LIQUIDAMBIN, AN ELLAGITANNIN FROM LIQUIDAMBAR FORMOSANA

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Abstract—A new ellagitannin, named liquidambin, which could be biogenetically closely correlated with casuarinin and pedunculagin, has been isolated from the leaves of *Liquidambar formosana*. Its structure was determined as 5-0-galloyl-2,3,4,6-di-O-(S)-hexahydroxydiphenoyl-D-glucose. The structural equilibration due to hydration of the aldehyde group of the glucose core in this tannin was shown from its ¹H and ¹³C NMR spectra.

INTRODUCTION

Previously, we reported remarkable changes in the tannin composition in the leaves of *Liquidambar formosana* Hance during the spring [1]. This change is in accord with the proposed biogenetic pathway [2] from galloylglucoses and tellimagrandin II to *C*-glucosidic casuarinin (1) via pedunculagin (2). We now describe the isolation and structure elucidation of an additional new tannin in the leaves, named liquidambin, which could be biogenetically closely correlated to both 1 and 2.

RESULTS AND DISCUSSION

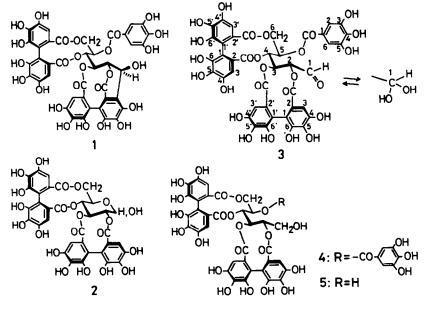
Liquidambin (3) was isolated from the n-BuOH soluble portion of the extract of L. formosana leaves by means of

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centrifugal partition chromatography (CPC) [3] in combination with CC over MCI gel CHP-20P.

The tannin, $C_{41}H_{28}O_{26} \cdot 7H_2O$, was obtained as an offwhite amorphous powder. Although liquidambin shows a complicated ¹H NMR spectrum (in Me₂CO-d₆) due to the formation of an equilibrium mixture, the signals of a galloyl group [δ 7.12 (2H, s)] and two hexahydroxydiphenoyl (HHDP) groups [δ 6.75, 6.74, 6.72 and 6.59 (1H each, s)] of the major component in the equilibrium mixture of 3, and those of the minor component [δ 7.15 (2H, s) (galloyl); 6.75, 6.65, 6.63 and 6.61 (1H each, s) (HHDP)] were observed.

Reduction of 3 by NaBH₄ afforded a product, $C_{41}H_{30}O_{26} \cdot 2H_2O$ (4). Its ¹H NMR spectrum (400 MHz, in Me₂CO-d₆) indicates the presence of a galloyl group [δ 7.12 (2H, s)], two HHDP groups [δ 6.74, 6.70, 6.58 and 6.57 (1H each, s)] and a glucitol core [δ 5.61 (dd, J = 1, 8.5 Hz, H-4), 5.60 (dd, J = 2.5, 8.5 Hz, H-5), 5.51 (dd, J



= 1, 9.5 Hz, H-3), 5.09 (*ddd*, J = 2, 5, 9.5 Hz, H-2), 4.88 (*dd*, J = 2.5, 13 Hz, H-6_a), 4.32 (*t*, J = 6 Hz, proton of the hydroxyl group at C-1), 4.10 (*d*, J = 13 Hz, H-6_b), 4.05 (*ddd*, J = 5, 6, 12 Hz, H-1_a) and 3.97 (*ddd*, J = 2, 6, 12 Hz, H-1_b)].

H-1_b)]. Although liquidambin is not identical with casuarictin [= 1-O-galloyl-2,3;4,6-di-O-(S)-hexahydroxydiphenoyl- β -D-glucose] [4] or potentillin [= 1-O-galloyl-2,3;4,6-di-O-(S)-hexahydroxydiphenoyl- α -D-glucose] [5], partial hydrolysis of liquidambin (3) with tannase gave ped-2,3;4,6-di-O-(S)-hexahydroxydiphenoyl-Dunculagin. glucose (2) [4]. Comparison of the chemical shifts of the glucitol protons of 4 with those of 2,3;4,6-di-O-(S)hexahydroxydiphenoyl-D-glucitol (5), which was produced from 2, shows a downfield shift of H-5 in 4 from that in 5 ($\delta 4.27 \rightarrow 5.60$). Therefore, the position of the galloyl group in 3 was assigned to be at O-5 of the glucose core. This assignment was further verified by treatment of 3 with p-toluenesulphonic acid, which afforded casuarinin (1).

Thus, liquidambin has structure 3, in which two (S)hexahydroxydiphenoyl groups are located at O-2-O-3 and O-4-O-6 of the glucose core, and the galloyl group is located at O-5. Although the presence of an aldehyde group at C-1 of the glucose core of 3 was shown by a signal at δ 9.77 in the ¹H NMR spectrum and shown by a signal at δ 194.61 in the ¹³C NMR spectrum of 3, the aldehyde group is to a large extent hydrated as shown by the signal at δ 87.99 in the ¹³C NMR spectrum. The ratio of the major and minor components was estimated to be 9:7, based on the peak areas of the aromatic protons in the ¹H NMR spectrum.

The facile chemical transformation of liquidambin into casuarinin suggests that liquidambin could be a precursor of casuarinin biosynthesis in *L. formosana*. Although the result of enzymatic transformation of liquidambin into pedunculagin suggests that biosynthesis of pedunculagin may occur by this route, pedunculagin will not be produced from liquidambin in most species, as liquidambin has not been found in the other plant species in spite of the wide distribution of pedunculagin [2, 6].

EXPERIMENTAL

¹H NMR and ¹³C NMR spectra were measured at 400 MHz and 100 MHz, respectively with TMS as int. std. TLC was carried out on Avicel SF cellulose plates (0.3 mm) using 7% HOAc as developer.

Isolation. Fresh leaves (3.5 kg) were collected in May, 1986, from a tree of L. formosana Hance grown at Okayama University, and were homogenized in Me₂CO-H₂O (7:3, 141). After filtration, the filtrate was coned in vacuo and extracted with Et2O, EtOAc and n-BuOH, successively. An aliquot (3 g) of the n-BuOH extract (36 g) was subjected to CPC on a Sanki L-90 centrifugal partition chromatograph [3] using n-BuOH-n-PrOH-H₂O (4:1:5) and reversed-phase development; 10 g portions of the eluate were collected. Combined fractions 16-30 were evapd and subjected to CPC using the same solvent system but normal-phase development: 5 g fractions were collected. Combined fractions 30-46 were evapd and dissolved in H₂O. The aq. soln was acidified to pH 2 with 10% HCl and subjected to CC over MCI gel CHP-20P (75-150 µm) with H₂O and then MeOH as eluants. The MeOH eluate was further purified by CC over MCI gel CHP-20P with 20% MeOH as eluant to afford liquidambin (20 mg).

Liquidambin (3). Pale yellow amorphous powder, R_f on TLC,

0.41. $[\alpha]_D + 69^\circ$ (Me₂CO; c 0.5). (Found: C, 46.43; H, 3.67. $C_{41}H_{28}O_{26}$ · 7 H_2O requires: C, 46.34; H, 3.98.) UV λ_{max}^{MeOH} nm $(\log \epsilon)$: 219 (4.88), 270 sh (4.55). IR ν_{max}^{KBr} cm⁻¹: 3400, 1740–1710, 1610, 1510, 1450, 1360-1310, 1230-1180, 1030. ¹H NMR (Me2CO-d6): 87.12 (2H, s, galloyl), 6.75, 6.74, 6.72, 6.59 (1H each, s, $2 \times$ HHDP), 5.62 [dd, J = 1.5, 9 Hz, glucose (glu) H-4], 5.57 (dd, J = 3.5, 9 Hz, glu H-5), 5.47 (dd, J = 1.5, 9 Hz, glu H-3), 5.05 (dd, J = 2,9 Hz, glu H-2), 4.87 (dd, J = 3.5, 13 Hz, glu H-6,), 4.12 (d, J = 13 Hz, glu H-6_b) [major (hydrated) form]; $\delta 9.77$ (1H, s, glu H-1), 7.15 (2H, s, galloyl), 6.75, 6.65, 6.62, 6.61 (1H each, s, $2 \times \text{HHDP}$ 5.69 (dd, J = 1, 9 Hz, glu H-4), 5.67 (dd, J = 3, 9 Hz, glu H₇5), 5.54 (dd, J = 1, 9.5 Hz, H-3), 5.52 (d, J = 9.5 Hz, glu H-2), 5.03 (dd, J = 3, 13 Hz, glu H-6_a), 4.09 (d, J = 13 Hz, glu H-6_b) [minor (aldehydo) form]. ¹³C NMR (Me₂CO- d_6): δ 64.79, 64.83 (glu C-6), 70.08, 70.16, 70.37, 70.75, 72.56, 74.98, 75.80, 77.92, (glu C-2-C-5), 87.99 (glu C-1), 107.40, 107.60, 107.64, 107.76, 107.83, 107.91, 108.28, 108.41 (HHDP C-3,3'), 110.35 (gailoyl C-2,6), 113.68, 113.97, 114.02, 114.07, 115.20, 115.35, 116.07, 116.40 (HHDP C-1,1'), 120.72, 120.89 (galloyl C-1), 124.88, 125.24, 125.97, 126.71, 126.79, 127.12, 127.31, 127.40 (HHDP C-2,2'), 136.06, 136.09, 136.14, 136.22, 136.30, 136.52, 136.92, 137.20 (HHDP C-5,5'), 139.36, 139.46 (galloyl C-4), 144.00, 144.31, 144.37, 144.50, 144.64, 144.70, 144.86, 144.97, 145.24, 145.31, 145.37, 145.41, 145.51 (HHDP C-4,4',6,6'), 146.02, 146.08 (galloyl C-3,5), 165.81, 165.88, 167.92, 168.01, 168.19, 168.38, 168.64, 168.72, 168.92, 168.98 (ester carbonyl), 194.61 (glu C-1).

Reduction of liquidambin. To a MeOH soln (1 ml) of liquidambin (12 mg) was added NaBH₄ (10 mg). The soln was left to stand for 10 min at room temp, and then it was acidified with HOAc (0.5 ml) and evapd. The residue was dissolved in H₂O (1 ml) and acidified to pH 2 with 10 % HCl. The aq. soln was passed through a SEP-PAK C18 cartridge (Waters Associates), and the cartridge was eluted with H₂O and then with MeOH. The MeOH eluate evapd to give 5-O-galloyl-2,3;4,6-di-O-(S)-hexawas hydroxydiphenoyl-D-glucitol (4) (11 mg). R_f on TLC, 0.46, $[\alpha]_D$ + 98° (MeOH; c 0.5). (Found: C, 50.28; H, 3.88. C41H30O26 · 2H2O requires C, 50.52; H, 3.52.) ¹H NMR: see text. Partial hydrolysis of liquidambin. An aq. soln (1 ml) of liquidambin (10 mg) was treated with tannase at 37° for 22 hr. The resulting soln was acidified to pH 2 with 10% HCl and then subjected to SEP-PAK C18 cartridge treatment eluting with H2O and then MeOH. The MeOH eluate was evapd and then subjected to prep. TLC to afford pedunculagin (2) (3 mg) [4].

Reduction of pedunculagin (2). A MeOH soln (1 ml) of pedunculagin (10 mg) was treated with NaBH₄ (10 mg) for 5 min. The soln was then acidified with HOAc (0.5 ml) and evapd. The residue was dissolved in H₂O (1 ml) and acidified to pH 2 with 10% HCl. The soln was subjected to CC on MCI gel CHP-20P using H₂O and then MeOH as eluants. The MeOH eluate was evapd to give 9 mg of 2,3;4,6-di-O-(S)-hexahydroxydiphenoyl-Dglucitol (5). R_f on TLC, 0.62, $[\alpha]_D$ + 186° (MeOH; c 0.5). (Found: C, 47.13; H, 3.85. C₃₄H₂₆O₂₂·4H₂O requires C, 47.56; H, 3.99.) ¹H NMR (Me₂CO-d₆ + D₂O): $\delta \delta \delta 5$ (s), $\delta \delta 4$ (2H, s), $\delta 52$ (s) (2 × HHDP), 5.61 (dd, J = 1.5, 9.5 Hz, glucitol H-3), 5.22 (dd, J = 1.5, 8.5 Hz, H-4), 5.05 (ddd, J = 2.5, 8.5 Hz, H-2), 4.70 (dd, J = 3, 12 Hz, H-6), 4.27 (dd, J = 2.5, 8.5 Hz, H-5), 3.97 (dd, J = 2.5, 13 Hz, H-1_a), 3.94 (d, J = 12 Hz, H-6_b), 3.94 (dd, J = 5, 13 Hz, H-1_b).

Transformation of liquidambin (3) into casuarinin (1). A mixture of liquidambin (10 mg), p-toluenesulphonic acid (5 mg) and dioxan (1 ml) was heated at 100° in a sealed tube for 1 hr and then evapd. The residue was dissolved in H_2O (1 ml) and subjected to a SEP-PAK C₁₈ cartridge treatment using H_2O and then MeOH as eluants. The MeOH eluate was further purified by prep. TLC to give casuarinin (1) (4 mg) [4].

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