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Domino Reaction for the Synthesis of Polysubstituted Pyrroles and Lamellarin R

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ABSTRACT: A three-component annulation reaction was developed for the synthesis of pyrroles, a class of compounds with various properties valuable to biomedical and polymer industries. Treatment of α -silylaryl triflates, Schiff bases, and alkynes generated polysubstituted pyrroles in good yields (61–86%) with regioselectivity. This domino reaction involved completion of five sequential steps in a single flask, which comprised arynes formation through 1,2-elimination, their alkylation by Schiff bases through 1,2-addition, 1,4-intramolecular proton transfer, Hüsgen 1,3-dipolar cycloaddition, and dehydrogenative aromatization. It was then successfully applied as the key step in the synthesis of the natural product lamellarin R. This new reaction represents an efficient, sustainable process for the production of chemical materials.

INTRODUCTION

Pyrrole is a five-membered heterocyclic aromatic compound, which darkens readily upon exposure to air. Although pyrrole is not naturally occurring, many of its derivatives are found in natural products.¹ Examples include chlorophylls, porphyrins, vitamin B_{12} , etc. Pyrrole-containing secondary metabolites include lamellarin, prodigiosin, rhazinilam, ryanodine, sceptrin, etc. It is of interest to note that pyrrole is also a constituent of tobacco smoke.²

As a flat and electron-rich ring, pyrrole is susceptible to electrophilic attack and can react with numerous biomolecules. Thus it becomes an important constituent of well-established drugs, such as atorvastatin, chlorfenapyr, premazepam, pyrvinium, roseophilin, tolmetin, and zomepirac.³ The pyrrole scaffold is used to construct therapeutic agents with antibacterial, anticancer, antifungal, anti-inflammatory, antimalarial, antimicrobial, antiprotozoal, antipsychotic, antiviral, antitubercular, and anxiolytic properties among others.^{4,5} This molecule also plays an active role as a component of polymers and indigoid dyes.⁵ In catalytic reactions, pyrroles are widely used for corrosion inhibition, metallurgy, polymerization, and preservation.⁵

Many methods have been developed for the synthesis of pyrroles,⁶⁻¹¹ such as the aza-Wittig reaction, the Hantzsch reaction, the Paal–Knorr condensation reaction, carbenoid insertions, the Friedel–Crafts acylation, the Heck coupling, hydroarylations, and the Michael addition. Zard et al.¹² provided elegant convergent routes to pyrroles by exploiting unusual radical chemistry.

Schiff bases have been used in pyrroles synthesis. Examples include Ag-catalyzed tandem 1,3-dipolar cycloaddition/aromatization by Hu and Wang,¹³ Cu-catalyzed three-component cyclization by Kostakis and Lykakis,¹⁴ Cu-catalyzed cascade process involving (3 + 2) cycloaddition of azomethine ylide by Lu and Wang,¹⁵ Li-mediated annulation,¹⁶ and phosphonite-mediated 1,3-dipolar cycloaddition by Arndtsen.¹⁷ Many other established methods for pyrrole generation require the use of expensive or toxic metal catalysts, some of which produce harmful waste streams.¹⁸ ²⁰ These concerns inspired us to develop a novel and efficient approach for the synthesis of pyrroles with various substituents through a domino reaction.

Implementation of our newly developed method as the key step in the synthesis of natural products (e.g., lamellarins) would affirm its efficiency and applicability. Lamellarins are pyrrole-derived marine alkaloids with diverse biological properties.²¹ This versatility makes lamellarins important subjects for research. They are often used as a starting point for the design of anticancer and antiviral compounds.²² In addition, efforts have been devoted to invention of new and improved methods for lamellarin synthesis.^{23,24}

Herein we report on a newly developed method for the direct synthesis of polysubstituted pyrroles from α -silylaryl triflates 1, Schiff bases 2, and alkynes 3. It involves a "single-flask" reaction. It possesses advantages of a few synthetic steps, a moderate reaction temperature, and metal-free conditions as well as provides good regioselectivity and yields. This domino reaction was applied as the key step in the synthesis of lamel-larin R, a tetrasubstituted natural pyrrole.

RESULTS AND DISCUSSION

The feasibility of the new method for the synthesis of the polysubstituted pyrroles in Scheme 1 was investigated. A mixture of (trimethylsilyl)aryl triflates 1 (1.0 equiv), Schiff bases 2 (1.0 equiv), and alkynes 3 (1.2 equiv) was treated with a fluoride (2.0 equiv) in acetonitrile at 25 °C. During the optimization of the new reaction conditions, different combinations of solvents, fluorides, additive of 18-crown-6, temperatures, and reaction times were studied for the synthesis of pyrrole **4acb**. The data in Table 1 reveal that the reaction could be conducted in THF (entries 1–5), acetonitrile (entries 6–8), and toluene (entries 9 and 10). Acetonitrile was the best solvent (entry 8). Among the three fluorides *n*-Bu₄NF, CsF, and KF, use of CsF without 18-crown-6 produced the highest yields of pyrroles (86% in entry 8). This annulation process proceeded well at 25 °C; elevation of the reaction temperature was unnecessary.

Scheme 1. A "Single–Flask" Synthesis of Pyrroles 4 from α -Silylaryl Triflates 1, Schiff Bases 2, and Alkynes 3



After 2.0 h of stirring under nitrogen atmosphere followed by 6.0 h of stirring in open air, pyrroles **4aab–dbb** were often isolated in 61–86% yields with purity >99.2%. Regarding green chemistry characteristics, the atom economy²⁵ was calculated to be 62.8%, whereas the atom efficiency²⁶ was 54.0% for compound **4acb**.

entry solvent E^- source (2.0 equiv) additive (2.0 equiv) temperature (°C) time (°h) 1 THF CsF - 25 10 2 THF KF 18-crown-6 25 10 3 THF KF 18-crown-6 60 10 4 THF n-Bu ₄ NF - 60 4.0 5 THF n-Bu ₄ NF - 60 4.0 6 CH ₃ CN KF 18-crown-6 25 10 7 CH ₃ CN KF 18-crown-6 25 10 7 CH ₃ CN KF 18-crown-6 600 8.0 8 CH ₃ CN CsF - 25 6.0 9 toluene CsF - 25 6.0 9 toluene KF 18-crown-6 80 8.0							
1 THF CsF $-$ 25 10 2 THF KF 18-crown-6 25 10 3 THF KF 18-crown-6 60 10 4 THF n -Bu ₄ NF $-$ 60 4.0 5 THF n -Bu ₄ NF $-$ 25 8.0 6 CH ₃ CN KF 18-crown-6 25 10 7 CH ₃ CN KF 18-crown-6 60 8.0 8 CH ₃ CN KF 18-crown-6 60 8.0 9 toluene CsF $-$ 25 6.0 9 toluene KF 18-crown-6 80 8.0	entry	solvent	F ⁻ source (2.0 equiv)	additive (2.0 equiv)	temperature (°C)	time (h)	yield (%)
2 THF KF 18-crown-6 25 10 3 THF KF 18-crown-6 60 10 4 THF n-Bu₄NF - 60 4.0 5 THF n-Bu₄NF - 25 8.0 6 CH ₃ CN KF 18-crown-6 25 10 7 CH ₃ CN KF 18-crown-6 60 8.0 8 CH ₃ CN CsF - 25 6.0 9 toluene CsF - 25 6.0 10 toluene KF 18-crown-6 80 8.0	1	THF	CsF	_	25	10	12
3 THF KF 18-crown-6 60 10 4 THF n-Bu ₄ NF - 60 4.0 5 THF n-Bu ₄ NF - 25 8.0 6 CH ₃ CN KF 18-crown-6 25 10 7 CH ₃ CN KF 18-crown-6 60 8.0 8 CH ₃ CN CsF - 25 6.0 9 toluene CsF - 25 6.0 10 toluene KF 18-crown-6 80 8.0	2	THF	KF	18-crown-6	25	10	20
4 THF n -Bu ₄ NF $-$ 60 4.0 5 THF n -Bu ₄ NF $-$ 25 8.0 6 CH ₃ CN KF 18-crown-6 25 10 7 CH ₃ CN KF 18-crown-6 60 8.0 8 CH ₃ CN CsF $-$ 25 6.0 9 toluene CsF $-$ 25 6.0 10 toluene KF 18-crown-6 80 8.0	3	THF	KF	18-crown-6	60	10	45
5 THF n -Bu ₄ NF $-$ 25 8.0 6 CH ₃ CN KF 18-crown-6 25 10 7 CH ₃ CN KF 18-crown-6 60 8.0 8 CH ₃ CN CsF $-$ 25 6.0 9 toluene CsF $-$ 25 6.0 10 toluene KF 18-crown-6 80 8.0	4	THF	n-Bu ₄ NF	_	60	4.0	18
6 CH ₃ CN KF 18-crown-6 25 10 7 CH ₃ CN KF 18-crown-6 60 8.0 8 CH ₃ CN CsF - 25 6.0 9 toluene CsF - 25 6.0 10 toluene KF 18-crown-6 80 8.0	5	THF	n-Bu ₄ NF	_	25	8.0	7.0
7 CH₃CN KF 18-crown-6 60 8.0 8 CH₃CN CsF - 25 6.0 9 toluene CsF - 25 6.0 10 toluene KF 18-crown-6 80 8.0	6	CH ₃ CN	KF	18-crown-6	25	10	_
8 CH ₃ CN CsF - 25 6.0 9 toluene CsF - 25 6.0 10 toluene KF 18-crown-6 80 8.0	7	CH ₃ CN	KF	18-crown-6	60	8.0	_
9 toluene CsF - 25 6.0 10 toluene KF 18-crown-6 80 8.0	8	CH ₃ CN	CsF	-	25	6.0	86
10 toluene KF 18-crown-6 80 8.0	9	toluene	CsF	_	25	6.0	23
	10	toluene	KF	18-crown-6	80	8.0	38

Various starting materials were used to evaluate the scope of this new reaction. The methoxy group was allowed to attach to the benzene nucleus of α -silvlaryl triflates 1. The benzene nucleus could be replaced by a pyridine ring (i.e., 1d). Methyl, methoxy, chloro, and bromo groups could be attached to the phenyl terminal (R³) in Schiff bases 2. This phenyl group could also be replaced by a naphthyl, furyl, and thienyl group (i.e., 2f, 2g, and 2h, respectively). The alkyl terminal group of the Schiff bases 2 must have a methylene unit that was attached to an electron-withdrawing alkoxycarbonyl group. The two substituents in alkynes 3 could be hydrogen atoms, alkyl, acetyl, alkoxycarbonyl or methoxyphenyl groups. Use of these starting materials often led to the desired pyrroles in 61-86% yields (Table 2). The exception involved the use of inactivated alkyne 3g, which had alkyl groups on both sides of the carbon to carbon triple bond. As a result, product 4acg was obtained only up to 15% vield.

To appraise the new reaction processes, we applied the new annulation reaction as the key step in the synthesis of lamellarin R (9, Scheme 2). Cesium fluoride (3.0 equiv) in acetonitrile was added to a solution containing (methoxy)silylphenyl triflate **1b** (1.2 equiv), Schiff base **2i** (1.0 equiv), and bis(methoxy-phenyl)acetylene (**3h**, 1.2 equiv). Reagents **1b** and **3h** are commercially available; Schiff base **2i** is readily available through the simple condensation of glycine methyl ester hydrochloride with paraformaldehyde.²⁷ After the annulation reaction proceeded at 0–10 °C for 6.0 h, the desired pyrroles para-**4bih** and its regioisomer meta-**4bih** were generated in a ratio of 2.3:1. Finally the isolated pyrrole para-**4bih** (61%) underwent demethylation with BBr₃ in methanol at room temperature to produce lamellarin R (9) in a high yield by use of Jia's method.²³

Table 1. Optimization of Yield for Reaction $1a + 2c + 3b \rightarrow$ 4acb by Use of Various Solvents, Fluorides, Temperatures, and Reaction Times

Table 2. Reactants 1

3 and Products 4 of the Newly

Developed Annulation Reaction along with their Isolated





^aThis Schiff base was generated *in situ*.

For verification of their configuration conclusively, the molecular framework of compound **4acb** was obtained through single crystal X-ray diffraction analysis. The data and ORTEP diagram in the Supporting Information reveal the relative positions of the four substituents and the vinylic proton of the pyrrole nucleus.







Our design and concerns in the development of a single-flask method for the synthesis of pyrroles 4 from triflates, Schiff bases, and alkynes is described in Scheme 1. The α-silylaryl triflates 1 first react with cesium fluoride to generate aryne intermediates 5 through 1,2-elimination (Step 1).28 Then Schiff bases 2 are added to arynes 5 in situ to produce iminium carbanions 6 through 1,2-addition (Step 2). The carbanion center in betaines 6 may trap an acidic proton nearby to generate azomethine ylides 7 through 1,4-intramolecular proton transfer (Step 3).²⁸ Subsequently, the Hüisgen 1,3-dipolar cycloaddition takes place between ylides 7 and alkynes 3 to give the (3 + 2) cycloadducts 8 (Step 4). Noteworthily, the activated acetylenes 3 as the starting material is expected to be more competitive than the solvent acetonitrile for reacting with azomethine ylides 7. Autoxidation in open air^{29,30} would then lead the 3-pyrrolines 8 to pyrroles 4 as the final products (Step 5). In a control experiment, a compound with a similar framework to intermediates 8 but with a CH_2 unit between the CO_2R^3 group and the pyrroline ring was produced. It was found inert to the dehydrogenative aromatization.

Results from our trials indicate that the dehydrogenative aromatization of pyrrolines 8 indeed proceeded gradually at 25 °C.³¹ The smooth conversion of $8 \rightarrow 4$ was facilitated by the presence of a CO₂R³ group attached directly to the C2 position of 3-pyrrolines 8. Activity of the allylic hydrogen at the α position to the CO₂R³ group increases significantly. Thus pyrrolines 8 with such a framework are well suited for dehydrogenative aromatization.

This approach to pyrrole synthesis accomplishes five steps sequentially in a single flask. This new domino reaction offers the following benefits: 1. The reaction generates the desired products in good-to-high yields at room temperature without any involvement of harsh conditions or reagents. 2. The entire "single-flask" reaction can be conducted with ease without isolation of the intermediates.

Application of this new reaction as the key step in the synthesis of natural product lamellarin R (9) allowed the establishment of one pyrrole nucleus, three anisole moieties, and one ester functionality in its scaffold in a single flask. In the reaction shown in Scheme 2, a byproduct meta-**4bih** was isolated in a 26% yield. Its generation was due to two possible 1,2-additions of Schiff bases **2i** to an unsymmetrical benzyne intermediate. Thus the addition may occur at both the meta and the para positions.

CONCLUSIONS

Polysubstituted pyrroles can be synthesized from silylaryl triflates, Schiff bases, and alkynes through an aryne-induced domino reaction. The overall pathway comprises 1,2-elimination, 1,2-nucleophilic addition, 1,4-proton transfer, (3 + 2) cycloaddition, and dehydrogenative aromatization. This reaction was applied successfully as the key step in the synthesis of lamellarin R.

Utilization of this new method in chemical synthesis minimizes waste production and reduces the need for extra reagents, solvents, and labor. Hence this three-component synthetic protocol is ecologically and environmentally benign for production of fine chemicals.

EXPERIMENTAL SECTION

General Information. All reactions were carried out in oven-dried glassware (120 °C) under an atmosphere of nitrogen unless as indicated otherwise. Acetonitrile, dichloromethane, ethyl acetate, hexanes, and toluene from Mallinckrodt Chemical Co. were dried and distilled from CaH2. Tetrahydrofuran (THF) from Mallinckrodt Chemicals Co. was dried by distillation from sodium and benzophenone under an atmosphere of nitrogen. The reagents purchased from Alfa Aesar included 4-bromobenzaldehyde, 3-butyn-2-one, 4-chlorobenzaldehyde, diethyl acetylenedicarboxylate, dimethyl acetylenedicarboxylate (DMAD), ethyl acetylenecarboxylate, glycine ethyl ester hydrochloride, glycine methyl ester hydrochloride, 4-methoxybenzaldehyde, methyl acetylenecarboxylate, and 4-methylbenzaldehyde. The reagents purchased from Sigma-Aldrich included benzaldehyde, bis(4-methoxyphenyl)acetylene, furan-2-carboxaldehyde, α -naphthaldehyde, 4-octyne, paraformaldehyde, thiophene-2-carboxaldehyde, and 3-(trimethylsilyl)pyridin-2-yl trifluoromethanesulfonate. Cesium fluoride (CsF), 18-crown-6, methyl 2-butynoate, potassium fluoride (KF), tetra-n-butylammonium fluoride (TBAF), and trimethylamine (Et₃N) were purchased from Acros. The compounds purchased from Tokyo Chemical Industry Co. included 4,5-dimethoxy-2-(trimethylsilyl)phenyl trifluoromethanesulfonate, 4-methoxy-2-(trimethylsilyl)phenyl trifluoromethanesulfonate, and 2-(trimethylsilyl)phenyl trifluoromethanesulfonate.

Analytical thin layer chromatography (TLC) was performed on precoated plates (silica gel 60 F-254). Purification by gravity column chromatography was carried out by use of Silicycle ultra-pure silica gel (particle size $40-63 \mu$ M, 230-400 mesh).

Infrared spectra (IR) were measured on a Fourier transform infrared spectrometer (FT-IR). Absorption intensities are recorded by the following abbreviations: s, strong; m, medium; and w, weak. Proton NMR spectra were obtained on a 400 MHz spectrometer by use of chloroform-d (CDCl₃) as the solvent. Proton NMR chemical shifts were referenced to the residual protonated solvent (δ 7.24 ppm for chloroform). Carbon-13 NMR spectra were obtained on a 100 MHz spectrometer by use of chloroform-d (CDCl₃) as the solvent. Carbon-13 chemical shifts were referenced to the center of the CDCl₃ triplet (δ 77.0 ppm). Multiplicities are recorded by the following abbreviations: s, singlet; d, doublet; dd, doublet of doublet; t, triplet; q, quartet; m, multiplet; and J, coupling constant (hertz). Highresolution mass spectra (HRMS) were measured on an instrument by use of a time-of-flight mass analyzer (TOF) with electrospray ionization (ESI).

59 60 Standard Procedure for the Single-Flask Synthesis of Pyrroles 4. To a stirred solution of 2-silylphenyl triflate 1 (1.0 equiv) in dry CH₃CN (2.0–2.5 mL) was added a Schiff base 2 (1.0 equiv), an alkyne 3 (1.2 equiv), and CsF (2.0–2.1 equiv) at room temperature under nitrogen atmosphere. After the reaction mixture was stirred at 25 °C for 6.0–8.0 h, the reaction mixture was quenched with water (10 mL) and then extracted with EtOAc (3×10 mL). The combined organic layers were dried over CaSO₄ (s), filtered, and concentrated under reduced pressure to afford the products. It was then purified by use of dry column chromatography on silica gel with a limited amount³² of EtOAc in hexanes as the eluent to give the desired pyrrole 4.

4-Ethoxycarbonyl-2-methoxycarbonyl-5-phenyl-N-phenylpyrrole (4aab). The Standard Procedure was followed by use of 2-silylphenyl triflate 1a (50.2 mg, 0.168 mmol, 1.0 equiv), Schiff base 2a³³ (29.8 mg, 0.168 mmol, 1.0 equiv), ethyl acetylenecarboxylate (3b, 19.8 mg, 0.202 mmol, 1.2 equiv), and CsF (51.2 mg, 0.336 mmol, 2.0 equiv) in CH₃CN (2.0 mL). After the reaction mixture was stirred for 6.0 h and then worked up, the residue was purified by use of silica gel column chromatography (10% EtOAc in hexanes as the eluent) to give the desired pyrrole 4aab (46.4 mg, 0.134 mmol) in 80% yield as a yellow liquid: TLC Rf 0.45 (15% EtOAc in hexanes as the eluent); ¹H NMR (CDCl₃, 400 MHz) δ 7.58 (s, 1H), 7.25–7.23 (m, 3H), 7.22–7.18 (m, 3H), 7.12–7.10 (m, 2H), 7.06–7.04 (m, 2H), 4.13 (q, 2H, J = 7.3 Hz), 3.69 (s, 3H), 1.13 (t, 3H, J = 7.3 Hz); ${}^{13}C{}^{1}H$ NMR (CDCl₃, 100 MHz) δ 163.8, 160.5, 144.5, 137.9, 130.5, 128.8, 128.3, 128.2, 128.1, 127.9, 127.1, 123.4, 119.5, 114.4, 59.9, 51.4, 14.2; IR (neat) 2924 (s), 1714 (s, C=O), 1598 (m), 1470 (m), 1235 (s, C-O), 1122 (m), 1039 (m), 759 (m) cm⁻¹; HRMS (ESI-TOF) m/z [M + H]⁺ calcd for C₂₁H₁₉NO₄ + H 350.1392, found 350.1385.

2,4-Dimethoxycarbonyl-5-(4-methylphenyl)-N-phenylpyrrole (4aba). The Standard Procedure was followed by use of 2silylphenyl triflate 1a (50.4 mg, 0.168 mmol, 1.0 equiv), Schiff base 2b³⁴ (32.1 mg, 0.168 mmol, 1.0 equiv), methyl acetylenecarboxylate (3a, 16.9 mg, 0.202 mmol, 1.2 equiv), and CsF (51.6 mg, 0.336 mmol, 2.0 equiv) in CH₃CN (2.0 mL). After the reaction mixture was stirred for 6.0 h and then worked up, the residue was purified by use of silica gel column chromatography (10% EtOAc in hexanes as the eluent) to give the desired pyrrole 4aba (47.3 mg, 0.136 mmol) in 81% yield as a yellow liquid: TLC Rf 0.45 (15% EtOAc in hexanes as the eluent); ¹H NMR (CDCl₃, 400 MHz) & 7.58 (s, 1H), 7.28-7.26 (m, 3H), 7.07-7.06 (m, 2H), 7.02-6.97 (m, 4H), 3.76 (s, 3H), 3.71 (s, 3H), 2.24 (s, 3H); ¹³C{¹H} NMR (CDCl₃, 100 MHz) δ160.2, 159.9, 144.3, 137.9, 137.7, 130.3, 128.3, 128.2, 128.1, 127.9, 127.1, 123.5, 119.4, 114.2, 51.3, 51.1, 21.3; IR (neat) 2923 (s), 1713 (s, C=O), 1598 (m), 1470 (s), 1235 (s, C-O), 1116 (s), 1039 (m), 759 (m) cm⁻¹; HRMS (ESI-TOF) *m/z* [M + H]⁺ calcd for C₂₁ $H_{19}NO_4$ + H 350.1392, found 350.1388.

4-Ethoxycarbonyl-2-methoxycarbonyl-5-(4-methylphenyl)-N-phenylpyrrole (4abb). (a) The Standard Procedure was followed by use of 2-silylphenyl triflate 1a (50.3 mg, 0.168 mmol, 1.0 equiv), Schiff base $2b^{34}$ (32.3 mg, 0.168 mmol, 1.0 equiv), ethyl acetylenecarboxylate (3b, 19.9 mg, 0.201 mmol, 1.2 equiv), and CsF (51.4 mg, 0.336 mmol, 2.0 equiv) in CH₃CN (2.1 mL). After the reaction mixture was stirred for 6.0 h and then worked up, the residue was purified by use of silica gel column chromatography (10% EtOAc in hexanes as the eluent) to give the desired pyrrole 4abb (50.2 mg, 0.139 mmol) in 83% yield as a yellow liquid: TLC R_f 0.45 (15% EtOAc in hexanes as the eluent); ¹H NMR (CDCl₃, 400 MHz) δ 7.57 (s, 1H), 7.26–7.24 (m, 3H), 7.07–7.05 (m, 2H), 7.01–6.95 (m, 4H), 4.15 (q, 2H, J = 6.9 Hz), 3.69 (s, 3H), 2.24 (s, 3H), 1.18 (t, 3H, J = 6.9 Hz); ¹³C {¹H} NMR (CDCl₃, 100 MHz) δ 163.6, 160.3, 144.4, 138.0, 137.8, 130.5, 128.2, 128.1, 128.1, 127.9, 127.0, 123.3, 119.4, 114.3, 59.9, 51.4, 21.4, 14.2; IR (neat) 2924 (m), 1715 (s, C=O), 1598 (w), 1470 (m), 1201 (s, C–O), 1116 (m), 1039 (m), 758 (m) cm⁻¹; HRMS (ESI-TOF) m/z [M + H]⁺ calcd for C₂₂H₂₁NO₄ + H 364.1548, found 364.1551.

(b) The same procedure was followed by use of 2-silylphenyl triflate **1a** (2.60 g, 8.71 mmol, 1.0 equiv), Schiff base **2b**³⁴ (1.67 g, 8.71 mmol, 1.0 equiv), ethyl acetylenecarboxylate (**3b**, 1.02 g, 10.5 mmol, 1.2 equiv), and CsF (2.65 g, 17.4 mmol, 2.0 equiv) in CH₃CN (100 mL). The desired pyrrole **4abb** (2.59 g, 7.14 mmol) was obtained in 82% yield as a yellow liquid.

2-Methoxycarbonyl-4-methylcarbonyl-5-(4-methylphenyl)-N-phenylpyrrole (4abc). The Standard Procedure was followed by use of 2-silylphenyl triflate 1a (50.1 mg, 0.168 mmol, 1.0 equiv), Schiff base **2b**³⁴ (32.2 mg, 0.168 mmol, 1.0 equiv), 3butyn-2-one (3c, 13.7 mg, 0.201 mmol, 1.2 equiv), and CsF (51.1 mg, 0.336 mmol, 2.0 equiv) in CH₃CN (2.2 mL). After the reaction mixture was stirred for 8.0 h and then worked up, the residue was purified by use of silica gel column chromatography (10% EtOAc in hexanes as the eluent) to give the desired pyrrole 4abc (38.1 mg, 0.114 mmol) in 68% yield as a yellow liquid: TLC $R_f 0.40$ (15% EtOAc in hexanes as the eluent); ¹H NMR (CDCl₃, 400 MHz) δ 7.33 (s, 1H), 7.26–7.23 (m, 3H), 7.07-7.05 (m, 2H), 7.01-6.95 (m, 4H), 3.69 (s, 3H), 2.39 (s, 3H) 2.24 (s, 3H); ${}^{13}C{}^{1}H$ NMR (CDCl₃, 100 MHz) δ 193.7, 160.3, 144.2, 138.4, 138.1, 130.3, 128.2, 128.1, 128.0, 127.8, 127.3, 123.3, 117.7, 114.2, 51.2, 28.7, 21.3; IR (neat) 2925 (m), 1717 (s, C=O), 1609 (m), 1438 (m), 1247 (s, C-O), 1122 (m), 1032 (s), 749 (w) cm⁻¹; HRMS (ESI-TOF) m/z [M + H^{+}_{1} calcd for $C_{21}H_{19}NO_3 + H 334.1443$, found 334.1447.

3,4-Diethoxycarbonyl-2-methoxycarbonyl-5-(4methylphenyl)-N-phenylpyrrole (4abf). The Standard Procedure was followed by use of 2-silylphenyl triflate 1a (50.5 mg, 0.168 mmol, 1.0 equiv), Schiff base **2b**³⁴ (32.2 mg, 0.168 mmol, 1.0 equiv), diethyl acetylenedicarboxylate (3f, 34.5 mg, 0.202 mmol, 1.2 equiv), and CsF (51.3 mg, 0.336 mmol, 2.0 equiv) in CH₃CN (2.2 mL). After the reaction mixture was stirred for 8.0 h and then worked up, the residue was purified by use of silica gel column chromatography (10% EtOAc in hexanes as the eluent) to give the desired pyrrole 4abf (52.6 mg, 0.121 mmol) in 72% yield as yellow solids: TLC R_f 0.40 (20%) EtOAc in hexanes as the eluent); mp (recrystallized from EtOH) 144.7-146.9 °C; ¹H NMR (CDCl₃, 400 MHz) *S*7.26-7.24 (m. 3H), 7.09–7.06 (m, 2H), 7.02–6.07 (m, 4H), 4.35 (q, 2H, J =7.0 Hz), 4.19 (q, 2H, J = 7.2 Hz), 3.75 (s, 3H), 2.24 (s, 3H), 1.34 (t, 3H, J = 7.0 Hz), 1.21 (t, 3H, J = 7.2 Hz); ¹³C{¹H} NMR (CDCl₃, 100 MHz) & 165.2, 164.1, 160.5, 144.6, 138.1, 137.8, 130.6, 129.1, 128.4, 128.2, 128.1, 127.1, 125.5, 119.6, 112.4, 61.3, 60.2, 52.3, 21.3, 14.2, 13.8; IR (neat) 2924 (s), 1714 (s, C=O), 1599 (w), 1470 (s), 1235 (s, C-O), 1116 (s), 1039 (m), 759 (m) cm⁻¹; HRMS (ESI-TOF) m/z [M + H]⁺ calcd for C₂₅H₂₅NO₆+H 436.1760, found 436.1770.

4-Ethoxycarbonyl-2-methoxycarbonyl-5-(4-methoxyphenyl)-N-phenylpyrrole (4acb). The Standard Procedure was followed by use of 2-silylphenyl triflate **1a** (301 mg, 1.01 mmol, 1.0 equiv), Schiff base $2c^{33}$ (209 mg, 1.01 mmol, 1.0 equiv), ethyl acetylenecarboxylate (3b, 119 mg, 1.21 mmol, 1.2 equiv), and

CsF (307 mg, 2.02 mmol, 2.0 equiv) in CH₃CN (15.0 mL). After the reaction mixture was stirred for 6.0 h and then worked up, the residue was purified by use of silica gel column chromatography (10% EtOAc in hexanes as the eluent) to give the desired pyrrole 4acb (328 mg, 0.867 mmol) in 86% yield as yellow solids: TLC Rf 0.45 (20% EtOAc in hexanes as the eluent); mp (recrystallized from EtOH) 137.6-139.8 °C; ¹H NMR (CDCl₃, 400 MHz) δ 7.56 (s, 1H), 7.26–7.23 (m, 3H), 7.06–7.03 (m, 4H), 6.69 (d, 2H, J = 8.8 Hz), 4.15 (q, 2H, J = 7.0 Hz), 3.71 (s, 3H), 3.69 (s, 3H), 1.19 (t, 3H, J = 7.0 Hz); $^{13}C{^{1}H}$ NMR (CDCl₃, 100 MHz) δ 163.6, 160.2, 159.2, 144.1, 137.8, 132.0, 128.2, 128.1, 128.0, 123.3, 122.1, 119.5, 114.2, 112.7, 59.9, 55.1, 51.4, 14.3; IR (neat) 2925 (s), 1714 (s, C=O), 1612 (m), 1471 (s), 1250 (s, C–O), 1118 (m), 1041 (m), 759 (m) cm⁻¹; HRMS (ESI-TOF) m/z [M + H]⁺ calcd for C₂₂H₂₁NO₅+ H 380.1498, found 380.1487.

2,4-Dimethoxycarbonyl-5-(4-methoxyphenyl)-3-methyl-Nphenvlpvrrole (4acd). The Standard Procedure was followed by use of 2-silylphenyl triflate 1a (50.2 mg, 0.168 mmol, 1.0 equiv), Schiff base 2c³³ (34.6 mg, 0.168 mmol, 1.0 equiv), methyl 2-butynoate (3d, 19.9 mg, 0.201 mmol, 1.2 equiv), and CsF (51.3 mg, 0.336 mmol, 2.0 equiv) in CH₃CN (2.1 mL). After the reaction mixture was stirred for 6.0 h and then worked up, the residue was purified by use of silica gel column chromatography (10% EtOAc in hexanes as the eluent) to give the desired pyrrole 4acd (45.4 mg, 0.119 mmol) in 71% yield as a yellow liquid: TLC $R_f 0.45$ (20% EtOAc in hexanes as the eluent); ¹H NMR (CDCl₃, 400 MHz) δ 7.27-7.24 (m, 3H), 7.06-7.02 (m, 4H), 6.69 (d, 2H, J = 8.4 Hz), 3.72 (s, 3H), 3.64(s, 3H), 3.62 (s, 3H), 2.62 (s, 3H); ¹³C{¹H} NMR (CDCl₃, 100 MHz) δ161.4, 158.9, 158.6, 143.5, 138.7, 131.8, 128.1, 128.0, 127.8, 123.1, 122.2, 117.1, 114.1, 112.6, 55.1, 51.1, 50.9, 12.5; IR (neat) 2925 (m), 1717 (s, C=O), 1598 (w), 1472 (m), 1240 (s, C–O), 1122 (m), 1032 (m), 836 (m) cm⁻¹; HRMS (ESI-TOF) m/z [M + H]⁺ calcd for C₂₂H₂₁NO₅ + H 380.1498, found 380.1484.

3,4-Di-n-propyl-2-methoxycarbonyl-5-(4-methoxyphenyl)-N-phenylpyrrole (4acg). The Standard Procedure was followed by use of 2-silylphenyl triflate 1a (101 mg, 0.336 mmol, 1.0 equiv), Schiff base 2c³³ (69.5 mg, 0.336 mmol, 1.0 equiv), 4octyne (3g, 44.6 mg, 0.403 mmol, 1.2 equiv), and CsF (103 mg, 0.672 mmol, 2.0 equiv) in CH₃CN (4.1 mL). After the reaction mixture was stirred for 8.0 h and then worked up, the residue was purified by use of silica gel column chromatography (10% EtOAc in hexanes as the eluent) to give the desired pyrrole 4acg (19.6 mg, 0.051 mmol) in 15% yield as a yellow liquid: TLC R_f 0.40 (15% EtOAc in hexanes as the eluent); ¹H NMR (CDCl₃, 400 MHz) & 7.38-7.18 (m, 3H), 6.99-6.95 (m, 4H), 6.67 (d, 2H, J = 8.4 Hz), 3.71 (s, 3H), 3.66 (s, 3H), 2.54 (t, 2H, J = 7.2 Hz), 2.31 (t, 2H, J = 7.0 Hz), 1.65–1.53 (m, 4H), 0.97–0.89 (m, 6H); ${}^{13}C{}^{1}H$ NMR (CDCl₃, 100 MHz) δ 160.4, 159.2, 144.3, 137.7, 131.8, 128.2, 128.1, 128.0, 124.3, 122.2, 116.2, 114.1, 112.6, 55.1, 51.3, 29.9, 28.8, 25.6, 25.4, 13.8; IR (neat) 2953 (m), 1716 (s, C=O), 1599 (m), 1446 (m), 1235 (s, C-O), 1095 (m), 1029 (m), 759 (m) cm⁻¹; HRMS (ESI-TOF) m/z [M + H]⁺ calcd for C₂₅H₂₉NO₃ + H 392.2225, found 392.2236

5-(4-Chlorophenyl)-4-ethoxycarbonyl-2-methoxycarbonyl-N-phenylpyrrole (4adb). The Standard Procedure was followed by use of 2-silylphenyl triflate **1a** (50.1 mg, 0.168 mmol, 1.0 equiv), Schiff base **2d**³⁴ (35.7 mg, 0.168 mmol, 1.0 equiv), ethyl acetylenecarboxylate (**3b**, 19.8 mg, 0.201 mmol, 1.2 equiv), and CsF (51.6 mg, 0.336 mmol, 2.0 equiv) in CH₃CN (2.1 mL). After the reaction mixture was stirred for 8.0 h and then worked up, the residue was purified by use of silica gel column chromatography (10% EtOAc in hexanes as the eluent) to give the desired pyrrole **4adb** (46.7 mg, 0.123 mmol) in 73% yield as a yellow liquid: TLC R_f 0.45 (20% EtOAc in hexanes as the eluent); ¹H NMR (CDCl₃, 400 MHz) δ 7.58 (s, 1H), 7.44–7.40 (m, 2H), 7.28–7.25 (m, 2H), 7.23–7.19 (m, 3H), 7.05–7.01 (m, 2H), 4.21 (q, 2H, *J* = 6.9 Hz), 3.73 (s, 3H), 1.17 (t, 3H, *J* = 6.9 Hz); ¹³C{¹H} NMR (CDCl₃, 100 MHz) δ 163.9, 161.0, 144.5, 140.2, 134.7, 130.2, 129.8, 128.5, 128.4, 128.3, 128.2, 124.7, 119.8, 114.6, 59.8, 51.2, 14.1; IR (neat) 2926 (m), 1716 (s, C=O), 1599 (w), 1490 (m), 1236 (s, C–O), 1092 (m), 1015 (m), 757 (m) cm⁻¹; HRMS (ESI-TOF) *m/z* [M + H]⁺ calcd for C₂₁H₁₈ClNO₄ + H 384.1002, found 384.1007.

5-(4-Bromophenyl)-4-ethoxycarbonyl-2-methoxycarbonyl-N-phenylpyrrole (4aeb). The Standard Procedure was followed by use of 2-silvlphenyl triflate 1a (50.3 mg, 0.168 mmol, 1.0 equiv), Schiff base 2e³⁵ (43.1 mg, 0.168 mmol, 1.0 equiv), ethyl acetylenecarboxylate (3b, 19.6 mg, 0.201 mmol, 1.2 equiv), and CsF (51.4 mg, 0.336 mmol, 2.0 equiv) in CH₃CN (2.3 mL). After the reaction mixture was stirred for 8.0 h and then worked up, the residue was purified by use of silica gel column chromatography (10% EtOAc in hexanes as the eluent) to give the desired pyrrole 4aeb (53.1 mg, 0.126 mmol) in 76% yield as a yellow liquid: TLC Rf 0.45 (20% EtOAc in hexanes as the eluent); ¹H NMR (CDCl₃, 400 MHz) δ 7.58 (s, 1H), 7.38–7.32 (m, 4H), 7.30 (t, 1H, J = 7.2 Hz), 7.15 (d, 2H, J = 8.0 Hz), 7.03 (d, 2H, J = 8.0 Hz), 4.17 (q, 2H, J = 7.2 Hz), 3.73 (s, 3H), 1.15 (t, 3H, J = 7.2 Hz); ¹³C{¹H} NMR (CDCl₃, 100 MHz) δ 163.4, 160.3, 143.9, 138.2, 132.2, 131.1, 130.3, 128.4, 128.2, 128.1, 123.5, 120.3, 119.7, 114.2, 59.8, 51.4, 14.4; IR (neat) 2925 (m), 1715 (s, C=O), 1599 (m), 1487 (m), 1236 (s, C-O), 1090 (m), 1011 (m), 759 (m) cm⁻¹; HRMS (ESI-TOF) m/z [M + H]⁺ calcd for C₂₁H₁₈BrNO₄+H 428.0497, found 428.0490.

4-Ethoxycarbonyl-2-methoxycarbonyl-5-(naphthalen-1-yl)-N-phenylpyrrole (4afb). The Standard Procedure was followed by use of 2-silylphenyl triflate 1a (50.2 mg, 0.168 mmol, 1.0 equiv), Schiff base $2f^{34}$ (38.3 mg, 0.168 mmol, 1.0 equiv), ethyl acetylenecarboxylate (3b, 19.7 mg, 0.201 mmol, 1.2 equiv), and CsF (51.1 mg, 0.336 mmol, 2.0 equiv) in CH₃CN (2.0 mL). After the reaction mixture was stirred for 8.0 h and then worked up, the residue was purified by use of silica gel column chromatography (10% EtOAc in hexanes as the eluent) to give the desired pyrrole 4afb (50.2 mg, 0.126 mmol) in 75% yield as a yellow liquid: TLC Rf 0.45 (20% EtOAc in hexanes as the eluent); ¹H NMR (CDCl₃, 400 MHz) δ 7.88 (d, 1H, J = 7.2 Hz), 7.79 (d, 1H, J = 8.0 Hz), 7.74 (d, 1H, J = 7.2 Hz), 7.63 (s, 1H), 7.54 (d, 2H, J = 7.6 Hz), 7.50-7.43 (m, 3H), 7.27 (t, 2H, J = 7.2 Hz), 7.13 (t, 2H, J = 7.4 Hz), 4.20 (q, 2H, J = 7.2 Hz), 3.78 (s, 3H), 1.29 (t, 3H, J = 7.2 Hz); ¹³C{¹H} NMR (CDCl₃, 100 MHz) δ164.1, 160.2, 147.3, 142.9, 139.0, 133.9, 131.4, 128.6, 127.9, 127.7, 127.6, 125.8, 125.5, 125.3, 124.8, 123.9, 123.4, 121.9, 121.3, 113.6, 60.6, 53.3, 14.4; IR (neat) 2953 (m), 1731 (s, C=O), 1599 (m), 1505 (m), 1446 (w), 1235 (s, C-O), 1029 (m), 778 (m) cm⁻¹; HRMS (ESI-TOF) m/z [M + H]⁺ calcd for $C_{25}H_{21}NO_4 + H$ 400.1548, found 400.1550.

4-Ethoxycarbonyl-5-(furan-2-yl)-2-methoxycarbonyl-N-phenylpyrrole (**4agb**). The Standard Procedure was followed by use of 2-silylphenyl triflate **1a** (50.4 mg, 0.168 mmol, 1.0 equiv), Schiff base $2g^{33}$ (28.4 mg, 0.168 mmol, 1.0 equiv), ethyl

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acetylenecarboxylate (3b, 19.9 mg, 0.202 mmol, 1.2 equiv), and CsF (51.8 mg, 0.336 mmol, 2.0 equiv) in CH₃CN (2.2 mL). After the reaction mixture was stirred for 8.0 h and then worked up, the residue was purified by use of silica gel column chromatography (10% EtOAc in hexanes as the eluent) to give the desired pyrrole 4agb (40.4 mg, 0.119 mmol) in 71% yield as a yellow liquid: TLC Rf 0.45 (20% EtOAc in hexanes as the eluent); ¹H NMR (CDCl₃, 400 MHz) δ 7.57 (s, 1H), 7.48 (d, 1H, J = 4.8 Hz), 7.41 (d, 2H, J = 6.8 Hz), 7.32 (t, 1H, J = 6.2 Hz), 7.08 (d, 2H, J = 6.8 Hz), 6.80 (d, 1H, J = 6.4 Hz), 6.46 (dd, 1H, J = 4.8, 1.6 Hz), 4.14 (q, 2H, J = 7.1 Hz), 3.71 (s, 3H), 1.15 (t, 3H, J = 7.1 Hz); ¹³C{¹H} NMR (CDCl₃, 100 MHz) δ 163.8, 160.1, 149.9, 148.4, 143.5, 139.6, 131.8, 129.4, 128.6, 124.0, 121.9, 116.5, 114.1, 111.6, 59.7, 51.1, 14.7; IR (neat) 2926 (s), 1716 (s, C=O), 1613 (m), 1472 (m), 1251 (s, C-O), 1117 (m), 1040 (m), 759 (m) cm⁻¹; HRMS (ESI-TOF) m/z [M + H]⁺ calcd for C₁₉H₁₇NO₅+H 340.1185, found 340.1181.

3-Ethoxycarbonyl-5-methoxycarbonyl-2-(thien-2-yl)-N-phenylpyrrole (4ahb). The Standard Procedure was followed by use of 2-silylphenyl triflate 1a (50.2 mg, 0.168 mmol, 1.0 equiv), Schiff base 2h³⁵ (30.6 mg, 0.168 mmol, 1.0 equiv), ethyl acetylenecarboxylate (3b, 20.1 mg, 0.202 mmol, 1.2 equiv), and CsF (51.6 mg, 0.336 mmol, 2.0 equiv) in CH₃CN (2.0 mL). After the reaction mixture was stirred for 8.0 h and then worked up, the residue was purified by use of silica gel column chromatography (10% EtOAc in hexanes as the eluent) to give the desired pyrrole 4ahb (43.6 mg, 0.123 mmol) in 73% yield as a yellow liquid: TLC R_f 0.45 (20% EtOAc in hexanes as the eluent); ¹H NMR (CDCl₃, 400 MHz) δ 7.58 (s, 1H), 7.41 (d, 2H, J = 6.4 Hz), 7.30–7.26 (m, 2H), 7.20 (d, 1H, J = 4.0 Hz), 7.12 (d, 2H, J = 6.4 Hz), 6.99 (dd, 1H, J = 7.2, 2.4 Hz), 4.20 (q, 2H, J = 6.6 Hz), 3.71 (s, 3H), 1.17 (t, 3H, J = 6.6 Hz); ¹³C{¹H} NMR (CDCl₃, 100 MHz) & 163.8, 160.1, 143.9, 139.9, 139.2, 132.1, 131.6, 129.1, 128.2, 127.9, 125.2, 123.4, 122.0, 114.1, 59.8, 51.3, 14.1; IR (neat) 2924 (m), 1715 (s, C=O), 1598 (w), 1470 (m), 1367 (m), 1234 (s, C–O), 1039 (m), 758 (m) cm⁻¹; HRMS (ESI-TOF) $m/z [M + H]^+$ calcd for C₁₉H₁₇NO₄S + H 356.0956, found 356.0947.

2-Ethoxycarbonyl-4-methoxycarbonyl-5-phenyl-N-phe-

nylpyrrole (4aja). The Standard Procedure was followed by use of 2-silylphenyl triflate 1a (50.3 mg, 0.168 mmol, 1.0 equiv), Schiff base 2j³³ (32.1 mg, 0.168 mmol, 1.0 equiv), methyl acetylenecarboxylate (3a, 17.1 mg, 0.202 mmol, 1.2 equiv), and CsF (51.7 mg, 0.336 mmol, 2.0 equiv) in CH₃CN (2.2 mL). After the reaction mixture was stirred for 6.0 h and then worked up, the residue was purified by use of silica gel column chromatography (10% EtOAc in hexanes as the eluent) to give the desired pyrrole 4aja (47.6 mg, 0.136 mmol) in 81% yield as a yellow liquid: TLC R_f 0.45 (15% EtOAc in hexanes as the eluent); ¹H NMR (CDCl₃, 400 MHz) δ7.57 (s, 1H), 7.24–7.23 (m, 3H), 7.20-7.16 (m, 3H), 7.14-7.12 (m, 2H), 7.07-7.03 (m, 2H), 4.06 (q, 2H, J = 7.2 Hz), 3.76 (s, 3H), 1.08 (t, 3H, J = 7.2 Hz); ${}^{13}C{}^{1}H$ NMR (CDCl₃, 100 MHz) δ 164.0, 160.2, 144.4, 138.0, 130.6, 128.9, 128.4, 128.2, 128.2, 127.8, 127.0, 123.5, 119.5, 114.3, 60.2, 51.1, 14.4; IR (neat) 2926 (m), 1714 (s, C=O), 1611 (m), 1470 (s), 1249 (s, C-O), 1118 (s), 1040 (m), 759 (m) cm⁻¹; HRMS (ESI-TOF) m/z [M + H]⁺ calcd for $C_{21}H_{19}NO_4 + H$ 350.1392, found 350.1383.

2,4-Diethoxycarbonyl-5-phenyl-N-phenylpyrrole (4ajb). (a) The Standard Procedure was followed by use of 2-silylphenyl triflate **1a** (50.5 mg, 0.168 mmol, 1.0 equiv), Schiff base $2j^{33}$ (32.3 mg, 0.168 mmol, 1.0 equiv), ethyl acetylenecarboxylate

(3b, 19.4 mg, 0.202 mmol, 1.2 equiv), and CsF (51.3 mg, 0.336 mmol, 2.0 equiv) in CH₃CN (2.0 mL). After the reaction mixture was stirred for 6.0 h and then worked up, the residue was purified by use of silica gel column chromatography (10% EtOAc in hexanes as the eluent) to give the desired pyrrole 4ajb (48.5 mg, 0.134 mmol) in 80% yield as a yellow liquid: TLC R_f 0.45 (15% EtOAc in hexanes as the eluent); ¹H NMR (CDCl₃, 400 MHz) δ 7.58 (s, 1H), 7.25–7.23 (m, 3H), 7.20-7.17 (m, 3H), 7.14-7.12 (m, 2H), 7.07-7.03 (m, 2H), 4.13 (q, 2H, J=7.1 Hz), 4.06 (q, 2H, J=7.1 Hz), 1.14 (t, 3H, J=7.1 Hz), 1.09 (t, 3H, J = 7.1 Hz); ¹³C{¹H} NMR (CDCl₃, 100 MHz) δ164.1, 163.8, 144.6, 137.8, 130.8, 128.9, 128.4, 128.2, 128.2, 127.9, 126.9, 123.4, 119.4, 114.3, 60.2, 59.8, 14.4, 14.2; IR (neat) 2924 (s), 1714 (s, C=O), 1598 (w), 1470 (m), 1235 (s, C–O), 1116 (s), 1039 (m), 759 (m) cm⁻¹; HRMS (ESI-TOF) m/z $[M + H]^+$ calcd for $C_{22}H_{21}NO_4 + H$ 364.1548, found 364.1552. (b) The same procedure was followed by use of 2-silylphenyl triflate **1a** (2.70 g, 9.05 mmol, 1.0 equiv), Schiff base **2i**³³ (1.73 g, 9.05 mmol, 1.0 equiv), ethyl acetylenecarboxylate (3b, 1.07 g, 10.9 mmol, 1.2 equiv), and CsF (2.74 g, 18.1 mmol, 2.0 equiv) in CH₃CN (100 mL). The desired pyrrole 4ajb (2.57 g, 7.06 mmol) was obtained in 78% yield as a yellow liquid.

3,4-Dimethoxycarbonyl-2-ethoxycarbonyl-5-phenyl-N-phenylpyrrole (4aje). The Standard Procedure was followed by use of 2-silylphenyl triflate 1a (50.7 mg, 0.168 mmol, 1.0 equiv), Schiff base 2j³³ (31.9 mg, 0.168 mmol, 1.0 equiv), dimethyl acetylenedicarboxylate (3e, 28.9 mg, 0.202 mmol, 1.2 equiv), and CsF (51.4 mg, 0.336 mmol, 2.0 equiv) in CH₃CN (2.5 mL). After the reaction mixture was stirred for 8.0 h and then worked up, the residue was purified by use of silica gel column chromatography (10% EtOAc in hexanes as the eluent) to give the desired pyrrole 4aje (50.4 mg, 0.124 mmol) in 74% yield as yellow solids: TLC Rf 0.40 (20% EtOAc in hexanes as the eluent); mp (recrystallized from EtOH) 141.2-143.5 °C; ¹H NMR (CDCl₃, 400 MHz) δ 7.25–7.24 (m, 3H), 7.20–7.17 (m, 3H), 7.13–7.11 (m, 2H), 7.06–7.03 (m, 2H), 4.18 (q, 2H, J = 7.0 Hz), 3.87 (s, 3H), 3.73 (s, 3H), 1.17 (t, 3H, J = 7.0 Hz); ¹³C{¹H} NMR (CDCl₃, 100 MHz) δ164.8, 162.2, 160.3, 144.5, 138.0, 130.5, 129.2, 128.5, 128.3, 128.2, 127.6, 127.2, 125.4, 121.2, 112.6, 60.1, 52.6, 51.5, 14.1; IR (neat) 2928 (m), 1738 (s, C=O), 1732 (s, C=O), 1599 (m), 1459 (m), 1235 (s, C=O), 1027 (m), 778 (m) cm⁻¹; HRMS (ESI-TOF) m/z [M + H]⁺ calcd for C₂₃H₂₁NO₆+H 408.1447, found 408.1442.

2-Ethoxycarbonyl-4-methoxycarbonyl-5-(4-methoxyphenyl)-N-phenylpyrrole (4aka). The Standard Procedure was followed by use of 2-silylphenyl triflate 1a (50.3 mg, 0.168 mmol, 1.0 equiv), Schiff base 2k³⁶ (37.1 mg, 0.168 mmol, 1.0 equiv), methyl acetylenecarboxylate (3a, 17.1 mg, 0.202 mmol, 1.2 equiv), and CsF (51.2 mg, 0.336 mmol, 2.0 equiv) in CH₃CN (2.1 mL). After the reaction mixture was stirred for 6.0 h and then worked up, the residue was purified by use of silica gel column chromatography (10% EtOAc in hexanes as the eluent) to give the desired pyrrole 4aka (54.4 mg, 0.142 mmol) in 85% yield as yellow solids: TLC Rf 0.45 (20% EtOAc in hexanes as the eluent); mp (recrystallized from EtOH) 138.6–140.9 °C; ¹H NMR (CDCl₃, 400 MHz) 87.57 (s, 1H), 7.26–7.23 (m, 3H), 7.07–7.03 (m, 4H), 6.70 (d, 2H, J = 7.2 Hz), 4.15 (q, 2H, J = 7.2 Hz), 3.72 (s, 3H), 3.68 (s, 3H), 1.16 (t, 3H, J = 7.2 Hz); ¹³C{¹H} NMR (CDCl₃, 100 MHz) δ 163.9, 160.1, 159.1, 144.2, 137.6, 131.9, 128.4, 128.3, 128.2, 124.0, 122.3, 119.6, 114.4, 112.8, 60.1, 55.2, 51.2, 14.1; IR (neat) 2925 (m), 1714 (s, C=O), 1611 (m), 1470 (m), 1249 (s, C-O), 1117 (m), 1040 (m), 759

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(m) cm⁻¹; HRMS (ESI-TOF) m/z [M + H]⁺ calcd for C₂₂H₂₁NO₅ + H 380.1498, found 380.1491.

2-Methoxycarbonyl-3,4-di(4-methoxyphenyl)-N-4-methoxyphenylpyrrole (4bih). To a stirred solution of glycine methyl ester hydrochloride (37.6 mg, 0.302 mmol, 1.0 equiv) in dry CH₃CN (2.5 mL) was added molecular sieves (4 Å, activated, 2.5 µm, powdered, 30.5 mg), Et₃N (30.7 mg, 0.302 mmol, 1.0 equiv), and paraformaldehyde (9.41 mg, 0.302 mmol, 1.0 equiv) to generate Schiff base 2i28 at 0 °C under nitrogen atmosphere. After the reaction mixture was stirred at 0-10 °C for 1.0 4-methoxy-2-(trimethylsilyl)phenyl h. trifluoromethanesulfonate³⁷ (1b, 101 mg, 0.302 mmol, 1.0 equiv), bis(4-methoxyphenyl)acetylene (3h, 86.5 mg, 0.362 mmol, 1.2 equiv), and CsF (134 mg, 0.906 mmol, 3.1 equiv) were added into the reaction mixture. The mixture was stirred at 0-10 °C for 8.0 h and then quenched with water (10 mL) and extracted with EtOAc (3×10 mL). The combined organic layers were dried over CaSO₄ (s), filtered, and concentrated under reduced pressure. It was then purified by use of silica gel column chromatography (20% EtOAc in hexanes as the eluent) to give the desired pyrrole 4bih (81.5 mg, 0.184 mmol) in 61% yield as a yellow liquid: TLC Rf 0.45 (25% EtOAc in hexanes as the eluent); ¹H NMR (CDCl₃, 400 MHz) δ 7.32 (d, 2H, J = 8.8 Hz), 7.23 (d, 2H, J = 8.8 Hz), 7.07 (d, 2H, J = 8.4 Hz), 7.06 (s, 1H), 6.95 (d, 2H, J = 8.8 Hz), 6.88 (d, 2H, J = 8.4 Hz), 6.77 (d, 2H, J = 8.8 Hz), 3.84 (s, 3H), 3.82 (s, 3H), 3.76 (s, 3H), 3.49(s, 3H); ${}^{13}C{}^{1}H$ NMR (CDCl₃, 100 MHz) δ 161.6, 158.9, 158.9, 158.0, 136.2, 132.4, 131.8, 129.1, 129.0, 128.5, 126.9, 125.7, 124.8, 124.5, 113.8, 113.6, 113.5, 55.4, 55.2, 55.1, 51.3; IR (neat) 2925 (m), 1715 (s, C=O), 1598 (w), 1464 (m), 1235 (s, C-O), 1120 (w), 1011 (w), 759 (m) cm⁻¹; HRMS (ESI-TOF) $m/z [M + H]^+$ calcd for $C_{27}H_{25}NO_5 + H = 444.1811$, found 444.1823.

4-Ethoxycarbonyl-2-methoxycarbonyl-N-3,4-dimethoxy-

phenvl-5-(4-methylphenvl) pyrrole (4cbb). The Standard Procedure was followed by use of 2-silylayl triflate 1c (50.4 mg, 0.141 mmol, 1.0 equiv), Schiff base 2b³⁴ (26.9 mg, 0.141 mmol, 1.0 equiv), ethyl acetylenecarboxylate (3b, 16.6 mg, 0.169 mmol, 1.2 equiv), and CsF (42.9 mg, 0.282 mmol, 2.0 equiv) in CH₃CN (2.1 mL). After the reaction mixture was stirred for 6.0 h and then worked up, the residue was purified by use of silica gel column chromatography (15% EtOAc in hexanes as the eluent) to give the desired pyrrole 4cbb (46.5 mg, 0.110 mmol) in 78% yield as yellow solids: TLC Rf 0.40 (20% EtOAc in hexanes as the eluent); mp (recrystallized from EtOH) 143.1-145.3 °C; ¹H NMR (CDCl₃, 400 MHz) *δ*7.54 (s, 1H), 7.02–6.96 (m, 3H), 6.71 (s, 1H), 6.68 (d, 2H, J = 6.4 Hz), 6.48 (d, 1H, J = 6.4 Hz), 4.14 (q, 2H, J = 6.9 Hz), 3.83 (s, 3H), 3.71 (s, 3H), 3.66 (s, 3H), 2.25 (s, 3H), 1.17 (t, 3H, J = 6.9 Hz); ¹³C{¹H} NMR (CDCl₃, 100 MHz) & 163.6, 160.2, 148.4, 148.0, 144.6, 138.0, 130.5, 130.4, 128.0, 127.2, 123.3, 120.6, 119.3, 114.2, 111.7, 109.7, 59.8, 55.9, 55.8, 51.4, 21.4, 14.3; IR (neat) 2930 (m), 1714 (s, C=O), 1515 (s), 1471 (m), 1237 (s, C-O), 1112 (s), 1029 (m), 760 (m) cm⁻¹; HRMS (ESI-TOF) m/z [M + H]⁺ calcd for C₂₄H₂₅NO₆+H 424.1760, found 424.1758.

4-Ethoxycarbonyl-2-methoxycarbonyl-5-(4-methylphenyl)-*N-2-pyridylpyrrole (4dbb).* The Standard Procedure was followed by use of 2-silylaryl triflate **1d** (50.1 mg, 0.167 mmol, 1.0 equiv), Schiff base **2b**³⁴ (31.8 mg, 0.167 mmol, 1.0 equiv), ethyl acetylenecarboxylate (**3b**, 19.6 mg, 0.201 mmol, 1.2 equiv), and CsF (50.8 mg, 0.334 mmol, 2.0 equiv) in CH₃CN (2.1 mL). After the reaction mixture was stirred for 8.0 h and then worked up, the residue was purified by use of silica gel column chromatography (10% EtOAc in hexanes as the eluent) to give the desired pyrrole **4dbb** (43.1 mg, 0.119 mmol) in 71% yield as a yellow liquid: TLC R_f 0.45 (20% EtOAc in hexanes as the eluent); ¹H NMR (CDCl₃, 400 MHz) δ 8.35 (d, 1H, *J* = 7.2 Hz), 7.58 (s, 1H), 7.55 (t, 1H, *J* = 7.0 Hz), 7.29–7.25 (m, 3H), 7.18 (t, 1H, *J* = 7.2 Hz), 7.03 (d, 2H, *J* = 7.6 Hz), 4.15 (q, 2H, *J* = 7.2 Hz), 3.72 (s, 3H), 2.26 (s, 3H), 1.16 (t, 3H, *J* = 7.2 Hz); ¹³C{¹H} NMR (CDCl₃, 100 MHz) δ 164.1, 160.8, 151.7, 148.8, 144.2, 140.1, 139.2, 137.8, 130.5, 128.2, 127.9, 127.0, 123.3, 119.4, 114.3, 59.9, 51.4, 21.4, 14.2; IR (neat) 2924 (m), 1715 (s, C=O), 1598 (w), 1470 (m), 1201 (s, C–O), 1120 (w), 1011 (w), 758 (m) cm⁻¹; HRMS (ESI-TOF) *m/z* [M + H]⁺ calcd for C₂₁H₂₀N₂O₄ + H 365.1501, found 365.1512.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website.

NMR spectra, IR spectra, and crystallographic data (PDF)

Crystallographic data for 4acb (CIF)

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REFERENCES

(1) Jusélius, J.; Sundholm, D. The aromatic pathways of porphins, chlorins and bacteriochlorins. *Phys. Chem. Chem. Phys.* **2000**, *2*, 2145–2151.

(2) Fowles, J.; Bates, M. *The Chemical Constituents in Cigarettes and Cigarette Smoke: Priorities for Harm Reduction*; Epidemiology and Toxicology Group ESR: Kenepuru Science Centre; March 2000.

(3) Domagala, A.; Jarosz, T.; Lapkowski, M. Living on pyrrolic foundations–Advances in natural and artificial bioactive pyrrole derivatives. *Eur. J. Med. Chem.* **2015**, *100*, 176–187.

(4) Gholap, S. S. Pyrrole: An emerging scaffold for construction of valuable therapeutic agents. *Eur. J. Med. Chem.* **2016**, *110*, 13–31.

(5) Bhardwaj, V.; Gumber, D.; Abbot, V.; Dhiman, S.; Sharma, P. Pyrrole: a resourceful small molecule in key medicinal hetero-aromatics. *RSC Adv.* **2015**, *5*, 15233–15266.

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(6) Organic Chemistry Portal. https://www.organic-chemistry.org/synthesis/heterocycles/pyrroles.shtm (accessed January 9, 2019).

(7) Vessally, E. A new avenue to the synthesis of highly substituted pyrroles: synthesis from *N*-propargylamines. *RSC Adv.* **2016**, *6*, 18619–18631.

(8) Chelucci, G. Metal-catalyzed dehydrogenative synthesis of pyrroles and indoles from alcohols. *Coord. Chem. Rev.* 2017, 331, 37–53.

(9) Zhou, N.-N.; Zhu, H.-T.; Yang, D.-S.; Guan, Z.-H. Recent developments in the group-1B-metal-catalyzed synthesis of pyrroles. *Org. Biomol. Chem.* **2016**, *14*, 7136–7149.

(10) Khajuria, R.; Dham, S.; Kapoor, K. K. Active methylenes in the synthesis of a pyrrole motif: an imperative structural unit of pharmaceuticals, natural products and optoelectronic materials. *RSC Adv.* **2016**, *6*, 37039–37066.

(11) Olivier, W. J.; Smith, J. A.; Bissember, A. C. Methods for the synthesis of annulated pyrroles *via* cyclisation strategies. *Org. Biomol. Chem.* **2018**, *16*, 1216–1226.

(12) Quiclet-Sire, B.; Zard, S. Z. Convergent Routes to Pyrroles Exploiting the Unusual Radical Chemistry of Xanthates – An Overview. *Synlett* **2017**, *28*, 2685–2696.

(13) Liu, Y.; Hu, H.; Wang, X.; Zhi, S.; Kan, Y.; Wang, C. Synthesis of Pyrrole via a Silver-Catalyzed 1,3-Dipolar Cycloaddition/Oxidative Dehydrogenative Aromatization Tandem Reaction. *J. Org. Chem.* **2017**, *82*, 4194–4202.

(14) Andreou, D.; Kallitsakis, M. G.; Loukopoulos, E.; Gabriel, C.; Kostakis, G. E.; Lykakis, I. N. Copper-Promoted Regioselective Synthesis of Polysubstituted Pyrroles from Aldehydes, Amines, and Nitroalkenes via 1,2-Phenyl/Alkyl Migration. *J. Org. Chem.* **2018**, *83*, 2104–2113.

(15) Hong, D.; Zhu, Y.; Li, Y.; Lin, X.; Lu, P.; Wang, Y. Three-Component Synthesis of Polysubstituted Pyrroles from α -Diazoketones, Nitroalkenes, and Amines. *Org. Lett.* **2011**, *13*, 4668–4671.

(16) Hu, Y.; Wang, C.; Wang, D.; Wu, F.; Wan, B. Synthesis of Tetrasubstituted Pyrroles from Terminal Alkynes and Imines. *Org. Lett.* **2013**, *15*, 3146–3149.

(17) Cyr, D. J. S.; Arndtsen, B. A. A New Use of Wittig-Type Reagents as 1,3-Dipolar Cycloaddition Precursors and in Pyrrole Synthesis. *J. Am. Chem. Soc.* **2007**, *129*, 12366–12367.

(18) Tietze, L. F. Domino Reactions in Organic Synthesis. *Chem. Rev.* **1996**, *96*, 115–136.

(19) Padwa, A. Domino reactions of rhodium(II) carbenoids for alkaloid synthesis. *Chem. Soc. Rev.* **2009**, *38*, 3072–3081.

(20) Ardkhean, R.; Caputo, D. F. J.; Morrow, S. M.; Shi, H.; Xiong, Y.; Anderson, E. A. Cascade polycyclizations in natural product synthesis. *Chem. Soc. Rev.* **2016**, *45*, 1557–1569.

(21) Fan, H.; Peng, J.; Hamann, M. T.; Hu, J.-F. Lamellarins and Related Pyrrole-Derived Alkaloids from Marine Organisms. *Chem. Rev.* **2008**, *108*, 264–287.

(22) Bailly, C. Lamellarins: A tribe of bioactive marine natural products. In *Outstanding Marine Molecules: Chemistry, Biology, Analysis;* La Barre, S., Kornprobst, J.-M., Eds.; Wiley-VCH Verlag GmbH & Co. KGaA.: Weinheim, 2014; pp 377–386.

(23) Li, Q.; Jiang, J.; Fan, A.; Cui, Y.; Jia, Y. Total Synthesis of Lamellarins D, H, and R and Ningalin B. Org. Lett. 2011, 13, 312–315.

(24) Fukuda, T.; Sudo, E.; Shimokawa, K.; Iwao, M. Palladiumcatalyzed cross-coupling of *N*-benzenesulfonyl-3,4-dibromopyrrole and its application to the total syntheses of lamellarins O, P, Q, and R. *Tetrahedron* **2008**, *64*, 328–338.

(25) Li, C.-J.; Trost, B. M. Green chemistry for chemical synthesis. *Proc. Natl. Acad. Sci. U. S. A.* **2008**, *105*, 13197–13202.

(26) Sheldon, R. A. Atom efficiency and catalysis in organic synthesis. *Pure Appl. Chem.* **2009**, *72*, 1233–1246.

(27) Rivera, D. G.; Wessjohann, L. A. Supramolecular Compounds from Multiple Ugi Multicomponent Macrocyclizations: Peptoid-based Cryptands, Cages, and Cryptophanes. *J. Am. Chem. Soc.* **2006**, *128*, 7122–7123.

(28) Swain, S. P.; Shih, Y.-C.; Tsay, S.-C.; Jacob, J.; Lin, C.-C.; Hwang, K. C.; Horng, J.-C.; Hwu, J. R. Aryne-Induced Novel Tandem 1,2-Addition/(3+2) Cycloaddition to Generate Imidazolidines and Pyrrolidines. *Angew. Chem. Int. Ed.* **2015**, *54*, 9926–9930.

(29) Asikainen, M.; Jauhiainen, O.; Aaltonen, O.; Harlin, A. Continuous catalyst-free aromatization of γ -terpinene using air as an oxidant. *Green Chem.* **2013**, *15*, 3230–3235.

(30) Hwu, J. R.; Swain, S. P. Silicon-Induced Phenanthrene Formation from Benzynes and Allenylsilanes. *Chem. Eur. J.* **2013**, *19*, 6556–6560.

(31) Ramaiah, D.; Muneer, M.; Gopidas, K. R.; Das, P. K.; Rath, N. P.; George, M. V. Phototransformations of *C*-Benzoylaziridines. Dipolarophilic Trapping of Photogenerated Azomethine Ylides. *J. Org. Chem.* **1996**, *61*, 4240 4246.

(32) Bohen, J. M.; Joullié, M. M.; Kaplan, F. A.; Loev, B. "Dry-Column" Chromatography. A new technique for the undergraduate laboratory. J. Chem. Educ. **1973**, *50*, 367–368.

(33) López-Pérez, A.; Adrio, J.; Carretero, J. C. Bis-Sulfonyl Ethylene as Masked Acetylene Equivalent in Catalytic Asymmetric [3 + 2] Cycloaddition of Azomethine Ylides. J. Am. Chem. Soc. 2008, 130, 10084–10085.

(34) Wang, C.-J.; Liang, G.; Xue, Z.-Y.; Gao, F. Highly Enantioselective 1,3-Dipolar Cycloaddition of Azomethine Ylides Catalyzed by Copper(I)/TF-BiphamPhos Complexes. *J. Am. Chem. Soc.* **2008**, *130*, 17250–17251.

(35) Robles-Machín, R.; López-Pérez, A.; González-Esguevillas, M.; Adrio, J.; Carretero, J. C. Pyrrole and Oligopyrrole Synthesis by 1,3-Dipolar Cycloaddition of Azomethine Ylides with Sulfonyl Dipolarophiles. *Chem. Eur. J.* **2010**, *16*, 9864–9873.

(36) Conde, E.; Rivilla, I.; Larumbe, A.; Cossío, F. P. Enantiodivergent Synthesis of Bis-Spiropyrrolidines via Sequential Interrupted and Completed (3 + 2) Cycloadditions. *J. Org. Chem.* **2015**, *80*, 11755–11767.

(37) Baumgärtner, K.; Chincha, A. L. M.; Dreuw, A.; Rominger, F.; Mastalerz, M. A Conformationally Stable Contorted Hexabenzoovalene. *Angew. Chem. Int. Ed.* **2016**, *55*, 15594–15598.

