Chemoselective Arylation of 2'-Deoxyguanosine at N-2 with Organoiron Complexes

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2'-Deoxyguanosine has been directly and chemoselectively arylated at the N^2 -amino function by treatment with tricarbonyl(η^5 -cyclohexadienylium)iron or tricarbonyl[η^5 -2-(n-butyl)cyclohexadienylium]iron cations, followed by decomplexation and dehydrogenation, providing a rapid route to N^2 -aryl-2'-deoxyguanosine nucleosides.

 N^2 -Aryl-2'-deoxyguanosine nucleosides have been found to be inhibitors of mammalian DNA polymerases¹ and viral thymidine kinases.² It is the potential of their 5'-diphosphates to inhibit DNA synthesis *via* interaction at the negative allosteric effector site of ribonucleotide reductase³ that attracted us to these compounds.⁴

The reported classical synthetic route¹ is a lengthy multistep process from primitive purine precursors that requires demanding control of both regioselectivity ($N^7 vs. N^9$) and stereoselectivity ($\alpha vs. \beta$ anomers) in the sugar coupling reaction as well as the need for masked functionality in the purine base and protecting groups on the ribose.

We report here a rapid route to these important compounds employing an organometallic aryl cation equivalent with 2'-deoxyguanosine itself as starting material, that avoids the use of protecting groups despite the multifunctional and highly polar nature of the nucleoside (Scheme 1).

In the crucial transformation, reaction of 2'-deoxyguanosine 1 with the electrophilic phenyl cation equivalent^{5.6} tricarbonyl(η^5 -cyclohexadienylium)iron tetrafluoroborate 2a (1.1 equiv.) in acetonitrile containing 2,6-di-*tert*-butyl-4-methylpyridine (1.1 equiv.) gave a respectable 37% yield (69% based on unrecovered 1) of the complex 3a‡ as pale-yellow crystals [m.p. 184 °C (decomp.)] in which the N^2 -H appears in the 1 H NMR spectrum at δ 6.46 as a doublet (J = 7.2 Hz) owing to coupling with the adjacent cyclohexadiene 1-H proton.

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[‡] New compounds gave satisfactory elemental analysis and spectral data; all of the nucleoside derivatives were obtained as hydrates.

Scheme 1 Reagents and conditions: i, 2,6-di-tert-butyl-4-methylpyridine, MeCN, 20°C; ii, Me₃NO, MeCONMe₂, 90°C; iii, (MeCN)₂PdCl₂, Et₃N, MeCONMe₂, 80 °C

The chemoselectivity for the N^2 -amino group, despite its very weak nucleophilicity, has been achieved by virtue of the soft, reasonably hindered, yet strong electrophilic character⁷ of the organoiron complex 2, which reacts preferentially with the softest unhindered nucleophile available. In this respect it is noteworthy that attempts to arylate 1 using tricarbonyl $(\eta^4$ -cyclohexa-2,4-dienone)iron or tricarbonyl $(\eta^5$ -1-ethoxycyclohexadienylium)iron tetrafluoroborate8 were unsuccessful, attributed to the insufficient electrophilicity of these complexes. The non-nucleophilic base used was necessary to prevent reversal of the reaction and to protect the acid-labile glycosidic bond from cleavage.

Oxidative decomplexation to the arene proved troublesome owing to sensitivity of both the ribose and purine groups to oxidation. Two steps were necessary. First, treatment with an excess of trimethylamine N-oxide (10 equiv.) in dimethylacetamide⁶ gave the diene $4a\ddagger$ as a white powder in 60% yield [m.p. 138 °C (decomp.)]. This was dehydrogenated with bis(acetonitrile)palladium(II) chloride9 (1 equiv.) in dimethylacetamide at 80 °C to give N²-phenyl-2'-deoxyguanosine 5a in 28% yield as white crystals, m.p. 230 °C (decomp.), (lit.2 m.p. 228-231 °C).

The greatest activity against mammalian enzymes assessed to date 1,3 has been observed not for the N^2 -phenylnucleoside but for the N^2 -[4-(n-butyl)phenyl] derivative. The appropriate cationic organoiron complex 2b was prepared (Scheme 2) from the commercially available tricarbonyl(2-methoxycyclohexadienylium)iron complex 6 so as to provide the necessary regiochemistry. Treatment of 6 with the low-order butylcopper(dimethyl sulfide) complex¹⁰ gave the butylated iron diene in over 90% yield as predominantly the required C-5 regioisomer (C-5:C-1 = 60:1), and the pure C-5 regioisomer 711 was obtained following chromatography in

Scheme 2 Reagents and conditions: i, BuⁿCu(Me₂S), tetrahydrofuran (THF); ii, H₂SO₄ then NH₄PF₆

86% isolated yield. On treatment of 7 with sulfuric acid diene relocation necessarily preceded methoxy elimination¹² to give, in 94% yield, the requisite 4-(n-butyl)phenyl cation equivalent tricarbonyl[η⁵-2-(n-butyl)cyclohexadienylium]iron hexafluorophosphate 2b‡ exclusively as the 2-butyl regioisomer [m.p. 190°C (decomp.)]. It is clear that other para-substituents could be incorporated using the appropriate low-order organocopper complex. 10

Reaction of 2b under the conditions described above gave **3b**‡ as yellow crystals, m.p. 128–130 °C [δ(dimethyl sulfoxide) of N²-H doublet = 6.36, J 7.4 Hz]. The lower yield of 24% obtained using 2b (46% based on unrecovered 2'-deoxyguanosine) is due presumably to the electron releasing inductive effect of the alkyl substituent decreasing the electrophilicity of this complex compared with 2a. Decomplexation (42% yield) and dehydrogenation (26% yield) afforded the target compound N^2 -[4-(n-butyl)phenyl]-2'-deoxyguanosine **5b** as white crystals, m.p. 193–196 °C (lit. 1 m.p. 196–197 °C).

Thus, N^2 -arylated nucleosides have been reached directly from the parent 2'-deoxyguanosine by a three-step sequence performed entirely on the unprotected nucleoside. Although yields remain to be improved, the overall yield compares very favourably with the multistep approaches used to date. The intermediate dienes 4 are not only precursors to the aromatic targets, but should also provide versatility for transformation into several unusual nucleoside analogues.

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