

## A Highly Regio- and Stereoselective C<sub>5</sub> Oxyfunctionalization of Coprostan Steroids by Dioxiranes: an Improved Access to Progesterone and Androgen Hormones.

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**Abstract:** Coprostan steroids are selectively oxyfunctionalized at C<sub>5</sub> by dimethyldioxirane and methyltrifluoromethyldioxirane to give useful intermediates for bioactive compounds.

Dioxiranes are known to be highly effective oxyfunctionalizing reagents<sup>1</sup>. At present, we are testing the reactivity of these oxidants towards steroidal molecules for the purpose of obtaining useful biological targets.

In this context, we recently reported<sup>2</sup> that estrone was selectively functionalized at C<sub>9</sub> benzylic carbon by dimethyldioxirane **1** to afford 9,α-hydroxy-estrone. Cholestan steroids were also selectively functionalized at C<sub>25</sub> isopropyl carbon to give intermediates of vitamin D metabolites<sup>3</sup>.

We decided to ascertain whether dioxiranes were equally effective in the functionalization of other steroid positions. Consequently, we investigated **1a** and **1b** reactivity towards coprostan steroids, which present the A/B-ring-cis junction and therefore the β-configuration at C<sub>5</sub> hydrogen.

When bile steroid lithocholic acid-3-acetate methyl ester **2** reacted with **1a** (D/S=2:1, solvent CHCl<sub>3</sub>/acetone, r.t., 24h), this produced a highly regio- and stereoselective oxyfunctionalization at C<sub>5</sub> carbon, whose only product was 5, β-hydroxy derivative **3** (30% conv., 85% isolated yield). When **1b** was employed, as selectivity was unchanged, we noted a considerable rise in conversion (70% conv., 85% isolated yield).

When **1a** (or **1b**) reacted with 5,β-cholestanic acid-3,α-12,α-diol-3,12-diacetate methyl ester **5**, we obtained the same regioselectivity, and the only product formed was 5,β-hydroxy derivative **6**<sup>5</sup>.

Since C<sub>5</sub> oxyfunctionalization of steroids represents a key step in the conversion of bile acids into androgenic and progesterone hormones<sup>6</sup>, these results therefore appear to be very useful.

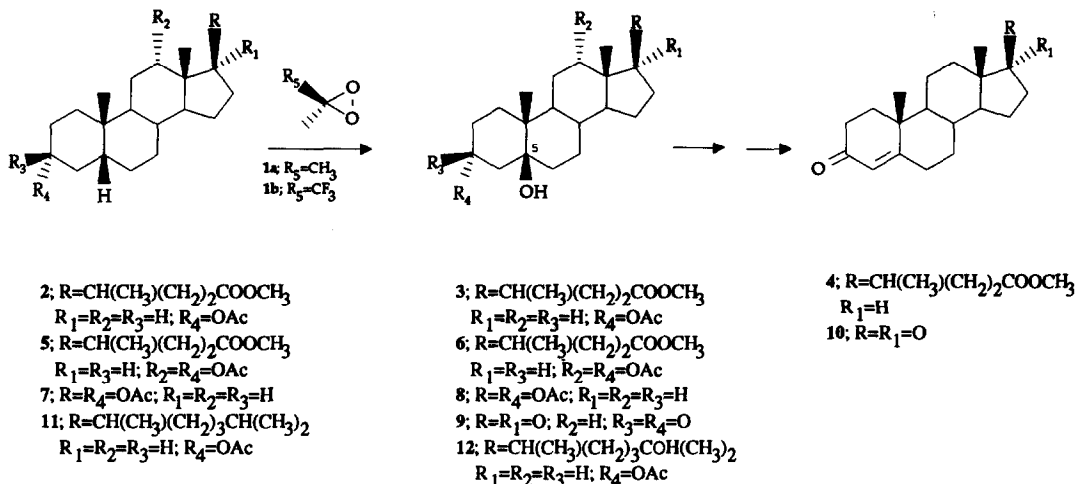
The recovery of unreacted starting material was performed by chromatographic separation, thus enabling us to increase the overall yield.

After this, we deacetylated **3** (K<sub>2</sub>CO<sub>3</sub> s.s./CH<sub>3</sub>OH 1:9, r.t., 3h), oxidized it at C<sub>3</sub> (PCC in CH<sub>2</sub>Cl<sub>2</sub>, r.t., 2h) and finally dehydrated it (H<sub>2</sub>SO<sub>4</sub> conc., 0-25°C, 1h) to give the enone **4**, an intermediate in progesterone synthesis<sup>6c</sup>. We would also like to point out that compounds such as **6**, which present both C<sub>5</sub> and C<sub>12</sub> oxyfunctionalized carbons, represent valuable intermediates in cortisone synthesis starting from cholic acid<sup>6d</sup>.

As a further application we oxyfunctionalized 5β-androstan-3α-17β-diacetoxy **7** by **1a** at C<sub>5</sub> (50% conv., 90% isolated yield). This gave **8**, which we then easily converted into the hormone androstenedione **10**.

We also submitted 5,β-cholestan-3,α-ol acetate **11** to **1a**, and achieved simultaneous double oxyfunctionalization at C<sub>5</sub> and C<sub>25</sub>, to yield 5,β-cholestan-3,α-5,α-25-triol-3-acetate **12**.

## SCHEME



The  $C_3$  oxyfunctionalization of coprostane steroids by dioxiranes is clearly due to the favoured steric environment at this position; the attack on  $C_{25}$  isopropyl carbon once again demonstrates the electrophilicity of dioxiranes.

We may conclude by suggesting that the stereoselective introduction of the  $\beta$ -hydroxyl moiety on  $C_3$  steroidal carbon may also allow a new entry to cardiac-active and ecdysonic compounds<sup>10</sup>.

## References and Notes

- For reviews, see: a) R. Curci in *Advances in Oxygenated Processes*; Baumstark A.L., Ed.; JAI: Greenwich, CT; Vol. 2, Ch. 1 (1990). b) W. Adam, R. Curci, J.O. Edwards, *Acc. Chem. Res.*, **22**, 205 (1989).
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- For a representative oxidation of steroids with dioxiranes see ref. (2).
- 3: m.p. = 167–168°C from Hexane/Et<sub>2</sub>O; <sup>1</sup>H-NMR  $\delta$ : 0.61 (3H, s,  $C_{18}$ -H), 0.87 (3H, s,  $C_{19}$ -H), 0.89 (3H, d,  $J = 6\text{Hz}$ ,  $C_{21}$ -H), 2.00 (3H, s,  $\text{CH}_3\text{COO}$ ), 3.64 (3H, s,  $\text{OCH}_3$ ), 5.05 (1H, t.t.,  $J_1 = 11.1$ ,  $J_2 = 4.7$ ,  $C_3$ -H). The observed  $C_{19}$  methyl chemical shift is consistent with a  $\beta$ -configuration of  $C_3$ -OH<sup>6</sup>. <sup>13</sup>C-NMR  $\delta$ : 71.41 ( $C_3$ ), 75.34 ( $C_2$ ), 170.80 ( $\text{CH}_3\text{COO}$ ), 175.02 ( $C_{19}$ ). 4: m.p. = 123–126°C from Hexane/Et<sub>2</sub>O; <sup>1</sup>H-NMR spectrum agrees with literature data<sup>7</sup>. 6: m.p. = 127–128°C from Hexane/Et<sub>2</sub>O; <sup>1</sup>H-NMR  $\delta$ : 0.69 (3H, s,  $C_{18}$ -H), 0.77 (3H, d,  $J = 6\text{Hz}$ ,  $C_{21}$ -H), 0.84 (3H, s,  $C_{19}$ -H), 1.99 (3H, s,  $\text{CH}_3\text{COO}-C_3$ ), 2.07 (3H, s,  $\text{CH}_3\text{COO}-C_{12}$ ), 3.63 (3H, s,  $\text{OCH}_3$ ), 5.03 (1H, t.t.,  $J_1 = 11.1\text{Hz}$ ,  $J_2 = 4.7\text{Hz}$ ,  $C_3$ -H), 5.07 (1H, m,  $C_{12}$ -H). <sup>13</sup>C-NMR  $\delta$ : 71.17 ( $C_3$ ), 75.14 ( $C_2$ ), 75.56 ( $\text{COOC}_3$ ), 170.63, 170.70 ( $\text{COO}-C_3 + \text{COO}-C_{12}$ ), 174.84 ( $C_{12}$ ). 8: m.p. = 220°–222°C from Hexane/Et<sub>2</sub>O (lit.<sup>8</sup> m.p. = 221–224°C); <sup>1</sup>H-NMR spectrum agrees with literature data<sup>6</sup>. 9: m.p. = 180–183°C from Hexane/Et<sub>2</sub>O (lit.<sup>9</sup> m.p. = 183–186°C). 10: m.p. = 171–173°C (lit.<sup>9</sup> 173–174°C). 12: <sup>1</sup>H-NMR  $\delta$ : 0.61 (3H, s,  $C_{18}$ -H), 0.87 (3H, s,  $C_{19}$ -H), 0.88 (3H, d,  $J = 6.5\text{Hz}$ ,  $C_{21}$ -H), 1.19 (6H, s,  $C_{26}$ -H +  $C_{27}$ -H), 2.00 (3H, s,  $\text{CH}_3\text{COO}$ ), 5.05 (1H, m,  $C_3$ -H). <sup>13</sup>C-NMR  $\delta$ : 71.09 ( $C_{23}$ ), 71.41 ( $C_3$ ), 75.40 ( $C_2$ ), 170.78 ( $\text{CH}_3\text{COO}$ ).
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- a) L.F. Fieser and M. Fieser, *Steroids*, Reinhold Publishing Corporation N.Y. p. 519 (1959); b) *ibid.* p. 545 and following; c) *ibid.* p. 546; d) *ibid.* p. 651 and following.
- It is known that oxyfunctionalization of saturated carbon atoms by dioxiranes proceeds with retention of configuration<sup>1</sup>.

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