

THE REACTION OF TRIBUTYLSTANNYLLITHIUM WITH 2-ETHOXY-3-ALKENE-
NITRILES. A NEW METHOD FOR THE PREPARATION OF γ -ETHOXYALLYL-
STANNANES AND DIENOL ETHERS

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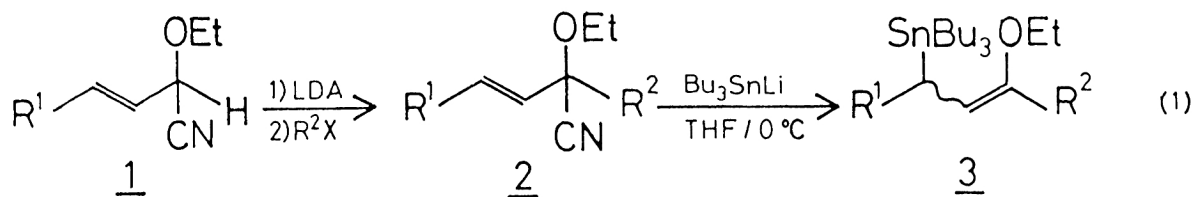
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The reaction of 2-ethoxy-3-alkenenitriles with tributylstannyl-
lithium gave the S_N2' product, γ -ethoxyallylstannanes (3),
in good yields. When 3 have a trimethylsiloxyl group at
 δ -position, the corresponding substituted 2-ethoxy-1,3-butadienes
were produced by the elimination of trimethylsiloxystannane.

The substitution reactions of allylic halides¹⁾ or phosphates²⁾ with
trialkylstannyl anion were reported as useful methods for the preparation of
substituted allylstannanes, which play an important role in organic synthesis. In
this communication, we wish to describe a new method for the preparation of
 γ -ethoxyallylstannanes (3) by the S_N2' reaction of 2-ethoxy-3-alkenenitriles
(2) with tributylstannyl lithium and the application of this reaction to the
synthesis of dienol ethers (5).

Recently, we showed that the alkylthio group of 2-alkylthioalkenenitrile was
reductively removed to give the corresponding α -anion of alkenenitrile by the
treatment of tributylstannyl lithium.³⁾ We examined the similar reductive
lithiation of 2-ethoxy-3-alkenenitrile using tributylstannyl lithium as a reducing
agent. The expected reductive removal of ethoxyl group, however, was not observed
but the nucleophilic displacement of cyano group with tributylstannyl anion
proceeded exclusively (Eq. 1).

2-Ethoxy-3-alkenenitriles (1) prepared from the corresponding acetals of
 α,β -unsaturated aldehyde⁴⁾ were treated with LDA (THF/ -78°C) and alkyl halides
(-78°C - r.t.) successively to give the α -alkylated products (2) in good yields
(2a; 74%, 2b; 54%, 2c; 86%, 2d; 67%, 2e; 63%).⁵⁾ When the alkenenitriles
2 were allowed to react with tributylstannyl lithium in THF at 0°C , γ -ethoxyallyl-



stannanes (3) were produced in good to high yields and no formation of the S_N2 product, α -ethoxyallylstannane, was observed (Table 1).

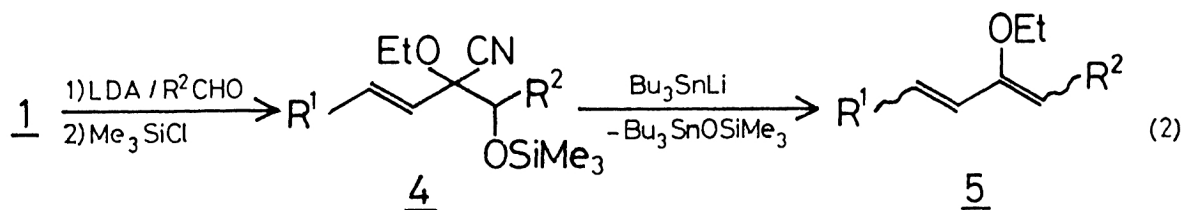
Table 1. Synthesis of γ -ethoxyallylstannane (3)^{a)}

	R ¹	R ²	Temp °C	Time h	Yield ^{b) 6)} %
a	CH ₃	PhCH ₂	0	2.5	87 ⁷⁾
b	CH ₃	CH ₃ (CH ₂) ₇	r.t.	1.5	97 ⁷⁾
c	CH ₃ (CH ₂) ₂	PhCH ₂	0	1.5	95
d	CH ₃ (CH ₂) ₂	CH ₃ (CH ₂) ₇	0	1.5	84
e	CH ₃ (CH ₂) ₂	CH ₃	0	1.5	89 ^{c)}

a) 1.2 equiv. of tributylstannyllithium were used. b) γ -Ethoxyallylstannane was isolated as a mixture of stereoisomers by column chromatography using aluminum oxide deactivated by addition of 6% of water. c) E:Z = 11 : 10. The ratio was determined by HPLC analysis (Merck Si 60; solvent, hexane).

Concerning the preparation of γ -ethoxyallylstannane (3), Quintard et al. reported that 3 was produced by the acid-catalyzed isomerization of α -ethoxyallylstannane which was synthesized by the reaction of chloroethoxymethylstannane with vinylmagnesium reagent.⁸⁾ It should be noted that the present method is apparently advantageous to the preparation of 3.

Next, the synthesis of dienol ethers (5) from 2-ethoxy-3-alkenenitriles (1) was examined on the basis of the above observation. The trimethylsiloxy-alkenenitriles (4) were synthesized by the reaction of lithium salts of 1 with aldehydes followed by the treatment with chlorotrimethylsilane. The substituted 2-ethoxy-1,3-butadienes (5) were obtained in good yields by the simple treatment of the trimethylsilyl ethers (4) with tributylstannyllithium in THF (Eq. 2).

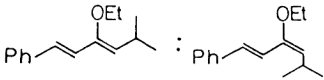


The following experimental procedure is representative: To a THF (1.5 ml) solution of LDA (diisopropylamine (132 mg, 1.3 mmol) and butyllithium (1.2 mmol)) was added a THF (1 ml) solution of 2-ethoxy-3-hexenenitrile (153 mg, 1 mmol) at -78 °C. After stirring for 30 min, 3-phenylpropionaldehyde (161 mg, 1.2 mmol) in THF (1 ml) was added at -78 °C and the reaction mixture was gradually warmed up to 0 °C. Then chlorotrimethylsilane (0.19 ml, 1.5 mmol) was added to the reaction mixture and the mixture was stirred overnight. The reaction was quenched by the addition of a phosphate buffer solution (pH 7) and organic material was extracted with AcOEt. The extract was dried (Na₂SO₄) and condensed under

reduced pressure. The residue was chromatographed on silica gel (hexane-AcOEt) and 4-ethoxy-1-phenyl-3-trimethylsiloxy-5-nonene-4-carbonitrile (4c, 288 mg) was isolated in 80% yield. A THF solution of tributylstannylolithium (2.1 mmol) was added to 4c (360 mg, 1 mmol) in THF (2 ml) at 0 °C and the mixture was stirred for 2 h. After addition of a 5% NaHCO₃ aqueous solution, organic material was extracted with ether and dried over Na₂SO₄. The solvent was removed and the crude dienol ether was purified by column chromatography (aluminum oxide deactivated by 6% of water / hexane) and 5-ethoxy-1-phenyl-4,6-decadiene (5c) (146 mg) was obtained in 60% yield.

Table 2. Preparation of dienol ether (5)

	R ¹	R ²	Yield/% ⁶⁾	
			<u>4</u>	<u>5</u> ^{a)}
a	CH ₃	(CH ₃) ₂ CH	67	73
b	CH ₃	Ph(CH ₂) ₂	74	86
c	CH ₃ (CH ₂) ₂	Ph(CH ₂) ₂	80	60
d	CH ₃ (CH ₂) ₂	CH ₃ (CH ₂) ₇	67	72
e	Ph	(CH ₃) ₂ CH	86	79 ^{b)}
f	Ph	Ph(CH ₂) ₂	83	63

a) Isolated as a mixture of stereoisomers. b)  = 6 : 5. The ratio was determined by NMR analysis.

In a similar manner, several dienol ethers (5) were synthesized as shown in Table 2.

It is known that the primary conventional route to 2-alkoxy-1,3-butadienes (5) is the pyrolysis of γ -alkoxyacetals.⁹⁾ However, it generally affords low yields because of the rather vigorous reaction conditions employed. The present reaction provides a convenient method for the preparation of 2-ethoxy-1,3-butadienes utilizing two different aldehydes (R¹CH=CHCHO and R²CHO).

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References

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