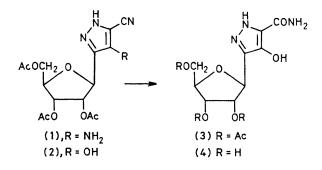
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## A New Synthesis of Pyrazofurin

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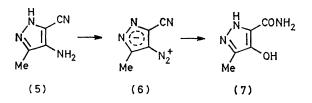
Summary Pyrazofurin (4) has been synthesised by a novel route, involving, as the key step, photochemical substitution of a diazopyrazole

THE C-nucleoside antibiotic pyrazofurin  $(pyrazomycin)^1$ (4), isolated from the culture filtrate of a strain of *Strepto-myces candidus*,<sup>2</sup> shows activity against a number of viruses<sup>3</sup> and tumours,<sup>4</sup> and is currently undergoing clinical trials



Pyrazofurin has previously been synthesised by two distinct routes,<sup>5,6</sup> and we now report the application of our general method for C-nucleoside synthesis,<sup>7</sup> involving acetylenic intermediates, to the synthesis of pyrazofurin, these methods have recently been successfully applied to the synthesis of the related antibiotic formycin <sup>8</sup>

Our earlier work<sup>8</sup> had led to the synthesis of the 4-aminopyrazole (1) Use of this intermediate for the synthesis of pyrazofurin clearly necessitates the replacement of nitrogen with oxygen at C-4 and a sequence involving diazotisation suggested itself Use of this procedure for the synthesis of 4-hydroxypyrazoles is, however, not well documented, owing presumably to the known stability of pyrazole diazonium salts <sup>9</sup> Furthermore, it would be necessary to carry out a diazotisation prior to conversion of the nitrile into an amide, in order to avoid cyclisation to a pyrazolo-[4,3-d]triazinone <sup>10</sup> Diazotisation (NaNO<sub>2</sub>, MeCO<sub>2</sub>H, 0 °C) of the model compound 4-amino-5-cyano-3-methylpyrazole (5)<sup>10</sup>, and subsequent neutralisation (NaHCO<sub>3</sub>), gave the



crystalline diazopyrazole (6) which was stable under a variety of hydrolytic conditions When a solution of (6) in aqueous acetone was subjected to u v photolysis<sup>11,12</sup> through Pyrex the 4-hydroxy compound (7) was produced in 62% overall yield When a similar sequence was applied to (1) the hydroxypyrazole (2) was obtained in 87% yield

Hydrolysis of the nitrile function was best achieved using nickel acetate tetrahydrate in boiling acetic acid,<sup>13</sup> to give pyrazofurin triacetate (3) (65%); treatment of (3) with ammonia in aqueous methanol produced crystalline pyrazofurin (4) (75%), indistinguishable from an authentic sample.

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- G. D. Daves, jun., and C. C. Cheng, Prog. Med. Chem., 1976, 13, 303.
   K. Gerzon, R. H. Williams, M. Hoehn, M. Gorman, and D. C. DeLong, Abstr., 2nd Int. Congr. Heterocycl. Chem., Montpellier, France, 1969, p. 131.
- France, 1969, p. 131.
  <sup>3</sup> G. E. Gutowski, M. J. Sweeney, D. C. DeLong, R. L. Hamill, K. Gerzon, and R. W. Dyke, Ann. N.Y. Acad. Sci., 1975, 255, 544.
  <sup>4</sup> M. S. Sweeney, F. A. Davis, G. E. Gutowski, R. L. Hamill, D. H. Hoffmann, and G. A. Poore, Cancer Res., 1973, 33, 2619.
  <sup>5</sup> J. Farkaš, Z. Flegelovà, and F. Šorm, Tetrahedron Lett., 1972, 2279.
  <sup>6</sup> S. DeBernardo and M. Weigele, J. Org. Chem., 1976, 41, 287.
  <sup>7</sup> J. G. Buchanan, A. R. Edgar, and M. J. Power, J. Chem. Soc., Perkin Trans. 1, 1974, 1943, and subsequent papers.
  <sup>8</sup> J. G. Buchanan, A. R. Edgar, R. J. Hutchison, A. Stobie, and R. H. Wightman, J. Chem. Soc., Chem. Commun., 1980, 237.
  <sup>9</sup> K. Schofield, M. R. Grimmett, and B. R. T. Keene, 'The Azoles,' Cambridge University Press, 1976, pp. 142, 214.
  <sup>10</sup> R. A. Long, J. F. Gerster, and L. B. Townsend, J. Heterocycl. Chem., 1970, 7, 863.

- D. G. Farnum and P. Yates, J. Am. Chem. Soc., 1962, 84, 1399.
   M. F. G. Stevens, Prog. Med. Chem., 1976, 13, 1303, and personal communication.
   J. Newcombe, J. R. Motes, and J. E. Kmiecik, U.S.P. 3,763,235; Chem. Abs., 1973, 79, 136860; R. Breslow, R. Fairweather, and J. S. Kumer, and J. K. Kmiecik, U.S.P. 3,763,235; Chem. Abs., 1973, 79, 136860; R. Breslow, R. Fairweather, and J. S. Kumer, and J. K. Kumer, and K J. Keana, J. Am. Chem. Soc., 1967, 89, 2135.