Note

Synthesis of 3,4-dideoxy-DL-hex-3-enopyranosides from 5-hydroxymethyl-2furaldehyde*

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Some years ago, we elaborated a method for the short conversion of 2-furaldehyde and its derivatives into various racemic monosaccharides^{1,2}, using the formyl group of furaldehyde as the reducing end of the sugar to be synthesised. Recently, a simple preparation of 5-hydroxymethyl-2-furaldehyde (1) has been described³ which made the compound available for the synthesis of the title compounds.

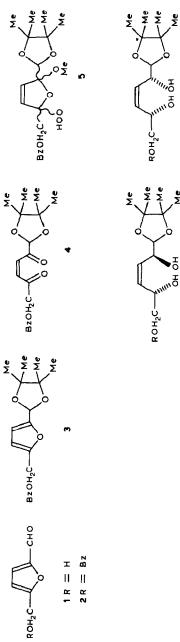
Isomeric methyl 3,4-dideoxy-DL-hex-3-enopyranosides have been synthesised⁴ in five steps from the butyl ester of *cis,trans*-5,6-dihydro-2-methoxy-2*H*pyrancarboxylic acid.

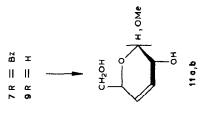
The hydroxyl group of 1 was protected by benzoylation (\rightarrow 2) and the formyl group by transformation into a tetramethyldioxolane derivative^{1,2} (3), which was stable under the acidic conditions necessary for the oxidative ring opening of the furan nucleus with bromine water at pH 3-4 to give 4. Reduction of 4 with sodium borohydride gave a 6:5 mixture of 6 and 7, which was fractionated by chromatography. Alternatively, the furan ring in 3 was oxidised with singlet oxygen and the resulting hydroperoxide 5 was reduced with methyl sulfide and sodium borohydride successively to give 6 and 7.

Treatment of 6 and 7 with methanolic sodium methoxide gave 8 and 9, respectively, which were methanolysed to give methyl 3,4-dideoxy- α -DL-threo-hex-3-enopyranoside (10) and methyl 3,4-dideoxy- α , β -DL-erythro-hex-3-enopyranosides (11ab), respectively. The structures of these products were elucidated by ¹H- and ¹³C-n.m.r. spectroscopy (see Table I).

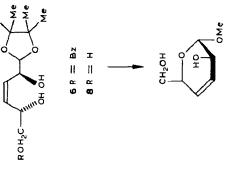
The chemical shifts and $J_{1,2}$ values for 10, 11a, and 11b accorded with pub-

^{*}Dedicated to the memory of Karl Freudenberg on the centenary of his birth.





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NOTE

lished data⁴. The preponderant conformer for each compound must be ${}^{\circ}H_1$, as demonstrated by the $J_{4,5}$ and $J_{2,5}$ values of 1.5 and 2.0 Hz. Calculated⁵ $J_{1,2}$ constants are also in agreement those observed (see Table I). The $J_{C\cdot1,H\cdot1}$ value is 168 Hz for the α anomers and 163 Hz for the β -erythro derivative. The ¹³C signals for C-1,5 are shifted downfield for the β anomer **11b** as compared to those of the α anomers **10** and **11a**.

TABLE I

Atom	a-threo (10)		a-erythro (11a)		β-erythro (11b)	
	δ (p.p.m.)	J (Hz)	δ	J (Hz)	δ	J (<i>Hz</i>)
H- 1	4.79	$J_{1,2} 1.5 (0.4)^a$	4.86	$J_{1,2}$ 4.0 (4.3) ^a	4.43	J _{1,2} 6.0 (8.5)
H-2	3.78	J _{1,3} 1.2	4.20	J _{3.4} 11.0	4.09	$J_{2,3}^{-1.7}$
H-3	6.04 ^b	$J_{2,3}^{1}5.0$	5.62	J _{5.64} 3.5	5.74	$J_{2.4}^{-1}$ 2.2
H-4	5.87 ^b	$J_{3.4}^{10.5}$	5.75 ^b	$J_{5,6b}^{,6b}$ 6.0	5.87	$J_{3,4}^{-1}$ 10.5
H-5	4.22	$J_{3,5}^{(1)} 2.0$	4.20	$J_{6a,6b}$ 12.0	4.40	$J_{3.5} 1.7$
H-6a	3.85	J _{4,5} 1.7	3.68	oute-	3.77	$J_{4.5}^{2.2}$
H-6b	3.66	$J_{2.5}^{2.0}$ 2.0	3.57		3.65	$J_{5,6a}^{(0)}$ 3.5
OH	1.72; 2.7	$J_{5.6a}^{-7}$ 2.7	1.67; 2.55		1.71; 2.2	$J_{5,6b}$ 6.0
ОМе	3.47	$J_{5,6b} 3.5 J_{6a,6b} 11.7$	3.48		3.58	J _{6a,6b} 11.5
C-1	101.6		98.0		103.5	
C-2	63.2		64.2		66.4	
C-3	125.8		126.4 ^b		127.5 ^b	
C-4	129.5		128.4 ^b		128.5 ^b	
C-5	68.6		69.0		75.3	
C-6	63.6		64.6		64.5	
ОМе	55.7		55.9		56.4	
¹ J _{C,H}		168 Hz		169 Hz		163 Hz

N.M.R. DATA FOR THE METHYL 3,4-DIDEON	XY-DL-HEX-3-ENOPYRANOSIDES
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^aCalculated value for the ${}^{\circ}H_1$ conformation. ^bInterchangeable assignments.

EXPERIMENTAL

General. — Column chromatography was performed on Kieselgel 60. ¹Hand ¹³C-n.m.r. spectra were recorded with a Bruker WP-200-SY instrument.

5-Benzoyloxymethyl-2-furaldehyde (2). — A stirred solution of 1 (6.45 g, 51 mmol) in pyridine (25 mL) was treated with benzoyl chloride (5.9 mL, 51 mmol) at 0° and then stored at room temperature for 16 h. The solvent was evaporated under diminished pressure and a solution of the residue in chloroform was washed successively with cold dilute sulphuric acid and aqueous NaHCO₃, then dried MgSO₄, and concentrated. The residue crystallised from ether–light petroleum, to give 2 (2.7 g, 84.2%), m.p. 51–52°.

Anal. Calc. for C₁₃H₁₀O₄: C, 67.83; H, 4.38. Found: C, 67.48; H, 4.70.

2-(5-Benzoyloxymethyl-2-furyl)-4, 4, 5, 5-tetramethyl-1, 3-dioxolane (3). — A mixture of 2 (1.05 g, 46 mmol), 2, 3-dimethylbutane-2, 3-diol (0.7 g, 59 mmol), toluene-p-sulphonic acid (50 mg), and benzene (100 mL) was boiled under reflux with azeotropic distillation of the water formed. After the end of the reaction (t.l.c.), the mixture was washed with aqueous NaHCO₃, dried (MgSO₄), clarified with charcoal and fuller's earth, and concentrated, to give 3 (1.1 g, 73.3%), m.p. 66.5–67.6° (from light petroleum). ¹H-N.m.r. data (CDCl₃): δ 1.28 (s, 6 H, 2 Me), 1.34 (s, 6 H, 2 Me), 5.29 (s, 2 H, CH₂), 5.90 (s, 1 H, dioxolane CH), 6.30 (s, 2 H, furan ring), 7.38–8.10 (m, 5 H, Ph).

Anal. Calc. for C₁₉H₂₂O₅: C, 69.07; H, 6.71. Found: C, 69.17; H, 6.86.

2-(DL-threo-5-Benzoyloxy-1,4-dihydroxy-cis-pent-2-enyl)-4,4,5,5-tetramethyl-1,3-dioxolane (6) and its erythro isomer (7). -(a) A solution of 3 (3.3 g, 10 mmol) in tert-butyl alcohol (130 mL) was diluted with a phosphate buffer (80 mL, pH 8; м $H_3PO_4 + M NaOH$). A solution of bromine (0.54 mL, 1.58 g, 10 mmol) in tert-butyl alcohol (20 mL) was added dropwise while the pH was maintained at 4 by continuous addition of 2M NaOH. The mixture was then neutralised and saturated with NaCl, the organic layer was separated, and the aqueous layer was extracted with tert-butyl alcohol. Water (50 mL) was added to the combined organic solutions, and ethanolic sodium borohydride (0.5 g, 13 mmol in 10 mL) was added dropwise with stirring. After 3 h, the mixture was acidified (pH 5) with dilute HCl, neutralised, and extracted with chloroform, and the extract was then dried and concentrated to give a mixture of 6 and 7. Column chromatography (light petroleumacetone, 9:1 then 8:2) gave, first, **6**, isolated as a syrup (0.90 g). ¹H-N.m.r. data $(CDCl_3)$: $\delta 1.25$ (s, 12 H, 4 Me), 2.62 (bs, 1 H, OH), 3.45 (bs, 1 H, OH), 4.39 (m, 2 H, H-5'a,5'b), 4.44 (m, 1 H, $J_{1',2'}$ 8.0, $J_{1',3'}$ 1.2 Hz, H-1'), 4.73 (m, 1 H, $J_{4',5',4}$ 6.0, $J_{4',5'b}$ 5.0 Hz, H-4'), 4.87 (d, 1 H, $J_{2,1'}$ 6.2 Hz, H-2), 5.65 (ddd, 1 H, $J_{2',3'}$ 11.5, $J_{2',4'}$ 1.0 Hz, H-2'), 5.87 (ddd, 1 H, J_{3',4'} 7.7 Hz, H-3'), 7.38-8.10 (m, 5 H, Ph).

Anal. Calc. for $C_{19}H_{26}O_6$: C, 64.81; H, 7.39. Found: C, 64.88; H, 7.50.

The tribenzoate of 6 was a syrup.

Anal. Calc. for C₃₃H₃₄O₈: C, 70.92; H, 6.09. Found: C, 70.50; H, 6.15.

Eluted second was 7, isolated as a syrup (0.75 g). ¹H-N.m.r. data (CDCl₃): δ 1.15 (s, 12 H, 4 Me), 3.23 (bs, 2 H, 2 OH), 4.29–4.50 (m, 3 H, H-1'.5'a,5'b), 4.75 (m, 1 H, H-4'), 4.89 (d, 1 H, $J_{2,1'}$ 6.5 Hz, H-2), 5.66 (dd, 1 H, $J_{1',2'}$ 7.5, $J_{2',3'}$ 12.1 Hz, H-2'), 5.80 (dd, 1 H, $J_{3',4'}$ 7.6 Hz, H-3'), 7.35–8.10 (m, 5 H, Ph).

Anal. Found: C, 64.70; H, 7.15.

The tribenzoate of 7 had m.p. 139-140° (from ethanol).

Anal. Found: C, 71.30; H, 6.20.

(b) To a solution of 3 (0.50 g) in dry methanol (20 mL) was added Methylene Blue (3 mg), and oxygen was bubbled through the solution for 1.5 h whilst it was irradiated with a 1000-W lamp. Methyl sulphide (0.22 mL) was then added, the mixture was kept in a refrigerator for 12 h and then concentrated, and the residue was reduced with sodium borohydride as in (a), to give 6 (50 mg) and 7 (40 mg) after chromatography.

4,4,5,5-Tetramethyl-2-(DL-threo-1,4,5-trihydroxy-cis-pent-2-enyl)-1,3-dioxolane (8). — Debenzoylation of **6** with methanolic sodium methoxide gave syrupy 8 (72.6%). ¹H-N.m.r. data (CDCl₃): δ 1.25 (s, 12 H, 4 Me), 3.00 (bs, 3 H, 3 OH), 3.57 (m, 1 H, $J_{5'a,5'b}$ 11.5 Hz, H-5'a), 3.65 (m, 1 H, H-5'b), 4.36 (m, 1 H, $J_{1',2'}$ 7.7, $J_{1',3'}$ 1.2 Hz, H-1'), 4.46 (m, 1 H, $J_{4',5'a}$ 6.0, $J_{4',5'b}$ 4.5 Hz, H-4'), 4.85 (d, 1 H, $J_{2,1'}$ 6.2 Hz, H-2), 5.61 (ddd, 1 H, $J_{2',3'}$ 11.5, $J_{2',4'}$ 1.0 Hz, H-2'), 5.76 (ddd, 1 H, $J_{3',4'}$ 7.7 Hz, H-3').

Anal. Calc. for C₁₂H₂₂O₅: C, 58.51; H, 8.94. Found: C, 58.55; H, 8.99.

4,4,5,5-Tetramethyl-2-(DL-erythro-1,4,5-trihydroxy-cis-pent-2-enyl)-1,3-dioxolane (9). — Debenzoylation of 7 led to 9 (67%). ¹H-N.m.r. data (CDCl₃): δ 1.25 (s, 12 H, 4 Me), 2.25, 2.55, 3.40 (bs, 1 H, OH), 3.60 (2 dd, 2 H, H-5'a,5'b), 4.40 (m, 2 H, H-1',4'), 4.85 (d, 1 H, $J_{2,1'}$ 7.5 Hz, H-2), 5.62 (ddd, 1 H, $J_{1',2'}$ 8.0, $J_{2',3'}$ 11.5, $J_{2',4'}$ 1.0 Hz, H-2'), 5.82 (ddd, 1 H, $J_{3',4'}$ 7.6, $J_{1',3'}$ 1.0 Hz, H-3').

Anal. Calc. for C12H22O5: C, 58.51; H, 8.94. Found: C, 58.45; H, 8.69.

Methyl 3,4-dideoxy- α -DL-threo-hex-3-enopyranoside (10). — A solution of 8 in methanolic 1% hydrogen chloride was boiled under reflux for 6 h, then neutralised, and concentrated. Column chromatography (chloroform-methanol, 9:1) of the residue gave 10 (206 mg, 63.3%). The amount of the β isomer was <5%. See Table I for the n.m.r. data.

Anal. Calc. for C₇H₁₂O₄: C, 52.47; H, 7.56. Found: C, 52.40; H, 7.42.

Methyl 3,4-dideoxy- α - (11a) and - β -DL-erythro-hex-3-enopyranoside (11b). — Methanolysis of 9, as for 8, gave a 1:4 mixture of 11a and 11b (150 mg, 77%). Column chromatography (chloroform-methanol, 95:5) of the mixture gave, first, the α anomer (11a) (30 mg) and then the β anomer (120 mg). See Table I for the n.m.r. data.

Anal. Calc. for C₇H₁₂O₄: C, 52.47; H, 7.56. Found: C, 52.37; H, 7.40.

ACKNOWLEDGMENTS

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