



Photocyclization of α,β -Unsaturated Amide Aldehydes Synthesis of Jatropham

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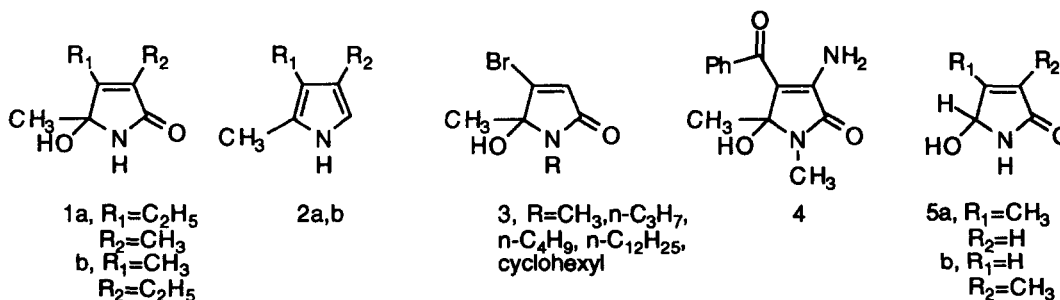
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Abstract: α,β -unsaturated amide aldehydes undergo photoisomerization and intramolecular cyclization to provide hydroxy substituted pyrrolidinone systems. A synthesis of the natural product jatropham was accomplished by application of this procedure.

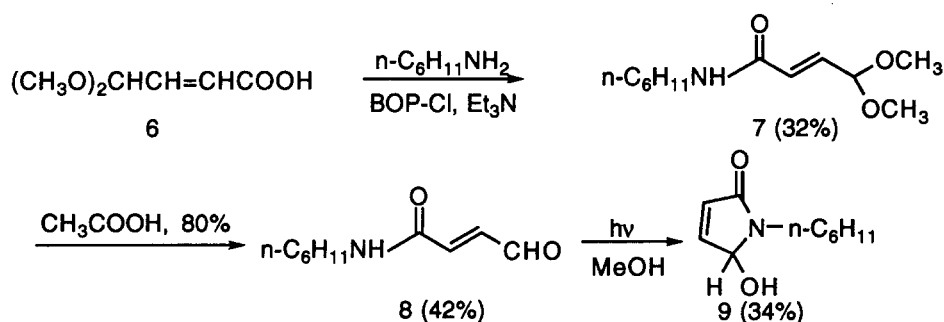
Derivatives of N-alkyl pyrrolidinones exhibit widespread and varied biological activity. Oxidized pyrroles **1** derived from kryptopyrroles **2a** reportedly induce behavioral, hypnotic and hypothermic changes. Compound **1a** is reported to induce porphyria and derivatives of hemopyrroles **2b** have been linked to psychiatric disorders. Thus compound **1b** was identified in the urine of schizophrenics and patients suffering from acute intermittent porphyria.¹ The structurally related 3-bromo-5-hydroxy-pyrrolin-2-ones **3** were evaluated for antineoplastic activity in tumor cells of white mice.² Derivatives such as **4** have been identified as metabolites of premarizepam, a pyrrolodiazepine which reportedly exhibits anxiolytic activity.³ The natural product jatropham, an antitumor alkaloid was isolated from *Jatropha macrorrhiza*, in 1973 by Cole *et al.* and assigned structure **5a**.^{4a} In 1980, Furukawa *et al.* suggested a revised structure for jatropham as 5-hydroxy-3-methyl-3-pyrrolin-2-one **5b**.^{4b} Subsequent reports on the syntheses of both 5-hydroxy-4-methyl-3-pyrrolin-2-one **5a** and 5-hydroxy-3-methyl-3-pyrrolin-2-one **5b** confirm the revised structure assignment.^{4c}



To date synthetic approaches to these 5-hydroxy-pyrrolin-2-ones have included sensitized photooxygenation of pyrroles⁵ and diazepines⁶, photooxidation of 2-furylcarbarnates⁷, conversion of lactones to lactams^{2a} and Grignard addition to maleimide derivatives.⁹ We recently reported that photolysis of γ -keto- α,β -unsaturated amides provides N-alkyl 5-methyl-5-hydroxy-3-pyrrolin-2-ones.¹⁰ Herein we detail an extension of

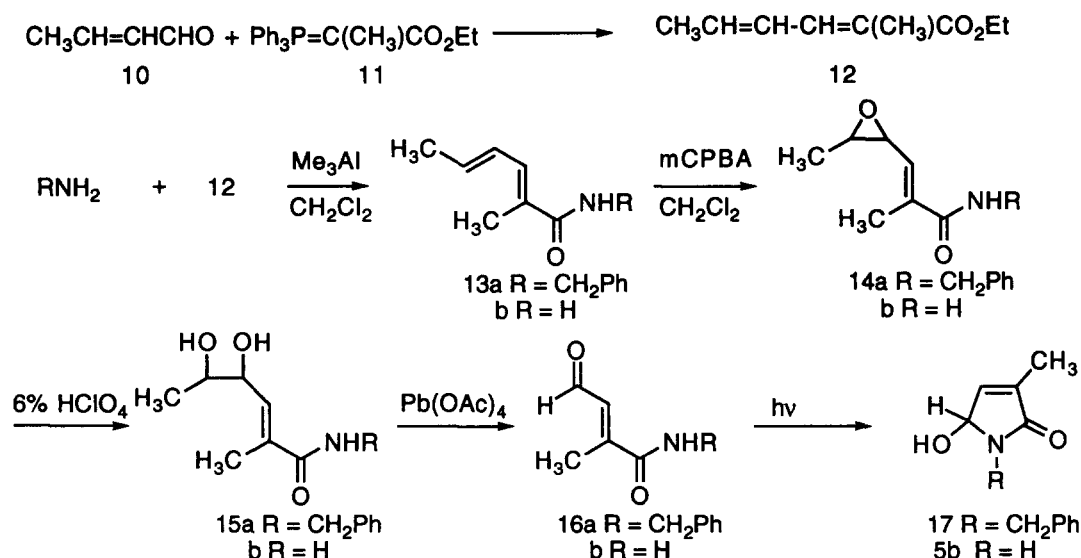
this procedure to include cyclization of α,β -unsaturated amide aldehydes. In demonstration of the utility of the photocyclization a total synthesis of the natural product jatropham has been completed.

Preparation of α,β -unsaturated amide aldehydes can be carried out from fumaraldehydic acid dimethyl acetal **6**. Compound **6** was prepared from furan via i) oxidative cleavage with bromine in methanol to provide fumaraldehyde bis(dimethyl acetal) (85 %) ii) hydrolysis with Amberlyst® 15 in acetone to give the monodimethyl acetal (99 %) and iii) oxidation with sodium chlorite-hydrogen peroxide to afford fumaraldehydic acid dimethyl acetal **6** (82 %).¹¹ Compound **6** was converted to the amide **7** in 32 % yield on treatment with *n*-hexylamine, triethylamine and bis(2-oxo-3-oxazolidinyl)-phosphinic chloride (BOP-Cl). Hydrolysis of the acetal **7** in 80 % acetic acid provided the amide aldehyde **8** (42 %). Irradiation of a solution of **8** in methanol through Pyrex provided the lactam **9** in 34 % yield after chromatography. Encouraged by the success of this photocyclization we turned our attention to the preparation of jatropham.



The synthesis of jatropham was approached via modification of ethyl 2-methyl sorbate **12** to provide the requisite unsaturated amide aldehyde photoprecursor. Compound **12** was prepared in 93 % yield via Wittig olefination of crotonaldehyde with **11**.¹² Aminolysis of **12** with trimethyl aluminum and benzyl amine provided **13a** (72 %). Selective epoxidation of **13a** was carried out with meta chloroperoxybenzoic acid in methylene chloride to provide **14a** (32 %). Hydrolysis of the epoxide with perchloric acid (6 %) provided the diol **15a** (71 %) which was treated with lead tetraacetate to afford aldehyde **16a** (90 %). Photolysis of aldehyde **16a** in methanol provided lactam **17** (50 %). However, attempts to convert **17** to **5b** via removal of the *N*-benzyl group were not successful. Accordingly **12** was transformed to the amide derivative **13b** with trimethylaluminum and ammonia (53 %). Epoxidation of **13b** with mCPBA gave **14b** (64 %) which was converted via the diol **15b** to the aldehyde **16b** (44 %). Photolysis of **16b** provided product (86 %) which had structural and physical data consistent with that reported for the natural product jatropham **5b**.¹³⁻¹⁴

In summary we have demonstrated that synthesis of 5-hydroxy-2-pyrrolidinones can be accomplished by photocyclization of either γ -keto- α,β -unsaturated amides¹⁰ or α,β -unsaturated amide aldehydes. The photocyclization can be carried out with either primary or secondary amides to provide either unsubstituted or *N*-alkyl substituted pyrrolidinone derivatives.



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13. All products gave spectral data (^1H NMR, IR, MS) which were consistent with the assigned structures. Satisfactory combustion analyses or high resolution mass spec data were obtained for all new products.
14. Physical and spectral data is provided for **16b** and **5b** (jatropham). Comparative literature data is also provided for jatropham:
Compound 16b white solid: mp 98-99°C; IR (film, CH_3CN) 3626, 3001, 2943, 1678, 1630 cm^{-1} ; ^1H NMR (CDCl_3 , 200 MHz) δ 2.37 (d, 3 H, $J = 1.5$ Hz), 5.75 (s, 2 H), 6.46 (dd, 1 H, $J = 7.3$ and 1.5 Hz), 10.16 (d, 1 H, $J = 7.3$ Hz); ^{13}C NMR (acetone- d_6 , 50.3 MHz) δ 13.6 (CH_3), 130.3 (CH), 151.4 (C), 170.6 (C=O), 193.0 (C=O); UV (EtOH) λ_{max} (e) 231 (4500), 204 (8100) nm; HRMS calcd for $\text{C}_5\text{H}_7\text{O}_2\text{N}$, 113.0475 found 113.0475.
Compound 5b (jatropham): mp 112-113 °C; IR (film): 3616, 3530, 3429, 1715, 1656 cm^{-1} ; ^1H NMR (acetone- d_6 , 200 MHz) δ 1.74 (m, 3 H), 4.85 (d, 1 H, $J = 9.0$ Hz), 5.45 (dd, 1 H, $J = 9.0$ and 1.6 Hz), 6.59 (m, 1 H), 7.45 (s, 1 H); ^{13}C NMR (acetone- d_6 , 50.3 MHz) δ 10.4 (CH_3), 79.1 (CH), 136.2 (C), 142.0 (CH), 173.4 (C=O); HRMS calcd for $\text{C}_5\text{H}_7\text{O}_2\text{N}$, 113.0475 found 113.0475.
Data reported for jatropham by Furukawa, H. et. al. Heterocycles 1981, 16, 1157: mp 115-118 °C; IR (CH_3CN) cm^{-1} 3600, 3520, 3420, 1715, 1645 cm^{-1} ; ^1H NMR (acetone- d_6) δ 1.78 (s, 3 H), 4.92 (d, 1H, $J = 8$ Hz), 5.48 (br d, $J = 8$ Hz), 6.59 (s, 1H), 7.54 (br, 1H); ^{13}C NMR (acetone- d_6) 10.41 (CH_3), 79.33 (CH), 136.25 (C), 142.16 (CH), 173.58 (C=O); UV (EtOH) λ_{max} 230 nm.

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