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Stereoselective synthesis of O-serinyl/threoninyl-2-acetamido-2-deoxy- α - or β -glycosides

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Abstract

General glycosidation methodology has been developed which can selectively provide 2acetamido-2-deoxy- α - or β -glycosides of β -hydroxy- α -amino acid derivatives [glucopyranoside-(8, 43), galactopyranoside- (9, 13), mannopyranoside- (10), lactoside analogs (11, 38) and 3-O- β -galactopyranosyl-mannopyranoside (12)] stereoselectively in excellent yield from the highly nucleophilic α -imino esters (Schiff bases) of L-serine and L-threonine. Various glycosides were converted via their amino and acetamido derivatives to Fmoc-protected serinyl- or threoninyl-glycosides (24–28, 37, 41, 46) which are all suitable building blocks for the solid-phase synthesis of O-glycopeptides. Complete ¹H- and ¹³C-NMR data are provided for all compounds.

Keywords: 2-Acetamido-2-deoxy-glycosides; Stereoselective synthesis

1. Introduction

In recent years there has been increasing interest in glycoproteins, due to the central role of their carbohydrate moieties in different types of cellular recognition processes [1], in intercellular and intracellular transport of the gene products [2], in the alteration of peptide backbone conformation [3], and in the control of membrane permeability and molecular recognition [4]. Additional roles are involved in numerous disease states as the modification of the τ protein in Alzheimer's disease [5] and the antigenic T- and T_N-epitopes associated with cancer [6]. The significance of *O*-glycopeptides might be further supported by the fact that L-serinyl- β -D-glucoside enkephalin analogs are transported across the blood-brain barrier [7].

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Despite the large number of naturally occurring glycoproteins, the types of covalent bonds between the protein and the saccharide moiety show limited variation. One of the most common is the O-linked 2-acetamido-2-deoxy-glycoside [8]. The synthesis of the O-glycoproteins is complicated by both acid lability of the glycosidic bond, and base sensitivity of the O-serinyl and O-threoninyl glycosides [9]. Additional complications arise due to the poor reactivity in Koenigs-Knorr reactions of the typical N-acylated (Boc, Cbz or Fmoc-protected) serine or threonine derivatives. Because of this poor reactivity, harsh reaction conditions are required to effect bond formation, and the yields suffer, as well as the anomer selectivity [9,10].

The poor reactivity of acyl-protected (cf. Boc, Z, etc.) serines and threonines or their analogs (cf. ceramides) is probably due to the unfavorable H-bonding pattern which arises from amide-type protecting groups. Replacement of the H-bond donor (amide) with an H-bond acceptor (imine), inverts the H-bonding pattern, and increases the nucleophilicity of the O-lone pair (Scheme 1). We have confirmed this H-bonding hypothesis by the synthesis of cerebrosides [11], O-serinyl and threoninyl α -or β -glycopyranosides, glycoconjugates and glycopeptides [12], as well as the related 2-deoxy-glucopyranosides [13]. We now report general synthetic methodology for Fmoc-protected O-serinyl/threoninyl 2-acetamido-2-deoxy- α - or β -glycopyranosides which are desirable building blocks for solid-phase O-linked glycopeptide synthesis [14].

2. Results and discussion

The crucial step in the 9-fluorenylmethoxycarbonyl (Fmoc)-protected serinate- or threoninate-2-acetamido-2-deoxy-glycoside derivatives is glycosylation. For successful reaction, the highly nucleophilic α -imino esters (O'Donnell's Schiff bases [15]), benzyl *N*-(diphenylmethylene)-L-serinate (6) [12], and diphenylmethyl *N*-(diphenylmethylene)-L-threoninate (7) [12] were chosen as the glycosyl acceptors (Scheme 2).



unfavorable H-bond



favorable H-bond

Scheme 1.





Chamical	chifte (<u></u>										· • •	
Number	8 ^b	թրու <i>)</i> 9	10 ^b	13	14	15	18	19	20	23	30	43	46
 H-1	4 58	5.03	4 56	5.27	4 97	4 51	4 81	4 83	4 32	4 66	4 81	4 66	4 51
H-2	2.77	3.63	3.50	3.65	3.30	3.57	3.49	3.25	3.43	3 39	4.50	3 53	3.83
H-3	5.63	5.24	5.56	5.26	5.43	5.47	5.22	5.40	5.44	5.21	5.04	5.06	5.08
H-4	5.12	5.32	5.71	5.38	5.04	5.61	5.38	4.98	5.58	5.38	5.26	4.94	5.01
H-5	3.81	4.00	3.88	4.22	4.04	3.98	4.18	3.94	3.95	4.26	4.00	3.54	3.68
H-6	4.26	4.01	4.30	4.05	4.26	4.27	4.03	4.18	4.26	4.43	4.00	4.13	4.19
H-6'	3.94	3.99	4.10	4.05	4.08	4.19	4.03	4.07	4.26	4.37	4.00	3.98	3.99
α-H	4.68	4.44	4.69	4.25	3.75	3.32	3.50	4.61	4.55	4.53	4.36	4.67	5.16
β-н	4.22	4.21	4.35	4.46	4.01	3.68	4.23	4.17	3.66	4.48	4.17	4.28	4.01
β'-Н	3.96	3.94	3.91		3.87	3.46		4.01	3.66		3.81	4.28	3.69
CH ₂ Ph	4.98	5.16	5.07		5.20	5.03		5.24	4.93		5.16	5.06	
CHPh ₂				6.88			6.96			6.97			
CH ₃				1.17			1.34			1.34			
NH								5.91	6.06	5.74	5.59	5.21	5.94
Fmoc ₁								4.23	4.10	4.12			4.20
Fmoc ₂₋₃								4.41	4.42	4.05			4.68
20										4.03			4.63
CH ₂												4.02	

Table 1 ¹H-NMR data for glycosides ^a

 a NMR data in CDCl_3 solution unless otherwise indicated. b Solution was $C_{6}D_{6}.$

Table 2			
¹ H-NMR	data	for	glycosides

Couplin	g constat	nts (H	[z)										
Ј _{Н,Н}	8	9	10	13	14	15	18	19	20	23	30	43	46
1,2	3.5	3.5	1.4	3.6	3.6	1.4	3.7	3.4	0.9	3.7	3.6	8.5	8.4
2,3	10.6	9.0	3.8	11.1	10.6	3.8	11.1	10.6	3.7	11.1	11.2	10.0	9.3
3,4	9.5	3.3	9.9	3.2	9.3	9.9	3.2	9.8	9.9	2.4	3.2	10.1	9.2
4,5	10.3	1.2	9.9	nd	10.2	9.9	0.9	10.3	9.8	nd	nd	9.8	9.4
5,6	4.3	5.9	4.5	6.8	4.6	4.4	6.1	5.2	3.5	7.6	nd	4.5	4.2
5,6'	2.2	nd	2.1	6.8	2.2	2.2	6.1	1.6	3.5	7.3	nd	2.2	2.0
6,6′	- 12.6	nd	- 12.3	nd	- 12.5	-12.3	nd	-12.3	nd	- 10.5	nd	-12.3	-12.2
α,β	4.3	4.5	4.6	7.6	3.7	4.5	3.0	2.5	4.3	1.9	5.2	4.6	5.9
α, β'	8.2	7.7	8.0		5.0	4.9		2.5	3.9		7.0	5.5	5.9
β, β'	9.7	9.8	9.5		9.9	9.6		10.9	11.5		9.8	10.4	11.0
β ,CH ₃				6.3			6.5			6.5			
NH-α								8.1	8.2	9.5			5.9
NH-2											9.8	8.6	9.1
Fmoc ₁₋₂	1							7.1	6.5	5.9			5.5
Fmoc ₁₋₂	.'							7.1	6.5	7.1			5.4
Fmoc ₂₋₂	2'							nd	nd	11.2			nd

¹ H-NMR o	lata for gly	cosides			
Chemical s	hifts (ppm))			
Number	11	12	16	17	21
H-1	4.98	4.80	4.92	4.68	4.79
H-2	3.21	3.89	3.12	3.91	3.08
H-3	5.38	5.32	5.40	5.31	5.40
H-4	3.68	3.94	3.71	3.92	3.69
H-5	3.83	3.75	3.95	3.85	3.90
H-6a	4.38	4.37	4.45	4.40	4.48
TT (1		1 4 0	1 1 0	4.4.4	4 4 7

Table 3

Number	11 11	12	16	17	21	22	38	39	40
H-1	4.98	4.80	4.92	4.68	4.79	4.59	4.43	4.37	4.31
H-2	3.21	3.89	3.12	3.91	3.08	3.66	3.33	3.37	3.35
H-3	5.38	5.32	5.40	5.31	5.40	5.31	4.95	4.96	4.96
H-4	3.68	3.94	3.71	3.92	3.69	3.88	3.67	3.69	3.67
H-5	3.83	3.75	3.95	3.85	3.90	3.79	3.45	3.51	3.46
H-6a	4.38	4.37	4.45	4.40	4.48	4.41	4.34	4.46	4.45
H-6b	4.11	4.10	4.13	4.11	4.17	4.07	4.04	4.08	4.07
H-1′	4.45	4.51	4.47	4.52	4.50	4.55	4.41	4.45	4.44
H-2′	5.12	5.14	5.10	5.12	5.13	5.13	5.07	5.08	5.09
H-3'	4.95	4.97	4.95	4.98	4.97	4.99	4.93	4.95	4.95
H-4′	5.35	5.37	5.34	5.36	5.36	5.37	5.34	5.35	5.35
H-5'	3.84	3.90	3.87	3.90	3.86	3.92	3.84	3.87	3.87
H-6'a	4.17	4.18	4.19	4.18	4.21	4.19	4.15	4.16	4.17
H-6'b	4.08	4.09	4.08	4.09	4.06	4.10	4.07	4.08	4.08
<i>α</i> -Η	4.44	4.40	3.83	3.72	4.62	4.61	4.45	3.72	4.62
β- Η	4.23	4.18	4.00	3.75	4.08	4.37	4.17	4.17	4.36
β'-Н	3.94	3.87	3.73	3.71	3.98	4.01	4.05	3.86	3.89
CH ₂ Ph	5.16	5.15	5.18	5.19		5.11	5.15	5.18	5.22
Fmoc ₁					4.25	4.26			4.23
Fmoc ₂					4.25	4.26			4.38

Table 4 ¹H-NMR data for glycosides

Coupling	constants	(Hz)							
J _{H,H}	11	12	16	17	21	22	38	39	40
1,2	3.6	1.5	3.5	1.6	3.6	2.3	8.0	8.0	8.0
2,3	10.7	3.9	10.6	3.8	10.7	3.7	10.2	10.3	9.5
3,4	9.0	9.1	9.4	8.6	9.8	8.6	10.0	9.9	9.6
4,5	9.6	9.5	9.4	9.1	9.5	9.3	9.7	9.6	9.6
5,6a	2.1	1.8	2.4	1.6	2.2	2.3	2.8	1.8	1.6
5,6b	4.8	4.9	4.7	5.4	4.2	4.7	4.9	5.0	4.7
6a,6b	-12.1	-11.9	- 12.1	- 12.0	-11.8	- 11.9	- 12.0	- 12.1	-12.4
1',2'	7.9	7.9	7.9	8.0	7.8	8.0	7.9	7.7	7.9
2',3'	10.4	9.3	10.2	10.4	10.4	10.5	10.4	10.3	10.4
3',4'	3.5	3.2	3.5	3.4	3.4	3.4	3.5	3.3	3.5
4',5'	0.9	0.7	nd	nd	nd	nd	1.3	nd	nd
5',6'a	6.2	6.6	6.0	6.6	6.0	6.6	6.5	6.2	6.3
5′,6′b	7.6	7.0	7.8	7.7	5.9	6.8	7.3	6.4	7.2
6'a,6'b	11.2	11.1	11.0	11.2	10.9	11.1	11.1	11.1	11.2
α,β	4.4	4.2	3.5	3.0		2.9	7.0	4.6	2.9
α,β'	8.1	8.5	4.9	5.1		3.0	6.2	4.5	3.0
β, β'	9.8	9.9	9.8	12.3		10.6	8.1	9.9	8.3
β ,CH ₃					_	8.3			8.3

For the synthesis of α -O-glycosides, per-O-acetyl 2-azido-2-deoxy-glycopyranosyl halides were selected as glycosyl donors [16], including non-participating groups at C-2. Thus, the Schiff bases 6 and 7 were treated with various acyl-protected 2-azido-glycosylbromides (1 [17], 2 [18], 3 [17], 4 [19], 5 [19]) in the presence of silver perchlorate [20] and 2,4,6-trimethylpyridine using dichloromethane as a solvent at room temperature to provide the desired α -glycosides 8-13 in excellent yield (73-87%). The glycosides were purified on SiO₂ by flash chromatography. The corresponding β -anomer could not be detected by either thin-layer chromatography or 250 MHz ¹H-NMR. The α -Othreoninyl-glycoside 7 was less reactive due to the secondary OH group. After azidonitration of the glucal [18], the two products, the gluco and the manno analogs, were not separated, but directly converted [17] to the corresponding bromides 1 and 3, and reacted with the serine Schiff base 6. The products, 8 and 10, were easily separated by column chromatography to give the pure glycosides. All of the ¹H- and ¹³C-NMR data (Tables 1-6) were consistent with the expected structures, and most of the chemical shift assignments were made by COSY or HETCOR methods. The anomeric ratios and assignments were made via ${}^{1}H^{-1}H$ spin-coupling constants (${}^{3}J_{H1,H2}$ 3–4 Hz for α -anomers). ¹³C-NMR chemical shifts, and proton-carbon one-bond coupling constants (for α -anomers, ${}^{1}J_{C1,H1} \sim 170$ Hz [21]) were also recorded. The last data confirmed the manno configurations as well.

The glycosylation of serinyl Schiff base 6 with hexa-O-acetyl-2-azido-2-deoxylactosyl bromide 4 [19] (Scheme 3) using Hanessian's modification of the Koenigs-Knorr reaction [22] provided a mixture (6:1) of the α -lactoside 11, and the 2-azido-2-deoxy- β lactoside 38 which were separable by flash chromatography on SiO₂.

In our first approach to the synthesis of Fmoc protected glycosides, the azido group was converted to the acetamido derivative by the Staudinger reaction with tri-



Table 5 ¹³ C-NMR sh	ifts in nn	m for olv	cosides ^a											
Number	8	6	10 ^b	11	12	13	14	15	16	17	18	19	20	21
C-1	97.46	98.10	97.80	97.58	66.79	98.45	98.26	98.05	98.25	97.70	99.05	99.02	98.67	00.66
C-2	61.10	57.57	61.30	61.11	61.27	57.80	60.55	60.79	60.75	60.69	57.74	60.76	60.94	60.80
C-3	70.36	67.57	70.97	69.81	70.96	67.13	69.88	70.49	69.56	70.39	67.44	69.97	70.59	69.52
0.4 1	67.79	66.77	65.65	76.09	73.76	66.26	67.58	65.47	75.91	73.44	66.71	67.22	65.67	76.09
C-5	68.22	68.10	68.80	68.42	69.21	68.11	68.05	68.38	68.42	68.90	68.41	68.14	68.87	68.72
C-6	61.25	61.63	61.45	61.61	62.01	61.29	61.43	61.70	61.54	61.66	61.71	61.65	61.88	61.74
C-1′				100.77	101.17				100.67	100.65				100.72
C-2'				68.87	69.01				68.82	68.64				68.88
C-3'				70.88	70.96				70.67	70.39				70.70
C-4′				66.42	66.71				66.35	66.47				66.39
C-5'				70.38	70.56				70.32	70.13				70.39
C-6'				60.56	61.01				60.53	60.81				60.54
C=N	172.68	172.82	172.12	173.15	173.04	171.56								
a-C	65.20	65.22	65.27	64.94	65.01	70.99	54.38	54.09	54.36	53.96	59.91	54.48	54,18	54.34
B-C	68.85	69.07	68.25	68.70	68.55	76.75 °	70.71	66.69	70.55	69.84	78.14 °	69.85	69.24	69.63
CH, Ph	66.25	66.87	66.35	66.67	66.76		66.94	66.73	66.87	66.38		68.29	66.91	67.08
CHPh_2						76.70 °					77.73 °			
СН, -						17.89					18.66			
CH ₂ Fmoc												68.06	67.20	67.52
CHFmoc												47.00	46.94	46.81
$J_{\rm CI-H1}$		171.5	173.9	172.8		174.1				172.0				
$J_{\rm Cl'-Hl'}$				160.1						158.3				
^a MNR data	in CDCI	³ solution	unless oth	erwise indi	cated.		ŗ							

^b Solution was $C_6 D_6$. ^c Assignments in the same column may be reversed.

Table 6	ļ													
Number	22	23	2	25	26 ª	27	28 ^a	30	34 b	37	38	39	40	41
<u>C</u> 1	98.67	98.38	98.2	98.7	97.72	99.58	98.78	98.30	66.76	98.55	100.70	100.52	100.62	100.49
C-2	60.73	57.53	51.9	53.1	50.88	49.97	46.76	47.58	47.20	47.03	63.86	63.66	63.64	53.59
C-3	70.70	67.23	72.5	68.5	69.61	70.77	67.23	67.04	66.72	65.08	71.76	71.40	71.39	71.49
C-4	74.05	66.78	67.1	62.4	76.43	74.55	66.44	67.37	67.61	67.19	75.67	75.55	75.61	76.46
C-5	69.41	68.21	6.69	67.2	68.22	69.10	67.67	68.42	68.05	68.08	72.41	72.35	72.47	72.38
C-6	61.96	61.56	61.7	60.9	62.00	62.52	61.69	61.90	62.11	61.88	61.61	61.53	61.60	62.09
C-1′	101.07				100.08	100.84					101.43	101.46	101.53	101.51
C-2'	68.89				68.99	68.91					68.83	68.73	68.83	68.80
C-3′	70.70				70.73	70.37					70.76	70.61	70.69	70.62
C-4′	66.64				67.63	66.60					66.44	66.38	66.46	66.44
C-5'	70.50				70.37	69.95					70.44	70.35	70.46	70.28
C-6′	60.98				60.83	60.80					60.70	60.70	60.80	60.67
C=N											172.26			
a-C	54.14	58.66	54.3	54.1	54.31	54.06	58.45	65.13	52.84	54.33	65.15	54.33	53.92	53.59
β-C	69.26	76.09 °	70.5	66.2	65.76	69.14	74.92	69.16	65.42	69.33	70.42	71.66	69.50	68.31
CH ₂ Ph	67.12							66.98			66.63	66.57	67.00	
CHPh ₂		78.39 °												
CH ₃		18.47					18.55							
CH ₂ Fmoc	67.26	67.44	67.9	60.0	67.05	67.40	68.46			67.19			67.14	65.84
CHFmoc	46.79	46.95	47.1	46.9	46.61	46.95	46.55			47.99			46.76	46.68
J _{С1-Н1}									172.0		160.5		159.3	
J _{Cl'-Hl}											161.5		158.3	
^a Solution w	as Me ₂ SC	Ċ							i i					

^b Solution was $D_2\tilde{O}$. ^c Assignments in the same column may be reversed.



phenylphosphine and acetic anhydride [23]. Since the Schiff base moiety is acid sensitive, the triphenylphosphineimine was cleaved in 1 M ammonium acetate buffer (Scheme 4) to give the 2-acetamido-2-deoxy analogues 29-32. In the next step, the Schiff base and the benzyl ester were reduced with Pd/C under an H₂ atmosphere, followed by Fmoc acylation in presence of sodium bicarbonate to provide the desired Fmoc acetamido-glycosides 24, 26, 27 and 37 [27]. The intermediate glycoside esters were all purified and fully characterized, except for the azido glucoside, which was converted in a one pot reaction to 24. This route was marginally satisfactory, but required long reaction times (several days), and the separation of the phosphine oxide was difficult. Thus, we developed a second, more efficient route to the Fmoc amino acid glycosides.

Following the improved route, the Schiff base was cleaved within 5–10 min using 10% trifluoroacetic acid to provide the amino glycosides 14–18 and 39 (Scheme 2). Tetrahydrofuran proved to be a better solvent than dichloromethane for this cleavage reaction. The free amino group was converted to the Fmoc-derivatives 19–23 and 40 in organic solvent in the presence of diisopropylethylamine or triethylamine within 15–30 min. The fluorenylmethoxycarbonyl group was very stable (despite expected cleavage [24]) during the hydrogenation of azido moiety of 19–23 and 40. The amines were converted to the acetamido derivatives 24–28 [27] and 41 by acetic anhydride in the presence of an organic base. The above four reaction steps were nearly quantitative, and



required only 4-5 h. The last two reactions were followed by TLC strictly to avoid the possible migration of the Fmoc group observed after longer reaction times. In one experiment the protected threoninyl-galactopyranoside 13 was transformed to the desired Fmoc-analog 28 without any intermediate chromatographic purification. The target compound 28 was isolated on a short column to give a 92% yield for the four steps. Thus, this method appears more practical than the Staudinger route.

For the synthesis of 1,2 *trans-O*-glycosides (β -anomer in the gluco series) the (2,2,2-trichloroethoxy)carbonyl (Troc) participating protecting group was used on the glycosyl donor 42 [25]. Glycosylation with the nucleophilic Schiff base [17] 6 using Hanessian's reaction [22] gave the β -serinyl-D-glucoside derivatives 43 an excellent yield as we have reported in one similar reaction [12] (Scheme 5). The conversion of 43 to (*N*-Fmoc-L-serin-3-yl)-3,4,6-tri-*O*-acetyl-2-acetamido-2-deoxy- β -D-glucopyranoside 46 has also been demonstrated. The Schiff base and the benzyl groups were cleaved from 43 by hydrogen in the presence of Pd/C, followed by re-protection of the amino group with Fmoc-Cl to provide 45. The Troc protecting group was transformed [26] to the unprotected amine with zinc/metal in acetic acid after subsequent treatment with acetic anhydride, the desired 46 was obtained in 62% overall yield.

3. Experimental

General methods.—All air and moisture sensitive reactions were performed under an argon atmosphere in flame-dried reaction flasks. THF was dried and de-oxygenated over $Ph_2C=O/Na^\circ-K^\circ$. CH_2Cl_2 and CH_3CN were dried over P_2O_5 and all solvents were freshly distilled under an argon atmosphere prior to use. For flash chromatography, 230–400 mesh silica gel 60 (E. Merck No. 9385) was employed. All compounds described were >95% pure by ¹H- and ¹³C-NMR, and purity was confirmed by elemental analysis in many cases. The ¹H- and ¹³C-NMR spectra were obtained on a

Bruker WM- or AM-250 MHz and a Gemini 200 MHz spectrometer. COSY and HETCOR spectra were obtained on a Bruker WM-500 spectrometer at 500 MHz or VARIAN UNITY 300 spectrometer at 300 MHz. Chemical shifts are reported in δ using Me₄Si as the standard reference in ¹H spectra and CDCl₃ in ¹³C spectra. Infrared spectra were obtained on a Perkin–Elmer 1600 Series FT-IR. All melting points were measured on a Hoover capillary melting point apparatus and are uncorrected. Optical rotations were measured on a Randolph Research AutoPol III polarimeter using the Na–D line. Elemental analyses were performed by Desert Analytics, Tucson, AZ 85719. Nominal and exact mass spectra were obtained on a JEOL JMS-01SG-2 mass spectrometer.

Reaction of acetobromoglycopyranose (1-5) and serine (6) or threonine Schiff base ester (7) using AgClO₄ (Method A —selective for α -glycosides).—A mixture of bromide (1 [17], 2 [18], 3 [17], 4 [19], or 5 [19]) (1.4 equiv), aglycone (5 [12] or 6 [12]) (1 mmol), and 2,4,6-collidine (1.7 equiv) in dichloromethane (5 mL) was dropped into a rigorously stirred suspension of AgClO₄ (1.7 equiv) in dichloromethane (10 mL) over a period of 40 min at room temperature. After the addition was complete the reaction was stirred for an additional 15 min, then quenched with Et₃N (2 equiv), diluted with dichloromethane (100 mL), and filtered through Celite. The filtrate was washed with 5% Na₂S₂O₃ solution (3 × 10 mL), saturated NaHCO₃ solution (3 × 10 mL), and water (3 × 10 mL). The colorless solution was dried over Na₂SO₄, evaporated under vacuum and separated by flash chromatography to provide glycosides 8–13.

N-Diphenylmethylene-O-(3,4,6-tri-O-acetyl-2-azido-2-deoxy- α -D-glucopyranosyl (8) and mannopyranosyl(10))-L-serine benzyl ester.—An unseparated mixture of 1 [17] and 3 [17] from the azidonitration [18] was reacted with 6 [12] as above to provide a mixture of gluco and manno products which could be easily separated on a SiO₂ column. The first fraction provided the manno product 10. Syrup, 24%, $[\alpha]_D + 34^\circ$ (c 1.9, chloroform), R_f 0.50 (hexane–ethylacetate 1:1). ¹H- and ¹³C-NMR data are provided in Tables 1–6. Anal. Calcd for C₃₅H₃₆O₁₀N₄: C, 62.48; H, 5.40; N, 8.33. Found: C, 62.37; H, 5.48; N, 8.28.

The second fraction provided the *gluco* product **8**. Syrup, 49%, $[\alpha]_D + 80^\circ$ (c 0.4, chloroform), R_f 0.45 (hexane-ethylacetate 1:1). ¹H- and ¹³C-NMR data are provided in Tables 1–6. Anal. Calcd for $C_{35}H_{36}O_{10}N_4$: C, 62.48; H, 5.40; N, 8.33. Found: C, 62.51; H, 5.52; N, 8.20.

N-Diphenylmethylene-O-(3,4,6-tri-O-acetyl-2-azido-2-deoxy- α -D-galactopyranosyl)-L-serine benzyl ester (9).—Syrup, 82%, $[\alpha]_D$ +38.2° (c 0.45, chloroform R_f 0.44 (dichloromethane–ethylacetate, 93:7). ¹H- and ¹³C-NMR data are provided in Tables 1–6. Anal. Calcd for C₃₅H₃₆O₁₀N₄: C, 62.48; H, 5.40; N, 8.33. Found: C, 62.44; H, 5.47; N, 8.42.

N-Diphenylmethylene-O-[O-(2,3,4,6-tetra-O-acetyl- β -D-galactopyranosyl)-(1 \rightarrow 4)-3,6-di-O-acetyl-2-azido-2-deoxy- α -D-glucopyranosyl]-L-serine benzyl ester (11) (Method A).—Foam, 81%, $[\alpha]_D$ +18.3° (c 1.8, chloroform), R_f 0.35 (hexane-ethylacetate 55:45). ¹H- and ¹³C-NMR data are provided in Tables 1–6. Anal. Calcd for C₄₇H₅₂O₁₈N₄: C, 58.73; H, 5.46; N, 5.83. Found: C, 58.69; H, 5.55; N, 5.68. N-Diphenylmethylene-O-[O-(2,3,4,6-tetra-O-acetyl- β -D-galactopyranosyl)-(1 \rightarrow 4)-3,6-di-O-acetyl-2-azido-2-deoxy- α -D-mannopyranosyl]-L-serine benzyl ester (12).— Foam, 87%, $[\alpha]_D + 12.7^\circ$ (c 0.7, chloroform), R_f 0.47 (toluene-ethylacetate 6:4). ¹Hand ¹³C-NMR data are provided in Tables 1–6. Anal. Calcd for C₄₇H₅₂O₁₈N₄: C, 58.73; H, 5.46; N, 5.83. Found: C, 58.80; H, 5.51; N, 5.72.

N-Diphenylmethylene-O-(3,4,6-tri-O-acetyl-2-azido-2-deoxy- α -D-galactopyranosyl)-L-threonine diphenylmethyl ester (13).—Syrup, 76%, $[\alpha]_D + 4.6^\circ$ (c 1.5, chloroform), R_f 0.28 (hexane-ethylacetate, 75:25). ¹H- and ¹³C-NMR data are provided in Tables 1–6. Anal. Calcd for C₄₂H₄₂O₁₀N₄: C, 66.13; H, 5.55; N, 7.34. Found: C, 65.94; H, 5.72; N, 7.20.

N-Diphenylmethylene-O-[O-(2,3,4,6-tetra-O-acetyl- β -D-galactopyranosyl)-(1 \rightarrow 4)-3,6-di-O-acetyl-2-azido-2-deoxy- α -(11) and β -(38)D-glucopyranosyl]-L-serine benzyl ester (Method B — preparation of authentic β -glycoside).—Benzyl N-(diphenylmethylene)-L-serinate (6) [12] (1.965 g), bromide (4) [19] (4.5 g), powdered, oven-dried 4 Å molecular sieves (5 g), and dichloromethane (25 mL) were stirred at 0°C under argon for 10 min. Silver triflate (1.8 g) was added in portions over 15 min, and stirring was continued for 14 h at room temperature. The reaction was quenched with Et₃N (2 mL), diluted with dichloromethane (150 mL), filtered through Celite and the organic layer was washed with saturated NaHCO₃ (3 × 15 mL), H₂O (3 × 15 mL), and dried (MgSO₄). Rotary evaporation and flash chromatography on SiO₂ with hexanes–ethylacetate, 55:45 (R_f 0.41) provided 320 mg pure **38** as a syrup, and a cofraction consisting of a mixture of **38** and **11**. The cofraction was re-chromotagraphed, to provide an additional 310 mg, thus giving 630 mg **38** (12%) in toto, $[\alpha]_D - 8.2^\circ$ (c 0.2, chloroform). ¹H- and ¹³C-NMR data are in Tables 1–6. Anal. Calcd for C₄₇H₅₂O₁₈N₄: C, 58.73; H, 5.46; N, 5.83. Found: C, 58.85; H, 5.60; N, 5.60.

The second fraction (R_f 0.35) provided 3.4 g 11 (71%) after two separations. Anal. Found: C, 58.68; H, 5.50; N, 5.76.

Removal of Schiff base from the serine / threonine-glycoside.—Glycoside 8, 10–13 or 38 (1 mmol) was dissolved in tetrahydrofuran (9 mL), and trifluoroacetic acid (1 mL) and water (0.2 mL) were added. The hydrolysis was complete within 5 min. The solution was diluted with dichloromethane (100 mL), washed by saturated NaHCO₃ solution (3×10 mL), water (3×10 mL), dried (Na₂SO₄), and evaporated in vacuo. The syrup was separated on a short SiO₂ column with a gradient of dichloromethane with 0.5% Et₃N to dichloromethane–acetone, 75:25 with 0.5% Et₃N to provide the amino esters 14–18 and 39. Since the benzyl esters are unstable as the free base [12] (diketopiperazine formation), no attempt was made to obtain elemental analyses on these intermediates.

O-(3,4,6-Tri-O-acetyl-2-azido-2-deoxy- α -D-glucopyranosyl)-L-serine benzyl ester (14).—Syrup, 94%, $[\alpha]_D$ +136° (c 0.25, chloroform). ¹H- and ¹³C-NMR data are provided in Tables 1–6. FAB-MS: Calcd for C₂₂H₂₈O₁₀N₄ 508.2, found m/z 509.2 (MH⁺).

O-(3,4,6-Tri-O-acetyl-2-azido-2-deoxy- α -D-mannopyranosyl)-L-serine benzyl ester (15).—Syrup, 92%, $[\alpha]_D$ +63.6° (c 0.27, chloroform). ¹H- and ¹³C-NMR data are provided in Tables 1–6. FAB-MS: Calcd for C₂₂H₂₈O₁₀N₄ 508.2, found m/z 509.2 (MH⁺).

O-[O-(2,3,4,6-Tetra-O-acetyl-β-D-galactopyranosyl)-(1 → 4)-3,6-di-O-acetyl-2azido-2-deoxy-α-D-glucopyranosyl]-L-serine benzyl ester (16).—Syrup, 97%, $[\alpha]_D$ +70.5° (c 1.6, chloroform). ¹H- and ¹³C-NMR data are provided in Tables 1–6. FAB-MS: Calcd for C₃₄ H₄₄O₁₈N₄ 796.2, found m/z 797.2 (MH⁺).

O-[(2,3,4,6-Tetra-O-acetyl- β -D-galactopyranosyl)-(1 \rightarrow 4)-3,6-di-O-acetyl-2-azido-2deoxy- α -D-mannopyranosyl]-L-serine benzyl ester (17).—Syrup, 96%, [α]_D + 33.9° (c 2.7, chloroform). ¹H- and ¹³C-NMR data are provided in Tables 1–6. FAB-MS: Calcd for C₃₄H₄₄O₁₈N₄ 796.2, found m/z 797.1 (MH⁺).

O-(3,4,6-Tri-O-acetyl-2-azido-2-deoxy- α -D-galactopyranosyl)-L-threonine diphenylmethyl ester (18).—Glassy solid, 91%, $[\alpha]_D$ +40.9° (c 1.1, chloroform). ¹H- and ¹³C-NMR data are provided in Tables 1–6. FAB-MS: Calcd for C₂₉H₃₄O₁₀N₄ 598.2, found m/z 599.2 (MH⁺).

O-[O-(2,3,4,6-Tetra-O-acetyl-β-D-galactopyranosyl)-(1 → 4)-3,6-di-O-acetyl-2azido-2-deoxy-β-D-glucopyranosyl]-L-serine benzyl ester (39).—Syrup, 96%, $[\alpha]_D$ +0.3° (c 3.0, chloroform). ¹H- and ¹³C-NMR data are provided in Tables 1–6. FAB-MS: Calcd for C₃₄H₄₄O₁₈N₄ 796.2, found m/z 797.1 (MH⁺).

Fmoc-protection of the serinate / threoninate glycoside esters.—Compound **14–18**, or **39** (1 mmol) was dissolved in dichloromethane (10 mL) and treated with 9-fluorenylmethylchloroformate or *N*-(9-fluorenylmethoxycarbonyloxy)succinimide (1 mmol) in the presence of *N*,*N*-diisopropylethylamine (2 mmol) for 45 min at room temperature. The solution was then diluted with dichloromethane (100 mL), washed with saturated NaHCO₃ solution (3×10 mL), water (3×10 mL), dried (Na₂SO₄), evaporated under vacuum, and separated on a SiO₂ column to provide the Fmoc-esters **19–23** or **40**.

N-(9-Fluorenylmethoxycarbonyl)-O-(3,4,6-tri-O-acetyl-2-azido-2-deoxy- α -D-glucopyranosyl)-L-serine benzyl ester (19).—Foam, 92%, [α]_D +92° (c 2.7, chloroform), R_f 0.25 (toluene–ethylacetate 2:1). ¹H- and ¹³C-NMR data are provided in Tables 1–6. FAB-MS: Calcd for C₃₇H₃₈O₁₂N₄ 730.2, found m/z 731.2 (MH⁺).

N-(9-Fluorenylmethoxycarbonyl)-O-(3,4,6-tri-O-acetyl-2-azido-2-deoxy- α -D-mannopyranosyl)-L-serine benzyl ester (20).—Foam, 87%, $[\alpha]_D$ +51.3° (c 0.15, chloroform), R_f 0.30 (toluene-ethylacetate 2:1). ¹H- and ¹³C-NMR data are provided in Tables 1–6. FAB-MS: Calcd for C₃₇H₃₈O₁₂N₄ 730.2, found m/z 731.2 (MH⁺).

N-(9-Florenylmethoxycarbonyl)-O-[O-(2,3,4,6-tetra-O-acetyl- β -D-galactopyranosyl)-(1 \rightarrow 4)-3,6-di-O-acetyl-2-azido-2-deoxy- α -D-glucopyranosyl]-L-serine benzyl ester (21).—Foam, 95%, $[\alpha]_D$ +62.8° (c 0.6, chloroform), R_f 0.30 (toluene-ethylacetate 6:4). ¹H- and ¹³C-NMR data are provided in Tables 1–6. FAB-MS: Calcd for C₄₉H₅₄O₂₀N₄ 1018.3, found m/z 1019.3 (MH⁺).

N-(9-Fluorenylmethoxycarbonyl)-O-[O-(2,3,4,6-tetra-O-acetyl-β-D-galactopyranosyl)-(1 → 4)-3,6-di-O-acetyl-2-azido-2-deoxy- α -D-mannopyranosyl]-L-serine benzyl ester (22).—Syrup, 96%, $[\alpha]_D$ +31.7° (c 1.2, chloroform), R_f 0.50 (toluene–ethylacetate, 6:4). ¹H- and ¹³C-NMR data are provided in Tables 1–6. FAB-MS: Calcd for C₄₉H₅₄O₂₀N₄ 1018.3, found m/z 1019.4 (MH⁺).

N-(9-Fluorenylmethoxycarbonyl)-O-(3,4,6-tri-O-acetyl-2-azido-2-deoxy- α -D-galactopyranosyl)-L-threonine diphenylmethyl ester (23).—Syrup, 91%, $[\alpha]_D$ +53.7° (c 0.5, chloroform), R_f 0.31 (hexane-ethylacetate 6:4). ¹H- and ¹³C-NMR data are provided in Tables 1–6. FAB-MS: Calcd for C₄₄H₄₄O₁₂N₄ 820.2, found m/z 821.3 (MH⁺).

N-(9-Fluorenylmethoxycarbonyl)-O-[O-(2,3,4,6-tetra-O-acetyl- β -D-galactopyranosyl)-(1 \rightarrow 4)-3,6-di-O-acetyl-2-azido-2-deoxy- β -D-glucopyranosyl]-L-serine benzyl ester (40).—Amorphous solid, 92%, [α]_D + 4.9° (c 2.9, chloroform), R_f 0.41 (toluene–ethylacetate, 6:4). ¹H- and ¹³C-NMR data are provided in Tables 1–6. FAB-MS: Calcd for C₄₉H₅₄O₂₀N₄ 1019.3, found m/z 1019.3 (MH⁺).

Fmoc-amino acid glycosides. Azide reduction / acylation, and ester hydrogenolysis. —Compound 19–23, or 40 (0.5 mmol) was dissolved in methanol (150 mL), and water (10 mL) and 5% palladium on activated carbon (200 mg) were added. This mixture was vigorously stirred under H₂ (1 atm) for 1–2 h (followed by TLC). The Pd/C was filtered from the suspension, and the solution was evaporated in vacuo. The residue was stirred in dioxane (20 mL) in the presence of acetic anhydride (1.3 equiv) and Et₃N (3 equiv) for about 30 min. at room temperature (followed by TLC). The mixture was concentrated in vacuo, dissolved in dichloromethane (100 mL). The pH was set to 3 with acetic acid and the solution was washed with water (3 × 10 mL), dried over MgSO₄, evaporated, and re-evaporated from toluene under vacuum. The residue was separated on a short SiO₂ column to provide 24–28, and 41. Since the Fmoc-amino acid derivatives exist as a mixture of rotamers, only characteristic ¹H-NMR data are presented.

N-(9-Fluorenylmethoxycarbonyl)-O-(3,4,6-tri-O-acetyl-2-acetamido-2-deoxy-α-Dglucopyranosyl)-L-serine (24).—Oil, 88%, $[\alpha]_{\rm D}$ +68° (c 2.2 chloroform), R_f 0.33 (dichloromethane-methanol 9:1). Characteristic ¹H-NMR data : δ 7.60-7.03 (m, 8 H, arom.), 6.35 (d, 1 H, J 9.8 Hz, NH), 6.24 (d, 1 H, J 9.6 Hz, NH), 1.98, 1.97, 1.90, 1.85 (4s, 12 H, 4 CH₃CO). ¹³C-NMR data are provided in Tables 5 and 6. Anal. Calcd for $C_{32}H_{36}O_{13}N_2$: C, 58.53; H, 5.53; N, 4.27. Found: C, 58.66; H, 5.45; N, 4.12.

N-(9-Fluorenylmethoxycarbonyl)-O-(3,4,6-tri-O-acetyl-2-acetamido-2-deoxy-α-D-mannopyranosyl)-L-serine (25).—Oil, 90%, $[\alpha]_{\rm D}$ +48.9° (c 0.55 chloroform), R_f 0.2 (dichloromethane-methanol 9:1). Characteristic ¹H-NMR data : δ 7.70–7.05 (m, 8 H, arom.), 6.26 (d, 1 H, NH), 5.62 (d, 1 H, NH), 4.92 (dd, 1 H, $J_{2,3} = J_{3,4}$ 10.4 Hz, H-3), 2.08, 1.97, 1.96, 1.95 (4s, 12 H, 4 CH₃CO). ¹³C-NMR data are provided in Tables 5 and 6. Anal. Calcd for $C_{32}H_{36}O_{13}N_2$: C, 58.53; H, 5.53; N, 4.27. Found: C, 58.40; H, 5.38; N, 4.32.

N-(9-Fluorenylmethoxycarbonyl)-O-[O-(2,3,4,6-tetra-O-acetyl-β-D-galactopyranosyl)-(1 → 4)-3,6-di-O-acetyl-2-acetamido-2-deoxy-α-D-glucopyranosyl]-L-serine (**26**).—Amorphous solid, 98%, $[\alpha]_D$ +55.5° (c 0.7 chloroform), R_f 0.5 (dichloromethane-methanol 85:15). Characteristic ¹H-NMR data (d_6 -Me₂SO) : δ 7.89–7.31 (m, 8 H, arom.), 5.21 (bd, 1 H, $J_{3',4'}$ 3.5 Hz, H-4'), 5.16 (dd, 1 H, $J_{2'3'}$ 10.3 Hz, H-3'), 5.00 (dd, 1 H, $J_{2,3}$ 9.9 Hz, H-3), 4.71 (d, 1 H, $J_{1',2'}$ 8.0 Hz, H-1'), 4.68 (d, 1 H, $J_{1,2}$ 3.3 Hz, H-1), 4.21 (bdd, 1 H, $J_{5',6'} = J_{5',6a'} = 6.7$ Hz, H-5'), 3.85 (ddd, 1 H, $J_{5,6}$ 1.8, $J_{5,6'}$ 4.2 Hz, H-5), 3.72 (dd, 1 H, $J_{4,5}$ 9.3 Hz, H-4), 2.08, 2.06, 2.02, 1.99, 1.98, 1.94, 1.79 (7s, 21 H, 7 CH₃CO). ¹³C-NMR data are provided in Tables 5 and 6. Anal. Calcd for C₄₄H₅₂O₂₁N₂: C, 55.93; H, 5.55; N, 2.96. Found: C, 55.78; H, 5.40; N, 2.91.

N-(9-Fluorenylmethoxycarbonyl)-O-[O-(2,3,4,6-tetra-O-acetyl-β-D-galactopyranosyl)-(1 → 4)-3,6-di-O-acetyl-2-acetamido-2-deoxy- α -D-mannopyranosyl]-L-serine (27) (Method A).—Amorphous solid, 93%, $[\alpha]_D$ +35.1° (c 0.5 chloroform), R_f 0.47 (toluene–ethanol 8:2). ¹³C-NMR data are provided in Tables 5 and 6. Anal. Calcd for C₄₄H₅₂O₂₁N₂: C, 55.93; H, 5.55; N, 2.96. Found: C, 56.21; H, 5.39; N, 2.74. N-(9-Fluorenylmethoxycarbonyl)-O-(3,4,6-tri-O-acetyl-2-acetamido-2-deoxy- α -D-galactopyranosyl)-L-threonine (28) (Method A).—Amorphous solid, 92%, $[\alpha]_D + 59^\circ$ (c 0.5 chloroform), R_c 0.55 (dichloromethane-methanol 85:15); lit [27]. $[\alpha]_D + 65.0^\circ$ (c 1.45, chloroform). ¹³C-NMR data are provided in Tables 5 and 6. Anal. Calcd for $C_{33}H_{38}O_{13}N_2$: C, 59.10; H, 5.71; N, 4.18. Found: C, 59.34; H, 5.60; N, 4.05.

N-(9-Fluorenylmethoxycarbonyl)-O-[O-(2,3,4,6-tetra-O-acetyl-β-D-galactopyranosyl)-(1 → 4)-3,6-di-O-acetyl-2-acetamido-2-deoxy-β-D-glucopyranosyl]-L-serine (41).—Amorphous solid, 95%, $[\alpha]_D$ +12.7° (c 0.8 chloroform), R_f 0.5 (dichloromethane-methanol 85:15). ¹³C-NMR data are provided in Tables 5 and 6. Anal. Calcd for C₄₄H₅₂O₂₁N₂: C, 55.93; H, 5.55; N, 2.96. Found: C, 55.99; H, 5.42; N, 2.73.

 $N-(9-Fluorenylmethoxycarbonyl)-O-[O-(2,3,4,6-tetra-O-acetyl-\beta-D-galactopyrano-acetyl-\beta-D-galactopyrano-beta acetyl-beta acet$ syl)- $(1 \rightarrow 4)$ -3,6-di-O-acetyl-2-acetamido-2-deoxy- α -D-glucopyranosyl]-L-serine (26) (direct route from Schiff base ester 11 without purification of intermediates).—Compound 11 (1.38 g) was dissolved in tetrahydrofuran (13.5 mL), and trifluoroacetic acid (1.5 mL) and water (0.3 mL) were added. Within 10 min the hydrolysis was complete. The solution was diluted with dichloromethane (150 mL), washed with saturated NaHCO₃ (3×15 mL), water (3×15 mL), dried (Na₂SO₄), and evaporated in vacuo. The syrup was dissolved in dichloromethane (10 mL) and treated with 9-fluorenylmethyl chloroformate (373 mg) in the presence of triethylamine (0.5 mL) for 50 min at room temperature. (The reaction was followed by TLC, toluene-ethylacetate 6:4.) The solution was then diluted with dichloromethane (150 mL), washed with saturated NaHCO₃ solution (3×15 mL), water (3×15 mL), dried (Na₂SO₄), and evaporated in vacuo. The residue was dissolved in methanol (300 mL) and water (30 mL), and 5% palladium on activated carbon (200 mg) was added. This mixture was vigorously stirred under H_2 (1 atm) for 2 h (followed by TLC). The Pd/C was filtered from the reaction, and the solution evaporated in vacuo. The residue was stirred in dioxane (25 mL) in the presence of acetic anhydride (0.2 mL) and Et_3N (0.6 mL) for 30 min at room temperature (followed by TLC). The mixture was concentrated in vacuo, and dissolved in dichloromethane (200 mL). The acidity of the solution was adjusted to pH 3 with acetic acid, and washed with water $(3 \times 25 \text{ mL})$, dried (MgSO₄), evaporated, then re-evaporated from dry toluene in vacuo. The residue was separated on a short column to give 26 in 88% overall yield. Amorphous solid, $[\alpha]_D + 54.8^\circ$ (c 1.0 chloroform), R_f 0.5 (dichloromethane-methanol 85:15). Anal. Found: C, 56.08; H, 5.67; N, 2.81.

Preparation of 2-acetamido-Schiff base glycosides 30-32. The Staudinger route.— 2-Azido-Schiff base glycosides (9-12) (0.1 mmol), triphenylphosphine (1.3 equiv), and acetic anhydride (1.5 equiv) were stirred in dry tetrahydrofuran (3 mL) for 16 h at room temperature. Into this mixture 1 M ammonium acetate in water (0.2 mL) was added, and the reaction was stirred for another 2-5 days (followed by TLC). The reaction mixture was evaporated, then re-evaporated from dry toluene in vacuo. The residue was dissolved in dichloromethane (30 mL), washed with water (3 × 5 mL), evaporated and separated on a column to give 30-32.

N-Diphenylmethylene-O-(3,4,6-tri-O-acetyl-2-acetamido-2-deoxy- α -D-galactopyranosyl)-L-serine benzyl ester (**30**).—Amorphous solid, 68%, $[\alpha]_{\rm D}$ +25° (c 0.18, chloroform), R_f 0.45 (dichloromethane-acetone 87:13). ¹H- and ¹³C-NMR data are provided in Tables 1–6. Anal. Calcd for C₃₇H₄₀O₁₁N₂: C, 64.53; H, 5.85; N, 4.07. Found: C, 64.31; H, 5.92; N, 4.18. N-Diphenylmethylene-O-[O-(2,3,4,6-tetra-O-acetyl-β-D-galactopyranosyl)-(1 → 4)-3,6-di-O-acetyl-2-acetamido-2-deoxy-α-D-glucopyranosyl]-L-serine benzyl ester (**31**).—Foam, 77%, $[\alpha]_D - 17^\circ$ (c 1.0, chloroform), R_f 0.37 (hexane-ethylacetatemethanol 5:5:1). Characteristic ¹H-NMR data : δ 7.78-7.11 (m, 15 H, arom.), 5.63 (d, 1 H, NH), 5.31 (dd, 1 H, $J_{3'-4'}$ 3.3 Hz, $J_{4'-5'}$ 1 Hz, H-4'), 4.93 (dd, 1 H, $J_{2'-3'}$ 10.3 Hz, H-3'), 4.71 (d, 1 H, J_{1-2} 3.5 Hz, H-1), 4.47 (d, 1 H, $J_{1'-2'}$ 7.8 Hz, H-1'), 2.13-1.95 (7s, 21 H, 7 CH₃CO). Anal. Calcd for C₄₅H₅₆O₁₉N₂: C, 60.24; H, 5.78; N, 2.87. Found: C, 60.01; H, 5.80; N, 2.65.

N-Diphenylmethylene-O-[O-(2,3,4,6-tetra-O-acetyl-β-D-galactopyranosyl)-(1 → 4)-3,6-di-O-acetyl-2-acetamido-2-deoxy-α-D-mannopyranosyl]-L-serine benzyl ester (32). —Foam, 61%, [α]_D + 27° (c 0.74, chloroform), R_f 0.45 (hexane–ethylacetate–methanol 5:5:1). ¹H-NMR characteristic data : δ 7.70–7.14 (m, 15 H, arom.), 5.45 (d, 1 H, J_{NH-2} 9.5 Hz, NH), 5.32 (dd, 1 H, $J_{3'-4'}$ 3.3 Hz, $J_{4'-5'}$ 1 Hz, H-4'), 5.15 (s, 2 H, CH₂Ph), 5.22 (dd, 1 H, J_{2-3} 4.3 Hz, J_{3-4} 8.8 Hz, H-3), 5.12 (dd, 1 H, $J_{1'-2'}$ 7.9 Hz, $J_{2'-3'}$ 10.5 Hz, H-2') 4.95 (dd, 1 H, $J_{2'-3'}$ 10.5 Hz, $J_{3'-4'}$ 3.3 Hz, $J_{3'-4'}$ 3.3 Hz, H-3'), 4.70 (d, 1 H, J_{1-2} 1.6 Hz, H-1), 4.53 (d, 1 H, $J_{1'-2'}$ 7.9 Hz, H-1'), 4.40 (dd, 1 H, $J_{\alpha-b}$ 8.9 Hz, $J_{\alpha-b'}$ 4.4 Hz, H-a) 2.18–1.95 (7s, 21 H, 7 CH₃CO). Anal. Calcd for C₄₉H₅₅O₁₉N₂: C, 60.30; H, 5.68; N, 2.87. Found: C, 60.33; H, 5.94; N, 2.55.

O-(3,4,6-Tri-O-acetyl-2-acetamido-2-deoxy- α -D-galactopyranosyl)-L-serine (34).— Compound 30 (230 mg) and 5% Pd/C (200 mg) were stirred in methanol (60 mL) under H₂ (1 atm) for 2 h. The Pd/C was filtered, and the solution evaporated in vacuo, dissolved in water (100 mL), and washed with toluene (3 × 15 mL). The inorganic phase was lyophilized to give the amorphous 34. 135 mg (93%), [α]_D + 95° (c 0.3, c water), R_f 0.4 (dichloromethane-methanol 65:35). ¹³C-NMR data are provided in Tables 5 and 6. Anal. Calcd for C₁₇H₂₆O₁₁N₂: C, 47.00; H, 6.03; N, 6.45. Found: C, 47.21; H, 5.87; N, 6.33.

N-(9-Fluorenylmethoxycarbonyl)-O-(3,4,6-tri-O-acetyl-2-acetamido-2-deoxy- α -D-galactopyranosyl)-L-serine (37).—Compound 34 (100 mg) was stirred in a mixture of 10% NaHCO₃ in water (10 mL) and dioxane (6 mL) at 0°C, and Fmoc-Cl (60 mg, 1 equiv) in dioxane (6 mL) was added over 30 min. After stirring at 0°C for 4 h and then at room temperature for 8 h, the reaction mixture was adjusted to pH 4 with 10% HCl, evaporated, dissolved in dichloromethane (100 mL), washed with water (3 × 15 mL), dried, evaporated, and chromatographed on SiO₂ to give 37. Amorphous solid,133 mg (88%), $[\alpha]_D + 87.5^\circ$ (c 2, chloroform), R_f 0.4 (dichloromethane–methanol 85:15); lit. [27] $[\alpha]_D + 89.9^\circ$ (c 1.0, chloroform.)

N-9-Fluorenylmethoxycarbonyl-O-(3,4,6-tri-O-acetyl-2-acetamido-2-deoxy- α -D-glucopyranosyl)-L-serine (24).—The Staudinger route as in $9 \rightarrow 30$ was followed, using 8 as starting material. The product was separated on a short SiO₂ column with dichloromethane-acetone 9:1 (R_f 0.35) to give 29. Without characterization the evaporated residue was hydrogenated and treated with Fmoc-Cl (cf. 43 \rightarrow 45, below) to give 24, in a 47% overall yield. Anal. Found: C, 58.41; H, 5.64; N, 4.24.

N-Diphenylmethylene-O-(3, 4, 6-tri-O-acetyl-2-deoxy-2-{[(2, 2, 2-trichloroethoxy)carbonyl]amino}- β -D-glucopyranosyl)-L-serine benzyl ester (43).—Glycosylation as in $6 \rightarrow 38$ (method B), using bromide 42 [25] (1.5 equiv) and silver triflate (1.5 equiv) to provide 43. Foam, 83%, [α]_D - 11.3° (c 0.8, chloroform), R_f 0.6 (hexane-ethylacetate 1:1). ¹H-NMR data are provided in Tables 1-4. Anal. Calcd for C₃₈H₃₉O₁₂N₂Cl₃: C, 55.52; H, 4.78; N, 3.41; Cl, 12.94. Found: C, 55.76; H, 4.81; N, 3.55; Cl, 13.05.

N-(9-Fluorenylmethoxycarbonyl)-O-(3,4,6-tri-O-acetyl-2-deoxy-2-{[(2,2,2-tri-chloroethoxy)carbonyl]amino}-β-D-glucopyranosyl)-L-serine (45).—Schiff base 43 (586 mg) and 5% Pd/C (200 mg) were stirred in methanol (300 mL) under H₂ (1 atm) for 3 h. The Pd/C was filtered off, the solvent evaporated in vacuo, and the residue was dissolved in water (250 mL) and washed with toluene (3×50 mL). The inorganic phase was evaporated to give 44. Without any purification the amino acid glycoside was stirred in a mixture of 10% NaHCO₃ in water (20 mL) and dioxane (12 mL) at 0°C, and Fmoc-Cl (259 mg, 1 equiv) in dioxane (10 mL) was added over 30 min. After stirring at 0° C for 4 h, and then at room temperature for 8 h, the pH was set to 4 with 10% HCl. The reaction mixture was evaporated, dissolved in dichloromethane (150 mL), washed with water $(3 \times 25 \text{ mL})$, dried, evaporated, and chromatographed to give 45 as a foam. 575 mg (73%), $[\alpha]_{\rm p}$ +17.9° (c 0.56, chloroform), R_f 0.2 (dichloromethane-methanol 9:1). Characteristic ¹H-NMR data : δ 7.76–7.27 (m, 8 H, arom.), 5.90 (d, 1 H, J 8.2 Hz, NH), 5.43 (d, 1 H, J 7.8 Hz, NH), 5.23 (dd, $J_{2-3} = J_{3-4} = 10.2$ Hz, H-3), 5.03 (dd, 1 H, J_{4-5} 10.0 Hz, H-4), 4.65 (d, 1 H, J_{1-2} 8.3 Hz, H-1), 3.63 (m, 2 H, H-2 and H-5), 2.06, 2.00, 1.98 (3s, 9 H, 3 CH₃CO). FAB-MS: Calcd for C₃₃H₃₅O₁₄N₂Cl₃ 788.1, found m/z 789.1 (MH⁺).

N-(9-Fluorenylmethoxycarbonyl)-O-(3,4,6-tri-O-acetyl-2-acetamido-2-deoxy-β-Dglucopyranosyl)-L-serine (46).—Compound 45 (0.8 g) was dissolved in acetic acid (20 mL) and Zn (dust, 1 g) was added. The mixture was stirred for 4 h at room temperature, filtered through Celite, and evaporated from toluene in vacuo. The residue was dissolved in dichloromethane (30 mL) and acetic anhydride (0.11 mL), and Et₃N (0.2 mL) were added. After 2 h, the solution was diluted with dichloromethane (100 mL), washed with water (3 × 10 mL) dried (MgSO₄), and concentrated in vacuo. Flash chromatography provided 46 as a oil. 415 mg (62%), $[\alpha]_D - 21.9^\circ$ (c 1.7, chloroform), R_f 0.2 (dichloromethane–methanol 9:1). ¹H-NMR data are provided in Tables 1–4. Anal. Calcd for C₃₂H₃₆O₁₃N₂: C, 58.53; H, 5.53; N, 4.27. Found: C, 58.58; H, 5.42; N, 4.10.

Frace glycosides (26 and 27) from 2-acetamido-glycosides (31 and 32).—Compounds 31-32 were treated as in $43 \rightarrow 45$ above, to provide 26 and 27.

N-(9-Fluorenylmethoxycarbonyl)-O-[O-(2,3,4,6-tetra-O-acetyl-β-D-galactopyranosyl)-(1 → 4)-3,6-di-O-acetyl-2-acetamido-2-deoxy- α -D-glucopyranosyl]-L-serine (26).—Amorphous solid, 90%, $[\alpha]_D$ +56.1° (c 0.5 chloroform), R_f 0.5 (dichloromethane-methanol 85:15). ¹³C-NMR data are provided in Tables 5 and 6. Anal. Found: C, 55.84; H, 5.69; N, 2.82.

N-(9-Fluorenylmethoxycarbonyl)-O-[O-(2,3,4,6-tetra-O-acetyl-β-D-galactopyranosyl)-(1 → 4)-3,6-di-O-acetyl-2-acetamido-2-deoxy- α -D-mannopyranosyl]-L-serine (27). —Amorphous solid, 91%, $[\alpha]_{\rm p}$ + 29.4° (c 0.8 chloroform), R_f 0.44 (dichloromethane-methanol 85:15). ¹³C-NMR data are provided in Tables 5 and 6. Anal. Found: C, 55.74; H, 5.67; N, 2.82.

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References

- [1] J. Montreuil, Adv. Carbohydr. Chem. Biochem., 37 (1980) 157-223.
- [2] H. Eylar, J. Theor. Biol., 10 (1965) 89-112.
- [3] (a) N.J. Maeji, Y. Inoue, and R. Chùjô, Biopolymers, 26 (1987) 1753-1767; (b) M. Hollosi, A. Perczel, and G.D. Fasman, Biopolymers, 29 (1990) 1549-1564; (c) H. Paulsen, R. Busch, V. Sinnwell, and A. Pollex-Krüger, Carbohydr. Res., 214 (1991) 227-234; (d) K. Dill, R.E. Hardy, M.E. Daman, J.M. Lacombe, and A.A. Pavia, Carbohydr. Res., 108 (1982) 31-40; (e) R. Shogren, T.A. Gerken, and N. Jentoft, Biochemistry, 28 (1989) 5525-5536; (f) T.A. Gerken, K.J. Butenhof, and R. Shogren, Biochemistry, 28 (1989) 5536-5543.
- [4] P.J. Winterburn and C.F. Phelps, Nature, 236 (1972) 147-151.
- [5] M. Hollósi, E. Kollát, I. Laczkó K.F. Medzihradszky, J. Thurin, and L. Otvös, Tetrahedron Lett., 32 (1991) 1531–1534.
- [6] (a) W. Kinzy and R.R. Schmidt, Carbohydr. Res., 164 (1987) 265-276; (b) Y. Nakahara, H. Iijima, S. Shibayama, and T. Ogawa, Carbohydr. Res., 216 (1991) 211-225; H. Kunz and S. Birnbach, Angew. Chem. Int. Ed. Engl., 25 (1987) 360-362, and references therein.
- [7] R. Polt, F. Porreca, L. Szabó, E.J. Bilsky, P. Davis, T. Davis, R. Horváth, T.J. Abbruscato, T.P. Davis, R. Horvath, H.I. Yamamura, and V.J. Hruby, Proc. Nat. Acad. Sci. U.S.A., 91 (1994) 7114-7118.
- [8] H. Kunz, Angew. Chem. Int. Ed. Engl., 26 (1987) 294-308; A. Vargas-Berenquel, M. Meldal, H. Paulsen, and K. Bock, J. Chem. Soc. Perkin Trans., 18 (1994) 2615-2619.
- [9] K. Wakabayashi and W. Pigman, Carbohydr. Res., 35 (1974) 3-14; H.G. Garg and R.W. Jeanloz, Carbohydr. Res., 49 (1976) 482-488; J.R. Vercellotti, R. Fernandez, and C.J. Chang, Carbohydr. Res., 5 (1967) 97-101; J.M. Lacombe and A.A. Pavia, J. Org. Chem., 48 (1983) 2557-2563.
- [10] F. Micheel and H. Köchling, Chem. Ber., 91 (1958) 673-676; N.K. Kochetkov, V.A. Derevitskaya, L.M. Likhosherstov, V.M. Kalinevich, and O.S. Novikova, Izv. Akad. Nauk. SSSR, Ser. Khim., (1969) 2509; H.G. Garg and R.W. Jeanloz, Carbohydr. Res., 52 (1976) 246-250; J.C. Jacquinet, S.E. Zurabyan, and A.Y. Khorlin, Carbohydr. Res., 32 (1974) 137-143; L.P. Egan, J.R. Vercellotti, and W.T. Lowry, Carbohydr. Res., 23 (1972) 261-273; P.J. Garegg and T. Norberg, Carbohydr. Res., 52 (1976) 235-240; J.M.Lacombe, A.A. Pavia, and J.M. Rocheville, Can. J. Chem., 59 (1981) 473-481; K. Higashi, K. Nakayama, T. Soga, U. Kouichi, and T. Kusama, Chem. Pharm. Bull., 38 (1990) 3280-3282.
- [11] M.A. Peterson and R. Polt, J. Org. Chem. 58 (1993) 4309-4314.
- [12] L. Szabó, Y. Li, and R. Polt, *Tetrahedron Lett.*, 32 (1991) 585-588; R. Polt, L. Szabó, J. Treiberg, Y. Li, and V.J. Hruby, J. Am. Chem. Soc., 114 (1992) 10249-10258; L. Szabó and R. Polt, *Carbohydr. Res.*, 258 (1994) 293-297.
- [13] J. Ramza, L. Szabó, E. Gross, C.D. Langdon, K. Pietrzak, and R. Polt, in preparation.
- [14] S. Peters, T. Bielfeldt, M. Meldal, K. Bock, and H. Paulsen, J. Chem. Soc. Perkin Trans. 1, (1992) 1163-71.
- [15] M.J. O'Donnell and R.L. Polt, J. Org. Chem., 47 (1982) 2663-2666.
- [16] H. Paulsen, C. Kolár, and W. Stenzel, Angew. Chem. Int. Ed. Engl., 15 (1976) 440-441.
- [17] H. Paulsen, J.P. Lorentzen, and W. Kutschker, Carbohydr. Res., 136 (1985) 153-176.
- [18] R.U. Lemieux and R.M. Ratcliffe, Can. J. Chem., 57 (1979) 1244-1251.
- [19] H. Paulsen and J.P. Hölck, Liebigs. Ann. Chem., (1982) 1121-1131.
- [20] V. Pavliak, P. Kovac, and C.P.J. Glaudemans, Carbohydr. Res., 229 (1992) 103-116; P. Kovac and L. Lerner, Carbohydr. Res., 184 (1988) 87-112.
- [21] K. Bock, I. Lundt, and C. Pedersen, Tetrahedron Lett., 13 (1973) 1037-1040.
- [22] S. Hanessian and J. Banoub, Carbohydr. Res., 53 (1977) C13-16.
- [23] M. Plewe, K. Sandhoff, and R.R. Schmidt, J. Carbohydr. Chem., 11 (1992) 881-890.
- [24] E. Atherton, C. Bury, R.C. Sheppard, and B.J. Williams, Tetrahedron Lett., 32 (1979) 3041-3042.
- [25] K. Higashi, K. Nakayama, T. Soga, E. Shioya, K. Uoto, and T. Kusama, Chem. Pharm. Bull., 38 (1990) 3280-3282.
- [26] T.B. Windholz and D.B.R. Johnston, Tetrahedron Lett., 27 (1967) 2555-2557.
- [27] H. Paulsen and K. Adermann, Liebigs Ann. Chem., (1989) 751-769.