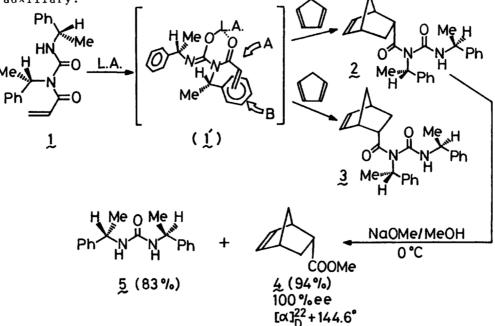
CHEMISTRY LETTERS, pp. 1623-1624, 1988.

Diels-Alder Reaction of Chiral Acrylamide and a Convenient Synthesis of Optically Pure Methyl (3R, 4R, 6R)-Bicyclo[2. 2. 1]heptene-4-carboxylate

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Asymmetric Diels-Alder reaction of the chiral acrylamide  $\underline{1}$  proceeds highly diastereoselectively and methanolysis of the adduct  $\underline{2}$  gives the optically pure (3R, 4R, 6R)-bicyclo[2. 2. 1]-heptene-4-carboxylate.

In recent studies of asymmetric Diels-Alder reactions<sup>1)</sup> some highly enantioselective reactions have been reported. However to isolate a pure enantiomer is often problematic. In our investigation of the asymmetric Diels-Alder reaction of acrylamide <u>1</u>, we have established a very convenient method to synthesize optically pure (100% ee) methyl (3R, 4R, 6R)-bicyclo[2. 2. 1] heptene-4-carboxylate (<u>4</u>) in an excellent yield. The advantages of our method over others are the commercial availability of the chiral sources (both (R)- and (S)-1-phenylethylamine), a simple separation of the primary diastereomeric cycloadducts <u>2</u> and <u>3</u> by conventional column chromatography and methanolysis under mild condition to remove the chiral auxiliary.



Scheme 1.

No.	Lewis Acid	Temp/°C	Time/h	Yield/%a)	endo/exo <sup>b)</sup>	<u>2/3</u> b)
1	no cat.	100	3	71	77:23	52:48
2	SnCl <sub>4</sub>	-78	1	87	97: 3	64 <b>:</b> 36
3	Et <sub>2</sub> AIC1	-115	1	60	97: 3	68:32
4	TiCl <sub>4</sub>	-78	1	85	98: 2	81:19
5	TiCl4	-115	1	93	98: 2	74:26

Table 1. Diels-Alder Reaction of the Chiral Acrylamide  $\underline{1}$  with Cyclopentadiene

a) Combined yields of all four Diels-Alder adducts. b) The ratios were determined by HPLC (Merck Lichrosorb Si 60; hexane:ethyl acetate=4:1) and  $^{1}$ H-NMR (270 MHz) analysis.<sup>2)</sup>

The acrylamide <u>1</u> was prepared from acrylic acid and N, N'-di-((S)-l-phenylethyl)urea.<sup>3)</sup> The Diels-Alder reaction of <u>1</u> (1 mmol) with cyclopentadiene (1.5 mmol) was performed in the presence of a Lewis acid (1.5 mmol) at 100 °C (in toluene), at -78 °C (in  $CH_2Cl_2$ ) or at -115 °C (in EtCl). Without catalyst the endo/exo ratio was 77:23. However in the presence of Lewis acid the endo-selectivity was significantly improved (over 97%). Especially TiCl<sub>4</sub> promoted the reaction with high stereoselectivity.

The major bicyclic acylurea  $\underline{2}$  was purified by column chromatography (silica gel, hexane-ethyl acetate) and treated with NaOMe (10 equiv.)/MeOH (0 °C, 8 h) to give the optically pure methyl ester  $\underline{4}^{(4)}$  (94%) and the chiral urea  $\underline{5}$  (83%).

The observed high selectivity can be explained by considering the complex l' depicted in Scheme 1. The  $\pi$ - $\pi$ -attractive interactions of the pendant vinyl group with the benzene ring must occur at the re-face of the olefine moiety due to the H-H repulsions between the olefin and the methyl group of the phenethyl moiety. Accordingly, the Diels-Alder reaction should occur stereoselectively at the siface giving the bicyclic acylurea  $\underline{4}$  as a major product.

These results indicate that it is also possible to obtain the optically pure (35, 45, 65)-isomer from (R)-l-phenylethylamine. Further, both the chiral auxiliaries can be efficiently recycled.

We thank the fund of Mitsui Petrochemical Industries, Ltd., for a fellowship (K. K.). Thanks are also given to Miss Kayoko Kogure for assistance in this work.

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- 2) The ratios (2/3) were determined based on the following signals. <sup>1</sup>H-NMR (§, J in Hz, CDCl<sub>3</sub>): <u>2</u>: 6.08(dd, J=5.6, 2.8, 1H, H-C(2)), 3.18(s, 1H, H-C(3)), 3.31 (qi, J=7.0, 1H, H-C(4)), <u>3</u>: 5.90(dd, J=5.8, 2.8, 1H, H-C(2)), 2.78(s, 1H, H-C(3)), 3.37(qi, J=4.2, 1H, H-C(4)).
- 3) K. Yamada, K. Kishikawa, S. Kohmoto, and M. Yamamoto, Chem. Lett., <u>1988</u>, 351.
- 4) The value observed was [𝒜]<sup>22</sup><sub>D</sub> +144.6° (EtOH). (lit. [𝒜]<sub>D</sub> +141° (95% EtOH): J. A. Berson and D. A. Ben-Efraim, J. Am. Chem. Soc., <u>81</u>, 4084 (1959).