

Synthesis of γ -Lactams via Pd(II)-Catalyzed C(sp³)–H Olefination Using a Self-Cleaving Polyfluoroethylsulfinyl Directing Group

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ABSTRACT: A monodentate directing group, 2-chlorotetrafluoroethylsulfinylmide (-NHSOCF₂CF₂Cl), for inert C(sp³)–H bond activation is reported. This directing group shows efficient ability in Pd(II)-catalyzed C(sp³)–H olefination. The desired olefination products undergo subsequent Michael addition and *in situ* expulsion of the auxiliary to provide the free NH γ -lactam products. Preliminary mechanistic studies reveal that the auxiliary group is crucial for C(sp³)–H activation.

ransition-metal-catalyzed C–H olefination has attracted a tremendous amount of interest, as this family of synthetic methods can enable rapid construction of bioactive molecules.¹ Nevertheless, the activation of aliphatic $C(sp^3)$ -H bonds remains challenging, due to the combination of high pK_{a} (pK_{a} > 40) and high bond dissociation energy (BDE > 400 kJ/ mol).² Moreover, selectively targeting only one C-H bond among many alternative C-H bonds in a similar chemical environment is difficult.³ In recent years, using a directing group strategy, several C(sp³)-H olefination precedents have been reported with Pd catalysts. Using heterocyclic directing groups (e.g., pyridine^{4a} or pyrazole^{4b}) or cyclic aliphatic amines^{4c,d} as directing groups, Sanford, Yu, and Gaunt developed straightforward $C(sp^3)$ -H olefinations that afforded open-chain or cyclized products (Figure 1a).⁴ A removable directing group strategy has also proven to be successful,⁵ as initially reported by Yu in the β -C(sp³)-H olefination of N-perfluorinated aryl amides.^{5a} Similar types of auxiliaries have also facilitated γ -C(sp³)-H olefination of amine/amide substrates with the aid of pyridine- or quinoline-based ligands, providing lactams or pyrrolidines after subsequent intra-molecular cyclization.^{Sb-d} Maiti achieved γ -C(sp³)–H olefination to give open-chain products with widely used bidentate 8aminoquinoline^o as the directing group (Figure 1b).⁵⁶

Though these seminal breakthroughs demonstrated the ability to achieve reactivity and selectivity in $C(sp^3)$ -H olefination, the removal of these exogenous auxiliaries is required, which generally involves tedious concession steps and/or harsh conditions.⁷ To overcome this problem, Yu reported a free carboxylic acid-directed ligand-enabled β - $C(sp^3)$ -H olefination in 2018. This transformation provided γ -lactones by the subsequent 1,4-addition (Figure 1c),⁸ but the synthesis of γ -lactams via facile $C(sp^3)$ -H activation is not compatible. Very recently, Yu reported that native amides are



Previous Work

a) Heterocyclic Directing Groups (DG)



Figure 1. Direct olefination of inert $C(sp^3)$ -H.

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also capable of directing β -C(sp³)–H olefination to provide open-chain products, but the removal of the directing group is still problematic.9 The application of a traceless directing group strategy, which is well established in aryl $C(sp^2)-H$ activation,¹⁰ would be highly enabling in $C(sp^3)$ -H functionalization but remains underdeveloped.¹¹ Herein, we report a self-cleaving amide auxiliary, 2-chlorotetrafluoroethylsulfinyl (SOR_f) , and its application in $C(sp^3)$ -H olefination. In this transformation, the auxiliary group (SOR_f) is automatically removed from the desired products during the course of the reaction, thus providing a straightforward protocol for synthesizing free NH γ -lactams, an important bioactive core structure.¹² The N-2-chlorotetrafluoroethylsulfinyl (NHSOR $_{t}$) auxiliary originates from an abundant industrial waste product, ICF₂CF₂Cl, and can thus be easily prepared on a large scale. Liu and Ellman have independently reported elegant uses of this auxiliary in asymmetric nucleophilic addition.

We began the study by examining the reaction of N-2chlorotetrafluoroethanesulfinyl pivalamide (1a) and ethyl acrylate (2a) (Scheme 1). When using 10 mol % Pd(OAc)₂ as the catalyst, 2 equiv of AgOAc as the oxidant, and hexafluoro-2-propanol (HFIP) as the solvent, the desired product 3a, resulting from C(sp³)-H olefination and the subsequent Michael addition, was afforded in 14% yield (entry 1). The addition of a base to the reaction mixture significantly increased the yield of 3a. For example, the use of 2 equiv of NaTFA gave 3a in 52% yield (entry 2), and the use of CsF afforded 3a in 72% yield (entry 3). To our delight, we isolated

Scheme 1. Optimization of the Reaction Conditions^a

ر		Pd(O/ CF ₂ CF ₂ CI Oxida Base Sol.,	Ac) ₂ (10 mo l%) nt (2 equiv.) (2 equiv.) 120 °C, 12 h	N-SCF ₂ CF ₂ CI 3a CO ₂ Et			
	Entry	ny Base Ov		Solvent	Yie	Yield (%) ^b	
	Linkiy		Oxidant	Contoint	3a	4a	
	1	None	AgOAc	HFIP	14	n.d.	
	2	NaTFA	AgOAc	HFIP	52	n.d.	
	3	CsF	AgOAc	HFIP	72	12	
	4	Na ₂ CO ₃	AgOAc	HFIP	n.d.	81(78)	
	5	K ₂ CO ₃	AgOAc	HFIP	n.d.	18	
	6	Cs ₂ CO ₃	AgOAc	HFIP	n.d.	10	
	7	NaHCO ₃	AgOAc	HFIP	5	38	
	8	Na ₂ CO ₃	Ag ₂ CO ₃	HFIP	n.d.	n.d.	
	9	Na ₂ CO ₃	Ag ₂ O	HFIP	n.d.	n.d.	
	10	Na ₂ CO ₃	AgTFA	HFIP	trace	n.d.	
	11	Na ₂ CO ₃	Cu(OAc) ₂	HFIP	n.d.	22	
	12 ^c	Na ₂ CO ₃	AgOAc	HFIP	n.d.	80	
	13	Na ₂ CO ₃	AgOAc	DCE	20	n.d.	
	14	Na ₂ CO ₃	AgOAc	MeCN	n.d.	n.d.	
	15	Na ₂ CO ₃	AgOAc	PhCH ₃	20	n.d.	
	16 ^d	Na ₂ CO ₃	AgOAc	HFIP	38	28	
	17 ^e	Na ₂ CO ₃	AgOAc	HFIP	30	n.d.	

^{*a*}Reaction conditions: **1a** (0.2 mmol), **2b** (0.6 mmol), $Pd(OAc)_2$ (10 mol %), oxidant (2 equiv), base (2 equiv), HFIP (2 mL). ^{*b*}Yields were determined by ¹H NMR analysis relative to the CH₂Br₂ internal standard. ^{*c*}Under a N₂ or an O₂ atmosphere. ^{*d*}100 °C. ^{*e*}80 °C.

Letter

lactam 4a as a second product in 12% yield, which indicated that the polyfluoroalkylsulfinyl auxiliary (SOR_{f}) was removed in situ under the reaction conditions. Next, we extensively screened various bases, hoping to achieve the convenient synthesis of γ -lactam by using this auxiliary as a self-cleaving directing group in $C(sp^3)$ -H olefination (for details, see the Supporting Information). It turned out that the yield of 4a could be increased to 81% (78% isolated yield) when using 2 equiv of Na₂CO₃ as the base (entry 4). Other bases, such as K₂CO₃, Cs₂CO₃, and NaHCO₃, were found not only to strongly inhibit the formation of 4a but also to depress alkenylation of 1a (entries 5–7). We examined various oxidants, only to find that AgOAc was optimal and that other Ag⁺ or Cu²⁺ salts were less effective or ineffective for the formation of 3a or 4a (entries 8-11). Under these conditions, the reaction was equally effective under either N2 or O2 atmospheres (entry 12), indicating that AgOAc is likely playing a key role in reoxidizing the Pd catalyst. Other solvents, such as MeCN, PhCH₃, and DCE, gave only 3a in low yields or did not lead to product formation (entries 13-15, respectively). The decreased reaction temperature resulted in lower yield of 4a or incomplete conversion of the starting material (entries 16 and 17). On the basis of these findings, we chose entry 4 as the optimal conditions.

Different polyfluoroalkylsulfinyl (SOR_f) auxiliaries were next tested under the optimal conditions (Scheme 2). Pivalamide with a -CF₃ auxiliary afforded a 35% yield of 4a, and the -C₄F₉ and -C₆F₁₃ auxiliaries provided 4a in 62% and 56% yields, respectively. All of these auxiliaries were less efficient than -CF₂CF₂Cl, demonstrating the unique proprieties of this group.

Scheme 2. Efficiencies of Different Auxiliaries

	+ O OEt 2b (3 equiv.)	Pd(OAc) ₂ (10 mol%) AgOAc (2 equiv.)	NH
1a (0.1 mmol)		Na ₂ CO ₃ (2 equiv.) HFIP, 120 °C, 10 h	
Entry		Rf	NMR Yield of 4a
1		-CF3	35%
2		-CF2CF2CI	76%
3		-C ₄ F ₉	62%
4			56%

Having optimized the conditions, we evaluated the substrate scope for this transformation. Various $\alpha_{,\alpha}$ -dialkyl-substituted propionic amides bearing the N-2-chlorotetrafluoroethylsulfinyl (NHSOR_i) auxiliary were examined, and the desired γ lactams were afforded in moderate to good yields via $C(sp^3)$ -H activation (Scheme 3). The linear substituents on the propionic amides, such as methyl, ethyl, 1-propyl, 2-propyl, and 1-butyl, were compatible with this protocol and afforded the desired γ -lactams in 56–78% yield (4a–4e). The connectivity of 4a was confirmed by X-ray diffraction. $\alpha_{,}\alpha_{-}$ Diethyl-substituted propionic amide provided a 50% yield of 4f. Spirocyclic propionic amide was tolerated under the reaction conditions, and the desired product 4g was achieved in 51% yield. A substrate bearing a methoxy group was also suitable, offering a 52% yield of the desired product 4h with a 1.5:1 diastereoselectivity. Phenyl substituents at the β , γ , or δ position of the propionic amides were well tolerated (4i-4n), and halogen substituents (i.e., -F, -Cl, and -Br) could be readily accommodated on these aryl species (4i-4k and 4m). Though

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Scheme 3. Substrate Scope for β -C(sp³)-H Olefination^{*a*}



"Reaction conditions: 1a (0.2 mmol), 2b (3 equiv), Pd(OAc)₂ (10 mol %), AgOAc (2 equiv), Na₂CO₃ (2.0 equiv), HFIP (2 mL), 120 $^{\circ}$ C, 12 h.

low yields were obtained with β -aryl-substituted substrates, moderate yields were provided when the aromatic rings were instead located at the γ or δ positions, and the desired products were obtained in 52–60% yields (41–4n). This *in situ* removable directing group could also be applied to aryl $C(sp^2)$ –H olefination, when α,α -dimethyl- α -phenyl amide was used as the starting material; under the optimal reaction conditions, C–H olefination took place at the aryl $C(sp^2)$ –H bond, providing the corresponding benzo-fused δ -lactam in 85% yield (4o).

The scope of the acrylate substrates was next examined with N-2-chlorotetrafluoroethanesulfinyl pivalamide (1a). Various acrylates were tolerated and provided the corresponding products in moderate to good yields (Scheme 4). For simple acrylates, such as methyl, tert-butyl, iso-butyl, cyclohexyl, and benzyl acrylate, $C(sp^3)$ -H olefination took place smoothly and provided the desired products in 66-75% yields (5a-5e, respectively). The reaction also performed well for acrylates that contain various functional groups; for example, methoxyl, 2-furanyl, and hydroxyl groups on the alkoxy fragment of the acrylates were compatible and provided the corresponding γ lactams in 70-73% yields (5f-5h). A cyano group on the acrylate was tolerated, albeit in a lower yield (5i). Hexafluoro-2-propanyl acrylate afforded the desired coupling product in 60% yield (5j). When phenyl acrylate was used as the starting material, the reaction delivered a 62% yield of 5j as the final product, rather than the expected phenyl ester product, most likely due to the alcoholysis with the HFIP solvent. Similarly, Scheme 4. Substrate Scope for Acrylates^a



^{*a*}Reaction conditions: **1a** (0.2 mmol), **2b** (3 equiv), $Pd(OAc)_2$ (10 mol %), AgOAc (2 equiv), Na_2CO_3 (2.0 equiv), HFIP (2 mL), 120 °C, 12 h. ^{*b*}2,2,2-Trifluoroethyl acrylate was used, resulting in a mixture of **5k** (35%) and **5j** (35%).

2,2,2-trifluoroethyl acrylate afforded a mixture of free NH γ -lactam 5j (35%) and 5k (35%) under these conditions.

Because self-cleaving directing groups have rarely been reported in C-H functionalization research, several experiments were conducted to shed light on the mechanistic details of this reaction. First, when the reaction was performed in the absence of acrylate, the starting material 1a was recovered in 81% isolated yield, and a trace amount of pivalamide 6 was detected (eq 1). This result indicates that **1a** is relatively stable under the reaction conditions. Furthermore, when pivalamide 6 was used as the starting material in lieu of 1a under the standard conditions, 6 was recovered, and no $C(sp^3)$ -H activation products were detected (eq 2). This observation suggests that amide 1a containing the polyfluoroalkylsulfinyl (SOR_f) auxiliary is the active substrate in $C(sp^3)$ -H activation, rather than pivalamide 6. On the basis of this insight, we reasoned that the cleavage of the auxiliary group likely takes place after the $C(sp^3)$ -H activation/olefination sequence, though we have never isolated the $C(sp^3)$ -H activation/ olefination product(s). Finally, when we used the optically enriched 1a (94% ee) as the starting material, we obtained the desired γ -lactam product in 47% ee, in 71% isolated yield (eq 3). This result also supports the notion that the auxiliary group is expelled after completion of the $C(sp^3)$ -H olefination/ cyclization cascade.

On the basis of these results, we propose a plausible reaction mechanism for this Pd-catalyzed self-cleaving polyfluoroalkyl-sulfinyl (SOR_f) auxiliary-directed $C(sp^3)$ -H olefination/



racemic 1a provided 4a in 73% vield and 0% ee

cyclization process (Scheme 5). The Pd^{II} catalyst coordinates to the polyfluoroalkylsulfinylamide and activates the proximal methyl C(sp³)-H bond to afford a five-membered palladacycle intermediate **A**. The coordination of acrylate and 1,2-migratory insertion of the Pd-C(sp³) bond into the olefin generate intermediate **C**. Subsequent β -H elimination provides olefinated intermediate **D**; the Pd^{II} catalyst is regenerated by Ag⁺ to close the catalytic cycle. Olefinated intermediate **D** undergoes intramolecular Michael addition and *in situ* cleavage of the auxiliary to give the final γ -lactam product.

Scheme 5. Proposed Mechanism



In summary, a new self-cleaving monodentate polyfluoroalkylsulfinyl (SOR_{*f*}) auxiliary for Pd(II)-catalyzed aliphatic $C(sp^3)$ -H bond olefination and tandem cyclization has been developed. This transformation provides a practical procedure for synthesizing NH free γ -lactams. Initial mechanistic investigations reveal that the polyfluoroalkylsulfinyl (SOR_{*f*}) auxiliary is crucial for $C(sp^3)$ -H activation.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/acs.orglett.0c00326.

¹H NMR, ¹³C NMR, ¹⁹F NMR, and HRMS spectra and the experimental procedure (PDF)

Accession Codes

CCDC 1974038 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge via www.ccdc.cam.ac.uk/data_request/cif, or by emailing data_request@ccdc.cam.ac.uk, or by contacting The Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK; fax: +44 1223 336033.

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Notes

The authors declare no competing financial interest.

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