Formation of 3H-1,3-Benzodiazepines from Quinoline N-Acylimides

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Summary Photolysis of the quinoline N-imides (3) having an electron-donating substituent in the 6- or 8-position

affords the corresponding 3H-1,3-benzodiazepines (4), whereas quinolines having an electron-donating group in

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the other positions or an electron-withdrawing group give no diazepines.

It is known that the quinoline and related fused pyridine N-imides (1a) and (2a) undergo photo-induced rearrangement to give the corresponding fused 1H-1,2-diazepines,1,2 whereas the N-acylimides (1b) and (2b) $(X = CO_2Et,$ COPh, or Ac) give only fused 2-aminopyridines and no diazepines.3 We have recently reported that photolysis of the 2-substituted fused pyridine N-acylimides (2c) gave the corresponding fused 3H-1,3-diazepines.4 However, the 2-substituted quinoline N-acylimides (1c) have been shown to undergo only N-N fragmentation to give the parent quinolines and no rearrangement products.3,5 We have also recently reported the conversion, by heating, of 1,2-diazepines, prepared from pyridine N-acylimides having an electron-donating group in the 3-position, into 1,3diazepines.6 These results prompted us to examine the photolysis of substituted quinoline N-acylimides in more

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detail and we now report the first examples of the formation of 3H-1,3-benzodiazepines from quinolines.

Irradiation (400 W, high-pressure Hg lamp; Pyrex filter) of the quinoline N-ethoxycarbonylimides (3a-e) having an electron-donating group in the 6- or 8-position resulted in the formation of the corresponding novel 3H-1,3-benzodiazepines (4a-e).† Treatment of the resulting diazepines (4) with 10% aqueous HCl at room temperature gave the ring-opened products (5), which were converted into the indoles (6) by further treatment with HCl at 70-80 °C; these results are analogous to those observed for fused 3H-1,3-diazepines obtained from (2c).4

(3)

R

NX

(4)

NX

(8)

R

(8)

R

(8)

R

NX

(8)

$$X = CO_2Et$$

a; R = H, b; R = OMe, c; R = NMe₂

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† Satisfactory elemental analyses and spectral data were obtained for the 1,3-diazepines (4), e.g., (4a): ca. 50 % yield; m.p. 62—64 °C; $\lambda_{\rm max}(\epsilon)$ (EtOH) 219 (18,000) and 256 nm (14,000); δ (CDCl₃) 2·44 (3H, s, 2-Me), 3·78 (3H, s, 7-OMe), 6·13 (1H, d, 5-H), 6·27 (1H, d, 4-H), 6·62 (1H, d, 6-H), 6·82 (1H, dd, 8-H), 7·16 (1H, d, 9-H), and 1·30 and 3·78 (3H, t, and 2H, q, CO₂Et), $J_{5,4}=7$, $J_{6,8}=3$, $J_{8,9}=9$ Hz; (4b): 25%, m.p. 115·5—117 °C; (4c): 8—10%, m.p. 110—111·5 °C; (4d): 8—10%, oil. The diazepine (4e) decomposed during separation to give the ring-opened product (5e): 5%, m.p. 106—107 °C.

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However, quinoline N-acylimides having an electronwithdrawing group (Ac, CO₂Me, NO₂, or Cl) in the 6- or 8-position gave the parent quinolines, but no diazepines, on irradiation. Similarly, quinolines with an electron-donating or withdrawing group in the other ring positions also gave no diazepines. These results clearly indicate that the presence of an electron-donating substituent in either the 6- or the 8-position is essential for the rearrangement of (3) to (4).

The formation of the 1,3-diazepines (4) from (3) may involve the diaziridine intermediate (7), which then rearranges to the aziridine (8) followed by ring-expansion to (4), analogous to the photolyses of (2c) and 1-substituted isoquinoline N-imides.7 The electron-donating groups may assist both the cleavage of the N-N bond in (7) and the cyclization of the resulting dipolar intermediate to give (8). Although electron-donating substituents at C-3 in the pyridine ring give no such assistance, this substituent effect is analogous to that observed for the thermal conversion of monocyclic 1,2-diazepines into 1,3-diazepines.⁶

In addition, a similar substituent effect was observed for the photo-induced rearrangement of the isoquinoline N-imides (9) into the 1H-1,3-benzodiazepines (11), for which the yields of (11b) (40%) and (11c) (35-40%) were higher than that of (11a) (15-20%).

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