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We discovered a new reaction in the pyrimidine series, namely the reaction of the anhydro-base 4-(3-indoly1)pyrimidine (I) with 1-pheny1-3-methylphrazolon-5-one:



The reaction proceeds by a double rearrangement scheme. First, as usual, the reagent adds to the $C_{(6)}$ -atom of the pyrimidine ring. This is followed by ring opening in adduct II, and closure of a new ring with the participation of an electron-excess pyrimidine nitrogen atom and the carbonyl group of pyrazolone, into betaine III. Then, there is no aromatization of betaine III to pyranolopyridine, as in several similar cases [1, 2], but a repeated cyclization by a scheme of a sigmatropic rearrangement, leading to oxime IV. This reaction is completely new, not only for pyrimidine bases, but also for nitrogen-containing heterocycles, in general. A very remote analogy can be found in one single source only [3].

Compound IV was obtained by reaction between equimolar amounts of anhydro-base I and 1phenyl-3-methylpyrazol-5-one in acetonitrile at room temperature for 3 h, yield 21%, mp 186-188°C (from acetone); IR spectrum: 1010 (NO); 1630 cm⁻¹ (C=O). In the mass spectrum, a peak of M⁺ with m/z 358 is recorded. The experimentally found isotopic correction (24.65%) corresponds well to the calculated one (25.05). The empirical formula was found from a high resolution mass spectrum (found 358.1143; calculated 358.1440). At the first stage of fragmentation, the OH particles are eliminated (found 341.1426, calculated for formula of ion $C_{21}H_{17}N_{4}O$ 341.1402). This phenomenon is typical of compounds containing an oxime group. In general, the character of the fragmentation strictly confirms the structure of the compound obtained.

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