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Concise Unified Access to (–)-8-Deoxy-13-dehydroserratinine, (+)-Fawcettimine, (+)-Fawcettidine, and (-)-8-Deoxyserratinine Using a Direct Intramolecular Reductive Coupling

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ycopodium alkaloids have been immensely appealing research topics since the isolation of lycopodine in 1881 due to their unique and diverse structures and attractive biological activities.¹ With over three hundred natural members, more than 90 Lycopodium alkaloids pertain to the subfamily of fawcettimine. Biosynthetically, the fawcettimine class is generated from lycopodine-type alkaloids via a C4-C13 to C4–C12 bond migration (Figure 1a).² The further division of fawcettimine-type alkaloids is based on the Nattachment; for example, the carbinolamine form commonly possesses a N-C₁₃ connection, whereas the keto-amine form contains either a $N-C_4$ bond or a $N-C_{17}$ bond. (Figure $1b).^{2a,h}$

The diverse and mutable structural features make the total synthesis of fawcettimine-type Lycopodium alkaloids challenging but intriguing work. Early in 1986, Heathcock and coworkers reported their pioneering synthesis of (\pm) -fawcettimine³ in which the closure of the nine-membered ring, which led to the famous 6/5/9 tricyclic Heathcock intermediate² from a 6/5 bicyclic precursor, was designed as a key step. Further epimerization at the C_4 position afforded (\pm)-fawcettimine via spontaneous hemiaminal formation. Heathcock's protocol for accessing fawcettimine-type Lycopodium alkaloids via the 6/5/9 tricyclic intermediates is significant effective in obtaining both carbinolamine and keto-amine forms and has been widely adopted in the past several decades (Figure 1c).⁴ However, a moderate or low efficiency often resulted when a medium-sized ring of Heathcock-type intermediates was formed due to unfavorable transannular interactions and entropic factors. Besides, the annulation of the medium-size rings needs to be carried out at a much lower concentration, which may further hinder the amplification and material





Figure 1. Protocols for the synthesis of fawcettimine-type Lycopodium alkaloids.

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accumulation. To avoid this negative influence on the synthesis of the nine-membered ring(s), Dake's team once elaborated the C and A rings of fawcettidine stepwise through a platinum(II)-catalyzed cyclization and Ramberg–Bäcklund reaction.⁵ Meanwhile, Tu and co-workers also accomplished a 6/9 spiro- β -hydroxylated precursor with a semipinacol rearrangement instead of the corresponding Heathcock intermediate in their syntheses of (–)-lycojapodine A and two other alkaloids.⁶ In addition, a 6/5/5 tricyclic intermediate was also employed in the syntheses of (+)-fawcettimine and three other alkaloids.⁷ Herein, we wish to report a unified, step-saving, and scalable synthetic route to several representative fawcettimine-type *Lycopodium* alkaloids that employs a dense multi-small-ring compound as the key intermediate.

Our retrosynthetic analysis is outlined in Figure 2. (-)-8-Deoxy-13-dehydroserratinine (11), a keto-amine alkaloid, was



Figure 2. Retrosynthetic analysis of the *Lycopodium* alkaloids (-)-8-deoxy-13-dehydroserratinine (11), (+)-fawcettimine (2), (+)-fawcettidine (3), and (-)-8-deoxyserratinine (4).

designed as a natural late-stage intermediate common to the other three fawcettimine-type Lycopodium alkaloids.^{4f,h,7-9} Contrary to the previous situations that use the Heathcock intermediates, the ring system combined with the A and C rings of 11 represent a perfect latent nine-membered ring, which can be immediately exposed through a potential C₄-N disconnection. Opening the piperidine ring (C ring) of 11 via a C₉-N disconnection gives a tricyclic spiro- α -cyclopentanone intermediate 12 with multiple small rings. However, the synthesis of such a fully functionalized spirocyclic pyrrolidine is thought to be challenging,¹⁰ and multiple steps (including several redox steps) or peculiar precursors are usually needed to establish the requisite functionalities, multiple dense stereochemistries, and continuous quaternary carbons. If the direct formation of a C_4 - C_5 bond (to form the D ring) could be achieved via the coupling of the imine and the ester carbonyl of 13, it would be a shortcut to approach the above polycyclic intermediate and would thus greatly improve the step and redox economies of the synthesis. Following this crucial disconnection, the resulting cyclic imine 13 can be further prepared from the linear primary amine 14 via dehydrative cyclization. Further modification of the functional groups delivers a multisubstituted cyclohexanone intermediate

15 that contains a chiral all-carbon quaternary carbon, which could be concisely prepared by a four-step sequence¹¹ starting from the commercially and easily available materials (*SR*)-methylcyclohex-2-en-1-one (**16**),¹² *N*,*N*-di-Boc-4-aminobutanal (**17**),¹³ and methyl acetate (**18**).

The synthesis commenced with a scalable preparation of the cyclohexanone **15**, which contains a chiral quaternary carbon.⁸ The Morita–Baylis–Hillman condensation of the commercially available cyclohexenone **16** and aldehyde **17** afforded the β -hydroxyketone **19** (38 g per batch) in an 84% yield in the presence of *n*-Bu₃P and 1,1'-2-naphthol (Scheme 1). The

Scheme 1. Synthesis of the Cyclic Imine 25



treatment of 19 with Dess-Martin periodinane and sodium bicarbonate in dichloromethane afforded the 1.3-diketone 20 (90% yield). Further reaction with the in situ-generated enolate 21, which was derived from methyl acetate and LiHMDS, provided 22 (79% yield) via a Michael addition from the opposite face of the chiral methyl group. The subsequent Michael addition of 22 to acrolein was carried out in MeCN with perfect stereochemical control by the steric hindrance of the ester. It is noteworthy here that use of acetonitrile as the solvent could further facilitate the selective reduction of the resulting aldehyde with sodium cyanoborohydride in the same flask after acidifying the first-stage reaction mixture with formic acid. This sequential Michael addition/NaBH₃CN reduction treatment finally provided a mixture of hemiketal 23a (major) and cyclohexanone 23b (minor) in an 81% combined yield, wherein the requisite C12 quaternary carbon was correctly established.

To prepare a substrate suitable for the designed reductive imine-ester coupling, the *N*-Boc groups of 23a and 23b were removed by a treatment with trifluoroacetic acid (Scheme 1). However, the reaction could not stop at the free-amine stage, and a substantial portion of the product underwent further hemiaminalization and dehydration to provide an enol-ether

24. Fortunately, the enol-ether functionality of 24 also can be regarded as a special protecting form for both the ketonecarbonyl group (of the six-membered ring) and the hydroxyl group (of the linear chain); therefore, it was suitable for the next transformation. Based on our observation, a continuously sequential operation was finally optimized for the preparation of 24 and 25. First, the mixture of 23a and 23b was treated with 0.3 equiv of boron trifluoride in DCM at 0 °C to rt. It was found that the enol-ether was formed completely while one of the two N-Boc groups of 23a or 23b was also removed. The removal of the remaining N-Boc group was carried out by a further treatment with trifluoroacetic acid. However, the purification of the free amine 24 via silica gel chromatography was found to be inefficient. To save the material, a solution of the unpurified amine 24 DCM was directly applied to the next step after neutralization and drying. The spontaneous formation of the cyclic imine in 25 took place slowly but efficiently when the above solution was heated at 40 °C. To avoid the side reactions brought by long-time heating, the reaction was interrupted after three days, and the expected imine 25 was separated. Usually, 56-66% of the amine 24 can be converted into the corresponding imine 25, with a recovery yield of 87-97%. The recovered amine 24 was redissolved in DCM and heated for the second batch of the crop. After two additional recycles, the combined overall yield of imine 25 could be increased to 82% over two steps.¹⁷

With the cyclic imine **25** in hand, we started to explore the crucial intramolecular reductive imine–ester coupling (Table 1). Although several electrochemical approaches were reported





^{*a*}Reactions were performed at a set concentration of **25** (0.020 M), and lithium arenide (0.25 M) in THF was added. ^{*b*}Isolated yields. ^{*c*}Lithium arenide (0.3 equiv) was added. ^{*d*}Lithium arenide (3.0 equiv) was added; n.d. stands for not detected, and n.r. stands for no reaction.

for similar intramolecular couplings of aromatic imines and esters,¹⁵ few have been explored on the corresponding aliphatic system, especially a complex aliphatic system that contains multiple continuous stereocenters. As a category of powerful reductants involving single-electron processes, lithium-arenides were employed in some challenging C–C bond formations.^{11,16} Due to the uncertainty of the reductive C–C coupling between an aliphatic imine and an ester, metal–arene-based conditions were chosen by us as a feasible starting

point for the examination. Lithium naphthalenide was first examined, but unfortunately the desired spiro- α -aminocyclopentanone 26 was not observed (Table 1, entry 1). To identify a matched combination, a variety of arenes were screened as parterns with lithium on this platform. To our delight, a small amount of the product 26 (7%) was detected when 2,6-di-tertbutylnaphthalene was applied (Table 1, entry 2). According to the general properties of lithium arenides, we suspected that a reductant milder than lithium naphthalenide might be preferred. The parallel tests of reductive couplings showed that lithium acenaphthalenide was too feeble to promote the reaction, and lithium 1.1'-binaphthylide seemed to be too powerful and resulted in messy reaction (Table 1, entries 3 and 4, respectively). Alternatively, the reaction with lithium diphenylide gave us a promising result and delivered an improved yield (37%) of the product 26 (Table 1, entry 5). A noticeable improvement was found in the reaction when using lithium 4,4'-di-tert-butylbiphenylide (LiDBB), affording a satisfying yield (74%) of the spiro- α -aminocyclopentaone 26 as a single diastereoisomer. It is worth noting here that the C_4 stereochemistry of 26 was indirectly determined by a comparison with the literature data of 11 in a late stage (Scheme 2). We speculate that the chelation of lithium ion

Scheme 2. Total Synthesis of the Fawcettimine Class



with the ester carbonyl and the steric effect of the α -amino radical finally led to the observed stereoselectivity. The detailed mechanism for the subsequent C–C bond formation, either by an anionic or free-radical process, remains to be explored. Further optimization showed that lower temperatures could further increase the yields (Table 1, entries 6–8), and the final reaction temperature was determined to be –98 °C.

To simplify the procedure for the scaled-up preparation, the purification of product **26** was waived after completing the reductive coupling of **25** under the optimized conditions. Instead, the resulting reaction mixture was treated with concentrated aqueous hydrochloric acid (Scheme 2). A mixture of **27a** and **27b** (4:1, 2.2 g) was afforded in an 84% yield. The subsequent reaction of **27a** and **27b** with methanesulfonyl chloride and $N_{,N}$ -diisopropylethylamine in DCM delivered the first natural fawcettimine-type alkaloid

(-)-8-deoxy-13-dehydroserratinine (11, 1.2 g, 91% yield). Undoubtedly, the accomplished eight-step synthesis of (-)-8deoxy-13-dehydroserratinine (11, 32% overall yield) shows excellent stereochemical control and much-improved step- and redox economies by using the developed direct intramolecular C-C bond formation between the imine and the ester. With an adequate quantity of the common intermediate 11 in hand, parallel one-step transformations of 11 were accomplished to three other fawcettimine-type alkaloids, (+)-fawcettimine (2), (+)-fawcettidine (3), and (-)-8-deoxyserratinine (5). ^{4f,h,7,9} The spectroscopic data of all four synthetic Lycopodium alkaloids are in good agreement with those reported previously (see the Supporting Information for the details).⁷ To the best of our knowledge, the above achievement represents a new shortest benchmark for the enantiospecific total syntheses of fawcettimine-type Lycopodium alkaloids in the current literature.3-

In summary, we have accomplished in this communication a short, scalable, and collective synthesis of four typical fawcettimine-type Lycopodium alkaloids, namely, (-)-8deoxy-13-dehydroserratinine, (+)-fawcettimine, (+)-fawcettidine, and (-)-8-deoxyserratinine, with much-improved stepand redox economies. The newly developed and successful LiDBB-mediated intramolecular reductive coupling of the alphalic imine and the ester-carbonyl served as a shortcut to the crucial fully functionalized and complex spiro- α -aminocyclopentanone intermediate. The concise formation and application of the dense small-ring intermediate also significantly improved the whole synthesis efficiency compared to those of traditional syntheses via Heathcock intermediates. Finally, a gram-scale synthesis of the natural alkaloid 11 was demonstrated within eight steps and a 32% overall yield. Further parallel transformations of 11 to three other natural Lycopodium alkaloids are also achieved within one step. We believe that the novel intermediates, key methodology, and overall strategy developed in this work will facilitate the synthesis of other Lycopodium alkaloids as well as the related pharmaceuticals.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/acs.orglett.1c00977.

Experimental procedures and NMR spectra for all compounds (PDF)

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Notes

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