

Spectral Assignments and Reference Data

¹H and ¹³C NMR spectral characterization of some novel 7H-1,2,4-triazolo[3,4-b][1,3,4]thiadiazine derivatives

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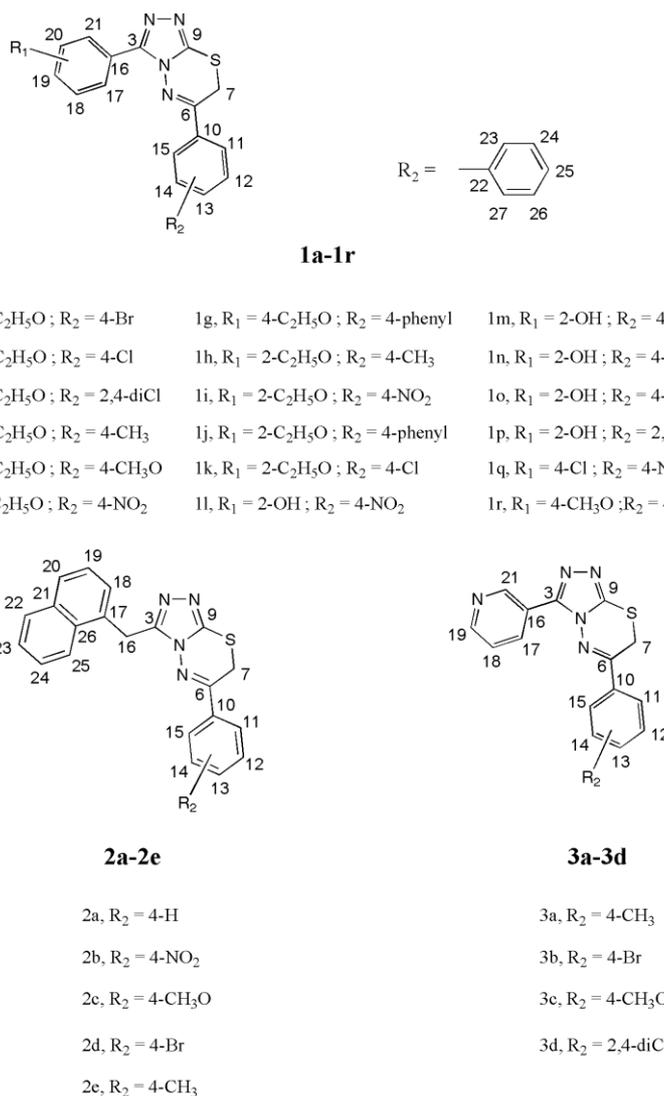
Some novel 1, 2, 4-triazolo[3,4-*b*][1,3,4]thiadiazines derivatives were synthesized. The complete ¹H and ¹³C NMR chemical shift assignments were analyzed on one- and two-dimensional NMR techniques, including DEPT, NOE-DIF, COSY, HMBC, and HSQC. Copyright © 2006 John Wiley & Sons, Ltd.

KEYWORDS: NMR; ¹H NMR; ¹³C NMR; 1,2,4-triazolo[3,4-*b*][1,3,4]thiadiazines; synthesis

INTRODUCTION

1,2,4-Triazoles and *N*-bridged heterocycles are found to be associated with diverse pharmacological properties.^{1–3} Broad biological and pharmacological properties of various thiadiazines fused with a triazole ring have been extensively studied.^{4–9} In particular, various substituted 7H-1,2,4-triazolo-[3,4-*b*][1,3,4] thiadiazines have been shown to possess diverse pharmacological properties, such as antimicrobial, bactericidal, anti-inflammatory, antiviral, antihypertensive, antifungal, anthelmintic, and analgesic effects.^{10–13} Prompted by the biological properties of 1,2,4-triazole derivatives and 1,3,4-thiadiazines, and in continuation of our studies on *N*-bridged heterocycles,^{14–15} we report here the synthesis and spectral characterization of some 7H-1,2,4-triazolo[3,4-*b*][1,3,4] thiadiazine derivatives (Scheme 1), of which compounds **1a**, **1b**, **1d**, **1e**, **1f**, **1o**, and **1q** were known.^{12,13}

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Scheme 1. Structure of various 3,6-disubstituted-7H-1,2,4-triazolo[3,4-*b*][1,3,4]thiadiazines.

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EXPERIMENTAL

Materials

Various 5-substituted-4-amino-2,4-dihydro-3*H*-1,2,4-triazole-3-thiones¹³ were prepared by methods given in the literature.^{16–18} All other chemicals and solvents used were of analytical reagent grade.

Preparation of various 3,6-disubstituted-7*H*-1,2,4-triazolo[3,4-*b*][1,3,4]thiadiazines

The mixture of 5-substituted-4-amino-2,4-dihydro-3*H*-1,2,4-triazole-3-thione (40 mmol), *o*-bromoacetophenones (42 mmol), ethanol (70 ml), and water (30 ml) was stirred and refluxed for 3 h. The solution was cooled and then filtered. The precipitate was washed with ether and dried. The crude material was recrystallized from 70% ethanol solution to obtain the pure products (**1a–1r**, **2a–2e**, and **3a–3d**) at a yield of 63–86%.

NMR spectra

All NMR experiments were performed at 293 K using a solution of 20 mg of the compound dissolved in 0.5 ml of dimethyl sulfoxide (DMSO)-*d*₆ on a Bruker AVANCE-300 instrument with a 5 mm BBFO probe head equipped with shielded Z-gradient coil. The gradient field was 50×10^{-4} T cm⁻¹. ¹H NMR spectra were recorded at a proton frequency of 300.13 MHz with a spectral width of 4.5 kHz and 11 μs (90°) pulse. The ¹³C NMR spectra were obtained using a spectral width of 20 kHz, a 3.9 μs (30°) pulse, and a 1.8 s acquisition time; 1024 scans with 32 768 data points each were used. Exponential multiplication was applied before Fourier transformation in both cases. The chemical shifts were referenced to tetramethylsilane

(TMS). The long-range ¹H–¹³C correlation (HMBC) spectra were obtained using the hmbcgpndqf program in the Bruker software. The spectra resulted from a 256 × 1024 data matrix size with 16 scans per *t*₁ increment, 200 μs delay for homospoil/gradient recovery (D₁₆), 1 ms for homospoil/gradient pulse (P₁₆), and 5:3:4 gradient combination using a pulse sequence optimized for 8 Hz. Spectral widths of 3.5 kHz in *f*₂ and 15.5 kHz in *f*₁ were used. The acquisition time was 0.57 s, the delay was set to 3.45 ms, the recycle time was 1.50 s, and the Fourier transformation was done on a 2k × 1k data matrix. The one-bond heteronuclear correlation (HSQC) spectra were obtained using the hsqcetgp program in the Bruker software. The spectra were measured with the same D₁₆ (200 μs) and P₁₆ (1 ms) and a different 4:1 gradient combination. The acquisition time was 0.14 s, the delay was set to 3.45 ms, and an average ¹J(C, H) of 145 Hz was used; the recycle time was 1.55 s. The Fourier transformation was done on a 2k × 1k data matrix.

RESULTS AND DISCUSSION

In order to unequivocally assign all NMR signals, we used 1D and 2D techniques such as DQF-COSY, HSQC, and HMBC.

In Scheme 1, the structures and numbering of the title compounds are presented. Their ¹H and ¹³C NMR chemical shifts are given in Tables 1 and 2, respectively. For example, assigning the spectra of **1k** is described as follows. H-7 (δ 4.41), H-11, H-15 (δ 7.90), H-12, H-14 (δ 7.61), methylene proton (δ 3.98), and methyl (δ 1.05) proton were assigned directly, and the ¹H signals of H-18~H-21 were determined by the combination of COSY, HSQC, and HMBC experiments. Seven quaternary carbons were observed by DEPT technique and assigned by HMBC experiments, the assignments were C-3 at

Table 1. ¹H NMR data for compounds **1a–2r**, **2a–2e**, and **3a–3d** (Solvent: DMSO-*d*₆)

Compound	7	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	26	27	R ₁	R ₂
1a	4.42	7.95	7.80		7.80	7.95		7.92	7.12		7.12	7.92							4.12, 1.36	
1b	4.42	8.01	7.66		7.66	8.01		7.93	7.12		7.12	7.93							4.12, 1.36	
1c	4.30		7.89		7.64	7.72		7.92	7.12		7.12	7.92							4.12, 1.35	
1d	4.40	7.96	7.39		7.39	7.96		7.91	7.12		7.12	7.91							4.12, 1.36	2.40
1e	4.38	7.95	7.30		7.30	7.95		7.92	7.12		7.12	7.92							4.12, 1.36	3.67
1f	4.50	8.25	8.40		8.40	8.25		7.94	7.13		7.13	7.94							4.13, 1.37	
1g	4.47	8.10	7.90		7.90	8.10		7.97	7.13		7.13	7.97	7.78	7.53	7.43	7.53	7.78		4.13, 1.37	
1h	4.37	7.77	7.34		7.34	7.77			7.18	7.08	7.54	7.52							3.98, 1.03	2.36
1i	4.50	8.11	8.37		8.37	8.11		7.19	7.11	7.56	7.53								4.01, 1.05	
1j	4.45	7.98	7.85		7.85	7.98		7.20	7.12	7.51	7.48		7.75	7.55	7.43	7.55	7.75		4.02, 1.08	
1k	4.41	7.90	7.61		7.61	7.90		7.18	7.10	7.55	7.52								3.98, 1.05	
1l	4.52	8.21	8.37		8.37	8.21		6.98	7.42	7.02	7.83								10.30	
1m	4.44	7.95	7.38		7.38	7.95		6.98	7.39	7.05	7.99								10.20	2.40
1n	4.46	8.07	7.58	7.58	7.58	8.07		6.98	7.41	7.04	7.91								10.20	
1o	4.44	7.99	7.80		7.80	7.99		6.96	7.42	7.05	7.92								10.20	
1p	4.32		7.90		7.65	7.73		6.95	7.41	7.06	7.88								10.30	
1q	4.52	8.25	8.39		8.39	8.25		8.01	7.67		7.67	8.01								
1r	4.47	8.10	7.89		7.89	8.10		8.00	7.17		7.14	8.00		7.77	7.51	7.40	7.51	7.77	3.85	
2a	4.39	7.97	7.47	7.52	7.47	7.97	4.74		7.53	7.50	7.85		7.95	7.57	7.60	8.29				
2b	4.46	8.21	8.38		8.38	8.21	4.76		7.50	7.47	7.85		7.95	7.57	7.59	8.29				
2c	4.41	7.97	7.14		7.14	7.97	4.75		7.47	7.44	7.79		7.96	7.54	7.56	8.27				3.85
2d	4.38	7.91	7.77		7.77	7.91	4.74		7.48	7.45	7.84		7.94	7.55	7.58	8.28				
2e	4.37	7.88	7.36		7.36	7.88	4.75		7.48	7.44	7.85		7.94	7.55	7.59	8.28				2.38
3a	4.46	7.94	7.40		7.40	7.94		8.73	7.94	8.90		9.32								2.40
3b	4.89	7.98	7.80		7.80	7.98		8.79	7.98	8.96		9.35								
3c	4.45	8.02	7.13		7.13	8.02		8.74	7.95	8.90		9.33								3.86
3d	4.48		7.91		7.67	7.74		8.74	7.94	8.91		9.34								

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Table 2. ¹³C NMR data for compounds **1a–2r**, **2a–2e**, and **3a–3d** (Solvent: DMSO-*d*₆)

Com- pound	3	6	7	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	26	27	R ₁	R ₂	
1a	151.65	155.03	22.73	141.95	132.92	132.23	129.57	125.79	129.57	132.23	118.26	129.60	114.79	160.17	114.79	129.60								63.45, 14.70	
1b	151.75	154.99	22.84	142.05	130.60	129.40	129.51	136.94	129.51	129.40	118.30	129.70	114.86	160.25	114.86	129.70								63.53, 14.75	
1c	151.60	156.10	22.80	141.98	132.81	129.92	127.85	136.48	132.47	133.43	118.36	129.75	114.78	160.11	114.78	129.75								63.46, 14.72	
1d	151.54	155.80	22.76	142.06	130.87	127.60	129.54	142.27	129.54	127.60	118.43	129.82	117.77	160.13	114.77	129.82								63.45, 14.71	21.16
1e	151.44	155.44	22.66	142.06	125.76	129.53	114.72	162.40	114.72	129.53	118.50	129.78	114.77	160.10	114.77	129.78								63.46, 14.72	55.71
1f	151.83	154.23	22.97	141.89	139.60	129.03	124.41	149.24	124.41	129.03	118.06	129.69	114.83	160.25	114.83	129.69								63.47, 14.68	
1g	151.62	155.52	22.80	142.07	132.55	128.30	127.36	143.43	127.36	128.30	118.39	129.59	114.79	160.15	114.79	129.59	138.93	126.99	129.23	128.44	129.23	126.99		63.45, 14.70	
1h	151.35	154.54	23.18	141.93	130.71	127.34	129.76	142.12	129.76	127.34	115.51	156.87	112.67	120.45	132.32	131.23								63.69, 14.43	21.11
1i	151.60	152.99	23.33	141.85	139.52	128.78	124.29	149.24	124.29	128.78	115.54	156.85	112.78	120.53	132.55	131.31								63.73, 14.49	
1j	151.34	154.46	23.25	142.07	139.39	128.08	127.34	143.36	127.34	128.08	115.29	156.94	112.76	120.52	132.50	131.33	138.87	126.97	129.22	128.45	129.22	126.97		63.80, 14.52	
1k	151.32	153.78	23.16	141.95	132.23	129.17	129.32	136.82	129.32	129.17	115.13	156.85	112.74	120.49	132.50	131.29								63.73, 14.40	
1l	151.30	154.15	23.13	142.22	139.50	129.10	124.28	149.34	124.28	129.10	111.66	156.77	116.78	132.18	119.35	129.50									21.20
1m	150.73	156.22	22.86	142.71	130.66	127.74	129.89	142.71	129.89	127.74	111.31	156.87	116.80	132.18	119.30	129.20									
1n	151.03	156.23	23.08	142.50	132.08	127.76	129.32	132.20	129.32	127.76	111.64	156.84	116.80	132.20	119.32	129.24									
1o	151.08	156.22	22.84	142.32	132.80	129.67	130.60	125.99	130.60	129.67	111.70	156.79	116.78	132.05	119.31	129.29									
1p	151.23	156.23	22.89	142.34	132.75	129.98	127.82	136.51	132.48	133.51	111.72	156.71	116.77	132.14	119.23	129.25									
1q	151.12	154.70	23.08	142.86	139.44	129.16	124.26	149.35	124.26	129.16	124.70	129.86	129.20	135.35	129.20	129.86									
1r	151.60	155.49	22.78	142.09	132.53	128.28	127.35	143.43	127.35	128.28	118.56	129.59	114.38	160.87	114.38	129.59	138.92	126.98	129.21	128.43	129.21	126.98		55.48	
2a	152.49	154.89	22.93	140.59	132.05	127.60	129.17	131.98	129.17	127.60	27.90	133.58	127.64	125.66	127.72	131.65	128.17	125.96	126.38	124.23	131.65				
2b	152.72	153.15	23.00	140.48	139.51	128.98	124.16	149.28	124.16	128.98	27.93	133.59	127.77	125.70	127.79	131.92	128.65	125.98	126.42	124.22	131.65				
2c	152.60	155.21	22.70	140.56	125.56	129.65	114.78	162.55	114.78	129.65	27.92	133.53	127.70	125.68	127.75	131.80	128.60	125.96	126.41	124.21	131.65				55.73
2d	152.52	153.98	22.75	140.52	132.79	132.20	129.55	125.88	129.55	132.20	27.90	133.58	127.69	125.67	127.74	132.03	128.64	125.97	126.40	124.22	131.65				
2e	152.42	155.09	22.82	140.91	130.65	127.62	129.76	142.43	129.76	127.62	27.87	133.58	127.72	125.66	127.77	131.90	128.64	125.97	126.40	124.20	131.65				21.18
3a	148.62	156.81	23.06	143.75	130.51	127.86	129.92	142.77	129.92	127.86	123.76	138.99	125.66	147.77	145.25										
3b	148.41	156.15	23.15	143.81	132.53	129.89	132.35	126.30	132.35	129.89	124.02	140.17	126.18	146.85	144.34										
3c	148.47	156.41	22.92	143.72	125.34	129.83	114.78	162.70	114.78	129.83	123.85	139.00	125.68	147.65	145.17										55.79
3d	148.43	156.08	23.18	143.70	132.38	129.58	127.92	135.94	132.60	133.32	123.85	139.32	125.84	146.69	145.12										

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δ 151.32, C-6 at δ 153.78, C-9 at δ 141.95, C-10 at δ 132.23, C-13 at δ 136.82, C-16 at δ 115.13, and C-17 at δ 156.85. All the other ^1H and ^{13}C NMR chemical shifts were determined completely by the same methods. Distinguishing H-18 from H-19 and H-22 from H-25 for compounds **2a–2e** was not easy and NOE difference spectra were applied. Because of the similarity in structures, we take **2e** as an example. The signals at δ 7.48 and δ 8.28 were increased by 5.5% and 9.6%, respectively, by radiating at δ 4.75 (H-16, methylene proton), and thus the signals of H-18 (δ 7.48) and H-25 (δ 8.28) were assigned.

From Table 1 it may be concluded that substitution on the aromatic ring moiety has little effect on the heterocyclic system H-7 proton. Thus, the H-7 proton resonates in the narrow range of 4.30–4.52 ppm, with the exception of compound **3b**. Generally, the electron-accepting group on the aromatic ring shifts the signal downfield, whereas the electron-donating substituents shift the signal upfield.

The ^{13}C NMR spectra exhibit characteristic signals for C-3, C-6, C-7, and C-9 of the fused ring at δ 148.41–152.72, 153.15–156.81, 22.66–23.33, and 140.48–143.81, respectively. As shown in Table 2, the signals corresponding to the heterocyclic system are relatively insensitive to the nature of the substituent on the aryl ring. C-3 in these systems resonates in the narrow range of 151–152 ppm, with the exception of compounds **3a–3d**.

Acknowledgements

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