NITROSYLATION OF ETHYL ESTERS OF  $\alpha$ -(3,3-DIMETHYL-

## 1,2,3,4-TETRAHYDRO-1-ISOQUINOLYLIDENE)CARBOXYLIC ACIDS

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We have discovered an unusual transformation of the ethyl esters of  $\alpha\text{-}(3,3\text{-}dimethyl-1,2,3,4-tetrahydro-1-isoquinolylidene)carboxylic acids (I) to the ethyl ester of (3,3-dimethyl-3,4-dihydro-1-isoquinolyl)hydroxyiminoacetic acid (IV), which is accompanied by cleavage of the <math display="inline">\text{C}_{alkyl} - \text{C}_{\alpha}$  bond in the case of (Ib-f).

 $R \,=\, H(a),\; CH_3(b),\; C_2H_5(c),\; n\text{-}C_3H_7(d), n\text{-}C_4H_9(e),\; i\text{-}C_5H_{11}\; (f).$ 

Upon the action of the nitrosylating mixture, esters (Ib-f) apparently initially are converted to C-nitroso compounds (IIb-f) which are viscous blue oils. The UV spectrum for (IIb):  $\lambda_{\text{max}}$  620 nm,  $\epsilon$  115. IR spectrum of (IIb) ( $\nu$ , cm<sup>-1</sup>): 1715 (C=0), 1615 (C=N), 1570 (N=0). PMR spectrum for (IIb) ( $\delta$ , ppm): 0.88-1.62 m (9H, OCH<sub>2</sub>CH<sub>3</sub> and C(CH<sub>3</sub>)<sub>2</sub>), 1.95 s (3H, CH<sub>3</sub>-C-), 2.42-2.98 m (2H, -C-CH<sub>2</sub>-C-), 4.18 q (2H, OCH<sub>2</sub>CH<sup>3</sup>), 6.68-7.75 m (4H, aromatic protons).

Nitroso compounds (IIb-f) readily dissolve in excess acid, and they turn yellow after 2-3 h, indicating their dimerization to (IIIb-f). After neutralization of the acid solutions of (IIIb-f) by the action of NH<sub>4</sub>OH or Na<sub>2</sub>CO<sub>3</sub>, oxime (VI) is obtained in 60-97% yield independently of the nature of R (the yield is dependent on the nature of R), mp 167-169°C (from acetone), hydrochloride salt, mp 218°C (dec.). PMR spectrum ( $\delta$ , ppm): 1.24 t (3H, OCH<sub>2</sub>CH<sub>3</sub>), 1.30 s (6H, C(CH<sub>3</sub>)<sub>2</sub>), 2.67 s (2H, -C-CH<sub>2</sub>-C-), 4.12 q (2H, OCH<sub>2</sub>CH<sub>3</sub>), 7.00-7.50

m (4H, aromatic protons), 12.35 s (1H, OH). IR spectrum ( $\nu$ , cm<sup>-1</sup>): 3450 (OH), 1725 (C=O), 1605 (C=N), 1565 (C=N). The structure of (IV) was indicated by elemental analysis and its identity to a sample obtained from (Ia).

Although the mechanism for the transformation of (Ib-f) to (IV) is not clear, we have established that the formation of (IV) occurs only after the addition of nucleophiles to the reaction mixture, and intermediate nitroso compounds (IIb-f) are not converted to oxime (IV) upon treatment by the same nucleophiles.

## LITERATURE CITED

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