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## Efficient Synthesis of 1-Ethyl-2,3,4,6,7,12-hexahydroindolo[2,3- $\alpha$ ] quinolizine: a Key Precursor to Eburnane Alkaloids

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Summary The title compound has been conveniently synthesized from the imine (3) by cyclisation with acrylic acid followed by reduction

RECENTLY the development of new synthetic routes to eburnane alkaloids has attracted interest culminating in several total syntheses of vincamine (1), an excellent cerebral vasodilator drug, and other pharmacologically active eburnamonine-like compounds  $^{\rm 1}$  As outlined in Wenkert's work, 1-ethyl-2,3,4,6,7,12-hexahydroindolo-[2,3-a]quinolizine (2) can serve as a precursor to the eburnane skeleton in some stereo- and enantio-selective approaches 2 We now describe a short and highly efficient synthesis of the 'Wenkert enamine' (2) starting from the easily available compound (3) 3

Utilizing the ability of the imine-enamine (3) to act as an ambident nucleophile (N- vs C-attack),4 we treated (3) with 1.1 equiv of acrylic acid in refluxing anhydrous p-xylene (1.5 h) under N2, to give the presumed intermediate (4) which underwent intramolecular cyclization to give the enamide (5) (91%), m p 235 °C [ $\lambda_{\text{max}}$  (MeOH) 229, 307, and 318 nm,  $\nu_{\rm max}$  (CHCl<sub>3</sub>) 3500, 1665, and 1645 cm<sup>-1</sup>,  $\delta$  (CDCl<sub>3</sub>) 1·26 (t, J 7 Hz, MeCH<sub>2</sub>), 2·24—2 60 (4H, m,  $C_2H_2 + C_3H_2$ ), 2.60 (q, J 7 Hz,  $MeCH_2$ ), 2.86 (t, J 6 Hz,  $C_7H_2$ ), and 4 06 (t, J 6 Hz,  $C_6H_2$ )] † Compound (5) was converted (75%) by L1AlH<sub>4</sub> in tetrahydrofuran (THF) at room temperature into the target molecule [isolated as its perchlorate, mp 175 °C,  $\lambda_{max}$  (MeOH) 207, 244, and 351 nm,  $v_{max}$  (Nujol) 3320 and 1635 cm<sup>-1</sup>] or, more efficiently, by sequential thionation (P4S10, refluxing  $C_6H_6)$  into (6)  $[\lambda_{max}~({\rm MeOH})~212,~248,~253,~{\rm and}$ 303 nm ,  $\delta$  (CDCl3) 1-28 (t, J 7 Hz,  $Me{\rm CH_2}$ ), 2 28 (t, J 7 Hz,  $C_2H_2$ ), 2 98 (t, J 5.5 Hz,  $C_7H_2$ ), 3.04 (t, J 7 Hz,  $C_3H_2$ ), and 4.76 (t, J 5 5 Hz,  $C_6H_2$ )] followed by desulphurisation with acetone-deactivated Raney nickel in MeOH at room temperature (81% yield) The suggested intermediacy of (4) was supported by  $N_b$ -acryloylation of (3) either with acrylic acid-diphenylphosphoryl azide5 [dimethylformamide

(DMF), room temperature] or with acryloyl chloride in the presence of 4-(dimethylamino)pyridine (MeCN, room temperature) 6‡ In both cases we obtained the enamide (5) as the sole isolable product in 95 and 63% yields, respectively

Alternatively, regiospecific C-alkylation of (3) was achieved by refluxing with methyl acrylate in C<sub>6</sub>H<sub>6</sub>-MeOH (1:1) (48 h) to give (7) [ $\nu_{max}$  (CHCl<sub>3</sub>) 3475, 3320, and 1725 cm<sup>-1</sup>,  $\delta$  (CDCl<sub>3</sub>) 0.90 (t, J 7 Hz, MeCH<sub>2</sub>) and 3.73 (s, CO<sub>2</sub>Me), devoid of olefinic protons] which, on further heating (96 h) in p-xylene, gave (5) in 52% overall yield §

These routes constitute a considerable improvement in the synthesis of (2) and appear to be flexible for the preparation of either ring D-functionalised eburnanes or corynantheme-related alkaloids

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- † The same product was recently observed by Le Men amongst the solvolysis products of the 7a-chloroindolenine from 1,12b-dhydro-(5), Y-Y Laronze, J Laronze, D Royer, J Levy, and J Le Men, Bull Soc chim France, 1977 1215
  - ‡ These conditions are known to enhance reactivity at the carbonyl group of the acrylic acid unit
- § By refluxing (3) and methyl acrylate in  $C_6H_6$ -MeOH (1–1) for 96 h Szantay (ref. 3) obtained in 12.6% yield a product to which structure (8) was assigned [m p 242—243 °C,  $\nu_{max}$  (KBr) 3310 and 1668—1620 cm<sup>-1</sup>,  $\delta$  (CDCl<sub>3</sub>) 1.26 (3H t Me), 4.05 (2H t COCH<sub>2</sub>), 6.96—7.58 (4H m aromatic H), and 8.38 (1H s indole NH)] Compound (8) was inefficiently converted into (2) by a harsh Wolff-Compound (8) compound (8) was inefficiently converted into (2) by a harsh Wolff-Compound (8) was inefficiently converted into (2) by a harsh Wolff-Compound (8) was inefficiently converted into (2) by a harsh Wolff-Compound (8) was inefficiently converted into (2) by a harsh Wolff-Compound (8) was inefficiently converted into (2) by a harsh Wolff-Compound (8) was inefficiently converted into (2) by a harsh Wolff-Compound (8) was inefficiently converted into (2) by a harsh Wolff-Compound (8) was inefficiently converted into (2) by a harsh Wolff-Compound (8) was inefficiently converted into (2) by a harsh Wolff-Compound (8) was inefficiently converted into (2) by a harsh Wolff-Compound (8) was inefficiently converted into (2) by a harsh Wolff-Compound (8) was inefficiently converted into (2) by a harsh Wolff-Compound (8) was inefficiently converted into (2) by a harsh Wolff-Compound (8) was inefficiently converted into (2) by a harsh Wolff-Compound (8) was inefficiently converted into (2) by a harsh Wolff-Compound (8) was inefficiently converted into (8) was inefficiently converted in It is especially difficult to check structure (8) in the absence of u v data and on the basis of the reported 4.05 Kishner reduction p p m triplet for COCH2
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  <sup>3</sup> Obtained by melting tryptamine and butyric acid at 190 °C followed by POCl₃-induced cyclization of the resulting N<sub>b</sub>-butyryltryptamine, G Kalaus, P Gyory, L Szabó, and C Szántay, Acta Chim Acad Sci Hung, 1978, 97, 429 (Chem Abs, 1979, 90, 39093)

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<sup>4</sup> Atta-ur-Rahman, J C S Perkin I, 1972, 94, 731

<sup>&</sup>lt;sup>5</sup> T Shioiri, K Ninomiya, and S Yamada, J Amer Chem Soc, 1972, 94, 6203 <sup>6</sup> G Hofle, W Steglich, and H Vorbruggen, Angew Chem Internat Edn., 1978, 17, 569