 8.1 \text{ (s, NH)}.$ Anal. Calcd for C₁₁H₁₁NOClF₃: C, 49.74; H, 4.17; N, 5.27. Found: C, 49.63; H, 4.12; N, 5.11. Compound **5h** (n = 3); mp 87 °C (from 2-propanol); yield 89%; ¹H NMR (CDCl₃); $\delta = 2.1$ (m, 2H), 2.3 (s, 3H, CH₃), 2.5 (t, 2H, J = 7 Hz), 3.6 (t, 2H, J = 7 Hz), 7.1 (d, 2H, J = 9 Hz), 7.35 (d, 2H, J = 9 Hz), 7.75 (s, NH). Anal. Calcd for C₁₁H₁₄NOCl: C, 62.40; H, 6.67; N, 6.62. Found: C, 62.32; H, 6.59; N, 6.56. Compound **5i** (n = 3): mp 151 °C (from 2-propanol); yield 80%; ¹H NMR (CDCl₃): $\delta = 2.2$ (m, 2H), 2.55 (t, 2H, J = 7 Hz), 3.65 (t, 2H, J = 7 Hz), 7.3-7.7 (m, 9H), 7.85 (s, NH). Anal. Calcd for C₁₆H₁₆NOCl: C, 70.20; H, 5.89; N, 5.12. Found: C, 70.32; H, 5.93; N, 4.95. Compound 5j (n = 3): mp 75 °C (from 2-propanol); yield 96%; ¹H NMR (CDCl₃): $\delta = 2.1$ (m, 2H), 2.45 (t, 2H, J = 7 Hz), 3.6 (t, 2H, J = 7 Hz), 3.8 (s, 3H, 3H) = 2.1 (m, 2H) = 2.45 OCH₃), 6.8 (d, 2H, J = 9 Hz), 7.35 (d, 2H, J = 9 Hz), 7.9 (s, NH). Anal. Calcd for $C_{11}H_{14}NO_2Cl: C$, 58.03; H, 6.20; N, 6.15. Found: C, 58.15; H, 6.28; N, 6.05.

1-[2-[(4-Fluorophenylamino)carbonyl]ethyl]-4-benzylpiperazine (4a)

Compound (**5a**) (20.15 g, 0.1 mol) in solution in 50 mL of DMF was added dropwise to a mixture of 1benzylpiperazine (35.2 g, 0.2 mol) and sodium iodide (1 g, 6.6 mmol) in 150 mL of DMF. The reaction mixture was stirred for 4 h and then for 2 h at 60-70 °C. After cooling, water was added and the mixture was extracted with ethyl acetate. The organic layer was washed several times with water, dried over MgSO₄ and then evaporated to give 29 g of crude base (yield 85%). This base was dissolved in ethanol, a solution of hydrogen chloride in ethanol was then added, the dihydrochloride crystallised out. Compounds **4b-j** were synthesised similarly. ¹H NMR spectra were carried out on the crude base. Compound **4a** (n = 2): mp 226 °C (from ethanol); yield 77%; ¹H NMR (CDCl₃): δ = 2.3-2.8 (m, 12H), 3.6 (s, 2H), 6.8-7.6 (m, 9H), 11.0 (s, NH). Anal. Calcd for C₂₀H₂₆N₃OCl₂F: C, 57.98; H, 6.32; N, 10.14. Found: C, 58.12; H, 6.12; N, 9.97. Compound **4b** (n = 2): mp 229 °C (from ethanol); yield 91%; ¹H NMR (CDCl₃): δ = 2.3-2.8 (m, 12H), 3.6 (s, 2H), 7.3 (m, 5H), 7.6 (m, 4H), 11.4 (s, NH). Anal. Calcd for C₂₁H₂₆N₃OCl₂F₃: C, 54.32; H, 5.64; N, 9.05. Found: C, 54.12; H, 5.53; N, 9.17. Compound **4c** (n = 2): mp 234 °C (from ethanol); yield 90%; ¹H NMR (CDCl₃): δ = 2.3 (s, 3H, CH₃), 2.4-2.6 (m, 12H), 3.55 (s, 2H), 7.1 (d, 2H, J = 9 Hz), 7.25-7.55 (m, 7H), 9.8 (s, NH). Anal. Calcd for C₂₁H₂₉N₃OCl₂: C, 61.46; H, 7.12; N, 10.24. Found: C, 61.31; H, 7.02; N, 10.07. Compound **4d** (n = 2): mp > 250 °C (from methanol); yield **88%**; ¹H NMR (CDCl₃): $\delta = 2.3-2.7$ (m, 12H), 3.55 (s, 2H), 7.2-7.7 (m, 14H), 11.1 (s, NH). Anal. Calcd for C₂₆H₃₁N₃OCl₂: C, 66.10; H, 6.61; N, 8.89. Found: C, 66.23; H, 6.54; N, 9.97. Compound **4e** (n = 2): mp 236 °C (from ethanol); yield 91%; ¹H NMR (CDCh): $\delta = 2.3-2.7$ (m, 12H), 3.55 (s, 2H), 3.8 (s, 3H, OCH_3), 6.8 (d, 2H, J = 9 Hz), 7.3 (m, 5H), 7.45 (d, 2H, J = 9 Hz), 10.8 (s, NH). Anal. Calcd for $C_{21}H_{29}N_3O_2Cl_2$: C, 59.16; H, 6.86; N, 9.85. Found: C, 59.12; H, 6.70; N, 9.95. Compound **4f** (n = 3): mp 245 °C (from ethanol); yield 82%; ¹H NMR (CDCl₃): $\delta = 2$ (m, 2H), 2.2-2.6 (m, 12H), 3.5 (s, 2H), 6.5-7.5 (m, 9H), 7.7 (s, NH). Anal. Calcd for C₂₁H₂₈N₃OCl₂F: C, 58.89; H, 6.59; N, 9.81. Found: C, 58.75; H. 6.42; N. 9.92. Compound 4g (n = 3): mp > 250 °C (from methanol); yield 92%; ¹H NMR (CDCl₃): δ = 1.95 (m, 2H), 2.3-2.6 (m, 12H), 3.55 (s, 2H), 7.3-7.7 (m, 9H), 9.2 (s, NH). Anal. Calcd for $C_{22}H_{28}N_3OCl_2F_3$: C, 55.25; H, 5.90; N, 8.78. Found: C, 55.10; H, 6.12; N, 8.71. Compound 4h (n = 3): mp 220 °C (from ethanol); yield 85%; ¹H NMR (CDCl₃): $\delta = 1.9$ (m, 2H), 2.3 (s, 3H, CH₃), 2.35-2.9 (m, 12H), 3.55 (s, 2H), 6.95 (d, 2H, J = 9 Hz), 7.2-7.5 (m, 7H), 9.2 (s, NH). Anal. Calcd for $C_{22}H_{31}N_3OCl_2$: C, 62.26; H, 7.36; N, 9.90. Found: C, 62.12; H, 7.12; N, 9.73. Compound 4i (n = 3): mp 230 °C (from ethanol); yield 77%; ¹H NMR (CDCl₃): $\delta = 1.9$ (m, 2H), 2.2-2.7 (m, 12H), 3.5 (s, 2H), 7.2-7.7 (m, 14H), 8.6 (s, NH). Anal. Calcd for C₂₇H₃₃N₃OCl₂: C, 66.65; H, 6.84; N, 8.84. Found: C, 66.50; H, 6.90; N, 8.71. Compound 4j (n = 3): mp 215 °C (from ethanol); yield 83%; ¹H NMR (CDCl₃): δ = 2.0 (m, 2H), 2.2-2.7 (m, 12H), 3.55 (s, 2H), 3.8 (s, 3H, OCH₃), 6.8 (d, 2H, J = 9 Hz), 7.25 (m, 5H), 7.6 (d, 2H, J = 9 Hz), 8.6 (s, NH). Anal. Calcd for C₂₂H₃₁N₃O₂Cl₂: C, 60.00; H, 7.09; N, 9.54. Found: C, 60.12; H, 7.18; N. 9.63.

4-[2-(4-Fluorophenylamino)carbonyl]ethylpiperazine (2a)

Compound (4a) (25 g, 0.06 mol) was dissolved in a mixture of 100 mL of water and 100 mL of ethanol. 3 g of 10% Pd/C were added. Hydrogenolysis was carried out until the absorption of the theoretical hydrogen volume. The reaction mixture was filtered and the filtrate was neutralised with 2N NaOH and evaporated to dryness. The residue was taken up with dichloromethane and filtered. The filtrate was dried over MgSO₄, filtered and concentrated under vacuum to give crude (2a).

Compounds (**2b-j**) were synthesised similarly. Compound **2a** (n = 2): mp 132 °C (from ethyl acetate); yield 93%; ¹H NMR (CDCl₃): δ = 2.3-2.6 (m, 8H), 2.8-3.0 (m, 4H), 3.7 (s, NH amine), 6.7-7.7 (m, 4H). 10.1 (s, NH amide). Anal. Calcd for C₁₃H₁₈N₃OF: C, 62.13; H, 7.22; N, 16.72. Found: C, 61.89; H, 7.12; N, 16.71. Compound **2b** (n = 2): mp 103 °C (from 2-propanol); yield 67%; ¹H NMR (CDCl₃): δ = 2.4-2.7 (m, 8H), 2.8-3.1 (m, 4H), 3.4 (s, NH amine), 7.3-7.6 (m, 4H), 11.3 (s, NH amide). Anal. Calcd for C₁₄H₁₈N₃OF₃: C, 55.80; H, 6.02; N, 13.95. Found: C, 55.72; H, 5.95; N, 13.78. Compound **2c** (n =

729

2): mp 116 °C (from ethyl acetate); yield 71%; ¹H NMR (CDCl₃): $\delta = 2.3$ (s, 3H, CH₃), 2.4-2.6 (m, 8H), 2.8-3.0 (m, 4H), 3.9 (s, NH amine), 7.1 (d, 2H, J = 9 Hz), 7.4 (d, 2H, J = 9 Hz), 10.6 (s, NH amide). Anal. Calcd for C14H21N3O: C, 67.98; H, 8.57; N, 17.00. Found: C, 67.85; H, 8.43; N, 16.86. Compound **2d** (n = 2); mp 159 °C (from 2-propanol); yield 87%; ¹H NMR (CDCl₃): δ = 2,3-2,6 (m, 8H), 2.8-3.0 (m, 4H), 3.75 (s, NH amine), 7.3-7.7 (9H), 10.8 (s, NH amide). Anal. Calcd for C19H23N3O; C, 73.76; H, 7.49; N, 13.58. Found: C, 73.63; H, 7.56; N, 13.47. Compound 2e (n = 2): mp 127 °C (from 2propanol): vield 84%; ¹H NMR (CDCl₃): $\delta = 2.4-2.6$ (m, 8H), 2.8-3.0 (m, 4H), 3.6 (s, NH amine), 3.8 (s, 3H, OCH₃), 6.8 (d, 2H, J = 9 Hz), 7.6 (d, 2H, J = 9 Hz), 10.0 (s, NH amide). Anal. Calcd for $C_{14}H_{21}N_3O_2$: C, 63.85; H, 8.04; N, 15.96. Found: C, 63.75; H, 7.90; N, 15.86. Compound **2f** (n = 3): mp 146 °C (from ethyl acetate); yield 89%; ¹H NMR (CDCl₃): $\delta = 1.8$ (m, 2H), 2.2-2.6 (m, 8H), 2.7-3.0 (m, 4H), 3.8 (s, NH amine), 6.8-7.6 (m, 4H), 9.0 (s, NH amide). Anal. Calcd for C14H20N3OF; C, 63.37; H, 7.60; N, 15.84. Found: C, 63.45; H, 7.46; N, 15.78. Compound 2g (n = 3): mp 132 °C (from 2propanol); yield 90%; ¹H NMR (CDCl₃): $\delta = 1.95$ (m, 2H), 2.2-2.6 (m, 8H), 2.7-3.1 (m, 4H), 3.2 (s, NH amine), 7.4-7.8 (m, 4H), 9.45 (s, NH amide). Anal. Calcd for C₁₅H₂₀N₃OF₃: C, 57.14; H, 6.39; N, 13.33. Found: C, 57.03; H, 6.25; N, 13.21. Compound **2h** (n = 3): mp 124 °C (from 2-propanol); yield 94%; ¹H NMR (CDCl₃): $\delta = 1.9$ (m, 2H), 2.3 (s, 3H, CH₃), 2.35-2.7 (m, 8H), 2.75-3.1 (m, 4H), 4.0 (s, NH amine), 7.0 (d, 2H, J = 9 Hz), 7.4 (d, 2H, J = 9 Hz), 8.65 (s, NH amide). Anal. Calcd for C15H23N3O; C, 68.93; H, 8.87; N, 16.08. Found: C, 68.72; H, 8.75; N, 15.92. Compound 2i (n = 3): mp 183 °C (from 2propanol); yield 91%; ¹H NMR (CDCl₃): $\delta = 1.9$ (m, 2H), 2.2-2.7 (m, 8H), 3.0 (m, 4H), 3.5 (s, NH amine), 7.3-7.8 (m, 9H), 10.1 (s, NH amide). Anal. Calcd for C₂₀H₂₅N₃O: C, 74.27; H, 7.79; N, 13.00. Found: C, 74.44; H, 7.67; N, 13.12. Compound 2j (n = 3): mp 115 °C (from ethyl acetate); yield 81%; ¹H NMR (CDCl₃): $\delta = 1.9$ (m, 2H), 2.2-2.6 (m, 8H), 2.7-3.0 (m, 4H), 3.8(s, 3H, OCH₃), 5.0 (s, NH amine), 6.8 (d, 2H, J = 9 Hz), 7.4 (d, 2H, J = 9 Hz), 8.4 (s, NH amide). Anal. Calcd for C₁₅H₂₃N₃O₂: C. 64.95; H, 8.36; N, 15.15. Found: C, 64.80; H, 8.23; N, 14.95.

1-Ethoxycarbonyl-4-(benzocycloalkyl)piperazines (9a-c)

A mixture of 1-chlorobenzocycloalkanes (3) (0.65 mol), ethyl *N*-piperazinecarboxylate (124.8 g, 0.79 mol), potassium carbonate (197.34 g, 1.43 mol) and sodium iodide (10 g, 0.06 mol) in 900 mL of acetonitrile, was refluxed for 24 h. After cooling, the mixture was filtered and the solvent was evaporated off. The residual oil was taken up with ice water and the product was extracted 3 times with ethyl acetate. An ethanolic solution of 2N hydrogen chloride was then added. The hydrochloride formed was filtered off. The base was freed with sodium carbonate in a dichloromethane / water mixture. The organic layer was separated, dried over MgSO₄ and evaporated off under vacuum to give compounds (**9a-c**) as an oil. Compound **9a** (m = 1): yield 70%; ¹H NMR (CDCl₃): $\delta = 1.25$ (t, 3H, J = 7 Hz), 2.1 (q, 2H, J = 7 Hz),

2.3-2.9 (m, 6H), 3.5 (t, 4H, J = 5 Hz), 4.15 (q, 2H, J = 7 Hz), 4.35 (t, 1H, J = 7 Hz), 7-7.7 (m, 4H). Anal. Calcd for $C_{16}H_{22}N_2O_2$: C, 70.05; H, 8.08; N, 10.21. Found: C, 69.90; H, 7.88; N, 10.06. Compound **9b** (m = 2): yield 63%; ¹H NMR (CDCl₃): δ = 1.25 (t, 3H, J = 7 Hz), 1.6-2.1 (m, 4H), 2.3-2.9 (m, 6H), 3.5 (t, 4H, J = 5 Hz), 3.8 (t, 1H, J = 7 Hz), 4.15 (q, 2H, J = 7 Hz), 7.0-7.75 (m, 4H). Anal. Calcd for $C_{17}H_{24}N_2O_2$: C, 70.80; H, 8.39; N, 9.71. Found: C, 70.92; H, 8.24; N, 9.58. Compound **9c** (m = 3): yield 61%, ¹H NMR (CDCl₃): δ = 1.25 (t, 3H, J = 7 Hz), 1.4-2.0 (m, 6H), 2.4-2.9 (m, 6H), 3.15 (t, 1H, J = 7 Hz), 3.45 (t, 4H, J = 5 Hz), 4.15 (q, 2H, J = 7 Hz), 7-7.2 (m, 4H). Anal. Calcd for $C_{18}H_{26}N_2O_2$: C, 71.49; H, 8.67; N, 9.26. Found: C, 71.38; H, 8.88; N, 9.10.

4-Benzocycloalkylpiperazines (10a-c)

Potassium hydroxide (300 g, 5.36 mol) was added slowly to a solution of 1-ethoxycarbonyl-4-(benzocycloalkyl)piperazine 9 (0.46 mol), water (100 mL) and methanol (400 mL). The reaction mixture was refluxed. The reaction was monitored by thin layer chromatography. After the starting material had disappeared, the mixture was cooled, filtered and extracted with dichloromethane. The organic layer was separated, dried over MgSO4 and evaporated. The residual oil was taken up with ethanol, and oxalic acid (39.33 g, 0.44 mol) dissolved in ethanol (700 mL) was added. The oxalate formed was filtered off and then freed with sodium carbonate in a dichloromethane / water mixture. After decantation, the organic layer was dried over MgSO₄ and the solvent was evaporated off to give compounds (10a-c) as an oil. Compound 10a (m = 1): yield 83%; ¹H NMR (CDCl₃): δ = 2.1 (t, 2H, J = 7 Hz), 2.4-2.85 (m, 7H), 2.9 (t, 4H, J = 5 Hz), 4.3 (t, 1H, J = 7 Hz), 7.0-7.7 (m, 4H). Anal. Calcd for $C_{13}H_{18}N_2$: C, 77.18; H, 8.97; N, 13.85. Found: C, 76.94; H, 8.88; N, 13.72. Compound 10b (m = 2): yield 84%; ¹H NMR (CDCl₃); $\delta =$ 1.5-2.1 (m, 4H), 2.0 (s, NH), 2.4-2.8 (m, 6H), 2.9 (t, 4H, J = 5 Hz), 3.75 (t, 1H, J = 7 Hz), 7.0-7.8 (m, 4H). Anal. Caled for C₁₄H₂₀N₂: C, 77.73; H, 9.32; N, 12.95. Found: C, 77.90; H, 9.18; N, 12.73. Compound 10c (m = 3); yield 70%; ¹H NMR (CDCh); $\delta = 1.3-2.7$ (m, 12H), 2.9 (t, 4H, J = 5 Hz), 3.2 (t, 1H, J = 7 Hz), 3.9 (s, NH), 7.1 (m, 4H). Anal. Caled for $C_{15}H_{22}N_2$: C, 78.21; H, 9.63; N, 12.16. Found: C, 78.06; H, 9.48; N, 12.05.

1-[2-[(4-Fluorophenylamino)carbonyl]ethyl]-4-(1,2,3,4-tetrahydronaphth-1-yl)piperazine dihydrochloride (1k)

Procedure A : A mixture of 4-[2-[(4-fluorophenylamino)carbonyl]ethyl]piperazine (**2a**) (4 g, 15.9 mmol), 1-chlorotetralin (**3b**) (2.65 g, 15.9 mmol), potassium carbonate (5.55 g, 40 mmol) and sodium iodide (0.5 g, 3.3 mmol) in 150 mL of acetonitrile was refluxed for 24 h. The solution was then evaporated to dryness, the residue was taken up with water and extracted with ethyl acetate. The organic layer was separated, dried over MgSO₄ and evaporated and the residue was chromatographied over a silica gel column (eluent, 0.5% Et₃N / ethyl acetate) to yield 3.3 g of pure product (65%).

Procedure B : A mixture of 4-(1,2,3,4-tetrahydronaphth-1-yl)piperazine (10b) (5.4 g, 25 mmol), ω bromo-*N*-(4-fluorophenyl)propionamide (5a) (5.04 g, 25 mmol) and 8 mL of triethylamine in 100 mL of DMF was heated at 60°C for 48 h. After cooling, the reaction mixture was poured all at once into 200 mL of water and then extracted with ethyl acetate. The organic phase was dried over MgSO₄ and evaporated under vacuum. The residue was purified by column chromatography (eluent, 0.5% Et₃N / ethyl acetate) to yield 3 g of a pure product (31%).

Compounds (1a-v) were synthesised similarly. Analytical and physicochemical data are shown in Tables 1 and 2.

Preparation of the dihydrochloride : The base was dissolved in ethanol, and an ethanolic solution of hydrogen chloride was added and the salt crystallised out.

CONCLUSION

Compounds (1a-v) were prepared by two procedures. The best yields were obtained using procedure A. This probably should be due to the fact that the « pseudobenzylic » halogen of the intermediates (3a-c) is more reactive than that the « alkyl type » halogen of arylalkanoylamides derivatives (5a-j).

ACKNOWLEDGEMENT

We thank Carole Lecinse for linguistic advice

	m	n	R	Yield (%)		mp	Molecular Formula	Microanalyses (%)		%)
				Α	В	°Ċ		C Calcd	H Caled	N Calcd
								Found	Found	Found
1 a	1	2	F	73	41	21 9	C22H28N3OCl2F	59.98	6.41	9.54
				_				60.11	6.47	9.61
16	1	2	CF3	54	36	249	C ₂₃ H ₂₈ N ₃ OCl ₂ F ₃	56.32	5.76	8.57
								56.47	5,68	8.72
lc	1	2	CH3	59	38	218	C ₂₃ H ₃₁ N ₃ OCl ₂	63.30	7.16	9.63
								63.42	7.03	9.59
ld	1	2	Ph	63	35	256	C28H33N3OCl2	67.45	6.67	8,43
		•	0.011				0 11 11 0 0	67.56	6.63	8,56
le	I	2	OCH3	71	40	214	C23H31N3O2Cl2	61.05	6.91	9.29
10		2	г	<i>(</i> 0)		217		60.92	7.05	9.21
11	1	3	F	69	ود	217	C23H30N3UCI2F	60.78	6.65	9.25
1_		2	CT-	40	26	254		60.69	6.76	9.33
Ig	I	2	Cr3	48	30	250	C24H30N3OCI2F3	57.14	5.99	8.33
16	1	2	CUa	57	20	222		57.03	0.10	8.21
10	1	3	Ch3	57	39	232	C24H33N3OCI2	03.98	7.38	9.33
1;	1	2	Dh	68	41	217		04.11 67.05	/.31	9.42
11	I	5	гu	00	41	217	C29H35N3OCI2	69.07	0.00	8.20
1;	1	3	OCH ₂	62	34	243		61 70	0.90	0.15
IJ	•	2	oeng	02	54	273	024113311302012	61.68	7.15	9.01
1k	2	2	F	65	37	246	CasHaoNaOClaF	60.78	6.65	0.25
•••	2	-	•	¢,	5,	240	0231130113000121	60.70	6 74	9.25
11	2	2	CF3	70	42	249	C24H30N3OCl2F3	57.14	5.99	833
			5				02419019001219	57.22	6.10	8 21
1m	2	2	CH3	68	36	221	C24H33N3OCl2	63.98	7.38	9.33
			-				5, 55 5 2	64.12	7,41	9.26
1n	2	2	Ph	64	32	215	C29H35N3OCl2	67.95	6.88	8.20
								68,07	6,79	8.34
10	2	2	OCH3	62	37	208	C24H33N3O2Cl2	61.79	7.13	9.01
								61.74	7.21	9.13
1p	2	3	F	55	31	204	C24H32N3OCl2F	61.53	6.89	8.97
								61.47	6.95	9.06
1q	2	3	CF3	66	38	227	C25H32N3OCl2F3	57.91	6.22	8.10
								58.11	6.36	8.03
1r	2	3	CH3	56	34	217	C25H35N3OCl2	64.64	7.60	9.05
								64.59	7.52	9.11
1s	2	3	Ph	58	32	194	C30H37Cl2N3O	68.43	7.08	7.98
_		-					_	68.52	7.16	8.10
lt	2	3	OCH ₃	51	30	202	C25H35N3O2Ci2	6 2. 49	7.34	8.75
-	•	-	-			a		62.61	7.27	8.89
Lu	ک	2	F	46	31	218	C24H32N3OCl2F	61.53	6.89	8.97
1	2	2	г	53	75	001	0 H N 60 -	61.67	6.73	9.06
1V	3	3	F	53	35	236	C25H34N3OCl2F	62.23	7.10	8.71
								62.33	7.21	8.65

Table 1. Analytical Data for Compounds (1a-v) Dihydrochlorides

Table 2. Spectral data for compounds (1a-v)

	¹ H NMR (base) (CDCl ₃) δ (ppm)	IR(v) (salt) cm ⁻¹
1a	2.1 (q, 2H, J = 7 Hz), 2.4-3.0 (m, 14H), 4.4 (t, 1H, J = 7 Hz), 6.8-7.6 (m, 8H), 11.1 (s, NH)	3261, 1685
1b	2.1 (q, 2H, J = 7 Hz), 2.4-3.1 (m, 14H), 4.4 (t, 1H, J = 7 Hz), 7.2-7.7 (m, 8H), 11.4 (s, NH)	3248, 1697
1c	2.1 (q, 2H, J = 7 Hz), 2.3 (s, 3H, CH ₃), 2.4-3.0 (m, 14H), 4.35 (t, 1H, J = 7 Hz), 6.9-7.45 (m, 8H), 10.9 (s, NH)	3261, 1696
1d	2.1 (q, 2H, J = 7 Hz), 2.4-3.1 (m, 14H), 4.4 (t, 1H, J = 7 Hz), 7.1-7.6 (m, 13H), 11.1 (s, NH)	3257, 1695
le	2.1 (q, 2H, J = 7 Hz), 2.4-3.0 (m, 14H), 3.8 (s, 3H, OCH ₃), 4.35 (t, 1H, J \approx 7 Hz), 6.7-7.5 (m, 8H), 10.8 (s, NH)	3254, 1678
1f	1.7-2.1 (m, 4H), 2.35-3.0 (m, 14H), 4.35 (t, 1H, J = 7 Hz), 6.8-7.6 (m, 8H), 9.0 (s, NH)	3251, 1688
1g	1.7-2.1 (m, 4H), 2.35-3.0 (m, 14H), 4.35 (t, 1H, J = 7 Hz), 7.1-7.6 (m, 8H), 9.2 (s, NH)	3246, 1696
1h	1.7-2.1 (m, 4H), 2.3 (s, 3H, CH ₃), 2.35-2.9 (m, 14H), 4.3 (t, 1H, J = 7 Hz), 6.9-7.5 (m, 8H), 8.8 (s, NH)	3258, 1687
1i	1.7-2.1 (m, 4H), 2.2-3.0 (m, 14H), 4.35 (t, 1H, J = 7 Hz), 7.1-7.6 (m, 13H), 8.8 (s, NH)	3316, 1669
1j	1.7-2.1 (m, 4H), 2.2-3.0 (m, 14H), 3.8 (s, 3H, OCH ₃), 4.35 (t, 1H, J = 7 Hz), 6.7-7.4 (m, 8H), 8.7 (s, NH)	3245, 1688
1 k	1.6-2.1 (m, 4H), 2.4-2.9 (m, 14H), 3.85 (t, 1H, J = 7 Hz), 6.8-7.7 (m, 8H), 11.1 (s, 1H)	3175, 1688
11	1.6-2.1 (m, 4H), 2.4-2.85 (m, 14H), 3.9 (t, 1H, J = 7 Hz), 7.0-7.7 (m, 8H), 11.5 (s, 1H)	3247, 1692
1m	1.6-2.1 (m, 4H), 2.3 (s, 3H, CH ₃), 2.4-2.8 (m, 14H), 3.8 (t, 1H, J = 7 Hz), 7-7.7 (m, 8H), 11.1 (s, NH)	3258, 1691
1n	1.6-2.1 (m, 4H), 2.4-2.9 (m, 14H), 3.8 (t, 1H, J = 7 Hz), 7.0-7.6 (m, 13H), 11.2 (s, NH)	3255, 1690
10	1.6-2.1 (m, 4H), 2.4-2.8 (m, 14H), 3.8 (s, 3H, OCH ₃), 3.85 (t, 1H, J = 7 Hz), 6.8-7.7 (m, 8H), 10.85 (s, NH)	3251, 1679
1p	1.5-2.1 (m, 6H), 2.3-2.8 (m, 14H), 3.8 (t, 1H, J = 7 Hz), 6.8-7.7 (m, 8H), 9.1(s, NH)	3245, 1684
1q	1.5-2.1 (m, 6H), 2.3-2.9 (m, 14H), 3.8 (t, 1H, J = 7 Hz), 7.0-7.8 (m, 8H), 9.25(s, NH)	3245, 1695
1r	1.6-2.1 (m, 6H), 2.3 (s, 3H, CH ₃), 2.4-2.8 (m, 14H), 3.8 (t, 1H, J = 7 Hz), 6.9-7.7 (m, 8H), 8.75 (s, NH)	3240, 1686
1s	1.6-2.1 (m, 6H), 2.3-2.8 (m, 14H), 3.8 (t, 1H, J = 7 Hz), 7.0-7.7 (m, 13H), 9.0 (s, NH)	3237, 1680
1t	1.6-2.1 (m, 6H), 2.3-2.8 (m, 14H), 3.7 (s, 3H, OCH ₃), 3.8 (t, 1H, J = 7 Hz), 6.7-7.7 (m, 8H), 8.8 (s, NH)	3248, 1688
1u	1.2-2.15 (m, 6H), 2.2-2.8 (m, 14H), 3.2 (t, 1H, J = 7 Hz), 6.8-7.6 (m, 8H), 11.05 (s, NH)	3276, 1689
1v	1.5-2.1 (m, 8H), 2.3-2.8 (m, 14H), 3.2 (t, 1H, J = 7 Hz), 6.85-7.7 (m, 8H), 9 (s, NH)	3249, 1688

REFERENCES

- 1. J. Heym and B. Kenneth Koe, J. Clin. Psychiatry, 1988, 49 (Suppl), 40.
- 2. D. N. Midbennis and J. R. Fozard, Eur. J. Pharmacol., 1983, 90, 151.
- (a) A. Buzas, R. Ollivier, Y. El Ahmad, and E. Laurent, PCT Int. Appl. WO 93 16,057 (Cl. C07D295/10), 1993, FR Appl. Feb. 92/1,843, 1992 (Chem. Abstr, 120, 11, 134523c). (b) Y El Ahmad, E. Laurent, P. Maillet, A. Talab, J.F. Teste, R. Dokhan, G. Tran, and R. Ollivier, J. Med. Chem. (in press)

- 4. K. P. Bogeso, J. Med. Chem, 1985, 26, 935.
- 5. M. Saxena, S. K. Agaruval, G. K. Patnaik, and A. K. Saxena, J. Med. Chem., 1990, 33, 2970.
- 6. W. H. Hartung, and R. Simonoff, Organic Reactions, 1953, VII, 263.

Received, 6th January, 1997