

LICORICE-SAPONINS F3, G2, H2, J2, AND K2, FIVE NEW OLEANENE-TRITERPENE OLIGOGLYCOSIDES FROM THE ROOT OF GLYCYRRHIZA URALENSIS

Isao KITAGAWA,* Jun Liang ZHOU, Masahiro SAKAGAMI, Emiko UCHIDA, and Masayuki YOSHIKAWA

Faculty of Pharmaceutical Sciences, Osaka University, 1-6 Yamada-oka, Suita, Osaka 565, Japan

Following the structure elucidation of licorice-saponins A3, B2, C2, D3, and E2, the structures of five more new oleanene-triterpene oligoglycosides named licorice-saponins F3 (1), G2 (5), H2 (7), J2 (10), and K2 (12), concomitantly isolated from the root of Glycyrrhiza uralensis Fischer (Tohoku-kanzo in Japanese), have been determined on the basis of chemical and physicochemical evidence. During the process, a facile chemical conversion method from olean-12-ene or olean-12-en-11-one oligoglycosides to an olean-11,13-diene oligoglycoside, has been found.

KEYWORDS Glycyrrhiza uralensis; Glycyrrhizae Radix; licorice-saponin F3; licorice-saponin G2; licorice-saponin H2; licorice-saponin J2; licorice-saponin K2; olean-11,13-diene oligoglycoside

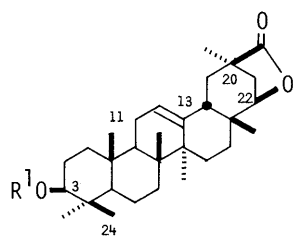
In our continuing chemical characterization studies of crude drug processing,¹⁾ we have initiated a chemical study of botanically identified licorice roots of various origins. Recently, we isolated ten oleanene-triterpene oligoglycosides and glycyrrhizin (4) from the root of Glycyrrhiza uralensis Fischer (Leguminosae)[Tohoku-kanzo (東北甘草) in Japanese] and elucidated the chemical structures of five of them named licorice-saponins A3, B2 (9), C2 (11), D3 (3), and E2.²⁾ Afterwards, we found two sweet oligoglycosides, apioglycyrrhizin and araboglycyrrhizin, from the root of G. inflata [Shinkyo-kanzo (新疆甘草) in Japanese].³⁾ In this paper, we describe the structure elucidation of licorice-saponins F3 (1), G2 (5), H2 (7), J2 (10),⁴⁾ and K2 (12),⁴⁾ which were isolated together with licorice-saponins A3, B2, C2, D3, and E2 from the root of G. uralensis previously.^{2, 5)} This paper also presents a facile chemical conversion method from an olean-12-ene or an olean-12-en-11-one oligoglycoside to an olean-11,13-diene oligoglycoside (i.e. 11 or 12).

Licorice-saponin F3 (1), mp 215-217 °C, $[\alpha]_D^{20}$ -20 °(MeOH), $C_{48}H_{72}O_{19}$,⁶⁾ IR (KBr, cm^{-1}): 3400, 1760, 1720, 1H NMR (500 MHz, pyridine- d_5 , δ): 5.36 (d, J = 7.6 Hz, 1'-H), 5.68 (d, J = 6.9 Hz, 1"-H), 6.10 (br.s, 1'''-H), gave deoxoglabrolide (2),⁷⁾ methyl glucuronide, and methyl rhamnoside, upon methanolysis. Ethereal CH_2N_2 treatment of 1 provided dimethyl ester 1a, mp 207-209 °C, $C_{50}H_{76}O_{19}$, IR (cm^{-1}): 3420, 1745, 1H NMR (δ): 4.92 (d, J = 7.2 Hz, 1'-H), 5.67 (d, J = 7.3 Hz, 1"-H), 6.20 (br.s, 1'''-H), positive FAB MS (m/z): 1003 ($M + Na$)⁺.

Methylation (CH_3I -DMSO-NaH) and subsequent $NaBH_4$ reduction of 1a followed by methanolysis, liberated methyl 3,4-di-O-methylglucopyranoside (a) and methyl 2,3,4-tri-O-methylrhamnopyranoside (b) in a 2:1 ratio. Comparison of ^{13}C NMR data for 1a (Table I) with those for D3 trimethyl ester (3a), led us to expect that licorice-saponins F3 (1) and D3 (3) have an identical trisaccharide moiety while F3 has a 30,22-lactone moiety in place of the 22 β -OAc and 20 β -COOH residues in D3. The expectation was verified by deacetylation (5% aq. KOH) of 3 and subsequent treatment with Dowex 50W $\times$ 8 (H^+) to furnish 1. Thus, the structure of licorice-saponin F3 (1) has been clarified as shown.

On ethereal CH_2N_2 treatment, licorice-saponin G2 (5), mp 229-230 °C, $[\alpha]_D^{20}$ +34 °(MeOH), $C_{42}H_{62}O_{17}$, UV (MeOH, nm, ϵ): 249 (10600), IR (cm^{-1}): 3350, 1720, 1648, 1H NMR (δ): 5.34 (d, J = 7.7 Hz, 1'-H), 5.59 (d, J = 6.7 Hz, 1"-H), ^{13}C NMR (125 MHz, pyridine- d_5 , δ): 63.1 (24-C), 104.4, 104.9 (1', 1''-C), positive FAB MS (m/z): 877 ($M + K$)⁺, 861 ($M + Na$)⁺, 839 ($M + H$)⁺, gave trimethyl ester 5a, mp 176-178 °C, $C_{45}H_{68}O_{17}$, IR (cm^{-1}): 3367, 1720, 1648, 1H NMR (δ): 4.94 (d, J = 7.9 Hz, 1'-H), 5.56 (d, J = 7.6 Hz, 1"-H).

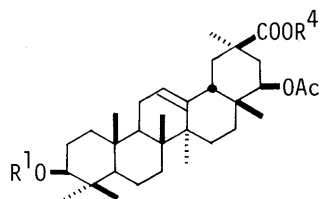
Methanolysis of 5a provided methyl 24-hydroxyglycyrrhetinate (6a)⁸⁾ and methyl glucuronide, while methanolysis, after $NaBH_4$ reduction and subsequent complete methylation of 5a, liberated methyl 2,3,4,6-tetra-O-methylglucopyranoside (c) and methyl 3,4,6-tri-O-methylglucopyranoside (d) in a 1:1 ratio. Based on these findings and examination of ^{13}C NMR data for 5a (Table I), the structure of licorice-saponin G2 has



1: R¹ = Sr
(licorice-saponin F3)

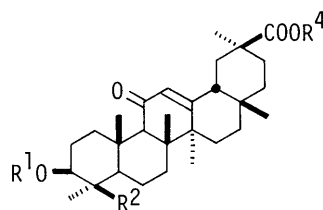
1a: R¹ = SrMe

2: R¹ = H (deoxoglabrolide)



3: R¹ = Sr, R⁴ = H
(licorice-saponin D3)

3a: R¹ = SrMe, R⁴ = CH₃

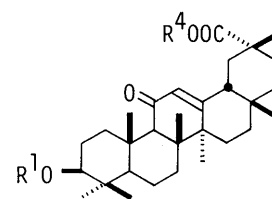


4: R¹ = S, R² = CH₃, R⁴ = H
(glycyrrhizin)

5: R¹ = S, R² = CH₂OH, R⁴ = H
(licorice-saponin G2)

5a: R¹ = SMe, R² = CH₂OH, R⁴ = CH₃

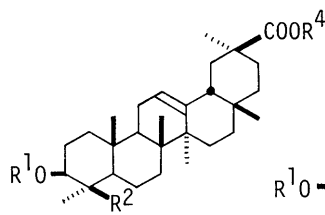
6a: R¹ = H, R² = CH₂OH, R⁴ = CH₃



7: R¹ = S, R⁴ = H
(licorice-saponin H2)

7a: R¹ = SMe, R⁴ = CH₃

8: R¹ = R⁴ = H
(liquiritic acid)

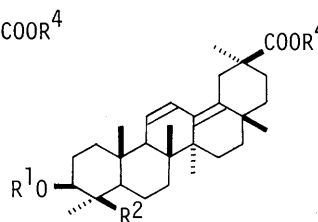


9: R¹ = S, R² = CH₃, R⁴ = H
(licorice-saponin B2)

9a: R¹ = SMe, R² = R⁴ = CH₃

10: R¹ = S, R² = CH₂OH, R⁴ = H
(licorice-saponin J2)

10a: R¹ = SMe, R² = CH₂OH, R⁴ = CH₃



11: R¹ = S, R² = CH₃, R⁴ = H (licorice-saponin C2)

11a: R¹ = SMe, R² = R⁴ = CH₃

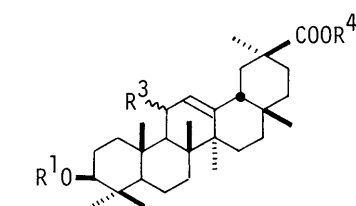
12: R¹ = S, R² = CH₂OH, R⁴ = H (licorice-saponin K2)

12a: R¹ = SMe, R² = CH₂OH, R⁴ = CH₃

Table I. ¹³C NMR Data for 1a, 5a, 7a, 10a, and 12a
(at 22.5 MHz in pyridine d₅, δc)

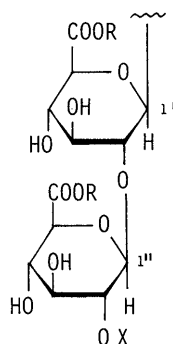
		1a	5a	7a	10a	12a
Sapogenol moiety	C-3	89.9	89.9	89.2	89.8	89.8
	C-11	47.5	198.9	199.0	47.5	125.3
	C-12	124.8	128.6	128.6	122.2	126.4
	C-13	140.6	168.7	168.4	144.3	135.2
	C-18	47.5	48.6	46.3	48.2	135.2
	C-22	84.3	38.2	37.1	36.2	36.2
	C-24	16.3	16.3	16.2	62.8	62.4
	C-29	28.0	28.0	178.1	28.1	28.1
	C-30	180.1	176.5	19.3	177.0	178.2
3-O-β-D-Glucurono-pyranosyl moiety	C-1'	104.7	104.1	104.9	104.0	104.0
	C-2'	79.1	81.9	84.3	81.1	80.8
	C-3'	76.3a)	75.2	76.4a)	76.3a)	76.6
	C-4'	72.2b)	72.2	72.5b)	72.1	72.1
	C-5'	77.9c)	77.2	77.3c)	77.4b)	77.7a)
	C-6'	170.1d)	169.6	170.1d)	169.6	169.8
2'-O-β-D-Glucurono-pyranosyl moiety	C-1''	102.4	105.5	106.8	104.7	104.5
	C-2''	78.2	76.5a)	76.6a)	76.6a)	76.6
	C-3''	76.6a)	76.8a)	77.4c)	77.0b)	77.0a)
	C-4''	72.8b)	72.2	72.9b)	72.1	72.1
	C-5''	77.6c)	77.2	77.6c)	77.4b)	77.4a)
	C-6''	169.8d)	169.6	170.2d)	169.6	169.8
2''-O-α-D-Rhamno-pyranosyl moiety	C-1'''	101.6				
	C-2'''	71.8				
	C-3'''	72.8				
	C-4'''	73.9				
	C-5'''	69.1				
	C-6'''	18.5				

a), b), c), d)
Assignments may be interchangeable within the same column.



13: R¹ = SMe, R³ = α-OAc & α-OCH₃, R⁴ = CH₃

14: R¹ = S, R³ = β-OH, R⁴ = H

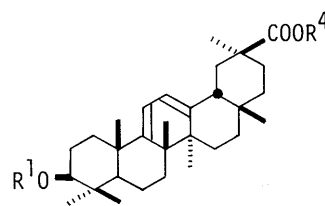


S: R = X = H

SMe: R = CH₃, X = H

Sr: R = H, X = α-L-rham(py).

SrMe: R = CH₃, X = α-L-rham(py).



15: R¹ = S, R⁴ = H

15a: R¹ = SMe, R⁴ = CH₃

been determined as 5.

Licorice-saponin H2 (7), mp 209-210 °C, [α]_D²⁵ +31 ° (MeOH), C₄₂H₆₂O₁₆, UV (MeOH, nm, ε): 248 (10700), IR (cm⁻¹): 3400, 1725, 1650, ¹H NMR (δ): 4.97 (d, J = 7.6 Hz, 1'-H), 5.35 (d, J = 7.6 Hz, 1''-H), ¹³C NMR (δc): 105.0 (1'-C), 106.8 (1''-C), 128.8 (12-C), 169.0 (13-C), 180.6 (29-C), 199.4 (11-C), positive FAB MS (m/z): 861 (M+K)⁺, 845 (M+Na)⁺, liberated liquiritic acid (8)⁹ and methyl glucuronide upon methanolysis.

Ethereal CH₂N₂ treatment of 7 afforded trimethyl ester 7a, mp 169 °C, C₄₅H₆₈O₁₆, IR (cm⁻¹): 3362, 1722, 1654, ¹H NMR (δ): 4.95 (d, J = 7.3 Hz, 1'-H), 5.33 (d, J = 7.7 Hz, 1''-H), positive FAB MS (m/z): 887 (M+Na)⁺,

which, upon methanolysis after NaBH_4 reduction and subsequent methylation, liberated methyl glycosides (c and d) as it did from 5a. These findings together with ^{13}C NMR data for 7a (Table I) led us to formulate licorice-saponin H2 as 7.

Licorice-saponin J2 (10), mp 263–265 °C, $[\alpha]_D^{25} +21^\circ$ (MeOH), $\text{C}_{42}\text{H}_{64}\text{O}_{16}$, IR (cm^{-1}): 3405, 1729, 1615, ^1H NMR (δ): 5.00 (d, $J=7.6$ Hz, $1'\text{-H}$), 5.65 (d, $J=7.0$ Hz, $1''\text{-H}$), was converted to trimethyl ester 10a, mp 198–199 °C, $\text{C}_{45}\text{H}_{70}\text{O}_{16}$, IR (cm^{-1}): 3445, 1730, by CH_2N_2 methylation. These findings together with the examination of ^{13}C NMR data for 10a (Table I) led us to expect that J2 (10) was an 11-deoxo analog of licorice-saponin G2 (5). The expectation was verified by converting 5 in high yield to J2 (10) by reduction with zinc amalgam and HCl.

Methylation with CH_2N_2 of licorice-saponin K2 (12), mp 207–209 °C, $[\alpha]_D^{25} +28^\circ$ (MeOH), $\text{C}_{42}\text{H}_{62}\text{O}_{16}$, UV (MeOH, nm, ϵ): 241 (13000), 249 (15000), 259 (9200), IR (cm^{-1}): 3460, 1690, 1629, ^1H NMR (δ): 5.04 (d, $J=7.6$ Hz, $1'\text{-H}$), 5.64 (d, $J=6.1$ Hz, $1''\text{-H}$), 5.54 (br.d, 11-H), 6.52 (br.d, 12-H), positive FAB MS (m/z): 845 ($\text{M}+\text{Na}$) $^+$ afforded trimethyl ester 12a, mp 179–181 °C, $\text{C}_{45}\text{H}_{68}\text{O}_{16}$, IR (cm^{-1}): 3424, 1731, 1621, positive FAB MS (m/z): 903 ($\text{M}+\text{K}$) $^+$. These physicochemical properties and the examination of ^{13}C NMR data for 12a led us to presume the structure of licorice-saponin K2 (12) with a heteroannular 11,13-diene structure. ¹⁰⁾

To prove the structure 12, we carried out a conversion from known oligoglycoside. Thus, licorice-saponin B2 trimethyl ester (9a) was subjected to constant-current electrolysis (Pt, 40 mA/cm², MeOH–AcONa) ¹¹⁾ to provide a mixture of 11 α -OAc and 11 α -OMe derivatives (13). Treatment of 13 with 1% aq. HCl furnished licorice-saponin C2 trimethyl ester (11a) in 61% yield from 9a. On the other hand, NaBH_4 reduction of glycyrrhizin (4) in EtOH–H₂O (1:1) at 90 °C yielded an 11 β -hydroxy derivative 14, mp 208–210 °C, $\text{C}_{42}\text{H}_{64}\text{O}_{16}$, ^1H NMR (δ): 4.45 (dd, $J=6.5, 6.5$ Hz, 11 α -H), ¹²⁾ in 95% yield. Treatment of 14 with 2% aq. HCl–dioxane (1:1) yielded a mixture of a 9,12-homoannular diene analog 15 and licorice-saponin C2 (11), ¹³⁾ whereas treatment of 14 in dioxane–H₂O (1:1) with heating provided 11 in an excellent yield (82% from 4). Finally, NaBH_4 reduction of licorice-saponin G2 (5) and subsequent treatment of the product with dioxane–H₂O as above furnished licorice-saponin K2 (12) in 76% yield, thus the structure 12 was substantiated.

ACKNOWLEDGEMENT The authors are grateful to the Ministry of Education, Science, and Culture of Japan for financial support (Grant No. 02403027).

REFERENCES AND NOTES

- 1) a) I.Kitagawa and M.Yoshikawa, J.Traditional Sino-Japanese Medicine, Vol.6, No.4, 101 (1985); b) Idem, *ibid.*, Vol.7, No.3, 55 (1986); c) I.Kitagawa, T.Taniyama, M.Yoshikawa, Y.Ikenishi, and Y.Nakagawa, *Chem. Pharm.Bull.*, **37**, 2961 (1989), and previous papers cited therein.
- 2) I.Kitagawa, J.L.Zhou, M.Sakagami, T.Taniyama, and M.Yoshikawa, *Chem.Pharm.Bull.*, **36**, 3710 (1988).
- 3) I.Kitagawa, M.Sakagami, F.Hashiuchi, J.L.Zhou, M.Yoshikawa, and J.Ren, *Chem.Pharm.Bull.*, **37**, 551 (1989).
- 4) The trivial names licorice-saponins I2 and J2 (obtained in 0.002% respectively from the root) in previous paper²⁾ were changed to licorice-saponins J2 and K2, respectively in this paper.
- 5) M.Yoshikawa, J.L.Zhou, M.Sakagami, F.Hashiuchi, and I.Kitagawa, presented at the 108th Annual Meeting of the Pharmaceutical Society of Japan, held in Hiroshima, Apr.4–6, 1988, Abstract Papers p.316.
- 6) The molecular composition of the compound given with the chemical formula was determined by elemental analysis and (or) high resolution mass-spectrometry.
- 7) L.Canonica, G.Russo, and A.Bonati, *Gazz.Chim.Ital.*, **96**, 772 (1966).
- 8) L.Canonica, B.Danieli, P.Manitto, G.Russo, and A.Bonati, *Gazz.Chim.Ital.*, **97**, 1359 (1967).
- 9) L.Canonica, G.Russo, and E.Bombardelli, *Gazz.Chim.Ital.*, **96**, 833 (1966).
- 10) a) S.Shibata, I.Kitagawa, and H.Fujimoto, *Chem.Pharm.Bull.*, **14**, 1023 (1966); b) M.Nose, S.Amagaya, T.Takeda, and Y.Ogihara, *ibid.*, **37**, 1293 (1989).
- 11) a) I.Kitagawa, T.Kamigauchi, H.Ohmori, and M.Yoshikawa, *Chem.Pharm.Bull.*, **28**, 3078 (1980); b) M.Yoshikawa, H.K.Wang, V.Tosirisuk, and I.Kitagawa, *Chem.Pharm.Bull.*, **30**, 3057 (1982).
- 12) S.Shibata, K.Takahashi, S.Yano, M.Harada, H.Saito, Y.Tamura, A.Kumagai, K.Hirabayashi, M.Yamamoto, and N.Nagata, *Chem.Pharm.Bull.*, **35**, 1910 (1987).
- 13) The separation of 11 and 15 was effected after CH_2N_2 methylation to provide 11a and 15a, mp 180 °C, $\text{C}_{45}\text{H}_{68}\text{O}_{15}$, IR (cm^{-1}): 3440, 1727, 1660, ^1H NMR (pyridine- d_5 + D_2O , δ): 5.30, 5.31 (both d, $J=7.3$ Hz, 11,12-H), ^{13}C NMR (22.5 MHz, pyridine- d_5 , δ c): 154.9, 115.7, 125.5, 146.0 (9,11,12,13-C).

(Received November 22, 1990)