

Preliminary communication

Diyne coordination chemistry: Reactions of $[\text{RuClH}(\text{CO})(\text{PPh}_3)_3]$ with diphenylbutadiyne and bis(phenylethynyl)mercury

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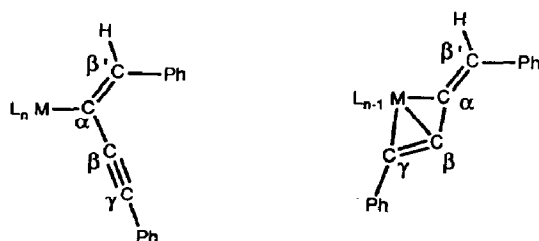
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Abstract

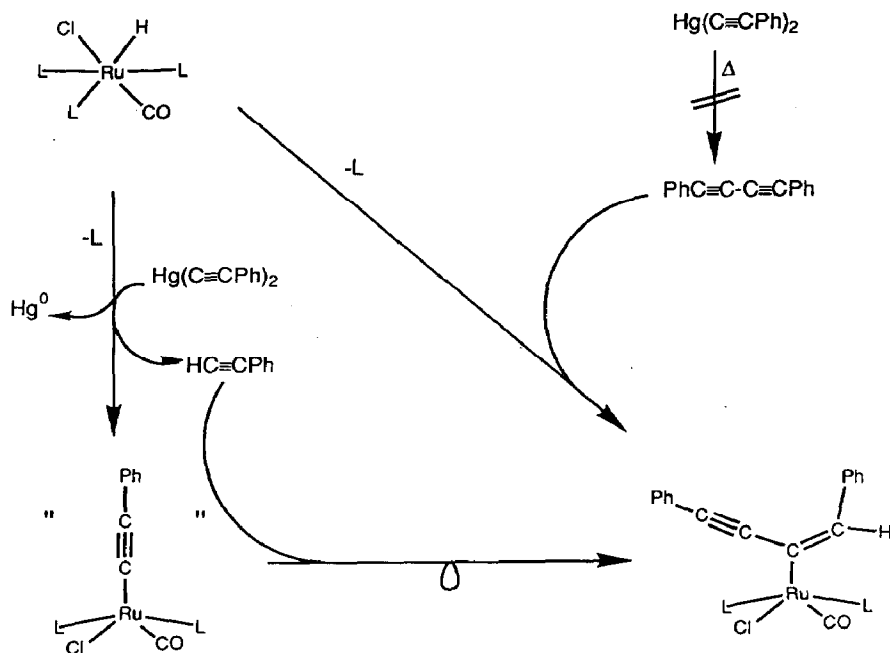
The reaction of the hydridometal complex $[\text{RuClH}(\text{CO})(\text{PPh}_3)_3]$ with 1,4-diphenylbut-1-en-3-yn-2-yl ligand has been investigated and found to proceed with monoinsertion to give a coordinatively unsaturated σ -vinyl complex $[\text{Ru}\{-\text{C}(\text{C}\equiv\text{CPh})=\text{CHPh}\}\text{Cl}(\text{CO})(\text{PPh}_3)_2]$, which is also the major product of the reaction of $[\text{RuClH}(\text{CO})(\text{PPh}_3)_3]$ with $[\text{Hg}(\text{C}\equiv\text{CPh})_2]$.

Recent interest in the unusual 1,4-diphenylbut-1-en-3-yn-2-yl ligand (Scheme 1) centres on its possible intermediacy in alkyne oligomerisation [1–3]. This highly unsaturated ligand may coordinate as either a one [1] or three [2,3] electron donor depending upon the requirements of the metal centre which, in the cases studied, are quite clear.

The complexes $[\text{RuRCl}(\text{CO})(\text{PPh}_3)_2]$ ($\text{R} = \text{aryl}$ [4], vinyl [5]) are stable 16-electron species even though the latter might be expected to coordinate in a bidentate manner as observed for σ - π -vinyl complexes of molybdenum [6]. It was therefore of interest to prepare such a complex with a potentially bidentate 1,4-diphenylbut-1-en-3-yn-2-yl ligand. Such a complex was the unexpected product of an unsuccessful attempt to obtain the corresponding hypothetical σ -alkynyl derivative



Scheme 1. Bonding modes for coordinated C_4HPh_2 .



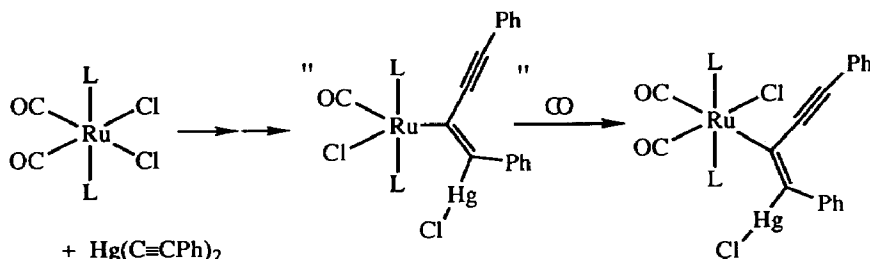
Scheme 2. Mechanisms for the formation of $[\text{Ru}\{\text{C}(\text{C}\equiv\text{CPh})=\text{CHPh}\}\text{Cl}(\text{CO})\text{L}_2]$, $\text{L} = \text{PPh}_3$.

$[\text{Ru}(\text{C}\equiv\text{CPh})\text{Cl}(\text{CO})(\text{PPh}_3)_2]$ from the reaction of $[\text{RuClH}(\text{CO})(\text{PPh}_3)_3]$ and di(phenylethynyl)mercury. Elemental mercury was deposited and a bright yellow compound obtained upon chromatography and characterised as $[\text{Ru}\{\text{C}(\text{C}\equiv\text{CPh})=\text{CHPh}\}\text{Cl}(\text{CO})(\text{PPh}_3)_2]$ [7*].

Two plausible explanations for the course of this reaction are (Scheme 2) (a) thermolysis of $[\text{Hg}(\text{C}\equiv\text{CPh})_2]$ to provide 1,4-diphenyl-butadi-1,3-yne, which subsequently reacts with the hydridoruthenium complex by insertion, or (b) reaction of $[\text{RuClH}(\text{CO})(\text{PPh}_3)_3]$ with the organomercurial to provide the desired complex $[\text{Ru}(\text{C}\equiv\text{CPh})\text{Cl}(\text{CO})(\text{PPh}_3)_x]$ ($x = 2$ or 3), which under the reaction conditions inserts the equivalent of phenylacetylene which is liberated in the initial reaction. Both these mechanisms could ultimately lead to the same product $[\text{Ru}\{\text{C}(\text{C}\equiv\text{CPh})=\text{CHPh}\}\text{Cl}(\text{CO})(\text{PPh}_3)_2]$. Treating $[\text{RuClH}(\text{CO})(\text{PPh}_3)_3]$ with one equivalent of the preformed diyne provides $[\text{Ru}\{\text{C}(\text{C}\equiv\text{CPh})=\text{CHPh}\}\text{Cl}(\text{CO})(\text{PPh}_3)_2]$ in high yield and a similar reaction has been reported for the trifluoroacetato complex $[\text{RuH}(\text{O}_2\text{CCF}_3)(\text{CO})(\text{PPh}_3)_2]$ [1]. However, heating $[\text{Hg}(\text{C}\equiv\text{CPh})_2]$ in tetrahydrofuran under reflux does not lead to deposition of mercury and therefore the extrusion of mercury must be in some way mediated by the ruthenium centre. In the absence of a preparative route to $[\text{Ru}(\text{C}\equiv\text{CPh})\text{Cl}(\text{CO})(\text{PPh}_3)_2]$, mechanism (b) remains unvalidated; however, the required second step, i.e., insertion reactions of acetylenes in this metal-ligand system, has a precedent in the reaction of $[\text{Ru}(\text{CH}=\text{CH}^t\text{Bu})\text{Cl}(\text{CO})(\text{PPh}_3)_2]$ with $\text{HC}\equiv\text{CCO}_2\text{Me}$ [8].

Mawby and co-workers have described a related ligand system which arises from the reaction of $[\text{RuCl}_2(\text{CO})_2(\text{PMe}_2\text{Ph})_2]$ with $[\text{Hg}(\text{C}\equiv\text{CPh})_2]$ via the proposed

* Reference number with asterisk indicates a note in the list of references.



Scheme 3. Chlororuthenation of bis(phenylethynyl)mercury (ref. 9); L = PMe_2Ph .

intermediacy of a 16-electron vinyl complex $[\text{Ru}\{\text{C}(\text{C}\equiv\text{CPh})=\text{C}(\text{HgCl})\text{Ph}\}\text{Cl}(\text{CO})(\text{PMe}_2\text{Ph})_2]$ related to $[\text{Ru}\{\text{C}(\text{C}\equiv\text{CPh})=\text{CHPh}\}\text{Cl}(\text{CO})(\text{PPh}_3)_2]$; however, in this case the mercury remains incorporated in the final ligand as a vinyl β -substituent (Scheme 3).

The complex $[\text{Ru}\{\text{C}(\text{C}\equiv\text{CPh})=\text{CHPh}\}\text{Cl}(\text{CO})(\text{PPh}_3)_2]$, being coordinatively unsaturated, reacts with a range of ligands, viz, carbon monoxide, isonitriles, pyrazole, 2,1,3-benzoselenadiazole, 2,2'-bipyridyl, 1,4,7-trithiacyclononane, and poly(azol-1-yl)chelates [7b]. We are currently investigating the generality of the insertion of diphenylbutadiyne into platinum-metal hydride bonds.

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References and notes

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- 7 (a) Data for $[\text{Ru}\{\text{C}(\text{C}\equiv\text{CPh})=\text{CHPh}\}\text{Cl}(\text{CO})(\text{PPh}_3)_2]$: IR $\nu(\text{CO})$ 1910 (Nujol), 1923 (CH_2Cl_2) cm^{-1} . ^{31}P - $\{^1\text{H}\}$ NMR (CDCl_3), δ 35.0 ppm. ^1H NMR (CDCl_3) δ 5.26 [s(br) 1 H, $\text{C}=\text{CHPh}$], 7.05–7.40, 7.70–7.80 [$m \times 2$, 40 H, PC_6H_5 and CC_6H_5]. FAB-MS (nitrobenzyl alcohol) m/z 890 [M] $^+$, 628 [$M - \text{PPh}_3$] $^+$, 626 [$\text{Ru}(\text{PPh}_3)_2$] $^+$, 600 [$\text{RuCl}(\text{PPh}_3)_2\{\text{C}(\text{C}\equiv\text{CPh})=\text{CHPh}\}$] $^+$, 565 [$\text{Ru}(\text{PPh}_3)\{\text{C}(\text{C}\equiv\text{CPh})=\text{CHPh}\}$] $^+$, 492 [$\text{Ru}(\text{PPh}_3)(\text{HCCCCPh})$] $^+$, 364 [RuPPh_3] $^+$. Yield, via PhC_4Ph route, 84%. (b) The characterisation of $[\text{Ru}\{\text{C}(\text{C}\equiv\text{CPh})=\text{CHPh}\}\text{Cl}(\text{CO})(\text{PPh}_3)_2]$ included its reaction with $\text{K}[\text{HB}(\text{pz})_3]$ (pz = pyrazol-1-yl) to provide $[\text{Ru}\{\text{C}(\text{C}\equiv\text{CPh})=\text{CHPh}\}(\text{CO})(\text{PPh}_3)\{\text{HB}(\text{pz})_3\}]$, the structure of which was determined by X-ray crystallography: N.W. Alcock and A.F. Hill, unpublished results.
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