

Synthesis of Some 1-Piperazinecarboxylic Acid Ethyl Ester Derivatives as Possible Antifilarial and Antihypertensive Agents¹

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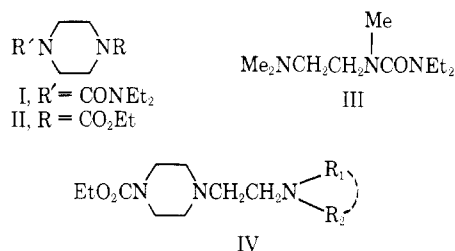
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N,N-Diethyl-4-substituted-1-piperazinecarboxamides (I) and 1-piperazinecarboxylic acid Et ester derivatives (II) exhibit pronounced antifilarial activity.^{3,4} Further, the ethylenediamine derivative III has also been found to be active against litomosoidal infections.⁵ In view of these observations we wished to build up a structural pattern of the type IV wherein 1-piperazinecarboxylic acid Et ester itself will form a part of the ethylenediamine chain.



Since some piperazine derivatives⁶ and also some ethylenediamine derivatives⁷ have been reported as potential antihypertensive agents, the antihypertensive effect of compounds IV, in which both terminal N atoms of the ethylenediamine chain form part of heterocyclic moieties (with the exception of 9) have been studied.

Chemistry.—The substituted ethylenediamine deriv-

atives IV were prepared by the interaction of 4-(β -chloroethyl)-1-piperazinecarboxylic acid Et ester·HCl⁸ and appropriate secondary amines in the presence of anhyd K₂CO₃ in abs EtOH. Compounds 1 and 2 have also been prepared by treating piperidinoethyl chloride·HCl⁹ and morpholinoethyl chloride·HCl,⁹ resp, with 1-piperazinecarboxylic acid Et ester by the same procedure.

Biological Testing. Antifilarial Activity.—Compounds 1–5, 7, and 9 that passed the short-term oral toxicity were screened on *L. carinii* infected albino rats generally at a dose level of 200 mg/kg once daily for 5 consecutive days, and the tail blood (4 mm³) was examined up to day 15. *N,N*-Diethyl-4-methyl-1-piperazinecarboxamide-treated and untreated albino rats were kept as control. Only 1 showed definite microfilaricidal activity even at 100 mg/kg.

Antihypertensive Effects.—All the compounds except 3 as listed in Table I were tested for hypotensive or hypertensive effect on pentobarbital-anesthetized cats of either sex at dose levels of 0.5 mg/kg, 5 mg/kg, and 25 mg/kg iv. The criterion of activity was taken to be a fall in blood pressure by 20 mm for at least 15 min. The hypo- or hypertensive effect is recorded in Table I. Of the 7 compounds tested for antifilarial activity, only one (1) was found to possess definite microfilaricidal activity at 100 and 200 mg/kg.

Encouraging antihypertensive activity was observed with a small number of compounds. Maximum activity was observed in 4, which embodies a 1,2,3,4-tetrahydroisoquinoline moiety at one end of the ethylenediamine chain. In a number of other tetrahydroisoquinoline derivatives, antihypertensive activity has been observed.¹⁰ 3,4-Dihydro-2(*H*)-isoquinoline carboxamide has been found clinically to be a potent antihypertensive agent.¹¹ The parent compound, 4-(β -chloroethyl)-1-piperazinecarboxylic acid Et ester·HCl also produced an hypotensive effect.

Experimental Section¹²

Intermediates.—The requisite 4-(β -chloroethyl)-1-piperazinecarboxylic acid Et ester·HCl, piperidinoethyl chloride·HCl, morpholinoethyl chloride·HCl, *N*-phenylpiperazine,¹³ *N*-benzylpiperazine,¹⁴ and *N*-(*p*/*m*-chlorophenyl)piperazine¹⁵ were prepd according to the methods available in the lit. 1,2,3,4-Tetrahydroisoquinoline has been prepd by a slight modification¹⁶ of the method of Pyman and coworkers.¹⁷ For the prepn of 1-piperazinecarboxylic acid Et ester, however, a simple method has been developed which has a definite advantage over the lit. proce-

(1) (a) Taken in part from the Ph.D. thesis submitted by Prabhask Chandra Das, November 1968, to the Calcutta University; (b) presented in part at the Annual Sessions of the Indian Science Congress: P. C. Das, B. B. Patra, and U. P. Basu, *Proc. Indian Sci. Congr.*, **57**, 131 (1970).

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(3) (a) S. Kushner, L. M. Brancione, R. I. Hewitt, W. L. McEwen, Y. Subbarow, H. W. Stewart, R. J. Turner, and J. J. Denton, *J. Org. Chem.*, **13**, 144 (1948); (b) R. I. Hewitt, S. Kushner, H. W. Stewart, E. White, W. S. Wallace, and Y. Subbarow, *J. Lab. Clin. Med.*, **32**, 1314 (1947).

(4) (a) R. I. Hewitt, E. White, W. S. Wallace, H. W. Stewart, S. Kushner, and Y. Subbarow, *ibid.*, **32**, 1304 (1947); (b) H. W. Stewart, R. J. Turner, J. J. Denton, S. Kushner, L. M. Brancione, W. L. McEwen, R. I. Hewitt, and Y. Subbarow, *J. Org. Chem.*, **13**, 134 (1948).

(5) P. Sewell and F. Hawking, *Brit. J. Pharmacol.*, **5**, 239 (1950).

(6) (a) S. M. Olin, U. S. Patent 2,858,313 (Oct 28, 1958); (b) Park Davis and Company, British Patent 803,403 (Oct 22, 1958); (c) R. M. Jacob, R. J. Harolais, and G. L. Regnier, British Patent 802,244 (Oct 1, 1958); (d) L. F. Bach, Jr., H. J. Barbander, and S. Kushner, U. S. Patent 2,909,523 (Oct 20, 1959), 2,837,522 (June 3, 1958); (e) Park Davis and Company, British Patent 850,662 (Oct 5, 1960); (f) B. Rudner, U. S. Patent 2,967,865 (Jan 10, 1961); (g) S. Hayao, R. N. Schut, and G. W. Strycker, *J. Med. Chem.*, **6**, 133 (1963); (h) J. R. Boissier and R. Ratouis, French Patent 1,318,449 (Feb 15, 1963); M1784 (May 20, 1963); M1749 (April 22, 1963); 1320235 (March 8, 1963); (i) R. Ratouis, J. R. Boissier, and C. Dumont, *J. Med. Chem.*, **8**, 271 (1965); (j) H. Howell, C. B. Pollard, L. B. Kier, and H. H. Sisler, *ibid.*, **6**, 604 (1963); (k) R. N. Prasad and J. Tietje, *ibid.*, **12**, 551 (1969).

(7) (a) A. P. Swain and S. K. Naegle, *J. Amer. Chem. Soc.*, **76**, 5089 (1954); (b) S. L. Shapiro, H. Soloway, E. Chodos, and L. Freedman, *J. Pharm. Soc.*, **50**, 1035 (1961).

(8) M. Harfenist, *J. Amer. Chem. Soc.*, **76**, 4991 (1954).

(9) P. S. Wadia, T. C. Asthana, and N. Anand, *J. Sci. Ind. Res., Sect. B*, **17**, 11 (1958).

(10) (a) A. G. Wander, British Patent 774,649 (May 15, 1957); (b) D. L. Cook, C. A. Lawler, and J. W. Cusic, *J. Pharmacol. Exp. Ther.*, **120**, 269 (1957); (c) K. Lempert and L. Karoly, *Magy. Chem. Foly.*, **63**, 84 (1957); (d) R. K. Bickerton, M. L. Lacquart, W. J. Kinnard, Jr., J. A. Bianculi, and J. P. Buckley, *J. Amer. Pharm. Ass. Pract. Pharm. Ed.*, **49**, 183 (1960); (e) J. W. Cusic, U. S. Patent 2,785,166 (March 12, 1957); (f) J. H. Biel, U. S. Patent 2,948,722 (Aug 1960); (g) W. Wenner, *J. Med. Chem.*, **8**, 125 (1965).

(11) F. A. Finnerty, Jr., *Med. Clin. N. Amer.*, **48**, 331 (1964).

(12) Melting points were detd in a Buchi melting point apparatus and are uncor.

(13) C. B. Pollard and L. G. Macdowell, *J. Amer. Chem. Soc.*, **56**, 2199 (1934).


(14) R. Baltzly, J. J. Buck, E. Lorz, and W. Schon., *ibid.*, **66**, 263 (1944).

(15) C. B. Pollard and T. H. Wicker, Jr., *ibid.*, **76**, 1853 (1954).

(16) S. S. Chakravorti, "Quinoline Derivatives as Possible Amoebicides," Ph.D. thesis, Calcutta University, 1963.

(17) R. Forsyth, C. I. Kelley, and F. L. Pyman, *J. Chem. Soc.*, **127**, 1659 (1925).

TABLE I: SUBSTITUTED ETHYLENEDIAMINES IV

No.		Bp (mm) or mp of base, °C	Mp of hydrochloride, °C	Crystn solvent	Formula ^b	—Pressure response, mm—		
						Dose, mg/kg	Fall	Rise
1	N-Piperidino	193–195 (4–6)	278–279	A ^c	C ₁₄ H ₂₇ N ₃ O ₂ ·2HCl	25	Nil	Nil
2	N-Morpholino	220–222 (10–12)	285–286	A	C ₁₃ H ₂₅ N ₃ O ₃ ·2HCl	25	Nil	Nil
3	N-Pyrrolidino	173–175 (4–6)	295–296	B	C ₁₃ H ₂₅ N ₃ O ₂ ·2HCl			
4	N-1,2,3,4-Tetrahydroisoquinolino	248–250 (8–10)	265–267	A	C ₁₈ H ₂₇ N ₃ O ₂ ·2HCl	10	62	Nil
5	4-Benzyl-1-piperazino	253–255 (4–6)	261–262	A	C ₂₀ H ₃₂ N ₄ O ₂ ·2HCl	25	41	Nil
6	4- <i>p</i> -Chlorophenyl-1-piperazino	97	264–265	B	C ₁₉ H ₂₀ ClN ₄ O ₂ ·2HCl	25	Biphasic	
7	4-Phenylpiperazino	<i>a</i>	264–266	A	C ₁₉ H ₂₀ N ₄ O ₂ ·2HCl	25	Nil	Nil
8	4- <i>m</i> -Chlorophenyl-1-piperazino	263–265 (4–6)	266–268	A	C ₁₉ H ₂₀ ClN ₄ O ₂ ·2HCl	10	55	Nil
9	<i>i</i> -Pr ₂ N	184–187 (4–6)	232–235	C	C ₁₅ H ₃₁ N ₃ O ₂ ·2HCl	25	Nil	Nil

^a Decompd during distn. ^b All HCl salts were analyzed for C, H, N, Cl, and the anal. results were within $\pm 0.4\%$ of the theoretical values. ^c A, MeOH; B, EtOH; C, EtOH–Et₂O.

dures.¹⁸ In these, 1,4-piperazinedicarboxylic acid Et ester is invariably formed along with 1-piperazinecarboxylic acid Et ester. Further the methods are tedious and work-up is difficult. In the present procedure, formation of the disubstituted product has been totally avoided. 1-Piperazinecarboxaldehyde¹⁹ is first converted to 4-formyl-1-piperazinecarboxylic acid Et ester²⁰ which on hydrolysis with NaOH (10%) for 4 hr gave 1-piperazinecarboxylic acid Et ester in 85–90% yield.

Substituted Ethylenediamines IV.—A mixt of 4-(β -chloroethyl)-1-piperazinecarboxylic acid Et ester·HCl (0.05 mole), the appropriate secondary amine (0.05 mole), anhyd K₂CO₃ (0.05 mole), and abs EtOH (50 ml) was refluxed for about 6 hr, and the solvent was removed by distn. The residual material was treated with H₂O and the aq soln after basification with 50% NaOH soln to pH 9 was extd with Et₂O. The ext was dried (Na₂SO₄) and concd to afford the desired product as liq which was distd *in vacuo*. In all cases the viscous liquids finally obt'd were converted into the corresponding hydrochlorides by passing dry HCl through an Et₂O soln. All compds were characterized as their hydrochlorides. Only 6 (see Table I) gave an anal. pure sample of the base on crystn from petr ether (bp 60–80°). The characteristics of IV have been recorded in Table I.

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(18) (a) T. S. Moore, M. Boyle, V. M. Thorn, *J. Chem. Soc.*, **39**, (1929); (b) H. K. Hall, Jr., *J. Amer. Chem. Soc.*, **78**, 2570 (1956); (c) K. R. Jacobi, *Ber.*, **B66**, 113 (1933).

(19) K. Fujii, K. Tomino, and H. Watanabe, *Yakagaku Zasshi*, **74**, 1049 (1954).

(20) (a) W. Logemann, D. Artini, and G. Tosolini, *Chem. Ber.*, **91**, 2566 (1958); (b) conversion of 1-piperazinecarboxaldehyde to 4-formyl-1-piperazinecarboxylic acid Et ester is more advantageous than to convert 1-piperazinecarboxylic acid Et ester to 4-formyl-1-piperazinecarboxylic acid Et ester according to the method of Logemann, *et al.*^{20a}

Optical Isomers of Mepivacaine and Bupivacaine

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Current interest in the potent local anesthetics mepivacaine and bupivacaine—*N*-methyl and *N*-butyl derivatives of (\pm)-2',6'-pipecoloxylidide—I prompted us to prepare and study the optical isomers. The parent (\pm)-I was resolved using dibenzoyl (+)-tartaric acid. Mepivacaine was resolved by crystallization of

its quinic acid salts.¹ Although a number of optically active acids were tried as resolving agents for (\pm)-bupivacaine, no separation of the isomers could be effected until seed crystals were made available by *N*-butylation of (–)-I and crystallization of its salt with (+)-tartaric acid.

An observation that (+)-mepivacaine·HCl and (–)-bupivacaine·HCl were significantly longer acting than their enantiomers has been reported in an earlier publication from this laboratory.² Thus it became of interest to establish their configuration. This was accomplished by preparing from (*R*)-(+)-methyl pipercolate³ and 2,6-xyldinomagnesium bromide⁴ the parent (*R*)-(–)-I identical with (–)-I by resolution of (\pm)-I. *N*-Butylation of a sample of this (*R*)-I gave (*R*)-(+)-bupivacaine and *N*-methylation of (*S*)-I (obtained from resolution of (\pm)-I) gave (*S*)-(+)-mepivacaine. Thus, the longer-acting (+)-mepivacaine and (–)-bupivacaine isomers are both of the (*S*) configuration.

Experimental Section

Resolution of 2',6'-Pipecoloxylidide (I).—To a soln of 42.0 g (0.15 mole) of (\pm)-I in 300 ml of boiling *i*-PrOH was added a soln of 38.0 g (0.10 mole) of dibenzoyl (+)-tartaric acid monohydrate (DBT) in 300 ml of boiling *i*-PrOH. Immediate crystn occurred which was completed by slow stirring while the mixt cooled to 35°. The ppt was collected, washed with *i*-PrOH, and dried at 70° to give 32 g of (+)-base DBT salt, mp 186–189°. This crop was converted to base by suspending in 300 ml each of H₂O and Et₂O and adding 8 ml of 28% NH₄OH. The Et₂O layer was sep'd, washed with H₂O, and concd *in vacuo*. The residue was crystd from boiling hexane to give a 12.0-g first crop of the base, mp 130–132°, [α]_D²⁵ +46.1° (c 2.3, 1 N HCl). This rotation was unchanged after recrystn from *i*-PrOAc.

The resoln liquor was evap'd *in vacuo*, and the residual crude (–)-base DBT salt was converted to base as above and recrystd twice from boiling hexane to give 11.1 g of base, mp 130–132°, [α]_D²⁵ –46.8° (c 2.3, 1 N HCl), [α]_D²⁵ –11.04 (c 5, MeOH).

Resolution of (\pm)-Mepivacaine.—A soln of 46.0 g (0.186 mole) of (\pm)-mepivacaine (mp 149–151°) with 38.4 g (0.2 mole) of quinic acid (Freas Bros.) and 400 ml of abs EtOH was seeded at 60° and stirred and cooled to 25°. The cryst ppt was collected and recrystd from 300 ml of 95% EtOH to give 34 g of (+)-base quinate, mp 192–195°. This salt was dissolved in 300 ml of H₂O and basified slowly with NH₄OH while rubbing and stirring to induce crystn. The pptd base was collected, washed with H₂O,

(1) B. T. Ekenstam, B. von Egner, and G. Petterson, *Acta Chem. Scand.*, **11**, 1183 (1957), who resolved mepivacaine "with the aid of tartaric acid" but gave no details.

(2) F. P. Luduena, *Annu. Rev. Pharmacol.*, **9**, 503 (1969).

(3) P. S. Portoghesi, T. L. Pazdernik, W. L. Kuhn, G. Hite, and A. Shafi'ee, *J. Med. Chem.*, **11**, 12 (1968).

(4) Thuresson and Egner, U. S. Patent 2,799,679. These authors used the Bodrax reaction to prepare several racemic 2,6-xyldides.