

mixture was allowed to stand overnight at room temperature, treated with MeOH, neutralized with 60% KOH, and extracted with Et₂O (5 × 200 ml). The combined Et₂O extracts were washed (H₂O), dried (MgSO₄), and evaporated to give 1.5 g of the piperidyl ketone 30a. The crude ketone was dissolved in 100 ml of THF and added to a freshly prepared CH₃MgI solution (0.1 mol) in 20 min. The mixture was stirred for 3 hr. After the excess Grignard reagent was consumed by H₂O, the Et₂O layer was separated. The aqueous phase was extracted with Et₂O (4 × 100 ml). The combined Et₂O solution was washed (H₂O), dried (MgSO₄), and evaporated to afford 1.2 g of 1-[3,6-bis(trifluoromethyl)-9-phenanthryl]-1-(4-piperidyl)ethanol (29c) as white crystals, mp 251–254°. Compound 29c was also prepared in 88% yield by catalytic hydrogenation (in PtO₂) of 29f. Treatment of 4.2 g of 29c with 100 ml of Ac₂O for 30 min gave 4.1 g (90% yield) of the acetamide 29d as white crystals, mp 218–220°.

A solution of 1 g of 29d in 20 ml of DMF was treated with 0.5 g of NaH in mineral oil and 3 ml of EtI. The mixture was stirred at room temperature for 24 hr and then diluted with H₂O (300 ml). The ether 29e was isolated by Et₂O extraction and then chromatographed over silica gel to separate the unreacted starting material. There was obtained 0.6 g of 29e, which slowly solidified on standing. This was refluxed overnight with a mixture of 20 ml of EtOH and 20 ml of 60% aqueous KOH. The mixture was diluted with H₂O (300 ml) and extracted with Et₂O (3 × 100 ml). The combined Et₂O extracts were washed (H₂O), dried (MgSO₄), and evaporated. The residue was dissolved in 1 ml of EtOH saturated with HCl. Upon addition of Et₂O, the solid product precipitated. It was collected by filtration and recrystallized from MeOH–Et₂O

to give 0.4 g of the HCl salt of 28, mp 195–198°. *Anal.* (C₂₅H₂₅F₆NO·HCl) C, H, N.

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References

- (1) A. Marxer, *Chimia*, **21**, 592 (1967).
- (2) A. Marxer, *Helv. Chim. Acta*, **52**, 262 (1969).
- (3) A. Marxer, British Patent 1,133,302 (1968); U. S. Patent 3,488,356 (1970) [*Chem. Abstr.*, **70**, 68187s (1969)].
- (4) W. Peters, *Ann. Trop. Med. Parasitol.*, **64**, 189 (1970).
- (5) C. C. Cheng, *J. Pharm. Sci.*, **60**, 1596 (1971).
- (6) P.-L. Chien, D. J. McCaustland, W. H. Burton, and C. C. Cheng, *J. Med. Chem.*, **15**, 28 (1972).
- (7) P.-L. Chien and C. C. Cheng, *J. Med. Chem.*, **16**, 1093 (1973).
- (8) C. J. Ohnmacht, A. R. Patel, and R. E. Lutz, *J. Med. Chem.*, **14**, 926 (1971).
- (9) L. F. Fieser and V. Desreux, *J. Amer. Chem. Soc.*, **60**, 2255 (1938).

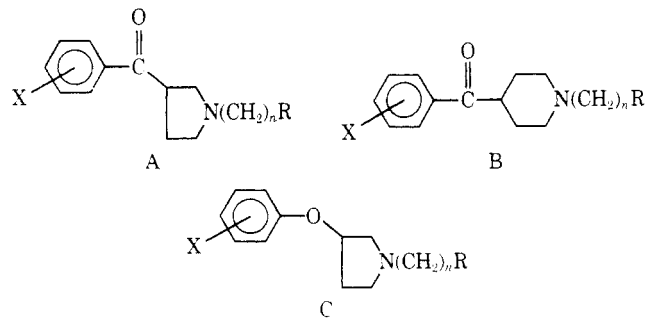
Synthesis of Some *N*-Carboxylic Acid Derivatives of 3-Phenoxypyrrolidines, 4-Phenoxypiperidines, and 3-Phenoxynortropans with Muscle Relaxant and Anticonvulsant Activities

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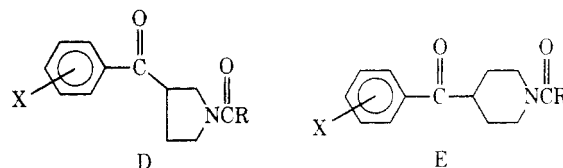
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The title compounds were prepared by reacting the intermediate phenoxypyrrolidines, phenoxypiperidines, and phenoxynortropans with cyanogen bromide, an isocyanate, a carbamoyl chloride, or phosgene and methylamine. Several of the intermediate ethers were prepared by a novel reaction in which an aromatic fluorine is displaced with a heterocyclic alkoxide ion. Anticonvulsant or muscle relaxant activities were observed for several of these compounds.

Previous reports from these laboratories described the preparation and CNS depressant activity of some *N*-alkyl derivatives of 3-benzoylpyrrolidines (A),¹ 4-benzoylpiperidines (B),¹ and 3-phenoxypyrrolidines (C).² *N*-Carboxylic



acid derivatives of the 3-benzoylpyrrolidines (D) and 4-benzoylpiperidines (E) were prepared in a study of structural modification and found to possess anticonvulsant and muscle relaxant activities.³ As an extension of this work we have prepared *N*-carboxylic acid derivatives of 3-phenoxypyrrolidines (F), 4-phenoxypiperidines (G), and 3-phenoxynortropans (H).



The *N*-carboxylic acid derivatives were prepared from the appropriate 3-phenoxypyrrolidine, 4-phenoxypiperidine, or 3-phenoxynortropane intermediate by reaction with nitrourea, an isocyanate, disubstituted carbamoyl chloride, or by treating the *N*-benzyl intermediates with cyanogen bromide (followed by hydrolysis) or phosgene

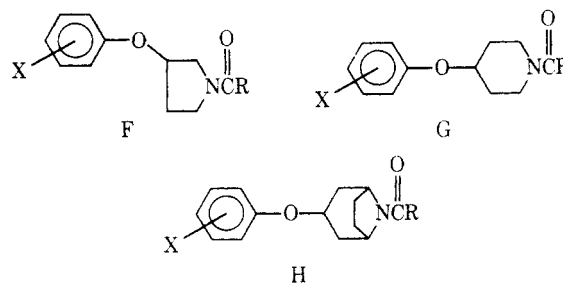
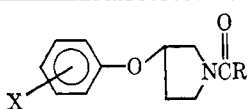


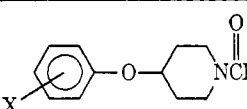
Table I



No.	X	R	Yield, %	Method	Recrystn solvent ^a	Mp or bp (mm), °C	Formula
1	H	NHCH ₃	81	E	I-B	122.5-123.5	C ₁₂ H ₁₆ N ₂ O ₂
2	3-Cl	NH ₂	14	D ₁	B-EA	160-163	C ₁₁ H ₁₃ ClN ₂ O ₂
3	3-Cl	NHCH ₃	78 (59.5)	E (C)	I-B	112-113	C ₁₂ H ₁₅ ClN ₂ O ₂
4	4-Cl	NH ₂	23	D ₁	EA	172-175	C ₁₁ H ₁₃ ClN ₂ O ₂
5	4-Br	NH ₂	12	D ₁	B-EA	167-169	C ₁₁ H ₁₃ BrN ₂ O ₂
6	4-F	NH ₂	62	G	EA-IE	166-167	C ₁₁ H ₁₃ FN ₂ O ₂
7	4-F	N(CH ₃) ₂	47	H		120-124 (0.04)	C ₁₃ H ₁₇ FN ₂ O ₂
8	3-CF ₃	NH ₂	44	G	IE-EA	145-147	C ₁₂ H ₁₃ F ₃ N ₂ O ₂
9	3-CF ₃	NHCH ₃	66	E	IE-B	102-103.5	C ₁₃ H ₁₅ F ₃ N ₂ O ₂
10	3-CF ₃	N(CH ₃) ₂	31	H		123-125 (0.08)	C ₁₄ H ₁₇ F ₃ N ₂ O ₂
11	3-CF ₃	NHC ₂ H ₅	42	E	IE-I	77-79	C ₁₄ H ₁₇ F ₃ N ₂ O ₂
12	3-CF ₃	NHC ₆ H ₅	70	E	B-I	150-152	C ₁₈ H ₁₇ F ₃ N ₂ O ₂
13	2-OCH ₃	NH ₂	57	G	EA-IP	145-147	C ₁₂ H ₁₆ N ₂ O ₃
14	2-OCH ₃	NHC ₆ H ₄ -4-OCH ₃	78	E	B-I	118-119.5	C ₁₉ H ₂₂ N ₂ O ₄
15	2-OCH ₃	NHCH ₃	71	E	IE-B	156-158	C ₁₃ H ₁₈ N ₂ O ₃
16	2-OCH ₃	N(C ₆ H ₅) ₂	73	H		^c	C ₂₄ H ₂₄ N ₂ O ₃
17	2-OCH ₃ , 4-COCH ₃	NH ₂	26	G	EA-IE	154-156	C ₁₄ H ₁₈ N ₂ O ₄
18	2-OCH ₃ , 4-COCH ₃	NHCH ₃	35	E	B	168-170	C ₁₅ H ₂₀ N ₂ O ₄
19	2-OC ₂ H ₅	NH ₂	8	G	IE	116-119	C ₁₃ H ₁₈ N ₂ O ₃
20	3,5-(CH ₃) ₂	NH ₂	41	G	EA	166-168	C ₁₃ H ₁₈ N ₂ O ₂
21	3,5-(CH ₃) ₂	NHCH ₃	52	E	EA	166-169	C ₁₄ H ₂₀ N ₂ O ₂
22	3-OCH ₃	NH ₂	51	G	IP	182-184	C ₁₂ H ₁₆ N ₂ O ₃
23	4-OCH ₃	NH ₂	51	G	IP	161-163	C ₁₂ H ₁₆ N ₂ O ₃

^aThe following solvent abbreviations are used in all tables: B = benzene; E = ethanol; EA = ethyl acetate; Et = diethyl ether; I = isooctane; IP = isopropyl alcohol; IE = isopropyl ether. ^bAll melting point and boiling point temperatures in all tables are uncorrected. ^cAnalytical sample molecularly distilled.

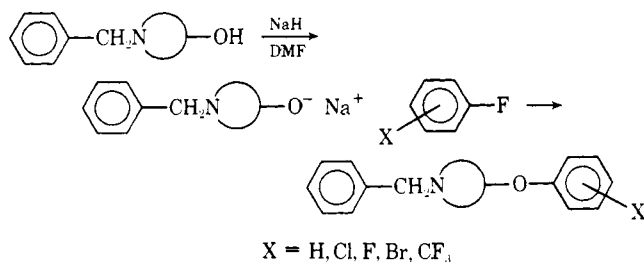
Table II



No.	X	R	Yield, %	Method	Recrystn solvent	Mp or bp (mm), °C	Formula
24	H	NHCH ₃	91	E	I-B	95-96	C ₁₃ H ₁₈ N ₂ O ₂
25	2-OCH ₃	NH ₂	37	I	EA	104-106	C ₁₃ H ₁₈ N ₂ O ₃
26	3-CF ₃	NH ₂	36	G	EA-IE	148-150	C ₁₃ H ₁₅ F ₃ N ₂ O ₂
27	3-CF ₃	NHCH ₃	67	E	IE-I	100-101	C ₁₄ H ₁₇ F ₃ N ₂ O ₂
28	3-CF ₃	N(CH ₃) ₂	54	H		121-124 (0.04)	C ₁₅ H ₁₉ F ₃ N ₂ O ₂
29	3-CF ₃	NHC ₄ H ₉	65	E			C ₁₇ H ₂₃ F ₃ N ₂ O ₂
30	3-CF ₃	NHC ₂ H ₅	66	E	I-B	78-79	C ₁₅ H ₁₉ F ₃ N ₂ O ₂
31	3-CF ₃	NHC ₆ H ₅	69	E	I-B	146.5-147.5	C ₁₉ H ₂₅ F ₃ N ₂ O ₂
32	3-CF ₃	NHC ₆ H ₄ -3-Cl	82	E	I-B	115-116	C ₁₉ H ₁₈ ClF ₃ N ₂ O ₂
33	4-CF ₃	NH ₂	56	G	B-I	176-178	C ₁₃ H ₁₅ F ₃ N ₂ O ₂
34	4-CF ₃	NHCH ₃	81	E	I-B	139-141	C ₁₄ H ₁₇ F ₃ N ₂ O ₂
35	4-CF ₃	N(CH ₃) ₂	53	H		142-145 (0.05)	C ₁₅ H ₁₉ F ₃ N ₂ O ₂

(followed by reaction with an amine). Details are given in Tables I-III and in the Experimental Section.

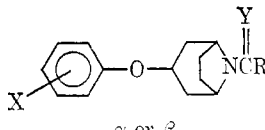
Most of the intermediates were prepared by nucleophilic displacement of a halogen or tosylate from 1-benzyl-3-bromo-, chloro-, or -tosylpyrrolidine or 1-benzyl-4-bromo-,



chloro-, or -tosylpiperidine with a phenoxide ion.^{2,4} The 8-benzyl-3-phenoxynortropanes⁵ and several of the 1-benzyl-3-phenoxypyrrolidines and 1-benzyl-4-phenoxypiperidines were prepared in reasonable yields by treating the sodium salt of 1-benzyl-3-pyrrolidinol, 1-benzyl-4-piperidinol, or 8-benzyl-3α- (or β-) nortropanol in DMF with fluorobenzene or a substituted fluorobenzene at 60-70°.

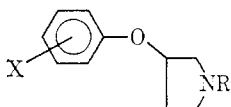
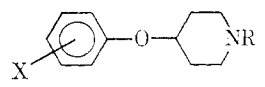
The resulting ethers are thought to be formed by direct nucleophilic displacement of an aromatic fluorine by the alkoxide ion. Positional isomers of substituents on the aromatic ring resulting from benzyne formation have not been detected. There was no evidence of chlorine or bromine displacement. The surprising ease with which the nonactivated fluorobenzene undergoes nucleophilic displacement by an alkoxide ion under the conditions described suggests great synthetic utility.

Table III

 α or β									
No.	X	α or β	Y	R	Yield, %	Method	Recrystn solvent	Mp or bp (mm), °C	Formula
36	3-CF ₃	α	O	NH ₂	62	G	I-B	106-109	C ₁₅ H ₁₇ F ₃ N ₂ O ₂ ^a
37	3-CF ₃	β	O	NH ₂	55	G	IE-EA	149-152	C ₁₅ H ₁₇ F ₃ N ₂ O ₂
38	3-CF ₃	α	O	NHCH ₃	57	E	I-B	155-157.5	C ₁₆ H ₁₉ F ₃ N ₂ O ₂
39	3-CF ₃	β	O	NHCH ₃	77	E	I-B	158-159	C ₁₆ H ₁₉ F ₃ N ₂ O ₂
40	3-CF ₃	β	O	N(CH ₃) ₂	59	H		146-148 (0.05)	C ₁₇ H ₂₁ F ₃ N ₂ O ₂
41	3-CF ₃	β	S	NHCH ₃	80	J	I-IE	133-135.5	C ₁₆ H ₁₉ F ₃ N ₂ O ₅
42	4-CF ₃	α	O	NH ₂	56	G	IE-EA	173-175	C ₁₅ H ₁₇ F ₃ N ₂ O ₂
43	4-CF ₃	α	O	NHC ₂ H ₅	71	E	IE	140-142	C ₁₇ H ₂₁ F ₃ N ₂ O ₂

^aC analyzed 0.43% high.

Table IV

							
No.	X	R	Yield, %	Method	Recrystn solvent	Mp or bp (mm), °C	Formula ^a
44	3-Cl	CH ₂ C ₆ H ₅	55 (91)	B (A)	IP-IE	123-124	C ₁₇ H ₁₉ Cl ₂ NO ₂
45	3-Cl	H	21	D	IP-IE	95-97 ^b	C ₁₀ H ₁₃ Cl ₂ NO ₂
46	4-Cl	CH ₂ C ₆ H ₅	59	B	IP-IE	158-159 ^c	C ₁₇ H ₁₉ Cl ₂ NO ₂
47	4-Cl	H	7	D	IP	135-138	C ₁₂ H ₁₄ ClNO ₅ ^d
48	4-Br	CH ₂ C ₆ H ₅	45	B	IP-IE	156-158 ^e	C ₁₇ H ₁₉ BrClNO
49	4-Br	H	12	D		144-145	C ₁₀ H ₁₃ BrClNO
50	2-OCH ₃ , 4-COCH ₃	H	7	B	IP	173-175	C ₁₃ H ₁₈ ClNO ₃
51	3,5-(CH ₃) ₂	CH ₂ C ₆ H ₅	24	B	IP-IE	158-160.5 ^f	C ₁₆ H ₂₄ ClNO
52	3,5-(CH ₃) ₂	H	86	F		133-135 ^g	C ₁₂ H ₁₈ ClNO
							
53	H	CH ₂ C ₆ H ₅	31	A	IE-IP	207-209	C ₁₅ H ₂₃ ClNO
54	H	H	22 (84)	B (F)	Not characterized ^h		C ₁₁ H ₁₅ NO ⁱ
55	4-Br	CH ₂ C ₆ H ₅	64	A	E-Et	215-217	C ₁₈ H ₂₁ BrClNO
56	2-OCH ₃	H	10	B, D		109-111 (0.07)	C ₁₂ H ₁₇ NO ₂ ⁱ
57	3-CF ₃	CH ₂ C ₆ H ₅	58	A	IE-IP	192-194	C ₁₉ H ₂₁ ClF ₃ NO
58	3-CF ₃	H	3 (94)	K (F)	IP	196-198	C ₁₂ H ₁₅ ClF ₃ NO
59	4-CF ₃	CH ₂ C ₆ H ₅	94	A	IE-IP	255-257	C ₁₉ H ₂₁ ClF ₃ NO ⁱ
60	4-CF ₃	H	78	F		43-47 ^j	C ₁₂ H ₁₄ F ₃ NO ⁱ

^aAll are HCl salts unless otherwise indicated. ^bBp (free base) 101-103° (0.07). ^cBp (free base) 140-145° (0.04). ^dOxalate salt. ^eBp (free base) 170-175° (0.07). ^fBp (free base) 145-148° (0.05). ^gBp (free base) 93-95° (0.05). ^hNmr and uv spectra are consistent with structure. ⁱFree base. ^jBp (free base) 74-76° (0.05).

Several of the intermediate 3-phenoxy-pyrrolidines and their *N*-benzyl derivatives have been reported previously by some of us.^{2,4} Chemical data for the intermediates not previously reported are given in Tables IV and V. Representative synthetic procedures are given in the Experimental Section.

Pharmacology. Compounds were tested for anticonvulsant activity in adult, female mice (ICR strain) using the methods of Swinyard, *et al.*,⁶ as modified by Helsley, *et al.*³ Prior to challenge by maximal electroshock or by pentylenetetrazole administration, behavioral effects in the animals were recorded. Most compounds which produced loss of righting in subtoxic doses were further evaluated for central muscle relaxant properties in acutely prepared cats.

For these studies, the patellar reflex (monosynaptic) was elicited every 2 sec by means of a solenoid which

pulled on an exposed patellar tendon with the resulting contraction recorded on a Grass polygraph. In the contralateral hind leg, the flexor reflex (polysynaptic) was obtained by electrical stimulation (100 Hz, 0.5-2 V intensity, 2 msec pulse width and 80 msec duration) of the central end of the sectioned tibial nerve, and the contraction of the tibial muscle was recorded on a second channel of the polygraph. In addition, carotid-arterial blood pressure was monitored. To eliminate supraspinal influences, the spinal cord was severed at C₁ and artificial ventilation instituted. Test compounds, dissolved in distilled water or polyethylene glycol-300, were administered slowly into a cephalic vein.

Although acute LD₅₀'s were not determined, most compounds were not lethal in mice in doses up to 200 mg/kg ip. Only compound 30 appeared very toxic, with an approximate LD₅₀ of 42 mg/kg ip. No muscle relaxant or an-

Table V

No.	X	α or β	R	Yield, %	Method	Recrystn solvent	Mp or bp (mm), °C	Formula
61	3-CF ₃	α	CH ₂ C ₆ H ₅	68	A	IP-E	204–206 ^a	C ₂₃ H ₂₄ F ₃ NO ₅ ^b
62	3-CF ₃	α	H	92	F	IP-IE	227–230.5	C ₁₄ H ₁₇ ClF ₃ NO ^c
63	3-CF ₃	β	CH ₂ C ₆ H ₅	66	A	IP-IE	148–150 ^d	C ₂₃ H ₂₄ F ₃ NO ₅ ^b
64	3-CF ₃	β	H	80	F	IP-IE	218–220	C ₁₄ H ₁₇ ClF ₃ NO ^c
65	4-CF ₃	α	CH ₂ C ₆ H ₅	71	A	IP-IE	235–238	C ₂₁ H ₂₃ ClF ₃ NO ^c
66	4-CF ₃	α	H	84	F	IP	283–285	C ₁₄ H ₁₇ ClF ₃ NO ^c

^aBp (free base) 153–156° (0.05). ^bOxalate salt. ^cHCl salt. ^dBp (free base) 156–158° (0.10).

ticonvulsant properties were seen for this compound in lower doses.

Pharmacological test results are summarized in Tables VI–VIII and compared with data for reference compounds (also in Table VIII). Several compounds (3, 9, 21, 26, 27, and 35) showed pronounced muscle relaxant activity in the range of mephenesin as determined by suppression of the flexor reflex. Unlike mephenesin, however, compound 3 did not appear selective, in that a dose-dependent reduction in patellar reflex activity was also recorded. None of the substituted phenoxynortropans appeared to possess muscle relaxant properties except compound 40, while several phenoxypyrrolidine and phenoxypiperidine derivatives were active.

While none of the test compounds appeared superior to diphenylhydantoin in suppressing electroshock-induced convulsions, several (33–39, 42, and 43) had protective ED₅₀'s against pentylenetetrazole lower than that of ethosuximide. With but three exceptions (33–35), all compounds in this latter class contained the substituted phenoxynortropane moiety.

Subsequent studies⁷ with compound 3 indicated that polysynaptic pathways in the spinal cord were selectively suppressed with doses as low as 10 mg/kg iv. This was ascertained by investigations of segmental action potentials in spinal cats. For these experiments, the lumbrosacral region of the spinal cord was exposed by laminectomy and a dorsal and ventral root on the same side of one segment, usually L₇ or S₁, were sectioned.

Stimulation of the dorsal root (0.1 Hz, 1 msec pulse width and 0.2–2 V intensity) produced an initial monosynaptic spike followed by a series of slower, lower amplitude polysynaptic action potentials. Compound 3, at 10 or 20 mg/kg iv, suppressed polysynaptic spike activity while having no effect on the monosynaptic spike, or producing a transient increase of up to 40% in the amplitude of the monosynaptic action potential.

The suppression of the patellar reflex by compound 3 was apparently due to peripheral blockade at, or beyond, the neuromuscular junction. This was realized using an *in situ* peroneal nerve–tibial muscle preparation. In these experiments, compound 3 produced a dose-dependent decrease in the amplitude of the muscle contraction in response to stimulation of the peroneal nerve.

Experimental Section

The procedures given below are representative for the preparation of the compounds listed in Tables I–VI. Yields and physical properties are recorded in the tables. Temperatures are uncorrected. Melting points were taken in a Thomas-Hoover capillary apparatus. All compounds were analyzed for C, H, and N and were within $\pm 0.4\%$ of the theoretical values except where noted.

Procedure A.⁵ 1-Benzyl-3-(*m*-chlorophenoxy)pyrrolidine (44). To a stirring suspension of 11.2 g (0.25 mol) of a 57% mineral oil dispersion of NaH in 200 ml of dry DMF was added a solu-

tion of 30.6 g (0.17 mol) of 1-benzyl-3-pyrrolidinol in 50 ml of dry DMF at a rate so as to maintain the temperature of the reaction mixture at ca. 32–35° and to maintain a steady evolution of H₂. After the addition was complete, the mixture was heated at about 50° until evolution of H₂ ceased. To the reaction mixture 27.6 g (0.213 mol) of *m*-fluorochlorobenzene was added at a rate so as to maintain a temperature of 50–60°. After the addition was complete the reaction mixture was stirred at 60–70° for 3 hr and then at 35° for an additional 12 hr. The mixture was cooled and a large excess of H₂O was added. The mixture was extracted with C₆H₆, the combined extracts were dried (Na₂SO₄), and the solvent was evaporated. The residue was dissolved in C₆H₆ and extracted with 6 N HCl. The hydrochloride separated from solution. The oily hydrochloride and the acid layer were combined and basified with NaOH solution. The aqueous mixture was extracted with C₆H₆, the combined extracts were dried (Na₂SO₄), and the solvent was evaporated. The residue weighed 44.7 g (91% yield). This product was considered pure enough to carry on to the next step (C₁).

Procedure B. 1-Benzyl-3-(*m*-chlorophenoxy)pyrrolidine (44). A stirred mixture of 302 g (1.55 mol) of 1-benzyl-3-chloropyrrolidine, 200 g (1.55 mol) of *m*-chlorophenol, 84 g (1.55 mol) of NaOCH₃, and 1 l. of DMF was heated at 110–113° for 16 hr, cooled, and treated with 1 l. of H₂O. The oil which separated was extracted with C₆H₆ and the combined extracts were washed successively with 10% NaOH solution and H₂O. After the solvent was evaporated the residual oil was distilled at reduced pressure, yielding 232 g (55% yield) of product boiling at 152–155° (0.07 mm).

Procedure C₁. 1-Chlorocarbonyl-3-(*m*-chlorophenoxy)pyrrolidine. Into a 500-ml three-necked flask containing 200 ml of C₆H₆ was bubbled 36.2 g (3.66 mol) of COCl₂. Under anhydrous conditions 84.1 g (0.294 mol) of 1-benzyl-3-(*m*-chlorophenoxy)pyrrolidine was added over a period of 1–2 hr. The temperature of the reaction was maintained between 20 and 25° using an ice bath. After the addition was completed the reaction mixture was stirred at room temperature for 16 hr. The reaction mixture was filtered and the filtrate concentrated under reduced pressure. The residue, a dark brown oil, weighed 93.7 g and the nmr showed the presence of about 0.5 equiv of benzyl chloride. The crude product was triturated with petroleum ether (bp 30–60°) and the petroleum ether decanted away from the insoluble oily carbamoyl chloride. The residual oil obtained weighed 75.4 g and the nmr indicated 0.25 equiv of benzyl chloride. The yield was theoretical.

Procedure C₂. 3-(*m*-Chlorophenoxy)-1-(*N*-methylcarbamoyl)-pyrrolidine (3). A mixture of 60 ml of THF and 40 ml of a 40% solution of CH₃NH₂ in H₂O was stirred and cooled to –5 to 0°. To the stirring mixture was added 49.3 g (0.143 mol) of 1-chlorocarbonyl-3-(*m*-chlorophenoxy)pyrrolidine at a rate so as to maintain the temperature at 0°. When the addition was completed, the reaction mixture was allowed to come to room temperature while stirring overnight. About 200 ml of H₂O was added to the reaction mixture and after stirring for 0.5 hr the reaction mixture was filtered. The solid residue was triturated in 40 ml of *i*-Pr₂O, filtered, and dried to give 30.7 g (89.5% yield) of crude product. The crude product was dissolved in 120 ml of CHCl₃ and washed through 10 g of Florisil in a fritted glass funnel under vacuum in 40-ml portions to remove an impurity. The collected CHCl₃ solution was evaporated and the solid residue triturated in *i*-Pr₂O, collected by filtration, and dried to give 21.6 g (59.5% yield) of off-white solid product.

Procedure D₁. 1-Carbamoyl-3-(*m*-chlorophenoxy)pyrrolidine (2). Over a period of 4 hr 204 g (0.70 mol) of 1-benzyl-3-(*m*-chloro-

Table VI. Pharmacological Data

No.	Electroshock			Anticonvulsant		Pentylentetrazole		Loss of righting (LRR)		Muscle relaxant activity		Spinal cat	
	Dose, mg/kg ip	% protected	ED ₅₀ (95% conf limits), mg/kg ip	Dose, mg/kg ip	% protected	ED ₅₀ (95% conf limits), mg/kg ip	Dose, mg/kg ip	% LRR	Lethal in 40%	ED ₅₀ (95% conf limits), mg/kg ip	Dose, mg/kg iv	% decrease flexor	% decrease patellar
1			50 (36.4-68.5)	100 150	0 100		100 150 200	0 100 Lethal in 40%					
2			66 (37.2-98.8)			111 (82.3-150)				134 (96.1-194.5)	5	60	10
3			54 (30.0-97.2)			88 (49.1-150)				113 (82.2-165.0)	10 20	100 100	25 40
4			35 (23.2-53.2)	67 100	0 67		67 100	0 20					
5			55 (37.8-83.3)	45 67	0 60		45 67 100	0 0 Lethal in 60%					
6			46 (35.1-60.3)	67 100	30 45		67 100 150	0 25 Lethal in 30%					
7			150 (62.8-360.1)	67 100 150	20 20 60		67 100 150	0 0 0					
8			35 (22.9-53.0)			89 (66.3-121)				136 (101.0-182.5)	10	0	0
9			80 (47.1-135.7)	100 200	0 80		50 100 200	0 30 100			5	90 90	40 0
10			47 (24.1-89.5)	100 200 100 200	0 80 0 40		100 200 50 100 200	0 10 100 80 100			20 5	90 30	0 0
11	50 100	40 80									10	40	36
12	50 100	0 20		100 200 100 200	0 20 20 40		100 200 50 100 200	0 0 0 10 40			25	0	0
13			109 (60.0-197.3)										
14	50 100	20 20		100	0		50	0					
15	100 200	60 100		100 200	0 0		100 200	0 0					
16	50 100	60 100		100	20		50 100 200	0 Lethal in 60%			10 ^a	0	0

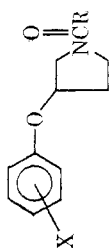
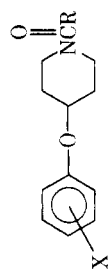


Table VII. Pharmacological Data

No.	Electroshock				Anticonvulsant		Pentylenetetrazole			Muscle relaxant activity			
	Dose, mg/kg		ED ₅₀ (95% conf limits), mg/kg		Dose, mg/kg	ip	Dose, mg/kg		ED ₅₀ (95% conf limits), mg/kg	Loss of righting (LRR)		Dose, mg/kg	Spinal cat
	ip	% protected	ip	ip			% protected	ip		% LRR	ED ₅₀ (95% conf limits), mg/kg	iv	% decrease flexor patellar
24	50 100	0 0		100 200			20 100			0 0 60			
25	50 100	20 0		100 200			0 0			0 0 0			
26			69 (47.6-101.0)								109 (72.7-164.1)	10	45
27			29 (15.6-52.8)								71 (49.6-100.1)	10	40
28			47 (34.0-65.3)								138 (109.1-173.6)	25	30
29	50 100	0 40		100 200			40 60	88 (55.4-137.3) 97 (63.4-148.0) 89 (62.1-120.1)		0 0 0			0
30	22 33	20 40		22			0						
31	50 100	0 40		100 200			0 40						
32	50 100	0 20		100 200			20 0						
33			32 (20.4-51.1)					37 (18.5-77.9)					
34			36 (21.0-62.0)					20 (10.5-38.0)					
35			28 (19.3-40.6)					64 (44.4-92.2)				20	50
												100	0



References

- (1) R. L. Duncan, Jr., G. C. Helsley, W. J. Welstead, Jr., J. P. DaVanzo, W. H. Funderburk, and C. D. Lunsford, *J. Med. Chem.*, **13**, 1 (1970).
- (2) W. J. Welstead, Jr., G. C. Helsley, R. L. Duncan, Jr., A. D. Cale, Jr., C. R. Taylor, J. P. DaVanzo, B. V. Franko, and C. D. Lunsford, *J. Med. Chem.*, **12**, 435 (1969).
- (3) G. C. Helsley, R. L. Duncan, Jr., W. H. Funderburk, and D. N. Johnson, *J. Med. Chem.*, **12**, 1098 (1969).
- (4) W. J. Welstead, Jr., J. P. DaVanzo, G. C. Helsley, C. D. Lunsford, and C. R. Taylor, Jr., *J. Med. Chem.*, **10**, 1015 (1967).
- (5) G. C. Helsley and R. F. Boswell, Jr., U. S. Patent 3,657,253 (1972).
- (6) E. A. Swinyard, W. C. Brown, and L. S. Goodman, *J. Pharmacol. Exp. Ther.*, **106**, 319 (1952).
- (7) D. N. Johnson, W. H. Funderburk, A. E. Hakala, and J. W. Ward, *Fed. Proc., Fed. Amer. Soc. Exp. Biol.*, **31**, 535 (1972).

Notes

Antihypertensive Activity of 1-Dimethylphosphinylmethyl-4-arylpiperazines

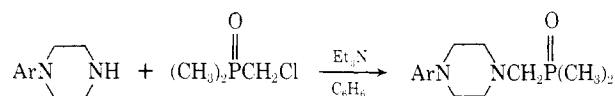
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1-Alkyl-4-phenylpiperazines have been shown to possess potent antihypertensive activity.¹ Our interest in phosphorus-containing molecules with pharmacological activity² has prompted the preparation of a series of arylpiperazines bearing an *N*-dimethylphosphinylmethyl moiety [$-\text{CH}_2\text{P}(\text{O})(\text{CH}_3)_2$] as potential antihypertensive agents.

The compounds were synthesized by the alkylation reaction shown below.



The structures and the physical and antihypertensive data for these novel 1-dimethylphosphinylmethyl-4-arylpiperazines are recorded in Table I.

Pharmacological Results. The compounds of Table I were initially screened for antihypertensive activity using spontaneous hypertensive rats (SHR) by a standard indirect tail-cuff method.³ In a standard 3-day test, systolic blood pressure readings were made at 0 time (control) on days 1 and 3, and at 2 hr after administration of the compound on days 1 and 3. Dosing was orally at 100 mg/kg at 0 hr on days 1, 2, and 3 on groups of six animals per test. Activity was determined by comparison of the treatment blood pressure values with the 0 time (control) blood pressure readings. Comparisons were made using the paired *t* test method for evaluation of statistical significance.⁴ A value of -15 mm or more is considered significant.

From Table I it will be seen that compounds where Ar = phenyl or substituted phenyl are active. Inserting a heteroatom in the Ar ring (2-pyridyl, compound 8) or separating the Ar group from the piperazine ring by a methylene bridge ($\text{C}_6\text{H}_5\text{CH}_2$, compound 9) abolished activity. Compounds 1, 4, and 7 showed especially marked reductions in blood pressure in the SHR screen and were there-

Table I

Compd	Ar	Yield, % ^a	Mp, °C	Recrystn solvent ^b	Formula ^c	Antihypertensive act. (Δ mm) ^d	
						Day 1	Day 3
1	C ₆ H ₅	79	149–151	A	C ₁₃ H ₂₁ N ₂ OP	–25	–43
2	2-ClC ₆ H ₄	88	222–223 dec	B	C ₁₃ H ₂₀ ClN ₂ OP · HCl	–12	–16
3	3-ClC ₆ H ₄	34	108–110	C	C ₁₃ H ₂₀ ClN ₂ OP	±	±
4	4-ClC ₆ H ₄	55	184–187	B	C ₁₃ H ₂₀ ClN ₂ OP	–10	–54
5	2-CH ₃ C ₆ H ₄	77	114–117	D	C ₁₄ H ₂₃ N ₂ OP	–20	–14
6	4-CH ₃ COC ₆ H ₄	82	170–172	A	C ₁₃ H ₂₃ N ₂ O ₂ P	±	±
7	3-CF ₃ C ₆ H ₄	74	217–218 dec	E	C ₁₄ H ₂₀ F ₃ N ₂ OP · 2HCl	–87	–66
8	2-C ₅ H ₄ N	61	116–120	A	C ₁₂ H ₂₀ N ₃ OP	–6	+1
9	C ₆ H ₅ CH ₂	34	110–113	C	C ₁₄ H ₂₃ N ₂ OP	–2	+3

^aIsolated yield of crude solid, fairly pure by melting point and tlc. ^bA = acetone, B = EtOH, C = cyclohexane, D = A + hexane, E = MeOH + Et₂O. ^cAll compounds were analyzed for C, H, and N within $\pm 0.4\%$ of the theoretical values. ^dThe experimental procedures are described in the text; \pm indicates marginal or transient activity.