

## Neolignans and flavonoids from the root bark of *Illicium henryi*

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### ABSTRACT

Two new neolignans (**1** and **2**) were isolated from the root bark of *Illicium henryi*, along with four known neolignans and seven known flavonoids (**3–13**). Their structures were elucidated on the basis of spectroscopic and chemical methods. The absolute configurations of compounds **1** and **2** were determined by the CD spectrum.

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### 1. Introduction

*Illicium henryi* belongs to Family Illiciaceae, and distributes mainly in the south of China. Most of *Illicium* species are considered to be toxic plant. The root bark of *I. henryi* is locally taken to dispel wind and cold and relieve pain in traditional Chinese medicine. Currently, the water-soluble extract from the root bark of *I. henryi* has been applied to clinical use as an analgesic agent by intramuscular injection in China. It has been reported that three sesquiterpene lactones were isolated from the fruits of *I. henryi*, one of them exhibits high toxicity in mice [1]. However, there are no previous reports concerning the phytochemistry of the root bark of *I. henryi* so far. For the purpose of finding analgesic compounds, we carried out systematic studies on the chemical constituents of the root bark of *I. henryi*. Two new neolignans, together with four known lignans, seco-isolariciresinol-*O*- $\alpha$ -*L*-rhamnoside (**3**) [2], 4-*O*-(2-hydroxymethylethyl)-dihydroconigeryl alcohol 6-(4'-hydroxy-3'-methoxyphenyl)-glucoside (**5**) [3], icaraside E4 (**6**)

[4], and massonioside B (**8**) [5] and seven known flavonoids, (2*R*,3*R*) taxifolin 3-*O*- $\beta$ -*D*-xyloside (**4**) [6], (2*S*,3*S*) taxifolin 3-*O*- $\beta$ -*D*-xyloside (**7**) [6], quercetin (**9**) [7], (2*R*,3*R*) catechin (**10**) [8], (2*R*,3*R*) taxifolin 3-*O*- $\beta$ -*D*-glucoside (**11**) [9], (2*S*,3*S*) taxifolin 3-*O*- $\beta$ -*D*-glucoside (**12**) [9], and quercitrin (**13**) [8], have been isolated from the root bark of *I. henryi*. The absolute configuration of **1** and **2** were deduced in the CD spectra.

### 2. Experimental

#### 2.1. General

Optical rotations were measured on a Perkin-Elmer 341 polarimeter. IR data were recorded on a Nicolet FIIIR 750 spectrophotometer. The CD spectra were recorded in MeOH on JASCO J-810 spectropolarimeter. <sup>1</sup>H, <sup>13</sup>C and 2D NMR data spectra were taken with a Bruker AMX 400 instrument. Chemical shifts are recorded in  $\delta$  (ppm). ESIMS were obtained on a Bruker Esquire 3000 Plus Spectrometer. HRESIMS were determined on a Micromass Q-Tof Global mass spectrometer. GC experiments were run on a GC-MS-QP5050A instrument (Shimadzu), using a db-1 column (0.25 mm i.d.  $\times$  30 m; column temperature, 200 °C; injection temperature 250 °C; carrier gas N<sub>2</sub> at flow of 32.2 mL/min; detector, EI-MS). Open column chromatography (CC) was performed with MCI gel

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CHP 20P (high porous polymer, 75–150  $\mu\text{m}$ , Mitsubishi chemical Ind.), silica gel (200–300 mesh, Qingdao Marine Chemical Ltd., China), RP-18 (20–45  $\mu\text{m}$ , Fuji Silysia Chemical Ltd.), and sephadex LH-20 (20–100  $\mu\text{m}$ , Pharmacia). TLC was carried out with GF-254 silica gel plates (Yantai Huiyou Inc., China). All solvents used were of chemical grade and purchased from the Shanghai Chemical plant, Shanghai, PR China.

## 2.2. Plant material

The root bark of *I. henryi* were collected in Kaihua County, Zhejiang Province, PR China in December 2009 and authenticated by Prof. Lihong Hu. A voucher sample of the plant (2009012003) was deposited at the Shanghai Institute of Materia Medica, Chinese Academy of Sciences, Shanghai, PR China.

## 2.3. Extraction and isolation

The air-dried root bark of *I. henryi* (1.0 kg) were extracted three times with 95% EtOH under reflux for 2 h. The extract was concentrated under reduced pressure to obtain a crude extract, which was suspended in  $\text{H}_2\text{O}$ , and then partitioned with petroleum ether, EtOAc and *n*-BuOH, respectively. The EtOAc layer residue (69.0 g) was subjected to silica gel CC with  $\text{CHCl}_3$ –MeOH gradient to give four fractions. Fr. 1 (1.227 g) was repeatedly applied to silica gel CC and eluted with  $\text{CHCl}_3$ –MeOH and further purified by sephadex LH-20 using MeOH to give **1** (15 mg). Fr. 2 (4.415 g) was fractionated on silica gel CC with  $\text{CHCl}_3$ –MeOH to give two subfractions. Fr. 2.1 was subjected to sephadex LH-20 CC using 1:1 MeOH– $\text{H}_2\text{O}$  and further chromatographed on RP-18 with 35:65 MeOH– $\text{H}_2\text{O}$  to yield **2** (11 mg), **3** (98 mg) and **4** (190 mg). The separation of Fr. 2.2 by sephadex LH-20 using 1:1 MeOH– $\text{H}_2\text{O}$  and RP-18 was eluted with 15:85 MeOH– $\text{H}_2\text{O}$  to afford **5** (20 mg) and **6** (13 mg). Fr. 3 (1.321 g) was separated by sephadex LH-20 column using MeOH, silica gel CC with a  $\text{CHCl}_3$ –MeOH gradient and further purified by sephadex LH-20 CC eluted with 1:1 MeOH– $\text{H}_2\text{O}$  to yield **7** (134 mg). Further purification of Fr. 4 (2.074 g) was achieved by sephadex LH-20 using 1:1 MeOH– $\text{H}_2\text{O}$  to give **8** (32 mg) and **9** (221 mg). The crude *n*-BuOH extract (128.6 g) was subjected to MCI CC and eluted with MeOH– $\text{H}_2\text{O}$  gradient to give three fractions, Fr. 5, 6 and 7. Fr. 5 (7.180 g) was successively purified on silica gel with  $\text{CHCl}_3$ –MeOH gradient and RP-18 eluted with 15:85 MeOH– $\text{H}_2\text{O}$  to afford **10** (48 mg). Fr. 6 (14.1 g) was submitted to CC: silica gel, sephadex LH-20 and RP-18 to yield **11** (40 mg) and **12** (15 mg). Fr. 7 (52.5 g) was shown the presence of one main compound signal by TLC analysis. A little portion of Fr. 7 (400 mg) was further purified by sephadex LH-20 and then RP-18 CC to give **13** (55 mg).

## 2.4. Acid hydrolysis of compound 1

Compound **1** (5 mg) was heated at 80 °C in 10% HCl–dioxane (1:1, 1 mL) for 4 h in water bath. The reaction mixtures were neutralized with  $\text{NaHCO}_3$  and then evaporated to dryness. The residue was partitioned between  $\text{CH}_2\text{Cl}_2$  and  $\text{H}_2\text{O}$  for three times. The aqueous phase was evaporated and

passed through a sephadex LH-20 column eluted with MeOH– $\text{H}_2\text{O}$  (1:1) to get a sugar fraction. The sugar portion was detected by TLC with authentic samples.

## 2.5. Determination of sugar components

The absolute configuration of xylose was determined according to a reported procedure [10]. The sugar portion and authentic sugar sample were derivatized with leucine reagent, and the leucine derivatives were subjected to GC (column temperature 200 °C; injection temperature 250 °C; carrier gas  $\text{N}_2$  at flow rate of 32.2 mL/min; derivatives D-xylose: 8.23 min).

Compound **1**, white, amorphous powder;  $[\alpha]_{\text{D}}^{20} + 13$  (c 0.11, MeOH); IR (KBr)  $\nu_{\text{max}}$  3421, 2929, 1602, 1517, 1457, 1274, 1035, 819, and 603  $\text{cm}^{-1}$ ; CD (MeOH)  $\lambda_{\text{max}}$  ( $\Delta\epsilon$ ) 212 (7.58), 235 (–8.75), and 274 (2.69) nm.  $^1\text{H}$  NMR and  $^{13}\text{C}$  NMR data, see Table 1; ESIMS  $m/z$  501.2  $[\text{M} + \text{Na}]^+$ ; HRESIMS  $m/z$  501.1733  $[\text{M} + \text{Na}]^+$  (calcd for  $\text{C}_{24}\text{H}_{30}\text{NaO}_{10}$ , 501.1737).

Compound **2**, white, amorphous powder;  $[\alpha]_{\text{D}}^{20} 0$  (c 0.11, MeOH); IR (KBr)  $\nu_{\text{max}}$  3396, 2927, 2879, 1700, 1608, 1511, 1452, 1263, 1135, and 1031  $\text{cm}^{-1}$ ; CD (MeOH)  $\lambda_{\text{max}}$  ( $\Delta\epsilon$ ) 227 (3.58), 242 (–0.10), and 261 (0.91) nm.  $^1\text{H}$  NMR and  $^{13}\text{C}$  NMR data, see Table 1; ESIMS  $m/z$  443.2  $[\text{M} + \text{Na}]^+$ ; HRESIMS  $m/z$  443.1686  $[\text{M} + \text{Na}]^+$  (calcd for  $\text{C}_{22}\text{H}_{28}\text{NaO}_8$ , 443.1682).

## 3. Results and discussion

Compound **1**, obtained as a white amorphous powder, has a molecular formula of  $\text{C}_{24}\text{H}_{30}\text{O}_{10}$  on the basis of a positive-ion HRESIMS at  $m/z$  501.1733 (calcd. for  $\text{C}_{24}\text{H}_{30}\text{O}_{10}\text{Na}$ ,

**Table 1**

$^1\text{H}$  (400 MHz) and  $^{13}\text{C}$  NMR (100 MHz) data of compounds **1** and **2** (in  $\text{CD}_3\text{OD}$ ).

Position	1		2	
	$\delta_{\text{C}}$	$\delta_{\text{H}}$ (J in Hz)	$\delta_{\text{C}}$	$\delta_{\text{H}}$ (J in Hz)
1	130.1		138.6	
2	112.9	7.12 d (2.1)	119.2	7.08 d (1.5)
3	149.5		152.4	
4	148.6		148.8	
5	116.7	6.87 d (8.4)	111.6	7.10 d (8.1)
6	122.0	6.97 dd (8.4, 2.1)	119.8	6.98 dd (8.1, 1.5)
7	77.7	5.07 d (8.4)	88.7	5.57 d (5.7)
8	79.1	4.16 m	56.4	3.46 m
9	69.8	4.04 dd (11.4, 2.1); 3.39 dd (11.4, 2.1)	65.6	3.75 m; 3.84 m
1'	137.0		137.3	
2'	122.8	6.72 dd (8.1, 2.4)	117.5	6.60 br, s
3'	118.1	6.85 d (8.1)	142.4	
4'	143.3		146.9	
5'	145.5		130.0	
6'	118.2	6.77 d (1.8)	117.1	6.62 br, s
7'	32.8	2.60 t (7.2)	33.1	2.58 t (6.6)
8'	36.0	1.81 m	36.2	1.81 m
9'	62.6	3.58 t (6.6)	62.7	3.58 t (6.6)
1''	106.0	4.08 d (7.8)	83.4	4.24 m
2''	75.3	3.23 m	62.4	3.74 m
3''	78.2	3.28 m	62.4	3.74 m
4''	71.6	3.46 m		
5''	67.4	3.11 m; 3.79 m		
OCH <sub>3</sub>	57.1	3.88 s	56.9	3.81 s

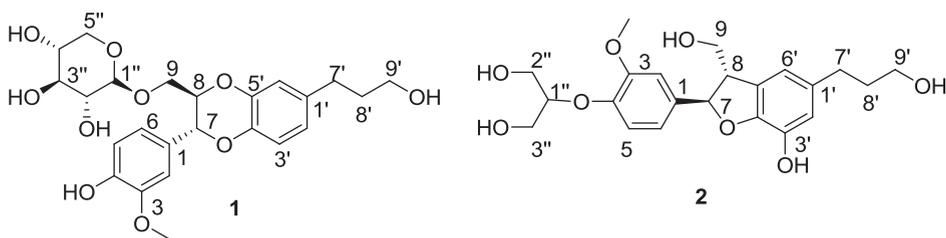


Fig. 1. The key HMBC correlations of compounds 1–2.

501.1737). The  $^1\text{H}$  NMR spectrum exhibited signals of a 1,3,4-trisubstituted aromatic ring [ $\delta$  7.12 (1H, d,  $J=2.1$  Hz, H-2),  $\delta$  6.87 (1H, d,  $J=8.4$  Hz, H-5),  $\delta$  6.97 (1H, dd,  $J=8.4, 2.1$  Hz, H-6)], a 1',4',5'-trisubstituted aromatic ring [ $\delta$  6.72 (1H, dd,  $J=8.1, 2.4$  Hz, H-2'),  $\delta$  6.85 (1H, d,  $J=8.1$  Hz, H-3'),  $\delta$  6.77 (1H, d,  $J=2.4$  Hz, H-6')], a hydroxypropyl group [ $\delta$  2.60 (2H, t,  $J=7.2$  Hz, H<sub>2</sub>-7'),  $\delta$  1.81 (2H, m, CH<sub>2</sub>, H<sub>2</sub>-8'),  $\delta$  3.58 (2H, t,  $J=6.6$  Hz, H<sub>2</sub>-9')], a moiety of (Ph)CH(O)-CH(O)-CH<sub>2</sub>O- [ $\delta$  5.07 (1H, d,  $J=8.4$  Hz, H-7),  $\delta$  4.16 (1H, m, H-8),  $\delta$  4.04 (1H, dd,  $J=11.4, 2.1$  Hz, H-9a),  $\delta$  3.39 (1H, dd,  $J=11.4, 2.1$  Hz, H-9b)], and a methoxyl group [ $\delta$  3.88 (3H, s)]. The carbon signals in the  $^{13}\text{C}$  NMR spectrum of **1** further confirmed the above units. The 1D and 2D NMR spectra showed the presence of a xylopyranosyl moiety. The anomeric proton signal appeared as a doublet at  $\delta$  4.08 (1H,  $J=7.8$  Hz) and permitted assignment of a  $\beta$  configuration of the xylose. On acid hydrolysis and GC analysis, the sugar part was determined to be a  $\beta$ -D-xylopyranosyl group. Significant HMBC correlations were observed between H-6/C-7, H-7/C-6, H-8/C-7, H-8/C-9, and OCH<sub>3</sub>/C-3 (Fig. 1). Thus the skeleton could be concluded as 3-methoxyl-4':7,5':8-diepoxynone-lignan-4,9,9'-triol. The  $\beta$ -D-xylopyranosyl group could be assigned at C-9 from the correlation between H-9 and C-1'' of xyl. This could be further confirmed by a downfield methylene signal at  $\delta$  69.8 (C-9) in the  $^{13}\text{C}$  NMR spectrum. The relative configuration of H-7 and H-8 was trans from the  $J_{7,8}$  value of 8.4 Hz in the  $^1\text{H}$  NMR. The absolute stereochemistries of C-7 and C-8 positions were determined by the CD spectrum (Fig. 2) with the comparison of model compounds. A negative cotton effect at the peak of 235 nm for the benzodioxane moiety in the CD spectrum indicated assignment of 7R,8R absolute configuration for **1** according to the study of a related benzodioxane system [11–14]. Therefore the structure of **1** was elucidated as (7R,8R)-3-

methoxyl-9-O- $\beta$ -D-xylopyranosyl-4':7,5':8-diepoxynone-lignan-4, 9'-diol.

Compound **2** gave the molecular formula C<sub>22</sub>H<sub>28</sub>O<sub>8</sub> base on an [M + Na]<sup>+</sup> ion at 443.1686 in the HRESIMS. The  $^1\text{H}$  NMR spectrum showed signals assigned to a 1,3,4-trisubstituted aromatic ring [ $\delta$  7.10 (1H, d,  $J=8.1$  Hz, H-5),  $\delta$  7.08 (1H, d,  $J=1.5$  Hz, H-2),  $\delta$  6.98 (1H, dd,  $J=8.1, 1.5$  Hz, H-6)] and a 1',3',4',5'-tetrasubstituted aromatic ring [ $\delta$  6.62 (1H, br, s, H-6'),  $\delta$  6.60 (1H, br, s, H-2')], in addition with a methoxyl group [ $\delta$  3.81 (3H, s, OCH<sub>3</sub>)]. The  $^1\text{H}$  NMR data also were in good agreement with a 1,2,3-propantriol moiety [ $\delta$  4.24 (1H, m, H-1''),  $\delta$  3.74 (4H, m, H-2'' and H-3''), a propan-3-ol moiety [ $\delta$  2.58 (2H, t,  $J=6.6$  Hz, H-7'),  $\delta$  1.81 (2H, m, H-8'),  $\delta$  3.58 (2H, t,  $J=6.6$  Hz, H-9')] and a moiety of [(CH(O)-CH(Ph)-CH<sub>2</sub>O)] [ $\delta$  5.57 (1H, d,  $J=5.7$  Hz, H-7),  $\delta$  3.46 (1H, m, H-8),  $\delta$  3.84 (1H, m, H-9a),  $\delta$  3.75 (1H, m, H-9b)]. The assignment of the  $^1\text{H}$  NMR and  $^{13}\text{C}$  NMR signals of **2** was based on its HSQC experiment. These units were assigned by the HMBC correlation from H-7 to C-6, H-6' to C-7', H-1'' to C-4 and OCH<sub>3</sub> to C-3. A trans configuration of **2** was confirmed by the  $J_{7,8}$  value of 5.7 Hz in the  $^1\text{H}$  NMR spectrum. It could be further determined on the basis of NOESY correlations between H-8 and H-2, H-6 protons; H-7 and H-9a protons. It has been reported that the configurations at C-7 and C-8 of the dihydrobenzofuran skeleton can be clearly distinguished from the 240–220 nm region. In the CD spectrum, a positive cotton effect at 227 nm indicated an 8S configuration [4,15,16]. Thus, the structure of **2** was determined to be 7R,8S-configuration as shown in Fig. 3. Therefore the structure of **2** was elucidated as (7R,8S)-3',9,9'-trihydroxyl-3-methoxyl-4-O-glycerol-7,8-dihydrobenzofuran-1'-propanolneoligan.

The other four known neolignans were identified as seco-solariciresinol-O- $\alpha$ -L-rhamnoside (**3**), 4-O-(2-hydroxy-methylethyl)-dihydro-conigeryl alcohol 6-(4'-hydroxy-3'-

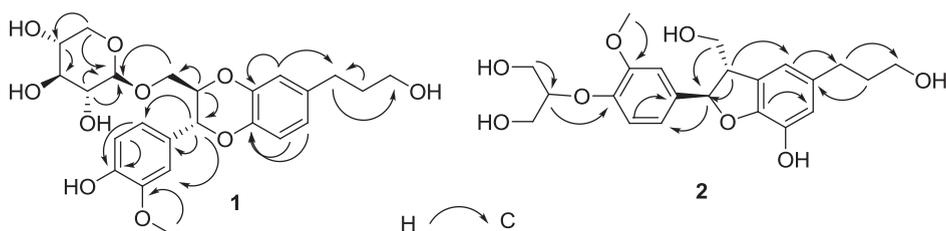


Fig. 2. The CD spectra of compounds 1 and 2.

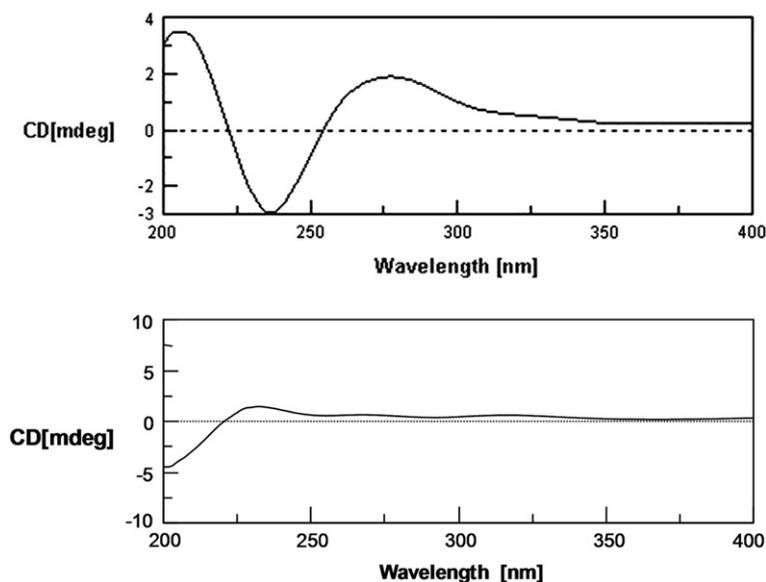


Fig. 3. Structures of compounds 1–2.

methoxyphenyl)-glucoside (**5**), icaraside E4 (**6**), and massonanoside B (**8**). The presence of neolignans was first reported in *I. henryi* and estimated as clue to the classification of genus *Illicium*. The flavonoids taxifolin 3-*O*- $\beta$ -D-xyloside (**4**), (2*S*,3*S*) taxifolin 3-*O*- $\beta$ -D-xyloside (**7**), quercetin (**9**), (2*R*,3*R*) catechin (**10**), (2*R*,3*R*) taxifolin 3-*O*- $\beta$ -D-glucoside (**11**), (2*S*,3*S*) taxifolin 3-*O*- $\beta$ -D-glucoside (**12**) and quercitrin (**13**) were also found in the water-soluble extract which was employed as a remedy against pain by intramuscular injection. Among them quercitrin (**13**) was the most abundant constituent by HPLC analysis. It has been demonstrated that some of flavonoids, such as quercetin, quercitrin and other quercetin glycosides showed pronounced antinociceptive properties [17–20]. The flavonoids isolated from the root bark of *I. henryi* were estimated as clue to analgesic activities, especially quercitrin appears to contribute for the antinociceptive property.

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