

# CONFIRMATION OF STRUCTURE AND ABSOLUTE STEREOCHEMISTRY OF 9-EPI-β-CARYOPHYLLENE FROM *DACRYDIUM CUPRESSINUM*

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Key Word Index—Dacrydium cupressinum; Podocarpaceae; rimu; absolute stereochemistry; molecular modelling; sesquiterpene; 9-epi- $\beta$ -caryophyllene;  $\beta$ -caryophyllene.

Abstract—The structure of the rare sesquiterpene 9-epi- $\beta$ -caryophyllene, isolated from the foliage of the New Zealand rimu tree, *Dacrydium cupressinum* was verified by 2D NMR techniques and the relative stereochemistry determined by NOE studies. The absolute stereochemistry was determined by conversion of both  $\beta$ -caryophyllene and 9-epi- $\beta$ caryophyllene into the same known [7,2,0<sup>1.6</sup>,0<sup>1.9</sup>] tricyclic compound. Molecular modelling results on the lowest energy conformations of  $\beta$ -caryophyllene and 9-epi- $\beta$ -caryophyllene were in agreement with experimental results.

#### INTRODUCTION

We have previously reported that the unusual sesquiterpene, 9-epi- $\beta$ -caryophyllene (1) is found at high levels in the foliage of some specimens of the New Zealand rimu tree, Dacrydium cupressinum Lamb. [1]. This compound differs from the widely distributed  $\beta$ -caryophyllene (2) only at the ring junction where the [7,2,0]bicyclo undecane ring system is cis-fused rather than trans-fused [2]. Compound 1 was reported previously by Bohlmann and Zdero [3] who isolated it from Euryops brevipapposus (Compositae, Tribus Senecioneae). Their structural assignment was on the basis of a comparison of the <sup>1</sup>H and <sup>13</sup>C NMR spectra of 1 and its monoepoxide (3) with the spectra of  $\beta$ -caryophyllene (2) and its oxide (4). An earlier report of a 'sesquiterpene of the caryophyllene type' from Abies magnifica (Pinaceae), which had spectral data (IR, MS, <sup>1</sup>H NMR) matching those of 1, may have been the first reported isolation of this compound [4]. Bohlmann and Ziesche [5] have also reported the isolation of the epoxide (3) from the roots of Senecio crassissimus (Compositae).

Barrero, Sánchez and Ferrol [6, 7] have more recently reported 15-hydroxy-9-epi- $\beta$ -caryophyllene (5) from the wood of Juniperus oxycedrus (Cupressaceae). This compound was clearly distinct from 15-hydroxy- $\beta$ -caryophyllene (6) which was synthesized from  $\beta$ -caryophyllene (2) by oxidation with selenium dioxide. It seemed strange that our NMR data for the geminal dimethyl grouping of 1 [1] were quite different to those quoted by Barrero *et al.* [6] for this group in 5. Comparison of NMR data for the related epoxides and keto epoxides derived from the parent hydrocarbons 1 and 2 seemed to reinforce this anomaly (Table 1). A further point of difference was that the <sup>1</sup>H NMR spectrum of 5 recorded at room temperature showed two conformations, present in approximately equal proportions [6]. Spectra of 1 had shown no evidence of more than one conformation [1]. These differences between 1 and 5 suggested that the structure of 1 should be re-examined.

None of the studies so far have determined the absolute stereochemistry for 1 or 5. However, optical rotation values indicate that the 9-epi- $\beta$ -caryophyllene (1) found in *D. cupressinum* was the same enantiomer as that found in *E. brevipapposus*. No optical rotation data were reported for the hydrocarbon from *A. magnifica*. The naturally occurring epoxide 3 isolated from *S. crassismus* is also of the same stereochemical series.

In this paper we report further spectroscopic evidence for the structure of 1 and describe experiments that show (+)-9-epi- $\beta$ -caryophyllene (1) is of the same absolute stereochemistry at C-1 as (-)- $\beta$ -caryophyllene (2). The conformations of caryophyllenes with both *cis*- and the *trans*-fused ring junctions are also examined by molecular modelling.

### **RESULTS AND DISCUSSION**

Diene 1 was isolated from *D. cupressinum* foliage as described previously [1]. Proton chemical shift values were determined by the use of homonuclear-2D-*J*resolved spectra, and a one-bond heteronuclear correlation experiment (HETCOR) was used to inter-relate proton signals with carbon signals. Carbon-carbon bond connectivities were established by the use of proton-proton homonuclear (COSY) and long-range heteronuclear correlation (LRHETCOR) experiments. The DEPT sequence was used to confirm carbon multiplicities. It was possible to determine non-ambiguous substructures from the LRHETCOR spectra even though

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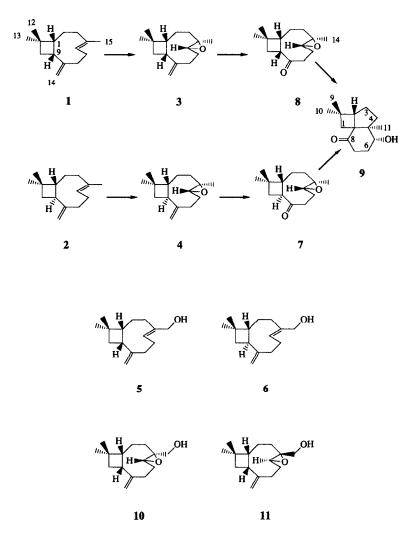


Table 1. NMR data for gem-dimethyl groupings of  $\beta$ -caryophyllenes and 9-epi- $\beta$ -caryophyllenes

	$\beta$ -Caryophyllenes		9-epi-β-Caryophyllenes		
	<sup>1</sup> H NMR	<sup>13</sup> C NMR		<sup>1</sup> H NMR	<sup>13</sup> C NMR
2*	0.97, 0.99	22.6, 30.1	1	0.90, 1.19	25.5, 29.9
6†	0.96, 0.98	22.8, 30.0	5†	0.98, 0.99,	22.1, 21.9,
				1.00	29.8, 30.0
4	0.97, 0.99	21.5, 29.8	3	0.90, 1.19	25.2, 29.8
7	1.04, 1.04	22.3, 29.4	8	0.99, 1.16	24.0, 30.2

\*Ref. [8].

†Ref. [6].

both two- and three-bond correlations were present. Complete NMR data for 1 are given in Table 2.

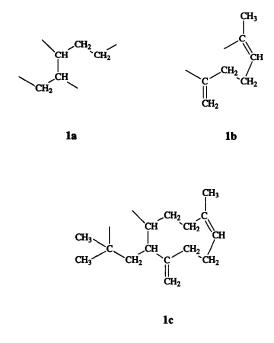
Correlations from the COSY spectrum (Table 2) established two sequences of protonated carbons as 1a and 1b. Although the picture was complicated by peak overlaps in the  $\delta 1.4-1.5$  region, single-frequency decoupling experiments and LRHETCOR correlations removed any ambiguities. Union of these two substructures at two points was enabled based on long-range COSY and LRHETCOR correlations involving the allylic methyl (C-15) and the exocylic methylene (C-14) groupings. This completed a nine-membered ring. Further correlations involving the remaining methyl groups established that these were geminally related. The quaternary carbon which bore these methyl groups was attached as shown in substructure 1c. This structure, on connection of the remaining two unfilled valencies led to the gross features of 1.

NOE experiments confirmed the key stereochemical points (Fig. 1). Evidence for a *cis*-ring junction was shown by an enhancement of the H-1 resonance upon irradiation of H-9, while the *trans*-geometry of the tri-substituted double bond was evident from an enhancement of one of the H-6 signals upon irradiation of H-15. NOE enhancements observed from H-12 to both ring junction methine protons enabled the assignment of the *gem*dimethyl resonances, while the mutual enhancements of H-5 and H-9 demonstrated the orientation of the trisubstituted double bond within the nine-membered ring. The sequence of enhancements involving H-13, H-10 $\alpha$ , the two C-14 protons and H-15 showed that the exocyclic

		Chemical shifts $(\delta)$	Correlations		
С	<sup>13</sup> C	<sup>1</sup> H*	LRHETCOR	COSY	
1	53.8	2.12 m	25.2, 41.6	1.47, 2.89	
2	25.2	1.44 m		2.07	
		1.47 m	_	2.12	
3	41.6	1.88 ddd $(J = 3, 12, 12) \beta$	18.7	1.53, 2.07	
		2.07 m α	—	1.44, 1.88	
4	136.6		_	_	
5	121.9	5.10 ddd $(J = 1, 3, 12) \beta$	18.7, 27.4, 41.6	1.53, 1.97, 2.32	
6	27.4	1.97 m β		1.74, 2.32, 5.10	
		$2.32 m (W_{1/2} = 13) \alpha$	_	1.74, 1.97, 2.45, 5.10	
7	39.2	1.74 m β	27.4, 112.8, 149.8	1.97, 2.32, 2.45, 4.79	
		2.45 dd $(J = 7, 12) \alpha$	27.4	1.74, 2.32	
8	149.7		_		
9	43.0	2.89 ddd $(J = 8, 8, 12)$	25.2, 112.8, 149.8	1.55, 1.70, 2.12	
10	38.2	1.70 dd $(J = 11, 11) \alpha$	25.5, 29.9, 32.9	1.19, 1.55, 2.89	
		1.55 m β		1.70, 2.89	
11	32.9	_		_	
12	29.9	1.19 s	25.5	0.90, 1.70	
13	25.5	0.90 s	32.9	1.19	
14	112.8	4.86 d (J = 1)	43.0	4.79	
		4.79 br s	39.2	1.74, 4.86	
15	18.7	1.53 br s	41.6, 136.6	1.88, 5.10	

Table 2. <sup>1</sup>H and <sup>13</sup>C NMR data for 1

\*Proton  $\delta$  values for overlapping peaks were deduced from HOMO-2D-J resolved spectra.



methylene grouping projects downwards. Thus the dominant conformation of 1 is that described as  $\alpha \alpha$  [8], with both the exocyclic methylene and the allylic methyl groups below the plane of the nine-membered ring.

The strategy used to determine the absolute stereochemistry of 1 was to convert both 9-epi- $\beta$ -caryophyllene (1), of unknown absolute stereochemistry, and (-)- $\beta$ caryophyllene (2), whose absolute stereochemistry has been determined [9], into a common chiral target compound. The success of this method depended critically on the direction by which an attacking reagent approached the tri-substituted double bond in 1 and 2. To achieve the same chirality at C-4 in each case required that the same face of this  $\pi$ -system be exposed.

Selective epoxidation of 2 by *m*-chloroperoxybenzoic acid (*m*-CPBA) gave the known  $\beta$ -epoxide (4). Although Warnoff and Srinivasan had generated both the  $\alpha$ - and  $\beta$ epoxides by using ethereal monoperphthalic acid at 5° [10], no  $\alpha$ -epoxide formation was observed in our work (GLC, <sup>1</sup>H NMR). NOE experiments on 1 (discussed above) had shown that its tri-substituted double bond was also disposed favourably to induce attack on the *re*face of C-4 to give the  $\beta$ -epoxide. Thus, epoxidation of 1, as previously described [3], gave epoxide 3. NOE experiments confirmed that the relative configuration was as expected with similar enhancements to those summarized for 1 in Fig. 1.

Ozonolysis of 4 generated the previously reported epoxy ketone, kobusone (7) which has been isolated from the essential oil of *Cyperus rotundus* (Cyperaceae) [11]. In a similar fashion, ozonolysis of 3 generated the new norsesquiterpene 9-epi-kobusone (8).

As Corey [12] has shown that the *cis*-caryophyllene ring junction is more strained than the *trans*-arrangement, it was hoped that epimerization at C-9 could be achieved in 8 to generate compound 7. Investigation of enolate formation in 7 by using NaOD-D<sub>2</sub>O-D<sub>1</sub>ethanol showed that the C-7 methylene ring protons were quickly exchanged. Although a longer reaction time did result in deprotonation at C-9, this led to formation of the

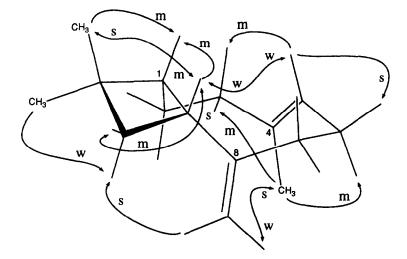


Fig. 1. Important observed NOEs for 1. w, Weak (1-2%); m, medium (2-5%); s, strong (5-10%).

tricyclic compound 9, di-deuterated at C-7. Under nondeuterating conditions, the previously reported tricycle (9) was obtained as the major product [11, 13]. Similar treatment of 8 also gave 9. Spectral data for the two samples were identical and specific rotations at five different wavelengths were very similar. Thus, the sample of 1 from *D. cupressinum* is (1R,4Z,9R)-(+)-9-epi- $\beta$ caryophyllene. This absolute configuration is consistent with biosynthesis of 1 by cyclization to the *si*-face of the 10,11-double bond of farnesyl pyrophosphate, as for the other rimu sesquiterpenes [1] and indeed for all the sesquiterpenes of higher plants [14]. The absolute stereochemistry of the sample of 1 from *E. brevipapposus* [3] and of the epoxide 3 from *Senecio crassissimus* [5], follows from this result.

The only reported molecular modelling study on the caryophyllene ring system is that by Shirahama et al. [8] on  $\beta$ -caryophyllene (2).  $\beta$ -Caryophyllene is known to exist as a mixture of two major conformations in solution at room temperature. The results of hydroboration [15] and epoxidation [10] experiments require the presence of at least one conformation with the allylic methyl group below the plane of the nine-membered ring, and at least one with it above. Shirahama et al. [8] later detailed the existence of two exchanging conformations from inspection of the <sup>13</sup>C NMR spectrum at different temperatures. Molecular models indicated four strain minimum conformations ( $\beta \alpha$ ,  $\beta \beta$ ,  $\alpha \alpha$ , and  $\alpha \beta$ ), distinguished by the positions of the exocyclic methylene (listed first) and the allylic methyl groups relative to the plane of the ninemembered ring [8]. Molecular modelling (MMI force field) on these suggested that the  $\beta \alpha$  and  $\beta \beta$  conformations would predominate in solution [8].

We now report molecular modelling studies on the caryophyllene and 9-epi-caryophyllene system using the systematic variant of the Monte Carlo search procedure, which has been reported as the most effective for searching cyclic flexible molecules [16]. These studies made use of MM2 force field parameters [17].

When this method was applied to  $\beta$ -caryophyllene (2), the results (Table 3) were in agreement with previous findings [8, 10, 15]. This MM2 modelling indicated that the  $\beta\alpha$  and  $\beta\beta$  conformations would predominate in solution, but the predicted populations (roughly 50:50) do not match the <sup>13</sup>C NMR data as closely as those predicted previously by MMI [8].

Conformational searches on 9-epi- $\beta$ -caryophyllene (1) predict that an  $\alpha\alpha$  conformation would be followed, 6.4 kJ mol<sup>-1</sup> higher in energy, by an  $\alpha\beta$  conformation (Table 3). This corresponds to a Boltzmann population distribution of 93:7  $\alpha\alpha$ :  $\alpha\beta$  (at 300 K). The prediction of a dominant  $\alpha\alpha$  conformation of 1 in solution is in agreement with the results from the NOE experiments (Fig. 1), with the production of only epoxide 3 by *m*-CPBA oxidation, and with the observation of only 15 peaks in

Table 3. Molecular modelling on  $\beta$ -caryophyllene and derivatives

	Conformer	Energy (kJ mol <sup>-1</sup> )
1	αα	223.34
	αβ*	229.71
	αβ	229.89
2	ββ	202.56
	βα	202.64
	άα	208.21
	αβ	221.18
5	aat	230.49
	aat	232.37
	aat	233.68
	aat	235.95
	αβ*	237.03
	αβ	237.30

\*Planar cyclobutane ring [20].

†These conformations are C4-C15 rotamers. the <sup>13</sup>C NMR spectrum over a temperature range of -60 to  $+70^{\circ}$ .

Alcohol 5 has been reported to be a mixture of two major conformations at room temperature in approximately equal proportions. As two mono epoxides, 10 and 11, were produced in the ratio 7:3 when 5 was epoxidized with m-CPBA [7], the major conformations of 5 were assigned as  $\alpha \alpha$  and  $\alpha \beta$ . Molecular modelling on alcohol 5 resulted in many discrete conformations (Table 3). The four lowest energy conformations were all aa, differing only in the orientation of the hydroxyl group. An  $\alpha\beta$ conformation was predicted to occur 6.5 kJ mol<sup>-1</sup> higher in energy above the global minimum. This energy difference corresponds to an  $\alpha\beta$ -population of 7%. This failure of the modelling to match reported experimental observations for 5 may indicate that some factor which is not recognized in the MM2 force field is at play. A possibility is intramolecular hydrogen bonding between the hydroxyl group and the exocyclic double bond [18].

We have thus confirmed that the sesquiterpene isolated from rimu foliage has the gross structure and absolute stereochemistry depicted in 1. NMR studies on 1 show that it exists as one major conformation in solution at room temperature, this being an  $\alpha\alpha$  form. Modelling studies on 1 and on  $\beta$ -caryophyllene (2) relate well to both the chemistry and to the NMR spectra, but fail to provide any insight into the discrepancies between the observed spectral data for 1 and its 15-hydroxy derivative 5.

#### **EXPERIMENTAL**

Molecular modelling. These studies were carried out on a Silicon Graphics Personal Iris computer using Macromodel V3.1X [19] with MM2 force field parameters [17]. Conformations were generated by opening the C2-C3 bond and varying torsions C10-C9-C8-C7, C9-C8-C7-C6, C8-C7-C6-C5 and C7-C6-C5-C4. Further C3-C4-C15-O and C4-C15-O-H torsions were included for compound 5 which contained a C-15 hydroxyl moiety. Five hundred conformations were generated and filtered to exclude geometries with energies greater than 50 kJ mol<sup>-1</sup> above the global (current) energy minimum, then minimized using the Polak-Ribiere conjugate gradient minimization procedure with the stopping criteria of 0.05 kJ mol<sup>-1</sup>Å. All minimized conformations were compared and only unique conformations within 50 kJ mol<sup>-1</sup> of the global (final) minimum were stored. Searches were repeatable, with search results independent of starting geometry or solvent treatment. Identical conformations were produced if the search was completed with CHCl<sub>3</sub> solvent treatment rather than in vacuo.

Diene 2. A  $\beta$ -caryophyllene sample (ex BDH) was purified on a silica column with hexane elution to remove humulene and oxidized material. Pure 2 was collected; <sup>1</sup>H NMR identical to an authentic sample;  $[\alpha]^{24} - 13.0^{\circ}$ (589 nm);  $-13.8^{\circ}$  (578 nm);  $-15.2^{\circ}$  (546 nm);  $-17.5^{\circ}$ (436 nm);  $-5.7^{\circ}$  (365 nm) (CHCl<sub>3</sub>; c 6.96).

Kobusone (7). Compound 7 was prepared from 2 by the published methods [3, 11]. Recrystallization from  $Et_2O$ 

gave crystals; mp 52–54° (lit. [11]: mp 60°); IR  $v_{max}^{\text{KBr}}$  cm<sup>-1</sup>: 1693 (C=O); <sup>1</sup>H NMR (299.99 MHz, CDCl<sub>3</sub>):  $\delta$  0.95 (1H, m, H-3), 1.04 (6H, s, H-12, 13), 1.31 (3H, s, H-14), 1.43 (1H, m, H-6), 1.52 (1H, m, H-2), 1.62 (1H, m, H-2), 1.67 (1H, m, H-10), 1.94 (1H, ddd, J = 1, 10, 10 Hz, H-1), 2.07 (1H, dd, J = 10, 10 Hz, H-10), 2.16 (1H, ddd, J = 4, 4, 13 Hz, H-3), 2.40 (1H, m, W<sub>1/2</sub>=30 Hz, H-6), 2.56 (2H, m, W<sub>1/2</sub>=15 Hz, H-7), 2.69 (1H, dd, J = 5, 10 Hz, H-5), 3.06 (1H, ddd, J = 9, 9, 9 Hz, H-9); <sup>13</sup>C NMR (74.9 MHz, CDCl<sub>3</sub>):  $\delta$ 16.3 (q, C-14), 22.3 (q, C-13), 24.9 (t, C-6), 26.6 (t, C-2), 29.4 (q, C-12), 34.5 (s, C-11), 35.4 (t, C-10), 37.8 (t, C-7), 39.1 (t, C-3), 51.4 (d, C-1), 52.7 (d, C-9), 58.0 (s, C-4), 61.7 (d, C-5), 214.3 (s, C-8).

Deuterated 7. Na metal (0.3 g) was dissolved in EtOD (5 ml) and the resultant soln added dropwise with vigorous stirring to a suspension of 7 (51.6 mg) in D<sub>2</sub>O (5 ml) under a N<sub>2</sub> atmosphere. EtOD (3 ml) was added to effect complete dissolution. After stirring at room temp. for 6.5 hr, H<sub>2</sub>O (10 ml) was added. Et<sub>2</sub>O extraction (3 × 10 ml), drying (MgSO<sub>4</sub>) and evapn yielded di-deuterated 7 (54.5 mg). A portion was chromatographed on basic Al<sub>2</sub>O<sub>3</sub> to give the pure compound; <sup>1</sup>H NMR (299.99 MHz, CDCl<sub>3</sub>):  $\delta$ 1.43 (dd, J = 10, 13 Hz, H-6), 2.40 (dd, J = 5, 13 Hz, H-6), 2.56 (no peak, H-7); <sup>2</sup>H NMR (46.04 MHz, CHCl<sub>3</sub>):  $\delta$ 2.6 (s, D-7); <sup>13</sup>C NMR (74.9 MHz, CDCl<sub>3</sub>):  $\delta$ 37.2 (quin, J = 20 Hz, C-7), 37.8 (no peak); MS m/z 224.1798 [M]<sup>+</sup>, C<sub>14</sub>H<sub>20</sub>D<sub>2</sub>O<sub>2</sub> requires 224.1745.

9-epi- $\beta$ -Caryophyllene (1). 9-epi- $\beta$ -Caryophyllene was isolated by extraction of rimu foliage as described in ref. [1]. The <sup>1</sup>H and <sup>13</sup>C NMR data are summarized in Table 2;  $[\alpha]^{24} + 186^{\circ}$  (589 nm); +188° (578 nm); +193° (546 nm); +211° (436 nm); +275° (365 nm) (CHCl<sub>3</sub>; c 0.75).

9-epi- $\beta$ -Caryophyllene oxide (3). Epoxidation of diene 1 as described in ref. [3] gave epoxide 3; <sup>1</sup>H NMR (299.99 MHz, CDCl<sub>3</sub>):  $\delta$ 0.90 (3H, s, H-13), 1.16 (3H, s, H-15), 1.19 (3H, s, H-12), 1.35 (1H, m, H-6), 1.52 (2H, m, H-2), 1.65 (1H, ddd, J = 3, 11, 11 Hz, H-10), 1.75 (1H, dd, J = 11, 11 Hz, H-10), 1.91 (1H, ddd, J = 2, 6, 13 Hz, H-7), 2.01 (1H, dd, J = 11, 11 Hz, H-6), 2.07 (1H, m, H-1), 2.12 (1H, m, H-3), 2.20 (1H, m, H-3), 2.61 (1H, m,  $W_{1/2} = 27$  Hz, H-7), 2.64 (1H, dd, J = 2, 11 Hz, H-5), 3.02 (1H, ddd, J = 8.5, 8.5, 11 Hz, H-9); <sup>13</sup>C NMR (74.9 MHz, CDCl<sub>3</sub>):  $\delta$ 18.5 (q, C-15), 21.3 (t, C-2), 25.2 (q, C-13), 27.2 (t, C-6), 29.8 (q, C-12), 33.0 (s, C-11), 36.5 (t, C-7), 38.2 (t, C-10), 41.4 (t, C-3), 42.3 (d, C-9), 53.5 (d, C-1), 60.1 (s, C-4), 61.8 (d, C-5), 114.5 (t, C-14), 148.4 (s, C-8).

9-epi-Kobusone (8). A soln of epoxide 3 (255 mg) in MeOH was treated with  $O_3$  at  $-78^{\circ}$  until a persistent blue colour remained. Dimethylsulphide (5 ml) was added after excess  $O_3$  had been removed by purging with  $N_2$ , and the reaction mixt. was stirred for 12 hr. Evapn, followed by washing with NaCl soln then chromatography (basic Al<sub>2</sub>O<sub>3</sub>) and recrystallization from Et<sub>2</sub>O gave [1*R*-(1*R*\*,4*R*\*,6*R*\*, 10*R*\*)]-4,12,12-trimethyl-5-oxatricyclo[8.2.0.0<sup>4.6</sup>]dodecan-9-one (8) as crystals (101 mg); mp < 50°; (found: C, 75.5; H, 9.7. C<sub>14</sub>H<sub>22</sub>O<sub>2</sub> requires: C, 75.6; H, 10.0%); IR  $\nu_{\text{max}}^{\text{MBr}}$  cm<sup>-1</sup>: 1698 (C=O); <sup>1</sup>H NMR (299.99 MHz, CDCl<sub>3</sub>):  $\delta$  0.91 (1H, m, H-3), 0.99 (3H, s, H-13), 1.14 (3H, s, H-14), 1.16 (3H, s, H-12), 1.52 (1H, dddd, J = 2, 2, 7, 15 Hz, H-2), 1.59 (1H, m, H-10), 1.63 (1H, m, H-6), 1.76 (1H, m, H-2), 2.09 (1H, m, H-10), 2.14 (1H, m, H-6), 2.18 (1H, dd, J = 7, 7 Hz, H-3), 2.27 (1H, br dd, J = 9, 9 Hz, H-1), 2.42 (1H, m, H-7), 2.79 (1H, m, H-7), 2.80 (1H, dd, J = 6, 6 Hz, H-5), 3.67 (1H, ddd, J = 8, 9, 9 Hz, H-9); <sup>13</sup>C NMR (74.9 MHz, CDCl<sub>3</sub>):  $\delta$ 17.3 (q, C-14), 21.1 (t, C-2), 23.8 (t, C-6), 24.0 (q, C-13), 30.2 (q, C-12), 34.1 (s, C-11), 35.1 (t, C-10), 40.4 (t, C-3), 41.5 (t, C-7), 45.7 (d, C-9), 52.5 (d, C-1), 60.0 (s, C-4), 61.0 (d, C-5), 213.4 (s, C-8).

Tricyclic keto alcohol 9. (a) Kobusone (7) (496 mg) was heated under reflux with KOH-EtOH (10%, 300 ml) for 2 hr. The mixt. was cooled and H<sub>2</sub>O (500 ml) was added. Et<sub>2</sub>O extraction and chromatography (basic Al<sub>2</sub>O<sub>3</sub>) followed by recrystallization  $(2 \times \text{ from } \text{Et}_2\text{O})$  gave 9 (247 mg); mp 148° (lit. [13]: 148–149°); IR  $v_{max}^{KBr}$  cm<sup>-1</sup>: 3504 (O-H), 1684 (C=O); <sup>1</sup>H NMR (299.99 MHz, CDCl<sub>3</sub>):  $\delta$  0.83 (3H, s, H-11), 0.89 (3H, s, H-10), 0.96 (3H, s, H-9), 1.38 (1H, d, J = 12 Hz, H-1), 1.64 (1H, m,  $W_{1/2}$ = 28 Hz, H-4), 1.77 (1H, m, H-3), 1.82 (1H, m, H-6), 2.00 (2H, m, H-3 and H-6), 2.05 (1H, m, H-4), 2.30 (1H, m, H-2a), 2.36 (1H, m, H-7), 2.48 (1H, m, H-1), 2.54 (1H, ddd, J =6, 14, 14 Hz, H-7, 3.87 (1H, dd, J=4, 12 Hz, H-5); <sup>13</sup>C NMR (74.9 MHz, CDCl<sub>3</sub>): δ11.9 (q, C-11), 24.6 (q, C-10), 25.3 (t, C-3), 29.9 (q, C-9), 30.5 (t, C-6), 31.3 (s, C-2), 33.3 (t, C-1), 36.4 (t, C-7), 37.5 (t, C-4), 53.3 (s, C-4a), 55.0 (d, C-2a), 58.2 (s, C-8a), 70.4 (d, C-5), 212.3 (s, C-8);  $[\alpha]^{24} - 32.4^{\circ}$ (589 nm);  $-33.4^{\circ}$  (578 nm);  $-36.6^{\circ}$  (546 nm);  $-35.9^{\circ}$  $(436 \text{ nm}); +77.4^{\circ} (365 \text{ nm}) (CHCl_3; c 1.16).$ 

(b) Treatment of 9-epi-kobusone (8) (35.3 mg) in an identical manner gave 9 (15.5 mg); identical mp and spectral data to those from 7;  $[\alpha]^{24} - 33.5^{\circ}$  (589 nm);  $-33.8^{\circ}$  (578 nm);  $-36.7^{\circ}$  (546 nm);  $-36.4^{\circ}$  (436 nm);  $+80.4^{\circ}$  (365 nm) (CHCl<sub>3</sub>; c 1.16).

(c) Treatment of kobusone (7) with NaOD-EtOD- $D_2O$  as described for the preparation of deuterated 7, with a reaction time of 72 hr, gave a sample of 9 which was di-deuterated at C-3; <sup>1</sup>H NMR (299.99 MHz, CDCl<sub>3</sub>):  $\delta$ 2.34 (no peak, H-7), 2.54 (no peak, H-7); <sup>2</sup>H NMR (46.04 MHz, CHCl<sub>3</sub>):  $\delta$ 2.33 (1D, s, D-7), 2.54 (1D, s, D-7); <sup>13</sup>C NMR (74.9 MHz, CDCl<sub>3</sub>):  $\delta \sim 35.6$  (m, C-7), 36.4 (no peak, C-7); MS m/z 224.1745 [M]<sup>+</sup>, C<sub>14</sub>H<sub>20</sub>D<sub>2</sub>O<sub>2</sub>, requires 224.1745.

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