One-pot synthesis of benzolactones and lactams via a cobalt-catalyzed regioselective [2 + 2 + 2] cocyclotrimerization of alkynyl alcohols and amines with propiolates†

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An efficient method for the synthesis of benzolactones and benzolactams via a cobalt-catalyzed [2 + 2 + 2] cocyclotrimerization of alkynyl alcohols and alkynyl amines with propiolates is described.

The transition metal catalyzed [2 + 2 + 2] cycloaddition reaction of unsaturated molecules is an efficient method to construct sixmembered cyclic compounds with the formation of three new carbon-carbon bonds.1 This cycloaddition reaction has been known for the past few decades. Of these cycloadditions, the intramolecular and partially intermolecular modes of cyclotrimerization have been established as powerful synthetic methods.² However, completely intermolecular versions of [2 + 2 + 2] cycloaddition generally have encountered difficulties in the control of chemo- and regioselectivity. When two or three different alkynes are used for the cycloaddition, in most cases, a mixture of several benzene derivatives was obtained.³ Therefore, the search for a highly chemo- and regioselective intermolecular [2 + 2 + 2] cycloaddition reaction of two or three different monoynes is greatly challenging. Recently, several groups have devoted substantial effort in this area.⁴ We demonstrated a nickel-catalyzed regio- and chemoselective [2 + 2 + 2] cocyclotrimerization of propiolates with allenes.5 Takeuchi and Nakaya reported a iridium-catalyzed [2 + 2 + 2] cycloaddition reaction of DMAD with substituted propargyl alcohols to afford substituted benzolactones, 4d but only few examples were studied. Our continuous interest in this type of cycloaddition reactions⁶ prompted us to explore the possibility of alkynyl alcohols with propiolates. Herein, we wish to report that cobalt catalyzed the [2 + 2 + 2] cycloaddition reactions of alkynyl alcohols and amines with propiolates to give various substituted benzolactones and lactams in a highly chemoand regioselective fashion.

Benzolactones and lactams are found in plants and they show several pharmacological effects, such as fungicidal, bactericidal, herbicidal and analgesic activities.⁷ The present catalytic reaction provides a convenient method for the synthesis of various substituted benzolactones and lactams in one pot with good to excellent yields.

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Substituted benzolactone 3a was successfully prepared in 91% of isolated yield from the reaction of propargyl alcohol (1a) (1.00 mmol) with DMAD (2) (2.2 mmol) in the co-solvent CH₃CN and THF (2.0 + 2.0 mL) in the presence of $CoI_2(dppe)$ (5.0 mol%)and zinc metal powder (2.75 mmol) (Scheme 1). It is necessary that the solution was initially stirred at room temperature for 1 h to reduce the Co(II) complex (pale green) to the active species (dark green) and then at 80 °C for 12 h to obtain high yield of product 3a. Presumably, the formation of 3a is via a [2 + 2 + 2]cocyclotrimerization of two molecules of 2 and a molecule of 1a followed by transesterification. Product 3a was thoroughly characterized by its ¹H, ¹³C NMR and mass spectral data. Control experiments revealed that in the absence of the cobalt catalyst or Zn powder, no 3a was obtained.

To understand the nature of the catalytic reaction and to determine optimal reaction conditions, the activities of various cobalt phosphine complexes and the effect of solvents on the [2 + 2 + 2]cocyclotrimerization of 1a with 2 were examined. Monodentate phosphine complexes CoCl₂(PPh₃)₂ and CoI₂(PPh₃)₂ exhibited low catalytic activity, affording 3a in 47 and 41% yields, respectively. Cobalt complexes with bidentate phosphine ligands dppm and dppp, gave 3a in 53 and 58% yields. The highest yield of 3a in 95% was obtained using CoI₂(dppe) as the catalyst. The solvent used is critical to the catalytic reaction. Of the solvents tested, the cosolvent acetonitrile and THF (2 + 2 mL) was most effective. Acetonitrile alone was also efficient, affording product 3a in 78% yield. The other solvents, THF, toluene, ethyl acetate and NMP, were not active for present catalytic reaction.

Under similar reaction conditions, substituted alkynyl alcohols **1b-d** also underwent the [2 + 2 + 2] cocyclotrimerization reaction with 2 to afford the corresponding benzolactone derivatives in excellent yields (Table 1). Thus, the reaction of 2 with 1-ethynylcyclohexanol (1b) afforded product 3b in 77% yield (entry 2), whereas but-2-yn-1-ol (1c) gave 3c in 85% yield (entry 3). Similarly, but-2-yne-1,4-diol (1d) undergoing reaction with 2 affording bislactone 3d in 50% yield (entry 4).

Scheme 1

[†] Electronic supplementary information (ESI) available: General experimental procedure, melting point, spectral data (IR, 1H, 13C NMR and HRMS) and copy of ¹H and ¹³C NMR spectra of all compounds. See http://dx.doi.org/10.1039/b509731b

Table 1 Results of cobalt-catalyzed cocyclotrimerization of alkynyl alcohols 1 and DMAD 2^a

Entry	1	2	Product	3	Yield ^b (%)
1	1a	2	MeO ₂ C CO ₂ Me O	3a	91 (95)
2	1b	2	MeO ₂ C	3b	77
3	1c	2	MeO ₂ C CO ₂ Me _O	3c	85
4	1d	2	MeO ₂ C CO ₂ Me	3d	50
5	1e	2	MeO ₂ C MeO ₂ C CO ₂ Me	3e	91
6	1f	2	MeO ₂ C MeO ₂ Me	3f	82
7	1g	2	MeO ₂ C CO ₂ Me MeO ₂ C CO ₂ Me	3g	80

^a All reactions were carried out using alkynyl alcohols 1 (1.00 mmol), DMAD (2) (2.2 mmol), CoI₂(dppe) (5.0 mol%), Zn (2.75 mol%) and CH₃CN and THF (2.0 + 2.0 mL) at room temperature for 1 h and then at 80 °C for 12 h. ^b Isolated yields: the yields in parenthesis was determined by ¹H NMR integration method.

The cobalt-catalyzed [2 + 2 + 2] cocyclotrimerization reaction can also be used for the preparation of six- and seven-membered benzolactones (Scheme 2). Thus, treating but-3-yn-1-ol (1e) and pent-4-yn-1-ol (1f) with 2 under the standard conditions afforded lactones 3e-f in 91 and 82% yields, respectively (Table 1, entries 5-6). However, an attempt to make an eight-membered lactone ring from the reaction of hex-5-yn-1-ol 1g and 2 afforded only the noncyclized product 3g in 80% yield (entry 7).

The present protocol can be further applied to propiolates 4a-c with alkynyl alcohols 1a, 1e and 1h under the similar reaction conditions to give highly regio- and chemoselective [2 + 2 + 2] cycloaddition and transesterification products 5a-f in good yields

Scheme 2

(Scheme 3). The reaction of 1a with pentylpropiolate 4a afforded highly substituted benzolactone 5a in 45% yield (Table 2, entry 1). The NOE data of product 5a support the proposed structure in entry 1 and exclude the other possible regioisomers. Similarly, phenylpropiolate 4b underwent cycloaddition reaction with 1a and 1h to afford products 5b,c in 40 and 35% yields, respectively. Again, NOE studies of these products were consistent with the proposed structures with the two phenyl groups ortho to each other.

Scheme 3

In a similar manner, methyl propiolate (4c) reacts smoothly with 1a, 1h and 1e catalyzed by the CoI₂(dppe)/Zn system to yield products that contain a five- and six-membered lactone ring **5d-f** in 70–80% yields (Table 2, entries 4–6). The reaction is completely regioselective; in each case only one regioisomer of the

Table 2 Results of cobalt-catalyzed cocyclotrimerization of alkynyl alcohols 1a, 1e and 1h and propiolates 4a-c^a

Entry	1	4	5		Yield ^b (%)
1	1a	4a	MeO ₂ C MeO ₃ C	5a	45
2	1a	4b	Ph O Ph O MeO ₂ C	5b	50
3	1h	4b	Ph O Ph O MeO ₂ C	5c	35
4	1a	4c	MeO ₂ C	5d	80
5	1h	4c	MeO ₂ C	5e	74
6	1e	4c	MeO ₂ C	5f	70

^a All reactions were carried out using alkynyl alcohols 1 (1.00 mmol), propiolate 4 (2.2 mmol), CoI₂(dppe) (5.0 mol%), Zn (2.75 mol%) and CH₃CN and THF (2.0 + 2.0 mL) at room temperature for 1 h and then at 80 °C for 16 h. b Isolated yields.

Scheme 4

cycloaddition product was observed and the structure was confirmed by NOE studies. In all of these reactions, the homo cyclotrimerization products of propiolates were observed in 5-20% yields.

In addition to propargyl alcohols, the reaction was further extended into propargyl amines (Scheme 4). The reaction of propiolate 4c with N-propargyl p-toluenesulfoneamide 6a in the presence of CoI₂(dppe)/Zn afforded benzolactam derivative 7a in 75% yield. Under similar reaction conditions, propargyl amine **6b** also underwent [2 + 2 + 2] cocyclotrimerization with 4c, but the expected product appears to react further with a propiolate molecule to give **7b** in 72% yield.

The regioselectivity of propargyl alcohol 1a in the present cobalt-catalyzed [2 + 2 + 2] cycloaddition was demonstrated with unsymmetrical divne 8 under the standard reaction conditions affording phthalide derivative 9 in 79% yield as the exclusive product. The other possible product 10 was not detected in the ¹H NMR spectrum of the crude reaction mixture indicating excellent regioselectivity of 1a in the reaction (Scheme 5).

The origin of formation of benzolactones 3, 5 and lactams 7 from the present cobalt-catalyzed cycloaddition of alkynyl alcohols or amines 1 or 6 with propiolates 2 or 4 is likely from the regioselective head to head (R⁴) oxidative cyclometalation of propiolates to give cobaltacyclopentadiene intermediate 11.¹ Insertion of a molecule of alkynyl alcohol (or amine) into a Co-carbon bond of 11 in which the two ester groups are next to the metal center, followed by reductive elimination and transesterification afforded the final product 3, 5 or 7.

It is known that nickel complexes also catalyze the [2 + 2 + 2]cycloaddition of alkynes. To compare the regioselectivity in the nickel- and cobalt-catalyzed reaction, the cycloaddition of 1a with 4a catalyzed by the NiBr₂(dppe)/Zn system was studied. As shown in Scheme 6, an entirely different product 13 in 62% yield was observed. In this product, the two ester groups are meta and para

Scheme 5

Scheme 6

to the hydroxymethyl group and thus, no transesterification could occur. The difference of regioselectivity in the oxidative cyclometalation of propiolates to nickel and cobalt complexes should account for these results. A tail-tail (pentyl group R⁴) oxidative cyclometalation of pentyl propiolate 4a with the nickel complexes to give intermediate 12 is known. 1,2,5,6 We have observed the same type of regioselective products in our pervious nickelcatalyzed [2 + 2 + 2] cycloaddition of propiolates and allenes.⁵

In conclusion, we have developed a cobalt-catalyzed [2 + 2 + 2]cyclotrimerization of alkynyl alcohols with propiolates. The reaction is highly regio- and chemoselective affording benzolactone derivatives in good to excellent yields. Further extension of this work into partially intermolecular version and a detailed mechanistic study is underway.

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Notes and references

- 1 (a) K. P. C. Vollhardt, Angew. Chem., Int. Ed. Engl., 1984, 23, 539; (b) N. E. Schore, Chem. Rev., 1988, 88, 1081; S. Saito and Y. Yamamoto, Chem. Rev., 2000, 100, 2901; (c) S. Ikeda, Acc. Chem. Res., 2000, 33, 511; (d) J. A. Varela and C. Saa, Chem. Rev., 2003, 103, 3787.
- 2 (a) S. Kotha and E. Brahmachary, Tetrahedron Lett., 1997, 38, 3561; (b) A. Jeevanandam, R. P. Korivi, I.-W. Huang and C.-H. Cheng, Org. Lett., 2002, 5, 807; (c) M. Shanmugasundaram, M.-S. Wu, M. Jeganmohan, C.-W. Huang and C.-H. Cheng, J. Org. Chem., 2002, 67, 7724; (d) M.-S. Wu, M. Shanmugasundaram and C.-H. Cheng, Chem. Commun., 2003, 718.
- 3 (a) J. C. Sauer and T. L. Cairns, J. Am. Chem. Soc., 1957, 79, 2659; (b) L. S. Meriwether, E. C. Colthup, G. W. Kennerly and R. N. Reusch, J. Org. Chem., 1961, 26, 5155; (c) A. F. Donda and G. Moretti, J. Org. Chem., 1966, 31, 985; (d) Y. Wakatsuki, T. Kuramitsu and H. Yamazaki, Tetrahedron. Lett., 1974, 4549.
- 4 (a) S. Ikeda, H. Watanabe and Y. Sato, J. Org. Chem., 1998, 63, 7026; (b) Y. Yamamoto, H. Kitahara, R. Ogawa and K. Itoh, J. Org. Chem., 1998, 63, 9610; (c) B. Witulski and C. Alayrac, Angew. Chem., Int. Ed., 2002, 41, 3281; (d) R. Takeuchi and Y. Nakaya, Org. Lett., 2003, 5, 3659; (e) D. Pena, D. Perez, E. Guitian and L. Castedo, J. Am. Chem. Soc., 1999, **121** 5827
- 5 M. Shanmugasundaram, M.-S. Wu and C.-H. Cheng, Org. Lett., 2001, 3, 4233
- 6 (a) D.-J. Huang, T. Sambaiah and C.-H. Cheng, New J. Chem., 1998, 22, 1147; (b) T. Sambaiah, L.-P. Li, D.-J. Huang, C.-H. Lin, D. K. Rayabarapu and C.-H. Cheng, J. Org. Chem., 1999, 64, 3663; (c) M.-S. Wu, D. K. Rayabarapu and C.-H. Cheng, Tetrahedron., 2004, **60**, 10005; (d) T. T. Jayanth, M. Jeganmohan and C.-H. Cheng, J. Org. Chem., 2004, 69, 8445; (e) J.-C. Hsieh, D. K. Rayabarapu and C.-H. Cheng, Chem. Commun., 2004, 532.
- 7 (a) T. Aoki, T. Furusho, T. Kimura, S. Satake and S. Funayama, Jpn. Pat., 7324724, 1973 (Chem. Abstr., 1974, 80, 129246); (b) M. Lacova, Chem. Zvesti., 1973, 27, 525, Chem. Abstr., 1974, 80, 59757; (c) R. C. Elderfield, Hetrocyclic Compounds., Wiley, New York, 1951, vol. 2, ch. 2.