

HYDROLYSIS AND SYNTHESIS OF BRANCHED CYCLOMALTO-HEXAOSSES WITH *Pseudomonas* ISOAMYLASE

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ABSTRACT

The action of *Pseudomonas* isoamylase on branched cyclomaltohexaoses (α -cyclodextrins, cG_6s) with side chains of various lengths and the reverse condensation reaction between malto-oligosaccharides (G_2 - G_7) and cG_6 have been studied. The rates of reaction for the liberation and the attachment of maltotriose were maximal in both the hydrolysis and the condensation reactions, and the activity decreased with increasing length of the side chain. The values of V_{max} (U/mg) for the hydrolytic reactions for G_2 -, G_3 -, G_4 -, and G_5 - cG_6 were 2.6, 690, 320, and 290, respectively, and the values of K_m (mM) were 72, 204, 92, and 47, respectively. The structures of the new branched cG_6s (G_4 - cG_6 and G_5 - cG_6), obtained through the condensation reaction, were identified by means of enzymic analyses, and ^{13}C -n.m.r. and f.a.b.-mass spectra. The haemolytic activities of these branched cG_6s are reported.

INTRODUCTION

Isoamylase (EC 3.2.1.68), which hydrolyses the (1 \rightarrow 6)- α -D-glucosidic linkages in amylopectin and glycogen, is an essential enzyme for analysis of the molecular structures of these polysaccharides and related oligosaccharides, and plays an important role in the industrial production of glucose and maltose from starch. *Pseudomonas* isoamylase¹ is commercially available, and its structure and action have been investigated in detail². However, the relationship between the activities and the structures of substrates has not been elucidated fully because suitable substrates are not readily available. This difficulty was overcome by

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following the reverse condensation action^{3,4} instead of the forward hydrolytic action. The main purpose of this study was to elucidate the action of the enzyme in relation to the length of the side chains of the substrate in both the condensation and hydrolytic reactions. The condensation reactions were carried out between cyclomaltose (cG₆) and a series of malto-oligosaccharides [maltose (G₂)-maltoheptaose (G₇)], and the hydrolytic reactions were performed on the products, *i.e.*, branched cG₆s with G₂-G₅ branches.

Branched cyclomalto-oligosaccharides (cyclodextrins, cG_ns) with glucose or malto-oligosaccharides attached at positions 6 of their glucosyl residue(s) are produced in minor quantities by the action of cyclodextrin glucanotransferase [(1→4)- α -D-glucan 4-D-glucosyltransferase (cyclising), EC 2.4.1.19] on starch⁵⁻⁷. Branched cG_ns have some superior properties compared to those of non-branched cG_ns, *i.e.*, they are highly soluble in water and form water-soluble complexes with hydrophobic materials⁸⁻¹¹. The preparation of maltosyl-(G₂-) and maltotriosyl-(G₃-)cG_ns by means of *Pseudomonas* isoamylase^{3,4} has been reported. Branched cG_ns were also synthesised by means of the reverse¹²⁻¹⁴ or transfer^{15,16} actions of pullulanase and isoamylase. Some properties of branched cG₆s are now described.

EXPERIMENTAL

Materials. — Purified *Pseudomonas* isoamylase, a series of malto-oligosaccharides (G₂-G₇), and cG₆ were the products of Hayashibara Biochemical Laboratories Inc. (Okayama). Beta-amylase¹⁷ and glucoamylase GIII¹⁸ were purified from sweet potato and a crude commercial preparation of *Rhizopus delemar*, respectively.

Analyses. — The products of reaction were quantified by h.p.l.c. on TSKgel NH₂-60 with acetonitrile-water (65:35) as the eluant, at 0.8 mL/min, with monitoring with a differential refractometer (RI-8000, Tosoh). Reducing power and total carbohydrate (as glucose) were assayed by the colorimetric method of Nelson¹⁹ and Somogyi²⁰, and an anthrone-sulfuric acid method²¹, respectively. The solubility and haemolytic activity of branched cG₆s were determined as described¹⁰.

¹³C-N.m.r. spectra (50.10 MHz) were recorded at 50° for 2-3% solutions in D₂O, using a JEOL JNM-FX200 spectrometer. The chemical shifts are expressed in p.p.m. downfield from the signal of Me₄Si, with reference to internal 1,4-dioxane (67.40 p.p.m.); the detailed conditions are given elsewhere³.

F.a.b.-mass spectra in the negative mode were obtained with a JEOL JMS-HX 110 mass spectrometer, using glycerol as the matrix⁴.

Preparation of branched cG₆. — G₃-, G₄-, and G₅-cG₆ were synthesised by condensation of the respective malto-oligosaccharides and cG₆ with isoamylase. The pH and temperature for the reaction were based on those for the preparation⁴ of G₃-cG₇, but the concentrations of the substrates were not optimised. The reaction mixtures, each containing 275mM cG₆, 605mM oligosaccharide, and 8.2 U/mL of isoamylase, in 50mM acetate buffer (pH 4.2) were incubated for 48 h at 58°. The

reaction was terminated by boiling, and the products were isolated by gel-permeation chromatography on Bio-Gel P-2 and Toyopearl HW40 and then purified by h.p.l.c. as described^{3,4,14}. $G\text{-}cG_6$ was produced from $G_3\text{-}cG_6$ by hydrolysis with glucoamylase (0.1 U/ μmol of $G_3\text{-}cG_6$) in 10mM acetate buffer (pH 4.5) for 24 h at 45°, and was isolated using a column (5.5 \times 46 cm) of Bio-Gel P-2 and elution with distilled water. $G_2\text{-}cG_6$ was prepared by the condensation of maltose and cG_6 with *Klebsiella aerogenes* pullulanase under conditions similar to those used for the preparation¹⁴ of $G_2\text{-}cG_8$. These products were purified by h.p.l.c. on ODS-Hypersil-5 (Shandon) with aqueous ~8% methanol as the solvent, each giving a single peak.

RESULTS AND DISCUSSION

Action of isoamylase. — *Pseudomonas* isoamylase releases maltotriose in preference to maltose from the beta-limit dextrin of amylopectin²², and causes³ the much faster condensation of maltotriose than of maltose to cG_n . The activities of the enzyme toward malto-oligosaccharides up to G_7 have now been examined (Table I). Experiments 1 and 2 were carried out with constant concentrations of the substrate (mol and weight, respectively). Experiment 2 avoided the possible decrease in the activity in the reactions of large substrates due to insufficient water⁴. In each experiment, maltose and maltotriose were the least and most effective substrates, respectively. Maltotetraose was a little less effective than maltotriose as substrate; for the higher malto-oligosaccharides, the enzyme showed less than half the activity for maltotetraose. The time courses of the formation of $G_2\text{-}$ to $G_5\text{-}cG_6$ (Fig. 1) showed that $G_2\text{-}cG_6$ was produced only in a small quantity even on prolonged reaction, and thus the isoamylase was unsuitable for the production of $G_2\text{-}cG_6$ as reported³.

The relationships between the rates of hydrolysis and the lengths of the side chain of the substrates are shown in Table II. The K_m for $G_2\text{-}cG_6$ was about one-third but the V_{\max} only 0.4% of those for $G_3\text{-}cG_6$. The V_{\max} for $G_3\text{-}cG_6$ was considerably higher than that for $G_2\text{-}cG_6$, but the affinity for this substrate was lower

TABLE I

CONDENSATION ACTIVITIES OF *Pseudomonas* ISOAMYLASE BETWEEN cG_6 AND MALTO-OLIGOSACCHARIDES

Malto-oligosaccharide	G_2	G_3	G_4	G_5	G_6	G_7
1 ^a Activity ($\mu\text{mol}/\text{min}/\text{mg}$)	0.097	2.57	1.96	0.97	n.d. ^c	n.d.
(relative)	1.0	26.5	20.2	10.0		
2 ^b Activity ($\mu\text{mol}/\text{min}/\text{mg}$)	0.15	2.80	2.44	0.97	0.91	0.86
(relative)	1.0	18.7	16.2	6.5	6.0	5.7

^aEach reaction mixture (1 mL) contained 275mM cG_6 , 605mM malto-oligosaccharide, and 8.5 U of isoamylase in 50mM acetate buffer (pH 4.2) and was incubated at 58°. ^bEach reaction mixture contained 550 mg/mL of oligosaccharide and cG_6 in the molar ratio 2:1, and the other conditions were as for Experiment 1. ^cNot determined.

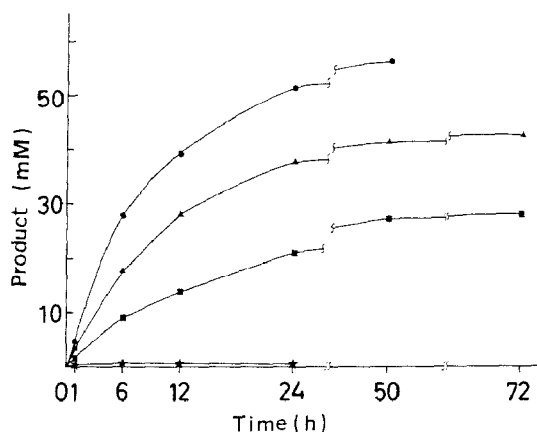


Fig. 1. Time courses of the formation of branched cG₆s. Each reaction mixture was at pH 4.2 and contained 275mM cG₆, 605mM G₂-G₅, and 8.2 U of isoamylase/mL, and was incubated at 58°: ★, G₂-cG₆; ●, G₃-cG₆; ▲, G₄-cG₆; ■, G₅-cG₆.

than that for G₂-cG₆ which was hydrolysed least. The values of K_m and V_{max} decreased with increasing chain length in the range of G₃-G₅. This finding implies that the enzyme has at least 5 binding sites for the side chains of the substrates, and the affinity is higher with the longer side-chains. It is reasonable, because of its marked debranching activity toward amylopectin and glycogen, that the enzyme shows similar activities toward the G₄ and G₅ side-chains. The activity appears to be modified by the structure of the main chain as reported³, and G₂-cG₇ and G₂-cG₈ may be better substrates than G₂-cG₆. Therefore, it should not be concluded that the G₂-stubs in the beta-limit dextrins of glycogen and amylopectin are not hydrolysed significantly by the enzyme. The relationships between the activities and the structures of the main chains remain to be determined.

Structures of the branched-cG₆. — The branched cG₆s, including the new compounds G₄-cG₆ and G₅-cG₆, were characterised by means of enzymic analyses, and ¹³C-n.m.r. and f.a.b.-mass spectra, G₃-, G₄-, and G₅-cG₆ in dilute solutions were hydrolysed into equimolar amounts of the respective malto-oligosaccharides

TABLE II

KINETIC PARAMETERS OF HYDROLYTIC ACTIVITIES FOR BRANCHED cG₆S

Substrate	K_m (mM)	V_{max} (U/mg)	V_{max}/K_m
G ₂ -cG ₆	72	2.6	0.04
G ₃ -cG ₆	204	690	3.4
G ₄ -cG ₆	92	320	3.5
G ₅ -cG ₆	47	290	6.2

and cG_6 by isoamylase. G_2 - G_5 - cG_6 gave 1-4 mol/mol glucose with glucoamylase (*Rhizopus delemar* GIII¹⁸) and an equimolar amount of G - cG_6 , respectively. G_4 - and G_5 - cG_6 were converted into maltose and G_2 - and G_3 - cG_6 , respectively, with sweet-potato beta-amylase. G_2 - and G_3 - cG_6 were resistant to this enzyme. These findings accord with the action of the enzyme on branched oligosaccharides²³.

The ^{13}C -n.m.r. spectra of G - to G_5 - cG_6 were assigned as shown in Table III by comparison with those of branched cG_7 ^{3,4} and cG_8 ¹⁴. The chemical shifts and intensities of the signals for C-1, C-4, and C-6 were consistent with the structures. The $J_{\text{C-1,H-1}}$ values for the side chains of G - to cG_5 - cG_6 involved in the linkages between the ring residues were ~ 170 Hz, indicating²⁴ the linkages between the malto-oligosaccharides and cG_6 to be α .

The f.a.b-mass spectra (negative mode) of G_4 - cG_6 and G_5 - cG_6 indicated the

TABLE III

CHEMICAL SHIFTS AND RELATIVE INTENSITIES OF THE SIGNALS FOR C-1, C-4, AND C-6 IN THE ^{13}C -N.M.R. SPECTRA OF SOLUTIONS OF THE BRANCHED cG_6 S IN D_2O

cG_6	C-1 ^a		C-4 ^a		C-6 ^a		
	<i>Rs and R</i>	<i>Sr</i>	<i>So and S</i>	<i>Rs and R</i>	<i>So</i>	<i>Sr and S</i>	<i>R, S, So, Rs and Sr</i>
G -	102.17	99.74		82.08	70.52		61.33
				82.30			61.48
				82.40			
	6:1 ^b			6:1			6:1
G_2 -	102.17	99.43	100.85	82.06	70.38	78.75	61.34
				83.18		78.76	61.36
				82.48			61.37
							61.50
	6:1:1			6:1:1			7:1
G_3 -	102.17	99.39	100.77	82.07	70.37	78.23	61.38
			100.84	82.15		79.24	61.45
				82.48			61.53
	6:1:2			6:1:2			8:1
G_4 -	102.19	99.42	100.58	82.13	70.38	78.23	61.46
			100.82	82.53		79.29	61.56
	6:1:3			6:1:3			9:1
G_5 -	102.19	99.43	100.53	82.12	70.38	78.07	61.47
			100.63	82.54		78.15	
			100.71			78.39	
			100.83			79.18	
	6:1:4			6:1:4			10:1

^aKey: Rs, ring Glc linked to a side chain; R, ring Glc other than Rs; Sr, side-chain Glc linked to a ring Glc; So, side-chain Glc with HO-4 unsubstituted (identical to Sr in G - cG_6); S, side-chain Glc other than Sr and So. ^bRelative intensity.

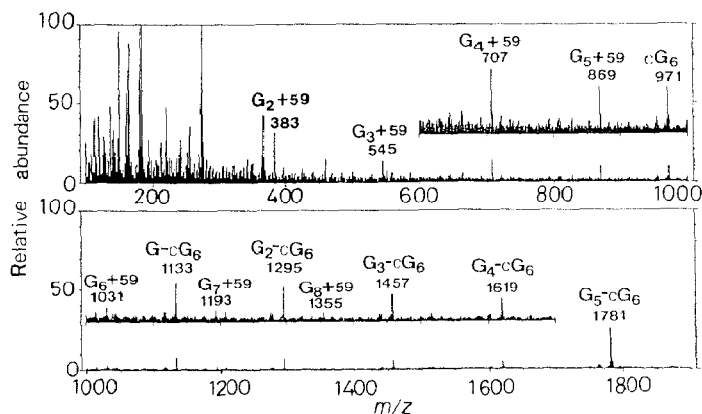


Fig. 2. F.a.b.-mass spectrum of $G_5\text{-cG}_6$ (mol. wt. 1782) in the negative mode.

molecular weights. The f.a.b.-mass spectrum of $G_5\text{-cG}_6$ (Fig. 2) shows a series of fragment ions corresponding to the shorter side-chains of similar abundance, suggesting a single and linear side-chain^{4,14}. Another series of fragment ions, with masses of $n\text{-glucosyl} + 59$, is due to the cleavage of the cG_6 ring and a glucosyl residue^{3,4}. A similar f.a.b.-mass spectrum was obtained for $G_4\text{-cG}_6$.

Properties. — The solubilities of cG_n in water were enhanced greatly by the attachment of glucose or malto-oligosaccharides¹⁰. cG_6 is moderately soluble in water at 25° , but its solubility is enhanced¹⁰ up to ~ 5 -fold on attachment of G_3 . Fig. 3 shows the solubilities of the series of branched cG_6 s at 25° , 45° , and 55° . The

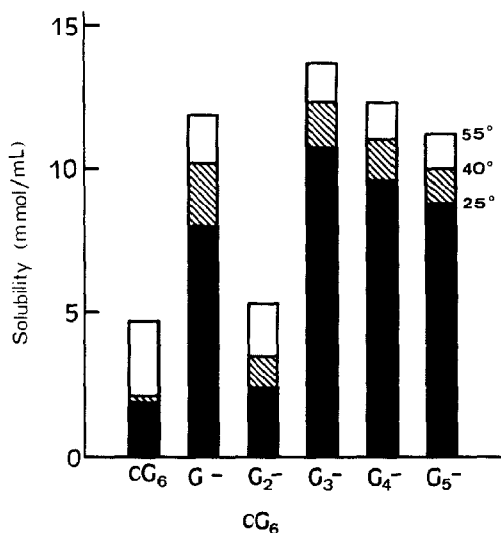


Fig. 3. Solubilities of cG_6 s in water at 25° , 40° , and 55° .

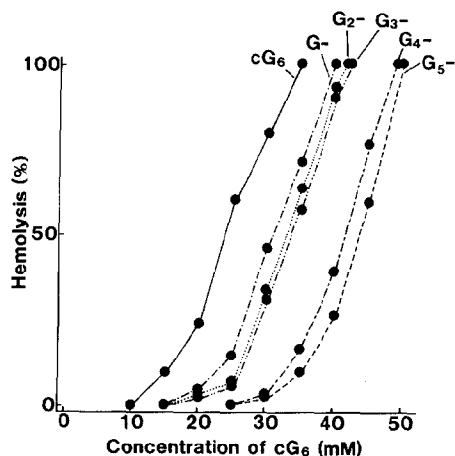


Fig. 4. Haemolytic activities of cG_6 s.

solubility of G_2 - cG_6 , which was the least soluble among the branched cG_6 s, was enhanced ~ 2.2 -fold by elevation of the temperature from 25° to 55° , and those of other, highly soluble, branched CDs by 1.3–1.5-fold. The low solubility of G_2 - cG_6 suggests that it may have a rigid, less hydrophilic structure, due to the formation of some intermolecular and/or intramolecular hydrogen bonds. The $[\alpha]_D^{22}$ values (water) were as follows: cG_6 , $+150^\circ$; G - cG_6 , $+157^\circ$; G_2 - cG_6 , $+165^\circ$; G_3 - cG_6 , $+167^\circ$; G_4 - cG_6 , $+170^\circ$; and G_5 - cG_6 , $+172^\circ$. The haemolytic activity toward human red cells of cG_6 , which is intermediate among cG_6 , cG_7 , and cG_8 , was considerably reduced on attachment of the long side-chains (Fig. 4).

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