

Short communication

## Synthesis and antifungal activity of some 3-benzylidenechroman-4-ones, 3-benzylidenethiochroman-4-ones and 2-benzylidene-1-tetralones

T Al Nakib<sup>1</sup>, V Bezjak<sup>1</sup>, MJ Meegan<sup>2\*</sup>, R Chandy<sup>1</sup>

<sup>1</sup>Department of Pharmacology and Microbiology, Faculty of Medicine, University of Kuwait, PO Box 24923, Safat 13110, Kuwait

<sup>2</sup>Department of Pharmaceutical Chemistry, Trinity College Dublin, 18 Shrewsbury Road, Dublin 4, Ireland

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**Summary** — A series of 3-benzylidenechroman-4-ones, 3-benzylidenethiochroman-4-ones and 2-benzylidene-1-tetralones have been synthesised and tested for their *in vitro* antifungal activity. Some of them were inactive; others displayed good activity against *Cryptococcus neoformans*, *Torulopsis glabrata* and *Trichosporon cutaneum*.

**Résumé** — Synthèse et activité fongicide de quelques 3-benzylidène-chroman-4-ones, 3-benzylidène-thiochroman-4-ones, et 2-benzylidène-1-tétralones. Une série de 3-benzylidène-chromanones-4, 3-benzylidène-thiochroman-4-ones et 2-benzylidène-1-tétralones diversement substituées a été synthétisée et essayée *in vitro* comme fongicides. Certains de ces produits ont montré une activité intéressante vis-à-vis de *Cryptococcus neoformans*, *Torulopsis glabrata* et *Trichosporon cutaneum* et semblent mériter une étude plus approfondie.

3-benzylidenechroman-4-ones / 3-benzylidenethiochroman-4-ones / 2-benzylidene-1-tetralones / antifungal activity

### Introduction

The most important antifungal agents currently available include flucytosine, griseofulvin, the polyene macrocycles nystatin and amphotericin B and the azole based antifungal agents of which miconazole [1] ketoconazole [2] and fluconazole [3] are examples [4, 5].

We wish to describe herein the preparation and antifungal activity of some 3-benzylidenechroman-4-ones, 3-benzylidenethiochroman-4-ones and 2-benzylidene-1-tetralones against 3 *Candida* species (*C. albicans*, *C. parapsilosis* and *C. tropicalis*), 3 *Cryptococcus* strains (Farwaria, 25 and 93/V), *Torulopsis glabrata* and *Trichosporon cutaneum*. These yeast-like fungi are important pathogens to man and cause a variety of chronic superficial and systemic diseases [6]. 3-Benzylidenechroman-4-ones (homo-isoflavones) have been isolated from members of the Liliaceae family [7, 8]. Révisé and Kirkiacharian have analysed the effect of a series of natural and synthetic 3-benzylidenechroman-4-ones on *Phytophthora parasitica* and on the activity of some of its enzymes. They have found inhibition of *in vitro* growth and sporo-

genesis of the micro-organism [9]. The enzyme  $\beta$ -glucosidase and 5 pectinolytic enzymes (*endo* PET/PATE and *endo* PMG/PG) which are directly involved in the infection mechanism are inhibited to a varying extent [8, 9].

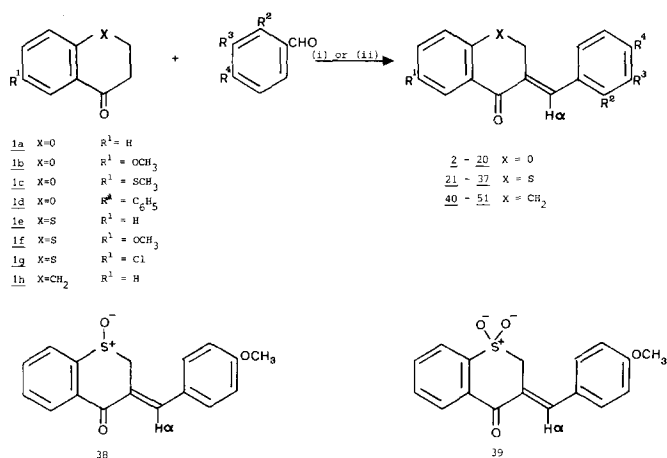
Analgesic, antiinflammatory and platelet anti-aggregating activities have been reported for some 3-benzylidenechroman-4-one derivatives [10].

### Chemistry

The chroman-4-ones **1a–d** and thiochroman-4-ones (**1e–g**) required for the present study were prepared using the available literature methods [11, 12].

The 3-benzylidenechroman-4-ones (**2–20**) were synthesised by acid catalysed condensation of the chroman-4-ones **1a–1d** with the appropriately substituted aryl aldehyde (scheme 1) [13, 14]. The yields and melting point data are displayed in table II. In the <sup>1</sup>H NMR spectrum of the 3-benzylidenechroman-4-ones (**2–20**) the vinyl proton H- $\alpha$  appears as a triplet in the range  $\delta$  7.81–8.12 which is coupled with the C-2 aliphatic protons which occur as a doublet in the

\*Correspondence and reprints



**Scheme 1.** Synthesis of 3-benzylidenechroman-4-ones, 3-benzylidenethiochroman-4-ones and 2-benzylidene-1-tetralones. Reagents (i): HCl, EtOH for compounds 2–37, (ii): NaOH, EtOH for compounds 40–51.

range  $\delta$  5.18–5.48 ( $J' = 1.5$ –2.0 Hz) [15, 18]. In the IR spectrum the  $\alpha$ ,  $\beta$ -unsaturated ketone is observed at 1635–1668  $\text{cm}^{-1}$ , typical of the *E* configuration [19].

The 3-benzylidenethiochroman-4-ones (21–37) were prepared by acid catalysed condensation of the thiochroman-4-ones (1e–1g) with the appropriately substituted aryl aldehyde [20]. The yield and melting point data for the 3-benzylidenethiochroman-4-ones prepared are displayed in table I. In the IR spectrum conjugated carbonyl absorption occurs at 1665–1675  $\text{cm}^{-1}$  typical of the *E* configuration [21]. The  $^1\text{H}$  NMR spectrum shows that only one isomer was obtained in each case and supports the proposed *E* configuration of the products. H- $\alpha$  appears downfield as a triplet at  $\delta$  7.71–7.93. The C-2 methylene protons occur as a doublet ( $\delta$  4.12–4.26 (S–CH<sub>2</sub>),  $J = 1.2$ –2.0 Hz).

The sulfoxide 38 was prepared by treatment of 22 with hydrogen peroxide (1.1 mol) at 0°C in glacial acetic acid. The IR spectrum of 38 showed strong sulfoxide bands at 996 and 1030  $\text{cm}^{-1}$ , and  $\alpha$ ,  $\beta$ -unsaturated carbonyl at 1670  $\text{cm}^{-1}$ . The  $^1\text{H}$  NMR spectrum displayed H- $\alpha$  as a double doublet ( $\delta$  7.80, 7.83  $J = 5.0, 1.2$  Hz). The C-2 protons appeared as 2 doublets at  $\delta$  4.40 (H-2a,  $J_{ab} = 1.9$  Hz,  $J_{a,\alpha} = 1.2$  Hz)  $\delta$  4.69 (H-2b,  $J_{ba} = 1.9$  Hz,  $J_{b,\alpha} = 5.0$  Hz) [22]. Oxidation of 3-(4'-methoxybenzylidene)thiochroman-4-one 22 with excess hydrogen peroxide in glacial acetic acid at 25°C afforded the sulfone 39 [23]. The IR spectrum showed asymmetric sulfone stretch at 1385, symmetric sulfone stretch at 1195  $\text{cm}^{-1}$  and  $\alpha$ ,  $\beta$ -unsaturated carbonyl absorption at 1670  $\text{cm}^{-1}$ . In the  $^1\text{H}$  NMR spectrum the H-2 protons were observed as a doublet at  $\delta$  5.39, while H- $\alpha$  appeared as a triplet at  $\delta$  7.86 ( $J = 4.8$  Hz).

The 2-benzylidene-1-tetralones 40–51 were prepared by base catalysed condensation of 1-tetralone 1h with the appropriately substituted aryl aldehydes (scheme 1) [24]. In the case of compounds 41, 44 and 50 the condensation reaction was carried out under acidic conditions to obtain reasonably good yields (table I). The IR spectra of the 2-benzylidene-1-tetralones 42–53 showed  $\alpha$ ,  $\beta$ -unsaturated carbonyl absorption at 1665–1675  $\text{cm}^{-1}$  [19]. In the  $^1\text{H}$  NMR spectrum H- $\alpha$  appears as a multiplet at  $\delta$  7.83–7.95 with long range coupling to the C-3 methylene protons. The H-8 signal appears as a multiplet at  $\delta$  8.10–8.25 due to the de-shielding influence of the carbonyl group. The C-3 and C-4 methylene protons appear as a complex multiplet centered at  $\delta$  2.85–3.14.

## Results and Discussion

The MIC values ( $\mu\text{g/ml}$ ) obtained for the 3-benzylidenechroman-4-ones 2–20, 3-benzylidenethiochroman-4-ones 21–39 and 3-benzylidene-1-tetralones 40–51 are displayed in tables II, III and IV, respectively. Of the 3-benzylidenechroman-4-ones examined, compounds 3, 8, 10, 11 and 12 showed good activity against *Cryptococcus neoformans* MIC  $\leq 25$   $\mu\text{g/ml}$  and indicated that the presence of a methoxy substituent at C-6 and methoxy, hydroxy or chloro substituents at C-4' were required for effective antifungal activity of these compounds. Compound 10 was also active against *Candida albicans* (MIC = 6  $\mu\text{g/ml}$ ), while compounds 5, 12 and 14 showed good activity against *Torulopsis glabrata* (MIC = 3  $\mu\text{g/ml}$ ) (table II).

Of the 3-benzylidenethiochroman-4-ones tested, only compounds 22 and 25 displayed significant activity against *Candida albicans* (MIC = 6  $\mu\text{g/ml}$ ). Many of the compounds displayed useful activity against *Cryptococcus neoformans*, particularly compounds 32, 33, 34 and 35 all of which contain a chloro substituent at C-6. The sulfoxide 38 together with compounds 22, 24 and 25 were active against *Torulopsis glabrata* (MIC = 6  $\mu\text{g/ml}$ ) while compounds 24, 33 and 38 showed useful activity against *Tri-chosporon cutaneum* (table III).

The 2-benzylidene-1-tetralones 40–51 were generally effective against *Torulopsis glabrata*; however only compounds 46, 48 and 51 showed good performance against *Cryptococcus neoformans* with MIC values of  $\leq 12.5$   $\mu\text{g/ml}$  (table IV). The MIC values ( $\mu\text{g/ml}$ ) obtained for the antifungal agents miconazole nitrate, tolnaftate, nystatin and amphotericin B against the organisms used in the study are also shown in table IV. These organisms were shown to be resistant to miconazole nitrate and tolnaftate (MIC = 100  $\mu\text{g/ml}$ ) but were susceptible to nystatin and amphotericin B.

**Table I.** 3-Benzylidenechroman-4-ones, 3-benzylidenethiochroman-4-ones and 2-benzylidene-1-tetralones.

Compound	X	R <sup>1</sup>	R <sup>2</sup>	R <sup>3</sup>	R <sup>4</sup>	Yield (%)	mp (°C)	Formula
<u>2</u>	O	H	OH	H	H	8.5	168-169	C <sub>16</sub> H <sub>12</sub> O <sub>3</sub>
<u>3</u>	O	H	H	H	OH	55.7	230-231 [25]	C <sub>16</sub> H <sub>12</sub> O <sub>3</sub>
<u>4</u>	O	H	OCH <sub>3</sub>	H	H	61.1	106-107 [26]	C <sub>17</sub> H <sub>14</sub> O <sub>3</sub>
<u>5</u>	O	H	H	H	OCH <sub>3</sub>	71.8	133-134 [14]	C <sub>17</sub> H <sub>16</sub> O <sub>4</sub>
<u>6</u>	O	H	H	OCH <sub>3</sub>	OH	53.8	124-125	C <sub>17</sub> H <sub>14</sub> O <sub>4</sub>
<u>7</u>	O	H	H	OCH <sub>3</sub>	OCH <sub>3</sub>	59.6	128-129 [27]	C <sub>18</sub> H <sub>16</sub> O <sub>4</sub>
<u>8</u>	O	H	H	H	Cl	81.3	167-169 [10]	C <sub>16</sub> H <sub>11</sub> ClO <sub>2</sub>
<u>9</u>	O	H	H	H	CH <sub>3</sub>	88.0	117-118 [10]	C <sub>17</sub> H <sub>14</sub> O <sub>2</sub>
<u>10</u>	O	OCH <sub>3</sub>	OCH <sub>3</sub>	H	H	52.9	145-146	C <sub>18</sub> H <sub>16</sub> O <sub>4</sub>
<u>11</u>	O	OCH <sub>3</sub>	H	H	OCH <sub>3</sub>	79.3	132-133 [28]	C <sub>18</sub> H <sub>16</sub> O <sub>4</sub>
<u>12</u>	O	OCH <sub>3</sub>	H	OCH <sub>3</sub>	OCH <sub>3</sub>	64.5	157-159 [28]	C <sub>19</sub> H <sub>18</sub> O <sub>5</sub>
<u>13</u>	O	OCH <sub>3</sub>	H	OCH <sub>2</sub> O		65.7	150-151	C <sub>18</sub> H <sub>14</sub> O <sub>5</sub>
<u>14</u>	O	OCH <sub>3</sub>	3-(2-naphthylidene)			65.8	130-131	C <sub>21</sub> H <sub>16</sub> O <sub>3</sub>
<u>15</u>	O	OCH <sub>3</sub>	3=CH-CH=CH-C <sub>6</sub> H <sub>5</sub>			61.3	149-150	C <sub>19</sub> H <sub>16</sub> O <sub>3</sub>
<u>16</u>	O	SCH <sub>3</sub>	H	H	OCH <sub>3</sub>	32.6	108-109	C <sub>18</sub> H <sub>16</sub> O <sub>3</sub> S
<u>17</u>	O	C <sub>6</sub> H <sub>5</sub>	OCH <sub>3</sub>	H	H	70.2	159-160	C <sub>23</sub> H <sub>18</sub> O <sub>3</sub>
<u>18</u>	O	C <sub>6</sub> H <sub>5</sub>	H	H	OCH <sub>3</sub>	74.3	178-179	C <sub>23</sub> H <sub>18</sub> O <sub>3</sub>
<u>19</u>	O	C <sub>6</sub> H <sub>5</sub>	H	OCH <sub>3</sub>	OCH <sub>3</sub>	60.1	171-172	C <sub>24</sub> H <sub>20</sub> O <sub>4</sub>
<u>20</u>	O	C <sub>6</sub> H <sub>5</sub>	H	OCH <sub>2</sub> O		61.7	181-182	C <sub>23</sub> H <sub>16</sub> O <sub>4</sub>
<u>21</u>	S	H	OCH <sub>3</sub>	H	H	63.6	101-102	C <sub>17</sub> H <sub>14</sub> O <sub>2</sub> S
<u>22</u>	S	H	H	H	OCH <sub>3</sub>	62.4	136-138	C <sub>17</sub> H <sub>14</sub> O <sub>2</sub> S
<u>23</u>	S	H	H	OCH <sub>3</sub>	OCH <sub>3</sub>	50.7	126-128 [29]	C <sub>18</sub> H <sub>16</sub> O <sub>3</sub> S
<u>24</u>	S	H	H	OCH <sub>3</sub>	OH	67.8	146-147	C <sub>17</sub> H <sub>12</sub> O <sub>3</sub> S
<u>25</u>	S	H	H	OCH <sub>2</sub> O		32.3	152-154	C <sub>17</sub> H <sub>14</sub> O <sub>3</sub> S

Table I. (Continued)

Compound	X	R <sup>1</sup>	R <sup>2</sup>	R <sup>3</sup>	R <sup>4</sup>	Yield (%)	mp (°C)	Formula
<u>26</u>	S	H	H	H	CH <sub>3</sub>	64.1	118-119 [29]	C <sub>17</sub> H <sub>14</sub> OS
<u>27</u>	S	H	H	H	C <sub>6</sub> H <sub>5</sub>	52.1	143-145	C <sub>22</sub> H <sub>16</sub> OS
<u>28</u>	S	H	3-(2-naphthylidene)			67.8	146-147	C <sub>20</sub> H <sub>14</sub> OS
<u>29</u>	S	H	H	H	Cl	66.3	138-139 [29]	C <sub>16</sub> H <sub>11</sub> Cl OS
<u>30</u>	S	OCH <sub>3</sub>	H	H	OCH <sub>3</sub>	41.3	169-170	C <sub>18</sub> H <sub>16</sub> O <sub>3</sub> S
<u>31</u>	S	OCH <sub>3</sub>	H	OCH <sub>2</sub> O		21.7	152-153	C <sub>18</sub> H <sub>14</sub> O <sub>4</sub> S
<u>32</u>	S	Cl	H	H	H	25.6	112-113	C <sub>16</sub> H <sub>11</sub> Cl OS
<u>33</u>	S	Cl	OCH <sub>3</sub>	H	H	41.6	145-147	C <sub>17</sub> H <sub>13</sub> ClO <sub>2</sub> S
<u>34</u>	S	Cl	H	H	OCH <sub>3</sub>	36.7	151-152	C <sub>17</sub> H <sub>13</sub> ClO <sub>2</sub> S
<u>35</u>	S	Cl	H	OCH <sub>3</sub>	OCH <sub>3</sub>	30.9	125-126	C <sub>18</sub> H <sub>15</sub> ClO <sub>3</sub> S
<u>36</u>	S	Cl	H	OCH <sub>2</sub> O		31.8	168-169	C <sub>17</sub> H <sub>11</sub> ClO <sub>3</sub> S
<u>37</u>	S	Cl	3-(2-naphthylidene)			46.9	165-167	C <sub>20</sub> H <sub>13</sub> Cl OS
<u>38</u>	SO	H	H	H	OCH <sub>3</sub>	49.2	199-201	C <sub>17</sub> H <sub>14</sub> O <sub>3</sub> S
<u>39</u>	SO <sub>2</sub>	H	H	H	OCH <sub>3</sub>	48.0	158-159 [23]	C <sub>17</sub> H <sub>14</sub> O <sub>4</sub> S
<u>40</u>	CH <sub>2</sub>	H	H	H	H	90.3	120-121 [19]	C <sub>17</sub> H <sub>14</sub> O
<u>41</u>	CH <sub>2</sub>	H	H	H	OH	9.3	196-197	C <sub>17</sub> H <sub>14</sub> O <sub>2</sub>
<u>42</u>	CH <sub>2</sub>	H	H	H	OCH <sub>3</sub>	78.3	87-88 [30]	C <sub>18</sub> H <sub>16</sub> O <sub>2</sub>
<u>43</u>	CH <sub>2</sub>	H	H	OCH <sub>3</sub>	OCH <sub>3</sub>	77.2	99-101 [31]	C <sub>19</sub> H <sub>18</sub> O <sub>3</sub>
<u>44</u>	CH <sub>2</sub>	H	H	OH	OH	20.5	225-127	C <sub>17</sub> H <sub>14</sub> O <sub>3</sub>
<u>45</u>	CH <sub>2</sub>	H	H	OCH <sub>2</sub> O		78.5	107-109	C <sub>18</sub> H <sub>16</sub> O <sub>3</sub>
<u>46</u>	CH <sub>2</sub>	H	H	H	OC <sub>2</sub> H <sub>5</sub>	52.1	128-129	C <sub>19</sub> H <sub>18</sub> O <sub>2</sub>
<u>47</u>	CH <sub>2</sub>	H	H	H	CH <sub>3</sub>	98.1	121-122 [32]	C <sub>18</sub> H <sub>16</sub> O
<u>48</u>	CH <sub>2</sub>	H	H	H	C <sub>2</sub> H <sub>5</sub>	61.0	75-76	C <sub>19</sub> H <sub>18</sub> O
<u>49</u>	CH <sub>2</sub>	H	NO <sub>2</sub>	H	H	1.7	120-122 [19]	C <sub>17</sub> H <sub>13</sub> NO <sub>3</sub>
<u>50</u>	CH <sub>2</sub>	H	H	H	F	87.3	110-112 [32]	C <sub>17</sub> H <sub>13</sub> F O
<u>51</u>	CH <sub>2</sub>	H	H	H	Cl	96.8	132-133 [34]	C <sub>17</sub> H <sub>13</sub> Cl O

**Table II.** MIC values ( $\mu\text{g/ml}$ ) of 3-benzylidenechroman-4-ones **2–20** against *Cryptococcus neoformans* (3 strains), *Candida albicans*, *Candida parapsilosis*, *Candida tropicalis*, *Torulopsis glabrata* and *Trichosporon cutaneum*.

Compound	Farwania	<i>Cryptococcus neoformans</i> 25	93/v	<i>Candida albicans</i>	<i>Candida parapsilosis</i>	<i>Candida tropicalis</i>	<i>Torulopsis glabrata</i>	<i>Trichosporon cutaneum</i>
<u>2</u>	12.5	12.5	25.0	12.5	50.0	25.0	6.0	25.0
<u>3</u>	6.0	6.0	6.0	12.5	25.0	25.0	12.5	12.5
<u>4</u>	25.0	25.0	50.0	50.0	50.0	25.0	12.5	50.0
<u>5</u>	100.0	50.0	50.0	50.0	50.0	50.0	3.0	25.0
<u>6</u>	25.0	12.5	25.0	50.0	50.0	50.0	6.0	50.0
<u>7</u>	25.0	25.0	50.0	50.0	50.0	50.0	6.0	50.0
<u>8</u>	6.0	6.0	12.5	25.0	100.0	50.0	25.0	50.0
<u>9</u>	12.5	12.5	25.0	12.5	50.0	50.0	12.5	50.0
<u>10</u>	3.0	3.0	6.0	6.0	50.0	50.0	50.0	50.0
<u>11</u>	6.0	6.0	25.0	25.0	50.0	50.0	12.5	25.0
<u>12</u>	1.5	1.5	1.5	25.0	25.0	25.0	3.0	50.0
<u>13</u>	12.5	25.0	25.0	25.0	50.0	50.0	50.0	50.0
<u>14</u>	12.5	12.5	12.5	12.5	25.0	25.0	3.0	25.0
<u>15</u>	12.5	25.0	25.0	25.0	50.0	50.0	25.0	50.0
<u>16</u>	12.5	25.0	50.0	25.0	50.0	50.0	25.0	50.0
<u>17</u>	50.0	25.0	50.0	50.0	50.0	50.0	50.0	25.0
<u>18</u>	25.0	25.0	50.0	25.0	50.0	25.0	12.5	25.0
<u>19</u>	100.0	50.0	50.0	50.0	100.0	25.0	6.0	50.0
<u>20</u>	50.0	100.0	50.0	50.0	50.0	50.0	25.0	50.0

**Table III.** MIC values ( $\mu\text{g/ml}$ ) of 3-benzylidenechroman-4-ones **21–39** against *Cryptococcus neoformans* (3 strains), *Candida albicans*, *Candida parapsilosis*, *Candida tropicalis*, *Torulopsis glabrata* and *Trichosporon cutaneum*.

Compound	Farwania	<i>Cryptococcus neoformans</i> 25	93/v	<i>Candida albicans</i>	<i>Candida parapsilosis</i>	<i>Candida tropicalis</i>	<i>Torulopsis glabrata</i>	<i>Trichosporon cutaneum</i>
<u>21</u>	25.0	25.0	25.0	25.0	50.0	50.0	12.5	12.5
<u>22</u>	25.0	3.0	6.0	6.0	25.0	25.0	6.0	12.5
<u>23</u>	100.0	50.0	100.0	100.0	100.0	50.0	25.0	12.5
<u>24</u>	50.0	25.0	6.0	50.0	25.0	25.0	6.0	1.5
<u>25</u>	12.5	1.5	6.0	6.0	50.0	50.0	6.0	25.0
<u>26</u>	100.0	50.0	50.0	50.0	100.0	50.0	12.5	25.0
<u>27</u>	25.0	12.5	6.0	25.0	50.0	50.0	12.5	25.0
<u>28</u>	12.5	12.5	25.0	25.0	100.0	50.0	25.0	25.0
<u>29</u>	12.5	3.0	12.5	50.0	50.0	50.0	25.0	25.0
<u>30</u>	12.5	12.5	25.0	25.0	100.0	50.0	12.5	12.5
<u>31</u>	25.0	12.5	25.0	50.0	50.0	50.0	25.0	12.5
<u>32</u>	6.0	6.0	50.0	50.0	50.0	50.0	25.0	50.0
<u>33</u>	12.5	1.5	12.5	12.5	25.0	12.5	12.5	3.0
<u>34</u>	6.0	6.0	12.5	25.0	50.0	50.0	25.0	25.0
<u>35</u>	6.0	6.0	12.5	50.0	50.0	50.0	25.0	50.0
<u>36</u>	12.5	12.5	12.5	25.0	50.0	50.0	25.0	25.0
<u>37</u>	12.5	50.0	50.0	50.0	50.0	50.0	50.0	25.0
<u>38</u>	50.0	50.0	25.0	25.0	25.0	50.0	6.0	6.0
<u>39</u>	25.0	25.0	25.0	12.5	50.0	50.0	12.5	25.0

**Table IV.** MIC values ( $\mu\text{g/ml}$ ) of 2-benzylidene-1-tetralones **40–51** against *Cryptococcus neoformans* (3 strains), *Candida albicans*, *Candida parapsilosis*, *Candida tropicalis*, *Torulopsis glabrata* and *Trichosporon cutaneum*.

Compound	<i>Cryptococcus neoformans</i>			<i>Candida albicans</i>	<i>Candida parapsilosis</i>	<i>Candida tropicalis</i>	<i>Torulopsis glabrata</i>	<i>Trichosporon cutaneum</i>
	Farwania	25	93/V					
<b>40</b>	25.0	25.0	25.0	50.0	50.0	50.0	12.5	25.0
<b>41</b>	100.0	100.0	100.0	100.0	100.0	50.0	12.5	25.0
<b>42</b>	100.0	100.0	100.0	50.0	100.0	50.0	25.0	25.0
<b>43</b>	25.0	25.0	6.0	25.0	50.0	50.0	6.0	25.0
<b>44</b>	100.0	100.0	100.0	50.0	100.0	50.0	25.0	50.0
<b>45</b>	100.0	50.0	50.0	50.0	100.0	50.0	6.0	50.0
<b>46</b>	12.5	12.5	25.0	50.0	50.0	12.5	12.5	25.0
<b>47</b>	100.0	50.0	25.0	50.0	100.0	50.0	3.0	25.0
<b>48</b>	12.5	12.5	12.5	25.0	50.0	50.0	12.5	50.0
<b>49</b>	100.0	100.0	100.0	100.0	100.0	100.0	12.5	50.0
<b>50</b>	25.0	25.0	50.0	50.0	100.0	50.0	12.5	50.0
<b>51</b>	6.0	6.0	6.0	12.5	50.0	50.0	12.5	25.0
Tolnaftate	100	100	100	100	100	100	100	100
Miconazole nitrate	100	100	100	100	100	100	100	100
Nystatin	50.0	3.0	25.0	12.5	1.5	3.0	6.0	6.0
Amphotericin B	6.0	1.5	3.0	3.0	50.0	25.0	3.0	25.0

In conclusion, many of the 3-benzylidenechroman-4-ones, 3-benzylidenethiochroman-4-ones and 2-benzylidene-1-tetralones synthesised were found to be active *in vitro* against the pathogenic fungi *Cryptococcus neoformans*, *Torulopsis glabrata* and *Trichosporon cutaneum*. Further modification of these compounds are in progress.

## Experimental protocols

### Chemistry

Melting points were determined on a Gallenkamp melting point apparatus and were not corrected. The infra-red spectra were recorded on a Perkin–Elmer SP3-300 spectrometer. The ultraviolet spectra were recorded on a Pye–Umicam SP8-100 spectrometer. The  $^1\text{H}$  NMR spectra were obtained on a Perkin–Elmer R12B spectrometer operating at 60 MHz with tetramethylsilane (TMS) as internal standard. Separations by column chromatography were carried out using Merck Kieselgel 60 silica gel while thin layer chromatography was carried out on pre-coated Merck HF silica-gel plates. Analyses of all new compounds in the text were within  $\pm 0.4\%$  of the theoretical values for C, H and N.

The chroman-4-ones **1a** [35], **1b** [36], **1c** [37], **1d** [38] and the thiochroman-4-ones **1e** [39], **1f** [40] and **1g** [40] were prepared according to the literature procedures [9–12].

The experimental procedures are exemplified as follows:

#### 3-(4'-Hydroxy-3'-methoxybenzylidene)chroman-4-one **6**

Dry hydrogen chloride gas was passed through a solution of 4-hydroxy-3-methoxybenzaldehyde (15.2 g, 0.1 mol) and

chroman-4-one (11.84 g, 0.08 mol) in ethanol (20 ml) for 15 min. The reaction mixture was allowed to stand at room temperature for 24 h. The precipitated product was then filtered, dried and recrystallized from ethanol (12.13 g, 43.8%) mp = 125°C.  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta$ : 8.10 (1H, m, H-5), 7.93 (1H, m, H- $\alpha$ ), 6.78–7.39 (6H, m, aromatic H), 5.40 (2H, d,  $J$  = 1.8 Hz, H-2), 3.87 (3H, s,  $-\text{OCH}_3$ ). IR (KBr)  $\text{cm}^{-1}$ : 3280 (OH), 1665 (C=O), 1595. Anal  $\text{C}_{17}\text{H}_{14}\text{O}_4$  = 282.28.

#### 6-Chloro-3-(2'-methoxybenzylidene)thiochroman-4-one **33**

Dry hydrogen chloride gas was passed through a solution of 2-methoxybenzaldehyde (0.68 g, 0.005 mol) and 6-chloro-thiochroman-4-one (0.99 g, 0.005 mol) in ethanol (20 ml) for 5 min. The solution was cooled, and the precipitated product was then filtered off, dried and recrystallised from ethanol (0.65 g, 41.6%) mp = 145–147°C.  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta$ : 8.41 (1H, d,  $J$  = 1.8 Hz, H-5), 7.86 (1H, m, H- $\alpha$ ), 6.86–7.65 (6H, m, aromatic H), 4.17 (2H, d,  $J$  = 2 Hz, H-2), 3.86 (3H, s,  $\text{OCH}_3$ ). IR (KBr)  $\text{cm}^{-1}$ : 1665 (C=O), 1585. Anal  $\text{C}_{17}\text{H}_{13}\text{ClO}_2\text{S}$  = 316.78.

#### 2-(3',4'-Methylenedioxybenzylidene)-1-tetralone **45**

A mixture of 3,4-methylenedioxybenzaldehyde (3.8 g, 0.02 mol) and  $\alpha$ -tetralone (2.92 g, 0.02 mol) was treated with ethanolic potassium hydroxide solution (20 ml, 4%) and stirred at 25°C for 16 h. The mixture was neutralised with acetic acid (20 ml) and water (10 ml) was added. The precipitated product was filtered off and recrystallised from ethanol. (4.4 g, 78.5%) mp = 107–108°C.  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta$ : 8.2 (1H, m, H-8), 7.95 (1H, m, H- $\alpha$ ), 6.8–7.5 (6H, m, aromatic H), 6.15 (2H, s,  $-\text{OCH}_2\text{O}-$ ), 2.9–3.1 (4H, m,  $-\text{CH}_2\text{CH}_2-$ ). IR (KBr)  $\text{cm}^{-1}$ : 1670, 1608. Anal  $\text{C}_{18}\text{H}_{16}\text{O}_3$  = 280.30.

#### 3-(4'-Methoxybenzylidene)thiochroman-4-one-1-oxide **38**

3-(4'-Methoxybenzylidene)thiochroman-4-one (0.71 g, 0.0025 mol) was dissolved in glacial acetic acid (5 ml) and the solution

cooled to 0°C. Hydrogen peroxide (0.2 ml, 35%, 0.0025 mol) was added dropwise over 30 min with stirring at 0–5°C. The mixture was stirred for 2 h at 0°C and then for 12 h at room temperature. The precipitated product was filtered off, and crystallised from benzene as needles, (0.37 g, 49.2%) mp = 199–201°C. <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ: 8.21 (1H, m, H-5), 7.81 (1H, dd, *J* = 5 Hz, 1.2 Hz, H-α), 7.38 (5H, m, aromatic H), 6.90 (2H, d, *J* = 7 Hz, H-3', H-5'), 4.69 (1H, dd, *J* = 1.2 Hz, 1.9 Hz, H-2 eq), 4.40 (1H, dd, *J* = 5 Hz, 1.9 Hz, H-2 ax), 3.94 (3H, s, 4'-OCH<sub>3</sub>). IR (KBr) cm<sup>-1</sup>: 1670 (C=O), 1601, 1030, 996. Anal C<sub>17</sub>H<sub>14</sub>O<sub>3</sub>S = 298.34.

### 3-(4'-Methoxybenzylidene)thiochroman-4-one-1,1-dioxide 39

This compound was prepared from 3-(4'-methoxybenzylidene)-thiochroman-4-one according to the literature procedure [23] mp = 158–159°C (lit [23] 161°C). <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ: 8.26 (1H, m, H-5), 7.86 (1H, d, *J* = 4.8 Hz, H-α), 7.41 (5H, m, aromatic H), 6.94 (2H, d, *J* = 7.3 Hz, H-3', H-5'), 5.39 (2H, d, *J* = 4.8 Hz, H-2), 3.95 (3H, s, -OCH<sub>3</sub>). IR (KBr) cm<sup>-1</sup>: 1670 (C=O), 1601, 1385, 1195 (SO<sub>2</sub>). Anal C<sub>17</sub>H<sub>14</sub>O<sub>4</sub>S = 314.34.

### Antimycotic activity

The *in vitro* antimycotic activity of tested compounds was determined against a series of yeasts and fungi and has been evaluated through the minimum inhibitory concentration (MIC) according to the method of progressive double dilutions in liquid Casitone medium [41].

The yeasts were originally clinical isolates obtained from Kuwait, and were typed by conventional methods [42, 43]. The cells were maintained by a periodic subculture on malt agar (Oxoid) slants. MIC determinations were performed by a microtitre technique on freshly subcultured 2-day-old cells from slants. One (1 mm) loopful of the freshly cultured cells was suspended in sterile distilled water (10 ml). The seeding rate was adjusted by successive dilution of this stock solution with sterile distilled water until the optical density reading of the solution is 0.05 (530 nm) [44, 45]. 25 µl of this diluted cell suspension is then added to the wells in the autotray-microtitre plate already containing liquid casitone medium (Difco; 20%, 0.1 mol). The compounds were then added to the wells in the concentration range of 0.1–100 µg/ml diluted with sterile distilled water from a 1 mg/ml stock solution of the pure compounds in DMSO (0.1 ml), with the highest content of DMSO in any well at 2.5%. The cells were incubated at 30°C for 24 h (for *Candida* spp and *Trichosporon cutaneum*) and 48 h for *Cryptococcus neoformans* strains and *Tolulopsis glabrata*) and the MIC reading recorded at 24 h and 48 h respectively using a colony reader compared to control cultures incubated on the same plate at the same conditions above, and in duplicate sets.

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