

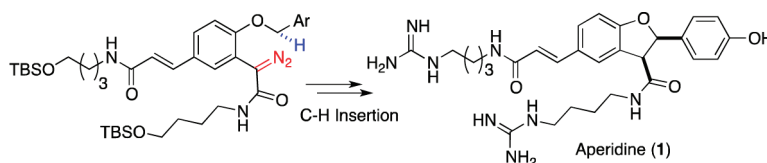
Enantioselective Total Synthesis  
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Received March 17, 2011

## ABSTRACT



An efficient total synthesis of aperidine was accomplished using a Rh-catalyzed C–H insertion of a *cis*-dihydrobenzofuran ring. To circumvent the facile epimerization of the *cis*-dihydrobenzofuran ring, we designed and prepared the C–H insertion precursor diazoamide by Raines' protocol. Finally, the efficient incorporation of a guanidine group and mild deprotection conditions yielded this labile natural product.

Aperidine (**1**)<sup>1</sup> and hordatine A (**2**)<sup>2</sup> were isolated from beer as muscarinic M<sub>3</sub> receptor antagonists (Figure 1). Recently, we reported their absolute structures and described their antagonist activity toward the  $\alpha_1$  adrenoceptor as well.<sup>3</sup> To evaluate their potential as lead compounds for drug development,<sup>4</sup> an ample supply and flexible preparation methods for **1** and **2** are strongly required. In previous studies of the enantioselective total synthesis of the *trans*-isomer **2**,<sup>3</sup> selective formation of the *cis*-isomer **1** was unfortunately prevented by the facile isomerization between these two diastereomers. In general, the higher thermodynamic stability of the *trans* configuration during isomerization renders the *trans*-isomer predominant in neolignans, and the *cis*-isomer is rarely observed in Nature.

Therefore, few reports have described the synthesis of *cis*-dihydrobenzofuran rings, reflecting the scarcity and instability of natural products containing *cis*-dihydrobenzofuran.<sup>5</sup>

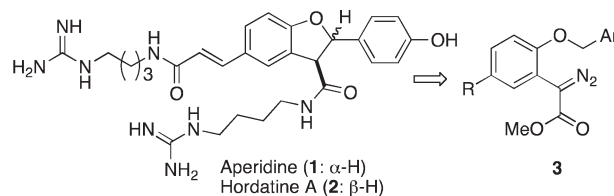


Figure 1. Structure and synthetic strategy for **1**.

Recently, we developed a novel methodology to construct optically active *trans*-dihydrobenzofuran rings via rhodium carbenoid-mediated intramolecular C–H insertion of aryl diazoesters in the presence of a chiral auxiliary.<sup>6</sup> Utilizing this methodology, we accomplished

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(1) (a) Yokoo, Y.; Fujii, W.; Hori, H.; Nagao, K.; Suwa, Y.; Taniyama, K.; Tsuji, K.; Yoshida, T.; Nukaya, H. *Alcohol.: Clin. Exp. Res.* **2004**, *28*, 129S–133S. (b) Yamaji, N.; Yokoo, Y.; Iwashita, T.; Nemoto, A.; Koike, M.; Suwa, Y.; Wakimoto, T.; Tsuji, K.; Nukaya, H. *Alcohol.: Clin. Exp. Res.* **2007**, *31*, S9–S14.

(2) Stoessl, A. *Tetrahedron Lett.* **1966**, *21*, 2287–2292.

(3) Wakimoto, T.; Nitta, M.; Chiba, T.; Yiping, Y.; Tsuji, K.; Kan, T.; Nukaya, H.; Ishiguro, M.; Koike, M.; Yokoo, Y.; Suwa, Y. *Bioorg. Med. Chem. Lett.* **2009**, *19*, 5905–5908.

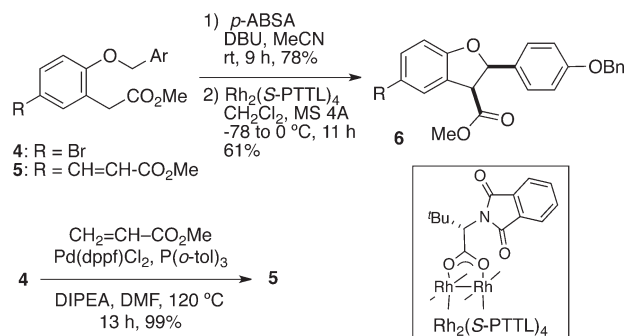
(4) Apers, D.; Vlietinck, A.; Pieters, L. *Phytochem. Rev.* **2003**, *2*, 201–217.

(5) Natori, Y.; Tsutsui, H.; Sato, N.; Nakamura, S.; Nambu, H.; Shiro, M.; Hashimoto, S. *J. Org. Chem.* **2009**, *74*, 4418–4421.

(6) For a review on C–H insertion reactions: (a) Doyle, M. P.; McKervey, M. A.; Ye, T. *Modern Catalytic Methods for Organic Synthesis with Diazo Compounds: From Cyclopropanes to Ylides*; Wiley: New York, 1998. (b) Davies, H. M. L.; Beckwith, R. E. J. *Chem. Rev.* **2003**, *103*, 2861–2904.

the total synthesis of several natural products.<sup>7</sup> The highly *cis*-selective construction from the diazoester itself using  $\text{Rh}_2(\text{S-PTAD})_4$  and  $\text{Rh}_2(\text{S-PTTL})_4$  was reported, respectively, by the Davies<sup>8</sup> and Hashimoto<sup>5</sup> groups. On the basis of these findings, we envisioned that optimization and refinement of the reaction conditions to **3** would provide an optically active *cis*-dihydrobenzofuran ring of **1**, as shown in Scheme 1. Herein, we describe an enantioselective synthesis of (–)-aperidine (**1**) with improvements to the synthetic route employed for **2**.<sup>3</sup>

**Scheme 1.** Construction of *cis*-Dihydrobenzofuran Ring of **6**

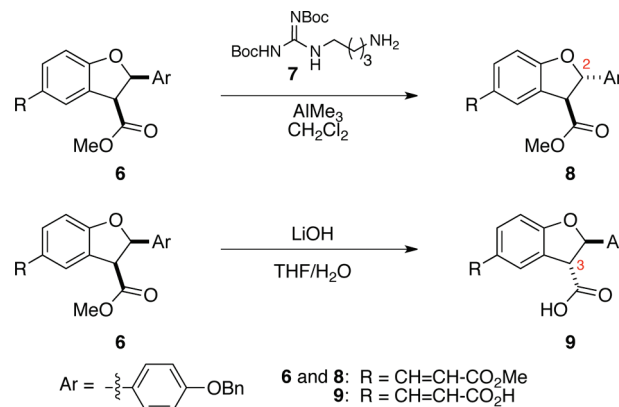


To construct the *cis*-dihydrobenzofuran ring, a precursor of the C–H insertion reaction was prepared from **4** by a previously described method.<sup>7a</sup> The side chain was incorporated using the Heck reaction of **4** with methyl acrylate to provide the *trans*-cinnamic ester derivative **5**. After installing a diazo group by treating with *p*-acetamidobenzenesulfonyl azide (*p*-ABSA) and DBU, a C–H insertion reaction was investigated using several rhodium catalysts and conditions. The rhodium carbenoid-mediated intramolecular C–H insertion reaction proceeded with an increasing *cis* selectivity upon decreasing the temperature. Furthermore, the Hashimoto catalyst  $\text{Rh}_2(\text{S-PTTL})_4$ <sup>9</sup> gave the best results among several catalysts tested to furnish predominantly a *cis*-dihydrobenzofuran ring **6** (Scheme 1).

With the desired dihydrobenzofuran **6** in hand, we turned our attention to incorporation of the agmatine unit **7**. Treatment of **6** with the di-Boc-agmatine **7**<sup>10</sup> and  $\text{AlMe}_3$ , however, resulted in epimerization at the C-2 position prior to the desired amide bond formation to give **8**. This reaction presumably proceeded through the *p*-quinone-methide intermediate, which was promoted by the Lewis acid  $\text{AlMe}_3$ . Next, hydrolysis of **6** by treatment with LiOH

proceeded, accompanied by the concomitant epimerization at the  $\alpha$ -position of the ester group, to give **9** (Scheme 2). Consequently, we were confronted with the formidable task of avoiding competitive epimerization, which generated the *trans* configuration in preference to the conversions required for introduction of the agmatine units. Therefore, construction of the *cis*-dihydrobenzofuran ring in a later stage of the total synthesis, at least after introduction of the side chains, was required.

**Scheme 2.** Epimerization of **6** under Acidic and Basic Conditions



A diazoamide with embedded side chains could potentially ameliorate the disadvantages of the above-described approach through a sequence of conversions after the C–H insertion reaction. However, attempts to promote C–H insertion of the diazoamide possessing protected guanidine groups were unsuccessful. These results suggested that a side chain could be incorporated as a protected butanol amine, i.e., **9** (Scheme 3). After hydrolysis of the diester **5**, condensation of the resultant carboxylic acid with **9** gave the amide **10**. Although the diazo transfer reaction proceeded smoothly in **5**, the same reaction with the amide **10** did not give the desired diazo compound due to the lower  $\text{p}K_a$  of the  $\alpha$  amide group. Recently, Raines and co-workers reported an efficient conversion to a diazo group from an azide functionality using a novel phosphine reagent **13**.<sup>11</sup> Considering its high reactivity and the mild required reaction conditions, this procedure would be suitable for incorporating the diazo group into **10**. After bromination of the ester **10** through the silylketene acetal intermediate, displacement of the bromide with  $\text{NaN}_3$  gave the desired  $\alpha$ -azide amide **12**. Upon treatment of **12** with **13**, phosphine-mediated activation of the azide group and amide bond formation proceeded smoothly to form an acyl triazene intermediate **14**. Without purification of **14**, subsequent transformation by treatment with aqueous  $\text{NaHCO}_3$  successfully furnished the diazoamide **15**.

The desired diazoamide **15** in hand, we then focused on constructing the *cis*-dihydrobenzofuran ring. Upon

(7) (a) Kurosawa, W.; Kan, T.; Fukuyama, T. *Synlett* **2003**, 1028–1030. (b) Kurosawa, W.; Kan, T.; Fukuyama, T. *J. Am. Chem. Soc.* **2003**, *125*, 8112–8113. (c) Koizumi, Y.; Kobayashi, H.; Wakimoto, T.; Furuta, T.; Fukuyama, T.; Kan, T. *J. Am. Chem. Soc.* **2008**, *130*, 16854–16855.

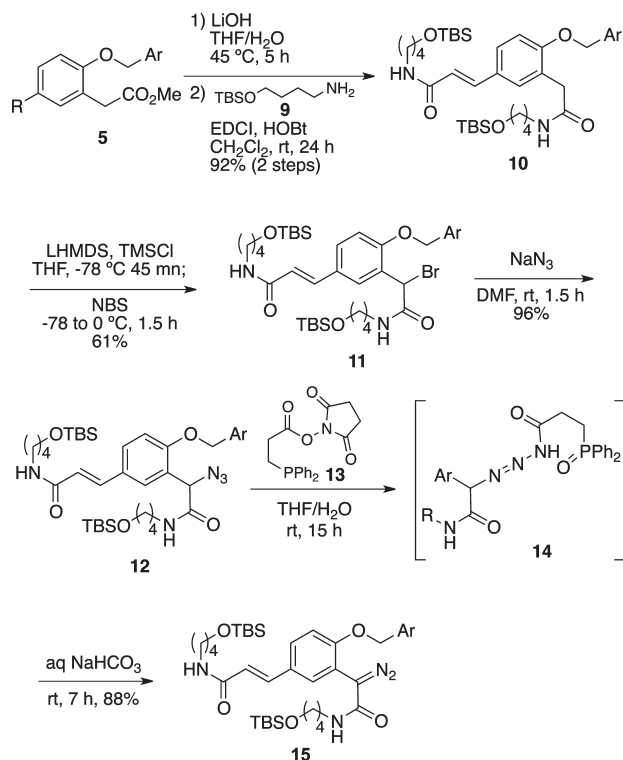
(8) Reddy, R. P.; Lee, G. H.; Davies, H. M. *Org. Lett.* **2006**, *8*, 3437–3440.

(9) Saito, H.; Oishi, H.; Kitagaki, S.; Nakamura, S.; Anada, M.; Hashimoto, S. *Org. Lett.* **2002**, *4*, 3887–3890.

(10) Drake, B.; Patek, M.; Lebl, M. *Synthesis* **1994**, 579–582.

(11) Myers, E. D.; Raines, R. T. *Angew. Chem., Int. Ed.* **2009**, *48*, 2359–2363.

**Scheme 3.** Preparation of C–H Insertion Precursor **15**



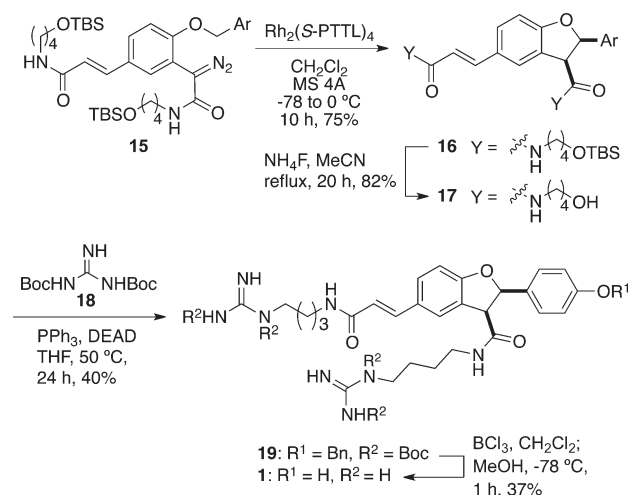
treatment of **15** with 4 mol % Hashimoto catalyst  $\text{Rh}_2(\text{S-PTTL})_4$  ( $\text{S-PTTL})_4$  at  $-78^\circ\text{C}$ , the C–H insertion reaction proceeded smoothly to exclusively afford the *cis*-dihydrobenzofuran **16** in 75% yield in a completely stereoselective manner (Scheme 4). The reaction proceeded even in the presence of the amide group, which could have decreased the activity of the catalyst. Because the labile *cis*-dihydrobenzofuran **16** can readily undergo epimerization under acidic as well as basic conditions, extra mild conditions were required for further conversion. After cleavage of both TBS ethers of **16** by  $\text{NH}_4\text{F}$ , a guanidine moiety was incorporated into **17** by the Mitsunobu reaction.<sup>12</sup> Upon treatment of the di-Boc guanidine **18** in the presence of the corresponding alcohol **17** with DEAD and  $\text{PPh}_3$ , smooth amination proceeded to provide the protected aperidine **19**.

After numerous attempts to circumvent the facile epimerization during the final deprotection step, we found that simultaneous cleavage of the Bn ether and Boc groups of **19** could be performed by treatment with  $\text{BCl}_3$ <sup>13</sup> at  $-78^\circ\text{C}$ , and careful workup provided **1**. Presumably, Lewis acid-mediated cleavage of the Bn ether proceeded by  $\text{BCl}_3$ , and HCl-mediated hydrolysis of the Boc group proceeded after addition of MeOH to the reaction mixture. After removal of the solvent and resultant  $\text{B}(\text{OMe})_3$ ,

(12) (a) Mitsunobu, O. *Synthesis* **1981**, 1–28. (b) Hughes, D. L. *Org. React.* **1992**, 42, 335–656.

(13) Williams, D. R.; Brown, D. L.; Benbow, J. W. *J. Am. Chem. Soc.* **1989**, 111, 1923–1925.

**Scheme 4.** Completion of the Total Synthesis of **1**



purification was performed by the sequential combination of gel filtration and HPLC separation.<sup>14</sup> All spectral data of **1** ( $^1\text{H}$ ,  $^{13}\text{C}$  NMR, IR, and HRMS) were identical to those of the natural product, and the optical rotation,  $[\alpha]_D^{25} = -54$  ( $c$  0.14, MeOH), agreed well with reported value of previously the (2*R*,3*S*)-**1**,  $[\alpha]_D^{25} = -57$  ( $c$  0.2, MeOH).

In conclusion, an enantioselective total synthesis of aperidine (**1**) was accomplished in 10 steps from the readily available arylacetic acid derivative **4**. Our stereoselective synthesis featured the *cis*-selective construction of a dihydrobenzofuran ring utilizing a Rh-catalyzed C–H insertion of the diazoamide **15**, which was efficiently synthesized by Raines' protocol. After incorporation of a guanidine group by the Mitsunobu reaction, all protecting groups were efficiently removed by treatment with  $\text{BCl}_3$  at low temperatures, even in the presence of an unstable *cis*-dihydrobenzofuran ring. The final deprotection step did not proceed with noticeable epimerization, suggesting that electrostatic interactions between the embedded guanidine and phenol of **1** may play a critical role in maintaining the *cis*-configuration despite an inherent preference for the epimer and generation of the *trans*-isomer **2**.<sup>3</sup> Thus, we achieved the first asymmetric synthesis of **1** using careful manipulations.

**Acknowledgment.** This work was financially supported in part by the Uehara Memorial Foundation and by a Grant-in-Aid for Scientific Research on Priority from the Ministry of Education, Culture, Sports, Science and Technology (MEXT), Japan.

**Supporting Information Available.** Detailed experimental procedures and spectroscopic data. This material is available free of charge via the Internet at <http://pubs.acs.org>.

(14) Detailed experimental procedures and spectral data are described in the Supporting Information.