

## TEN AZULENES FROM *PLAGIOCHILA LONGISPINA* AND *CALYPOGEIA AZUREA*\*

U. SIEGEL, R. MUES, R. DÖNIG,† TH. EICHER,† M. BLECHSCHMIDT‡ and H. BECKER‡

Fachbereich 13 Botanik, Universität des Saarlandes, D-6600 Saarbrücken 11, F.R.G.; †Fachbereich 11 Organische Chemie, Universität des Saarlandes, D-6600 Saarbrücken 11, F.R.G.; ‡Fachbereich 12 Pharmakognosie und Analytische Phytochemie, Universität des Saarlandes, D-6600 Saarbrücken 11, F.R.G.

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**Key Word Index**—*Plagiochila longispina*; *Calypogeia azurea*; Jungermanniales; Hepaticae; azulenes; synthesis.

**Abstract**—From gametophytic tissues of the naturally grown foliose liverworts *Plagiochila longispina* and *Calypogeia azurea* and from *in vitro* cultures of *C. azurea*, 10 azulenes have been isolated, of which eight proved to be new natural products. Their structures were elucidated by means of spectroscopic methods and by independent synthesis.

### INTRODUCTION

Azulenes are known from tracheophytes, mainly angiosperms, where they often occur together with terpenoids, being synthesized in the plant by dehydrogenation of sesquiterpenes [1]. Liverworts are the only bryophyte class from which azulenes have been reported: 1,4-dimethylazulene and 1-methoxycarbonyl-4-methylazulene are the only known such constituents so far detected in liverwort species. In all liverworts, where it occurs, 1,4-dimethylazulene is either the main or the only azulene. Huneck [2] reported, for the first time, the occurrence of azulenes in the blue oil bodies of liverworts: he isolated both compounds from *Calypogeia azurea*. Takeda and Katoh [3] also detected 1,4-dimethylazulene in extracts of cell suspension cultures of *C. granulata*. Katoh and Takeda [4] and Nagashima *et al.* [5] found the same azulene in *C. peruviana*, *C. tosona*, *Macrolejeunea pallescens* and *Plagiochila micropterys*.

In the course of a chemical study of the liverwort *Plagiochila longispina* [6], we noticed the occurrence of a blue azulene in its methylene chloride extract. A closer investigation of this extract revealed, besides the two main azulenes, the occurrence of further minor azulenes. A reinvestigation of *Calypogeia azurea* from the natural habitat and from *in vitro* cultures surprisingly revealed even more azulenes as minor constituents in the methylene chloride and MeOH extracts from both sources.

### RESULTS AND DISCUSSION

#### *Azulenes from Plagiochila longispina*

A *n*-hexane–methylene chloride extract (see Experimental) of *Plagiochila longispina* revealed, after chromatographic separation, the presence of one blue and three purple compounds. The blue compound was isolated

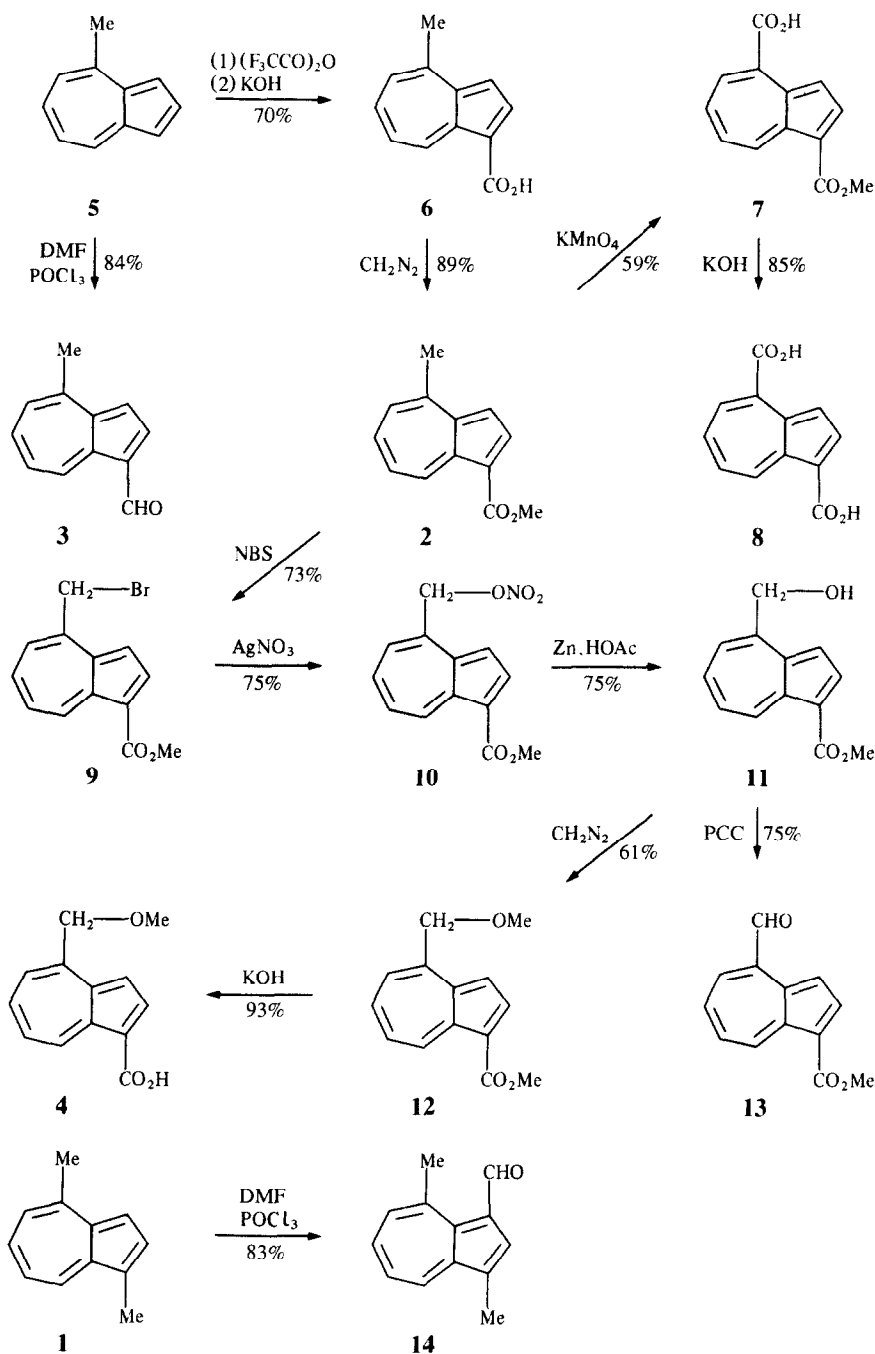
by column chromatography on silica gel with *n*-hexane, whereas this solvent did not elute the three further ones from the column. The mass spectrum of **1** showed  $m/z$  156 as  $[M]^+$  and the  $^1H$  NMR spectrum revealed the same signal pattern and chemical shifts as published by Meuche and Huneck [7] for 1,4-dimethylazulene. This compound (*ca* 5 mg) was the main azulene type of the lipophilic extract; the three further ones could be isolated only in traces. The real content of **1** in the liverwort is certainly much higher, because it is very volatile and most of it evaporated from the air-dried plant material after its collection in the Ecuadorian Andes. The purple colour of the three further constituents indicated the presence of an oxygen substituted azulene skeleton [1]. Compound **2** was isolated from the silica gel column by a *n*-hexane–ethyl acetate gradient, eluted with 5% ethyl acetate together with the chlorophylls and was separated from them by column chromatography on RP 18 with 80% aqueous methanol.

Compounds **3** and **4** eluted from the silica gel column with more than 5% ethyl acetate, **4** only after addition of 5% acetic acid to the solvent. Both compounds were finally separated by column chromatography on silica gel with *n*-hexane–10% ethyl acetate as solvent (see Experimental).

The spectroscopic data of **2** are in accord with those published by Meuche and Huneck [7] for 1-methoxycarbonyl-4-methylazulene and identical in all respects with an authentic sample synthesized according to the literature procedure [8] from 4-methylazulene (**5**). For the other two compounds, the structures of 1-formyl-4-methylazulene (**3**) and 1-carboxy-4-methoxymethylazulene (**4**) can be assigned mainly on the basis of mass and  $^1H$  NMR spectroscopic data (Table 1). An unambiguous structure proof for **3/4** is provided by the independent synthesis of **3** from 4-methylazulene **5** and of **4** from ester **2**. These are new natural products.

After this first isolation of oxygen-containing azulenes from *Plagiochila longispina*, we reinvestigated *Calypogeia azurea* to see if it contained any new azulenes [2]. We analysed both gametophytic fresh material of *C. azurea*

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collected from the natural habitat (see Experimental) and air-dried material from *in vitro* cultures.

#### *Azulenes from Calypogeia azurea*

As the *in vitro* cultures generally contained more compounds than the naturally grown plants, we analysed the *in vitro* cultures. The lipophilic extracts were separated with a column chromatography system (see Experimental) by which seven compounds were eluted almost in the front, followed by two others. The  $^1H$  NMR,

mass spectra and chromatographic data (Table 1) proved the blue compound to be the 1,4-dimethylazulene (**1**). With 90 mg (=0.05% of dry weight of extracted plant material) this was the main azulene of *C. azurea* from *in vitro* cultures. The second main constituent was shown to be 1-methoxycarbonyl-4-methylazulene (**2**) by its spectroscopic data (Table 1). The yield was 28 mg (=0.02% of dry weight of extracted plant material).

The first seven compounds were separated by column chromatography with a gradient system, in which three azulenes eluted first, followed by four others (see Experi-

mental). The yield of these and the following pure compounds was 1 to 2 mg or less. Compounds **8** ester and **11** are a red purple and **7** a deep purple colour. In solid state they are relatively sensitive to oxidation, thus they are better stored in solution, e.g. in acetone. These compounds are 1,4-disubstituted azulenes according to their spectroscopic data (Table 1): one proved to be 4-hydroxymethyl-1-carboxymethyl-azulene (**11**) and another 4-carboxy-1-methoxycarbonylazulene (**7**), as shown by their independent synthesis. The ester of **8** was rather unstable being easily hydrolysed to the stable azulene-1,4-dicarboxylic acid (**8**), obtained independently by saponification of the monoester **7**. We assume that in the natural product the carboxy group at position 4 is esterified with a hitherto unknown alcohol, since  $^1\text{H}$ NMR spectroscopic data indicate a free carboxy group at position 1.

The purification of the second fraction with four compounds resulted in the isolation of azulenes **3** and **6** as main compounds and of **13** and **14** as trace constituents. The  $^1\text{H}$ NMR, mass spectra and chromatographic data of **3** are identical with the same compound isolated from *Plagiochila longispina* thus identifying it as 4-methylazulene-1-aldehyde. As **8** ester and **7**, **6** shows the typical chromatographic behaviour of an acid. By direct comparison of its spectroscopic data with an authentic sample synthesized according to ref. [8] it was shown to be 4-methylazulene-1-carboxylic acid.

Although **13** and **14** were only isolated in minute amounts, the structure of **14** could be completely elucidated, whereas for **13** full spectroscopic characterization was not possible. Compound **14** appeared chromatographically as a carboxy-free blue-purple azulene. Among all azulenes discussed in this paper the  $^1\text{H}$ NMR spectrum of **14** exhibits for the first time a structure type different from 1,4-disubstitution: the  $^1\text{H}$ NMR and mass spectral data show two methyl groups and an aldehyde function arranged in the pattern of a 1,3,4-trisubstituted azulene skeleton. The independent synthesis by formylation of 1,4-dimethylazulene (**1**) proves its structure to be 1,4-dimethyl-3-formyl-azulene.

Compound **13** could not be purified enough to get suitable  $^1\text{H}$ NMR and mass spectra because of the minute amount available. It contained, as a chromatographically detectable impurity, 4-methylazulene-1-carboxylic acid (**6**). For **13** the structure of a 4-formyl-1-methoxycarbonylazulene seems to be likely, since its fragmentation pattern in the mass spectrum (Table 1) is identical with that of the product **13** synthesized by oxidation of the hydroxymethyl compound **11**. Further information on **13** is not available. A formyl methoxycarbonylazulene was found in *Helichrysum* and *Ixiolaena* species (Compositae) [9–11], the physical and spectroscopic data of which are not in accord with our findings on **13**. Compounds **6**, **7**, **8** ester, **11**–**14** are new natural products.

From the material of the natural habitat we isolated the azulenes **1**–**3** and **14**. With 2300 mg (=0.45% of extracted air-dried plant material) the yield of the main compound **1** was significantly higher than for the cultures. The second main compound **2** was obtained in an amount of 52 mg (=0.01%), almost the same yield as for the cultured material, and the two remaining compounds were isolated only in traces. Thus, besides compound **1**, which was found in almost 10 times higher yield in the naturally grown plants than in those from *in vitro* cul-

tures, the quantity of the isolated azulenes is comparable in both sources. From *in vitro* cultures, nine azulenes have been isolated, whereas only four azulenes, structurally identical to those from *in vitro* cultures, were detected in the plants collected in the field.

The independent syntheses were performed as outlined in the Experimental. Thus, aldehyde **3** was obtained by Vilsmeier formylation of 4-methylazulene (**5**) with DMF-POCl<sub>3</sub>, the monoester **7** by oxidation of known **2** [8] with KMnO<sub>4</sub> and the diacid **8** by saponification of **7** with KOH in methanol-H<sub>2</sub>O. The ester **2** also served as starting material for the synthesis of **4** by functionalization at the 4-methyl group through NBS-bromination (**2**→**9**), nucleophilic exchange of bromine by nitrate (**9**→**10**) and reductive cleavage of the nitrate **10** by Zn in acetic acid to give the 4-hydroxymethyl ester **11**. The ester **11** was successively *O*-methylated with diazomethane (to yield **12**) and saponified with KOH in methanol-H<sub>2</sub>O to give rise to the acid **4** with an overall yield of 22% (based on **2**, in a five-step sequence). Oxidation of the hydroxymethyl ester **11** by PCC leads to the aldehyde ester **13**. Finally, formylation of **1** (DMF-POCl<sub>3</sub>) gave rise to the aldehyde **14**.

## EXPERIMENTAL

**Plant material.** Fresh gametophytes of *Plagiochila longispina* Lindenb. et Gott. were collected in October 1988 in the Ecuadorian Andes, Prov. Napo, road Quito-Baeza, subalpine disturbed rainforest, 3400 m, by J.-P. Frahm and S. R. Gradstein (No. 6929). Gametophytic material of *Calypogeia azurea* Stotler et Crotz was collected from the natural habitat in August 1990 in the Bernese Alps, near Handegg, Hasli-valley, Switzerland, at about 1500 m by U.S. and R.M. Voucher specimens are deposited in the herbarium of the Fachrichtung Botanik, Universität des Saarlandes, Saarbrücken (SAAR). The plants were identified by S. R. Gradstein (Utrecht; P.I.) and R.M.(C.a.). An axenic culture of *C. azurea* was obtained in 1985 from Dr. Vandekerckhove, University of Mainz. The cultures were grown in 200-ml flasks with 50 ml solid modified B5 (pH 5.7) medium [12], containing 20 g l<sup>-1</sup> sucrose. The flasks were kept under constant illumination (2000 lux) at 20°.

### Extraction and isolation.

*Plagiochila longispina*. After careful cleaning, air-dried gametophytes (30 g) were ground in a coffee-mill and extracted with CH<sub>2</sub>Cl<sub>2</sub>. This extract was evapd to a small vol. and chromatographed by CC over silica gel with a *n*-hexane-Me<sub>2</sub>CO gradient. Compound **1** was eluted by *n*-hexane and purified again by a silica gel column with *n*-hexane. Compounds **2**–**4** eluted from the silica gel column with a gradient of *n*-hexane-EtOAc starting from *n*-hexane up to *n*-hexane-EtOAc (17:3). Chlorophyll and other lipids were separated from the azulenes by CC on RP 18 with 80% aq. MeOH, in which the azulenes eluted, and compound **2** was separated from **3** and **4**. Final purification of **2** was achieved by CC on silica gel with *n*-hexane-EtOAc (9:1). The yield was less than 1 mg.

Compounds **3** and **4** were separated from each other on a silica gel column with *n*-hexane-EtOAc (9:1). With this solvent **3** eluted and was finally purified by prep. HPLC on a Spherisorb ODS-2 column with MeCN-H<sub>2</sub>O (1:1; R, 20 min) in about 1 mg amount. Addition of 5% HOAc to the last silica gel column eluted **4** in traces. It was finally purified two times by CC on Sephadex LH-20 with MeOH (Uvasol)-H<sub>2</sub>O (bidest.; 2:3).

*Calypogeia azurea*, *in vitro* cultured material. Air-dried gametophytes (160 g) were ground in a coffee-mill and extracted first with *n*-hexane, followed by CH<sub>2</sub>Cl<sub>2</sub> and EtOAc. The combined extracts were evapd to a small vol. and chromatographed by CC

Table 1. Spectroscopic data of azulenes 1–4/6–14\*

Compound	IR $\nu_{\text{max}}^{\text{KBr}}$ $\text{cm}^{-1}$	UV $\lambda_{\text{max}}^{\text{CH}_2\text{Cl}_2}$ nm (log $\epsilon$ )	$^1\text{H NMR}$ (400 MHz, $\text{CDCl}_3$ )	$^{13}\text{C NMR}$ (100 MHz, $\text{CDCl}_3$ )	CIMS (probe) 120 eV, $m/z$ (rel. int.)
1	3020, 2920	722 (2.90)	$\delta$ 8.21 (1H, d, $J=9.5$ Hz, H-8)	$\delta$ 145.4, 141.2,	157 ( $[\text{M}+1]^+$ , 89)
	1595, 1460	625 (3.33)	7.64 (1H, d, $J=4.7$ Hz, H-2)	139.9, 138.6,	156 ( $[\text{M}]^+$ , 100)
		597 (3.42)	7.43 (1H, t, $J=9.9$ Hz, H-6)	137.4, 126.8,	141 (18)
		362 (4.31)	7.31 (1H, d, $J=3.7$ Hz, H-3)	126.2, 126.0,	115 (4)
		347 (4.63)	7.05–6.97 (2H, m, H-5/H-7)	124.8, 116.3,	
2		337 (4.80)	2.84, 2.65 (3H, s, Me)	24.3, 19.8	
	2945, 1690 (CO),	602 (3.33)	$\delta$ 9.69 (1H, d, $J=9.8$ Hz, H-8)	$\delta$ 166.0 (CO), 148.5,	201 ( $[\text{M}+1]^+$ , 11)
	1460, 1145	553 (3.58)	8.31 (1H, d, $J=4.3$ Hz, H-2)	142.6, 140.4,	200 ( $[\text{M}]^+$ , 21)
		348 (4.68)	7.69 (1H, t, $J=9.9$ Hz, H-6)	139.0, 137.9,	185 (7)
		332 (4.55)	7.51–7.39 (2H, m, H-5/H-7)	132.7, 130.1,	142 (27)
3			7.30 (1H, d, $J=4.3$ Hz, H-3)	126.4, 117.2,	115 (100)
			3.94 (3H, s, $\text{CO}_2\text{Me}$ )	115.1, 51.0 ( $\text{CO}_2\text{Me}$ ),	
			2.94 (3H, s, Me)	25.0 (Me)	
	2945, 2725,	630 (2.42)	$\delta$ 10.35 (1H, s, CHO)	$\delta$ 186.8 (CO),	171 ( $[\text{M}+1]^+$ , 10)
	1650 (CO), 1265	533 (2.90)	9.64 (1H, d, $J=9.6$ Hz, H-8)	149.9, 144.3,	170 ( $[\text{M}]^+$ , 64)
4		384 (4.18)	8.21 (1H, d, $J=4.2$ Hz, H-2)	141.3, 139.8,	169 (100)
		370 (4.17)	7.76 (1H, t, $J=10.1$ Hz, H-6)	138.6, 137.8,	115 (58)
		305 (4.87)	7.60–7.50 (2H, m, H-5/H-7)	131.5, 128.3,	
			7.36 (1H, d, $J=4.2$ Hz, H-3)	126.4, 116.7,	
			2.97 (3H, s, Me)	25.1 (Me)	
6	3415 (OH),	640 (2.15)	$\delta$ 9.69 (1H, d, $J=9.6$ Hz, H-8)	$\delta$ 165.3 (CO),	217 ( $[\text{M}+1]^+$ , 5)
	2920,	402 (3.67)	8.30 (1H, d, $J=4.3$ Hz, H-2)	148.7, 142.3,	216 ( $[\text{M}]^+$ , 8)
	1710 (CO),	385 (4.36)	7.86 (1H, t, $J=9.6$ Hz, H-6)	140.8, 140.0,	172 (100)
	1030	321 (4.68)	7.63–7.49 (2H, m, H-5/H-7)	139.7, 138.1,	126 (21)
		301 (4.54)	7.24 (1H, d, $J=4.3$ Hz, H-3)	133.9, 130.4,	115 (34)
7			3.61 (3H, s, OMe)	126.3, 117.6,	
			3.59 (2H, s, $\text{CH}_2$ )	69.3, 52.5	
	3445 (OH),	548 (3.40)	$\delta$ 9.64 (1H, d, $J=9.8$ Hz, H-8)	$\delta$ 166.2 (CO),	187 ( $[\text{M}+1]^+$ , 11)
	2915,	347 (4.51)	8.24 (1H, d, $J=4.2$ Hz, H-2)	148.9, 141.8,	186 ( $[\text{M}]^+$ , 100)
	1640 (CO),	330 (4.36)	7.86 (1H, t, $J=9.6$ Hz, H-6)	139.4, 138.8,	169 (81)
7	1250	300 (4.93)	7.62–7.54 (2H, m, H-5/H-7)	138.3, 137.3,	144 (38)
			7.39 (1H, d, $J=4.2$ Hz, H-3)	130.1, 126.1,	115 (91)
			2.94 (3H, s, Me) <sup>†</sup>	117.2, 114.9,	
				24.5 (Me) <sup>†</sup>	
	3310 (OH),	621 (2.98)	$\delta$ 9.80 (1H, d, $J=9.9$ Hz, H-8)	$\delta$ 168.4 (CO),	230 ( $[\text{M}]^+$ , 26)
7	2915,	399 (3.88)	8.21 (1H, d, $J=4.2$ Hz, H-2)	165.9 (CO),	199 (45)
	1720/1695 (CO),	315 (4.61)	7.86 (1H, t, $J=9.9$ Hz, H-6)	148.9, 141.5,	186 (100) <sup>†</sup>

8	1215, 1040	309 (4.47) 301 (4.46)	7.73–7.59 (2H, <i>m</i> , H-5/H-7) 7.41 (1H, <i>d</i> , <i>J</i> = 4.2 Hz, H-3) 3.95 (3H, <i>s</i> , CO <sub>2</sub> Me)	140.1, 138.8, 138.2, 137.2, 131.5, 130.6, 127.9, 117.4, 51.0 (Me)	216 ([M] <sup>+</sup> , 16) 198 (7) 172 (11) 115 (100) <sup>‡</sup>
	3390, 2980, 1750/1695 (CO), 1460, 1035	633 (2.71) 501 (2.99) 399 (4.01) 336 (4.48) 300 (4.85)	δ9.86 (1H, <i>d</i> , <i>J</i> = 9.8 Hz, H-8) 8.81 (1H, <i>s</i> , CO <sub>2</sub> H) 8.35 (1H, <i>d</i> , <i>J</i> = 4.4 Hz, H-2) 7.84 (1H, <i>t</i> , <i>J</i> = 9.7 Hz, H-6) 7.61–7.49 (2H, <i>m</i> , H-5/H-7) 7.40 (1H, <i>d</i> , <i>J</i> = 4.4 Hz, H-3)	δ168.5 (CO), 166.3 (CO), 141.6, 140.5, 139.8, 138.8, 137.1, 136.9, 129.6, 128.3, 124.4, 117.2	
9	2940, 1695 (CO), 1440, 1095	662 (3.82) 361 (4.51) 349 (4.76) 314 (4.96)	δ9.76 (1H, <i>d</i> , <i>J</i> = 9.3 Hz, H-8) 8.38 (1H, <i>d</i> , <i>J</i> = 4.4 Hz, H-2) 6.76 (1H, <i>t</i> , <i>J</i> = 10.0 Hz, H-6) 7.54–7.46 (2H, <i>m</i> , H-5/H-7) 7.26 (1H, <i>d</i> , <i>J</i> = 4.4 Hz, H-3) 4.47 (2H, <i>s</i> , CH <sub>2</sub> ) 3.94 (3H, CO <sub>2</sub> Me)	δ164.5 (CO), 147.5, 145.4, 142.2, 139.5, 138.4, 134.5, 132.3, 128.1, 119.4, 117.4, 51.4 (Me), 34.0 (CH <sub>2</sub> )	
	2930, 1690 (CO), 1635/1450 (NO <sub>2</sub> ), 1265, 1075	656 (3.79) 363 (4.46) 346 (4.72) 310 (4.90)	δ9.71 (1H, <i>d</i> , <i>J</i> = 9.4 Hz, H-8) 8.32 (1H, <i>d</i> , <i>J</i> = 4.4 Hz, H-2) 7.71 (1H, <i>t</i> , <i>J</i> = 9.9 Hz, H-6) 7.60–7.49 (2H, <i>m</i> , H-5/H-7) 7.30 (1H, <i>d</i> , <i>J</i> = 4.4 Hz, H-3) 4.62 (2H, <i>s</i> , CH <sub>2</sub> ) 3.94 (3H, <i>s</i> , CO <sub>2</sub> Me)	δ165.0 (CO), 148.4, 145.6, 143.0, 140.2, 139.5, 135.7, 132.4, 129.0, 121.0, 117.7, 68.3 (CH <sub>2</sub> ), 51.0 (Me)	
11	3310 (OH), 2915, 1695 (CO), 1465, 1030	657 (2.31) 355 (3.81) 343 (4.31) 318 (4.22)	δ9.79 (1H, <i>d</i> , <i>J</i> = 9.7 Hz, H-8) 8.36 (1H, <i>d</i> , <i>J</i> = 4.4 Hz, H-2) 7.74 (1H, <i>t</i> , <i>J</i> = 9.9 Hz, H-6) 7.71–7.59 (2H, <i>m</i> , H-5/H-7) 7.31 (1H, <i>d</i> , <i>J</i> = 4.4 Hz, H-3) 3.92 (3H, <i>s</i> , CO <sub>2</sub> Me) 3.75 (2H, <i>s</i> , CH <sub>2</sub> ) 3.90 (1H, <i>s</i> , OH)	δ166.5 (CO), 148.6, 143.0, 141.0, 140.9, 139.3, 137.8, 132.7, 130.9, 126.4, 117.5, 62.2 (CH <sub>2</sub> ), 51.0 (Me)	217 ([M + 1] <sup>+</sup> , 8) 216 ([M] <sup>+</sup> , 21) 198 (100)
	2930, 1690 (CO), 1250, 1035	675 (2.81) 411 (4.07) 382 (3.99) 305 (4.68)	δ9.70 (1H, <i>d</i> , <i>J</i> = 9.8 Hz, H-8) 8.41 (1H, <i>d</i> , <i>J</i> = 4.4 Hz, H-2) 7.77 (1H, <i>t</i> , <i>J</i> = 9.8 Hz, H-6) 7.69–7.55 (2H, <i>m</i> , H-5/H-7) 7.29 (1H, <i>d</i> , <i>J</i> = 4.4 Hz, H-3) 3.93 (3H, <i>s</i> , CO <sub>2</sub> Me)	δ167.3 (CO), 149.1, 142.9, 141.7, 141.0, 139.6, 137.3, 133.0, 130.9, 126.7, 117.4,	231 ([M + 1] <sup>+</sup> , 11) 230 ([M] <sup>+</sup> , 16) 186 (100) 128 (27)

Table 1. *Continued*

Compound	IR $\nu_{\text{max}}^{\text{KBr}}$ $\text{cm}^{-1}$	UV $\lambda_{\text{max}}^{\text{CH}_2\text{Cl}_2}$ nm (log $\epsilon$ )	$^1\text{H}$ NMR (400 MHz, $\text{CDCl}_3$ )	$^{13}\text{C}$ NMR (100 MHz, $\text{CDCl}_3$ )	CIMS (probe) 120 eV, $m/z$ (rel. int.)
13	2930, 2720,	599 (2.72)	3.79 (3H, s, OMe)	69.3 ( $\text{CH}_2$ ),	
	1700/1690 (CO),	394 (4.16)	3.51 (2H, s, $\text{CH}_2$ )	51.2, 50.9 (Me)	
	1450, 1030	331 (4.53)	$\delta$ 10.31 (1H, s, CHO)	$\delta$ 205.4 (CO),	214 ( $[\text{M}]^+$ , 21)
		309 (4.61)	9.66 (1H, d, $J=9.2$ Hz, H-8)	166.4 (CO),	186 (43)
			8.19 (1H, d, $J=4.6$ Hz, H-2)	148.2, 141.3,	185 (27)
14			7.75 (1H, t, $J=9.2$ Hz, H-6)	140.8, 139.9,	170 (51)
			7.66–7.51 (2H, m, H-5/H-7)	139.6, 138.0,	159 (31)
			7.31 (1H, d, $J=4.6$ Hz, H-3)	133.2, 130.6,	157 (20)
			3.95 (3H, s, $\text{CO}_2\text{Me}$ )	125.4, 118.1,	156 (14)
				52.4 (Me)	127 (29)
					115 (100) <sup>†</sup>
	2950, 2725,	675 (2.71)	$\delta$ 10.66 (1H, s, CHO)	$\delta$ 195.4 (CO),	185 ( $[\text{M}]^+ + 1$ , 7)
	1645 (CO),	506 (3.42)	8.32 (1H, d, $J=9.4$ Hz, H-8)	148.6, 146.7,	184 ( $[\text{M}]^+$ , 67)
	1465, 1220,	384 (4.45)	8.23 (1H, s, H-2)	139.4, 137.3,	156 (17)
	1050	320 (4.83)	7.62 (1H, t, $J=9.6$ Hz, H-6)	136.5, 136.2,	153 (21)
		312 (4.83)	7.45–7.33 (2H, m, H-5/H-7)	135.9, 127.0,	115 (100)
		306 (4.80)	3.15, 2.65 (3H, s, Me)	125.9, 119.4,	
				24.8 (Me),	
				16.4 (Me)	

\* Full spectroscopic characterization is also given for the known compounds **1**, **2** and **6**.<sup>†</sup> Measured in DMSO as solvent.<sup>‡</sup> EIMS (probe), 70 eV,  $m/z$  (rel. int.).

over RP 18 with 90% aq. MeOH. The minor azulenes eluted in the front, followed by the first main compound **2** and the second main compound **1**, which eluted just before the chlorophyll and the lipids. Compound **1** was removed from the eluent by solid-phase extraction on RP 18, eluted with Me<sub>2</sub>CO and finally purified on a silica gel column with *n*-hexane. Final purification of **2** was achieved by CC on silica gel with *n*-hexane–EtOAc (9:1). The other compounds were separated on a RP 18 column (Lobar B, Merck, Darmstadt) with a gradient of 50% aq. MeOH to pure MeOH in presence of 5% HOAc. Among the 10 collected fractions, **2** and **3** contained the azulenes **7**, **8** ester and **11**, **4** contained **3**, **6**, **13** and **14**. The first compounds were separated by CC on silica gel with *n*-hexane–EtOAc (3:2). With this solvent **11** eluted; after addition of 5% aq. HOAc first **8** ester and later **7** eluted too. Final purification was performed in each case by CC on silica gel with *n*-hexane–EtOAc–HOAc (16:3:1) followed by CC on Sephadex LH-20 with MeCN–5% aq. HOAc (3:2). After evapn to dryness only about half of **8** ester was still soluble in Me<sub>2</sub>CO. The other half could only be dissolved in MeOH or H<sub>2</sub>O. This behaviour indicated a chemical change to a more polar compound, which was later found to be identical with the dicarboxylic acid **8**.

Compound **7** also decomposed partly on the Sephadex LH-20 column to a polar product, later proved to be identical with **8**. Fraction **4** with four compounds was shaken with *n*-hexane to separate the azulenes from any more polar constituents. After evapn to a smaller vol. the *n*-hexane phase was separated on a silica gel column with *n*-hexane–EtOAc–HOAc (17:2:1) resulting in the pure compounds **6** and **13**, **3** and **14** still being in a mixture. Both compounds were finally separated on a silica gel column with *n*-hexane–EtOAc (17:3).

*Calypogeia azurea* from the natural habitat. After cleaning, fresh gametophytes (5 kg) were ground and extracted with Me<sub>2</sub>CO. The extract was evapd to the H<sub>2</sub>O phase and shaken against EtOAc until the H<sub>2</sub>O phase was free of azulenes. The EtOAc phase was evapd to a small vol. and chromatographed by CC over silica gel with a gradient from *n*-hexane up to a mixture of *n*-hexane–EtOAc (4:1). Compound **1** eluted with *n*-hexane and was finally purified by two silica gel columns with *n*-hexane (2300 mg). The oxygen-containing azulenes **2**, **3** and **14** were separated from chlorophyll and the lipids by CC on RP 18 with MeOH–H<sub>2</sub>O–HOAc (18:1:1). The azulene fraction was subsequently chromatographed on a silica gel column with a gradient of *n*-hexane up to *n*-hexane–EtOAc (7:3). With this column **2** was separated from the last 2 compounds **3** and **14**. Final purification of **2** was achieved by CC on silica gel with *n*-hexane–EtOAc (9:1; 28 mg). Compounds **3** and **14** were separated by CC on silica gel with *n*-hexane–EtOAc (17:3), **14** eluting first. Compound **3** was finally purified by CC on RP 18 (Lobar B, Merck, Darmstadt) with MeOH–H<sub>2</sub>O (3:2) (ca 2 mg). Because of the minute amount (<1 mg), **14** had to be purified by micro-prep. HPTLC on RP 18 (Merck, Darmstadt, 10 × 10 cm) with 90% aq. MeOH.

TLC: silica gel; *n*-hexane; *n*-hexane–EtOAc (19:1); 2D-TLC on RP 18 (Merck, Darmstadt, 5 × 5 cm); 1: *n*-hexane–propionic acid (9:1); 2: MeOH–H<sub>2</sub>O–propionic acid (14:5:1). CC: silica gel; 25 μ for MPLC (Merck, Darmstadt); RP 18; 25–40 μ Baker (for flash chromatogr.); for solvents see above. HPLC: 1. prep. column: Spherisorb ODS-2; 5 μ; 250 × 20 mm<sup>2</sup>, flow rate 10 ml min<sup>-1</sup>; UV: 280 nm; solvents see above; 2. separation of isolated azulenes was achieved on Superspher RP-Select B (Merck, Darmstadt); 5 μ; 250 × 4 mm; flow rate 1 ml min<sup>-1</sup>; UV: 280 nm; solvent: linear gradient of 50% to 90% aq. MeOH in 1% phosphoric acid within 20 min, followed by 8 min isocratic flow with the last solvent.

1-Methoxymethyl-4-methylazulene (**2**). The acid **6** (0.50 g,

2.69 mmol) in MeOH (20 ml) is esterified with CH<sub>2</sub>N<sub>2</sub> in Et<sub>2</sub>O {prepared from *N*-nitrosomethyl urea (2.00 g, 19.4 mmol) according to the procedure in [13]}. Usual workup and purification by CC (SiO<sub>2</sub>, eluent Et<sub>2</sub>O) yields the product as violet crystals; 0.48 g **2** (89%), mp 49–51°. Found: C, 78.11; H, 6.09; O, 15.98. C<sub>13</sub>H<sub>12</sub>O<sub>2</sub> (200.2); requires: C, 77.98; H, 6.04; O, 16.12. The spectroscopic data (see Table 1) are in accord with the data reported in ref. [8].

1-Formyl-4-methylazulene (**3**). 4-Methylazulene **5** [14] (1.50 g, 10.5 mmol), dimethylformamide (20 ml) and POCl<sub>3</sub> (1.84 g, 12.0 mmol) were reacted according to the procedure in ref. [15]. After the usual workup and purification by CC (neutral Al<sub>2</sub>O<sub>3</sub>, activity grade I, 150 g; eluent Et<sub>2</sub>O) the product was obtained as deep-red oil, 1.50 g **3** (84%). Found: C, 84.60; H, 5.99; O, 9.29. C<sub>12</sub>H<sub>10</sub>O (170.2); requires: C, 84.68; H, 5.92; O, 9.40. The spectroscopic characterization is given in Table 1.

1-Carboxy-4-methoxymethyl azulene (**4**). The ester **12** (0.25 g, 1.09 mmol) was saponified with KOH (0.55 g, 9.80 mmol) in MeOH–H<sub>2</sub>O (35 ml–35 ml) according to the procedure in ref. [16]. Usual workup yields the product as violet crystals, which are recrystallized from EtOH; 0.22 g **4** (93%), mp 128–129°. Found: C, 72.00; H, 5.40; O, 22.8. C<sub>13</sub>H<sub>12</sub>O<sub>3</sub> (216.2); requires: C, 72.21; H, 5.59; O, 22.20. The spectroscopic characterization is given in Table 1.

1-Carboxy-4-methylazulene (**6**). 1-Trifluoroacetyl-4-methylazulene [8] (1.00 g, 4.20 mmol) was reacted with KOH (2.35 g, 41.9 mmol) in EtOH–H<sub>2</sub>O (15 ml–15 ml) according to the procedure in ref. [8]. Usual workup yields the product as violet needles, which are recrystallized from Et<sub>2</sub>O–*n*-hexane 1:1; 0.69 g **6** (88%) mp 189–190°C. Found: C, 77.21; H, 5.30; O, 17.09. C<sub>21</sub>H<sub>10</sub>O<sub>2</sub> (186.2); requires: C, 77.40; H, 5.41; O, 17.18. The spectroscopic characterization is given in Table 1.

4-Carboxy-1-methoxycarbonylazulene (**7**). The ester **2** (0.35 g, 1.75 mmol) was oxidized by KMnO<sub>4</sub> (0.85 g, 5.38 mmol) in MeOH (50 ml) according to the procedure in ref. [16]. After the usual workup the product was obtained as blue-violet crystals, which are recrystallized from EtOH; 0.24 g **7** (59%), mp 159–161°. Found: C, 67.70; H, 4.30. C<sub>13</sub>H<sub>10</sub>O<sub>4</sub> (230.2); requires: C, 67.82; H, 4.38. The spectroscopic characterization is given in Table 1.

1,4-Dicarboxyazulene (**8**). The monoester **7** (0.20 g, 0.87 mmol) was saponified with KOH (0.50 g, 8.91 mmol) in MeOH–H<sub>2</sub>O (35 ml–35 ml) according to the procedure in ref. [16]. Usual workup gives the product as violet crystals, which are recrystallized from EtOH; 0.16 g **8** (85%), mp 120–122°. Found: C, 66.48; H, 3.63. C<sub>12</sub>H<sub>8</sub>O<sub>4</sub> (216.2); requires: C, 66.67; H, 3.73. The spectroscopic characterization is given in Table 1.

4-Bromomethyl-1-methoxycarbonylazulene (**9**). The ester **2** (1.00 g, 5.00 mmol), *N*-bromosuccinimide (0.92 g, 5.16 mmol) and azoisobutyronitrile (0.20 g) in CCl<sub>4</sub> (120 ml) were reacted according to the procedure in ref. [16]. Usual workup yields the product as black-blue platelets, which are recrystallized from petrol ether (40–60°); 1.02 g **9** (72%), mp 113–114°. Found: C, 55.79; H, 3.90. C<sub>13</sub>H<sub>11</sub>BrO<sub>2</sub> (279.1); requires: C, 55.94; H, 3.97. The spectroscopic characterization is given in Table 1.

1-Methoxycarbonyl-4-nitromethylazulene (**10**). A soln of the bromide **9** (0.88 g, 3.16 mmol) in MeCN (100 ml) was added slowly with stirring to the soln of AgNO<sub>3</sub> (0.56 g, 3.28 mmol) in MeCN (20 ml). After 1 hr at reflux temp. and 2 hr at 20° the precipitate (AgBr) was filtered off and the filtrate evapd to dryness. The product was obtained as black-blue needles, which are recrystallized from EtOH; 0.62 g **10** (75%), mp 138–140°. Found: C, 59.63; H, 4.20; N, 5.28. C<sub>13</sub>H<sub>11</sub>NO<sub>2</sub> (261.2); requires: C, 59.77; H, 4.24; N, 5.36. The spectroscopic characterization is given in Table 1.

4-Hydroxymethyl-1-methoxycarbonylazulene (**11**). To a suspension of zinc powder (0.70 g) in HOAc (30 ml) a soln of **10**

(0.65 g, 2.49 mmol) in HOAc (5 ml) was added at 0° slowly and with vigorous stirring. After 2 hr at 0° the reaction mixture was filtered, the filtrate was diluted with H<sub>2</sub>O (50 ml) and extracted with Et<sub>2</sub>O (× 3, 50 ml). The combined extracts were dried (Na<sub>2</sub>SO<sub>4</sub>), the solvent removed *in vacuo* and the crude product purified by CC (neutral Al<sub>2</sub>O<sub>3</sub>, activity grade I; eluent CH<sub>2</sub>Cl<sub>2</sub>); 0.44 g **11** (81%), dark-blue oil. Found: C, 72.09; H, 5.44; O, 22.01. C<sub>13</sub>H<sub>12</sub>O<sub>3</sub> (216.2); requires: C, 72.21; H, 5.59; O, 22.20. The spectroscopic characterization is given in Table 1.

**1-Methoxycarbonyl-4-methoxymethylazulene (12).** The hydroxymethyl ester **11** (0.40 g, 1.85 mmol) in MeOH (20 ml) was reacted with CH<sub>2</sub>N<sub>2</sub> in Et<sub>2</sub>O {prepared from *N*-nitrosomethylurea (1.00 g, 9.71 mmol) according to the procedure in ref. [13]}. After the usual workup and purification by CC (neutral Al<sub>2</sub>O<sub>3</sub>, activity grade III; eluent CH<sub>2</sub>Cl<sub>2</sub>) the product was obtained as blue-violet oil; 0.26 g **12** (61%). Found: C, 72.88; H, 6.00; O, 20.69. C<sub>14</sub>H<sub>14</sub>O<sub>3</sub> (230.2); requires: C, 73.03; H, 6.13; O, 20.84. The spectroscopic characterization is given in Table 1.

**4-Formyl-1-methoxycarbonylazulene (13).** The hydroxymethyl ester **11** (0.50 g, 2.31 mmol) was oxidized with PCC on Al<sub>2</sub>O<sub>3</sub> (4.00 g ≈ 4.00 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (25 ml) according to the procedure in ref. [16]. After usual workup and purification by CC (neutral Al<sub>2</sub>O<sub>3</sub>, activity grade I; eluent CH<sub>2</sub>Cl<sub>2</sub>) the product was obtained as red-violet oil; 0.37 g **13** (75%). Found: C, 73.00; H, 4.82; O, 22.50. C<sub>13</sub>H<sub>10</sub>O<sub>3</sub> (214.2); requires: C, 72.89; H, 4.71; O, 22.41. The spectroscopic characterization is given in Table 1.

**1,4-Dimethyl-3-formylazulene (14).** 1,4-Dimethylazulene (**1**) (1.00 g, 6.40 mmol), dimethylformamide (20 ml) and POCl<sub>3</sub> (1.00 g, 6.52 mmol) were reacted according to the procedure in ref. [15]. After the usual workup and purification by CC (neutral Al<sub>2</sub>O<sub>3</sub>, activity grade I; eluent Et<sub>2</sub>O) the product was obtained as violet crystals; 0.98 g **14** (83%), mp 73–74°. Found: C, 84.71; H, 6.50. C<sub>13</sub>H<sub>12</sub>O (184.2); requires: C, 84.75; H, 6.57. The spectroscopic characterization is given in Table 1.

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