

## Note

### Synthesis of *N*-(*L*-glutam-5-oyl)- $\alpha$ -D-glucopyranosylamine and *O*- $\alpha$ -D-glucopyranosyl-(1 $\rightarrow$ 6)-*O*- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 6)-*N*-(*L*-glutam-5-oyl)- $\alpha$ -D-glucopyranosylamine\*

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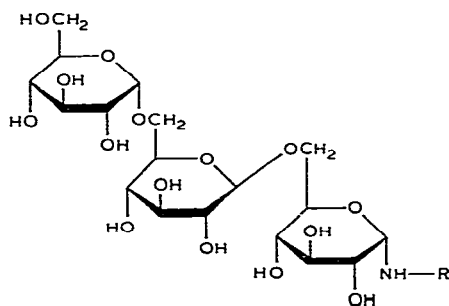
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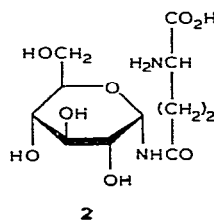
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Shibata *et al.*<sup>1</sup> isolated and purified from the glomerular basement membrane of rats a new glycopeptide that has activity for the induction of glomerulonephritis in homologous animals<sup>2</sup>. From a consideration of <sup>13</sup>C-n.m.r. data in comparison with those of related synthetic glycosylamine derivatives<sup>3</sup>, Shibata and Nakanishi<sup>4</sup> proposed **1** as the structure of the nephritogenoside. This is the first example of an *N*- $\alpha$ -D-glycosyl linkage in a natural D-glucose-containing glycopeptide (or glycoprotein)<sup>5</sup>.

In a previous paper<sup>6</sup>, we reported the synthesis of an *N*-glycosyl linkage between the trisaccharide glycosylamine, *O*- $\alpha$ -D-glucopyranosyl-(1 $\rightarrow$ 6)-*O*- $\beta$ -D-glucopyrano-



**1** R = Peptide



\*The nephritogenic glycopeptide from rat glomerular-basement membrane. Part 3.

syl-(1→6)-*O*- $\alpha$ -D-glucopyranosylamine and L-aspartic acid. As it is not clear whether the amino acid linked to the glycosylamine residue by an *N*-glycosyl linkage is an L-aspartic acid or L-glutamic acid residue, we synthesized *N*-(L-glutam-5-oyl)- $\alpha$ -D-glucopyranosylamine (**2**) and *O*- $\alpha$ -D-glucopyranosyl-(1→6)-*O*- $\beta$ -D-glucopyranosyl-(1→6)-*N*-(L-glutam-5-oyl)- $\alpha$ -D-glucopyranosylamine (**3**) as models of corresponding derivatives possibly present in the glomerular basement-membrane of rats, and studied the  $^{13}\text{C}$ -n.m.r. spectra of these model compounds.

2,3,4,6-Tetra-*O*-acetyl- $\alpha$ -D-glucopyranosylamine (**4**) was obtained by the procedure described in a previous paper<sup>3</sup>. 1-Ethyl *N*-(benzyloxycarbonyl)-L-glutamate, readily prepared in high yield by the procedure of Wünsch and Zwick<sup>7</sup> from *N*-(benzyloxycarbonyl)-L-glutamic acid, was condensed with **4**, by use of the diethyl phosphorocyanidate method of Shioiri *et al.*<sup>8</sup>, to give, in 82% yield, the *N*-(L-glutam-5-oyl)-D-glucopyranosylamine derivative **5**. Removal of the benzyloxycarbonyl group by catalytic hydrogenation afforded 2,3,4,6-tetra-*O*-acetyl-*N*-(1-ethyl-L-glutam-5-oyl)- $\alpha$ -D-glucopyranosylamine (**6**), which was *O*-deacetylated and

TABLE I

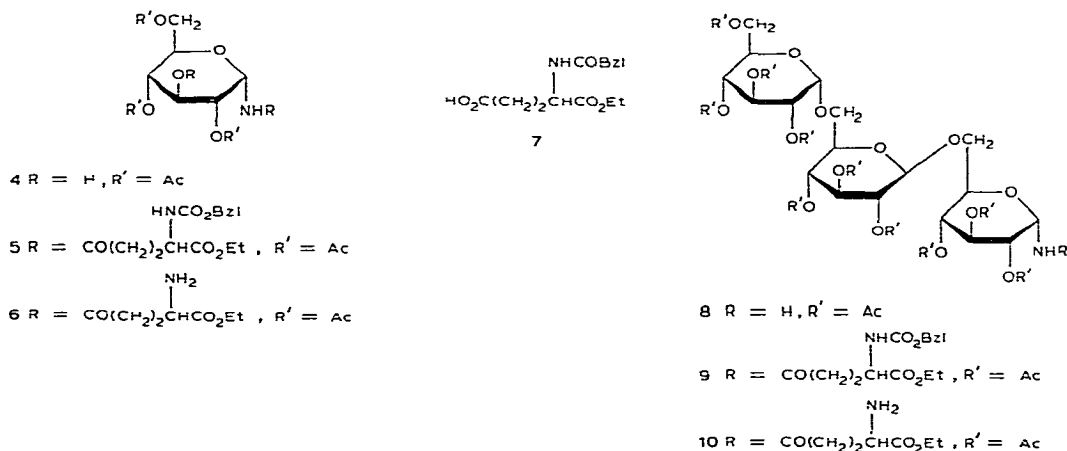
$^{13}\text{C}$ -N.M.R. DATA ( $\delta$ ) FOR COMPOUNDS **1-3**, **5**, **6**, **9**, AND **10**<sup>a</sup>

Carbon atom	Compound						
	1	2	3	5	6	9	10
<i>Sugar residues</i>							
C-1	77.6	77.4	77.7	73.9	73.3	73.5	73.4
C-2	70.8	70.3	70.5	68.5	68.3	69.1	68.9
C-3	72.4	73.5	72.7	69.2	68.3	70.0	69.5
C-4	70.8	70.3	70.5	68.2	68.3	67.4	67.8
C-5	74.1	73.9	73.0	70.3	68.3	70.7	70.4
C-6	70.4	61.4	69.5	61.9	62.0	67.4	67.2
C-1'	104.1		103.8			99.9	100.1
C-2'	74.8		74.2			71.3	71.8
C-3'	77.2		77.0			72.8	73.1
C-4'	71.2		70.6			68.6	68.9
C-5'	77.2		75.4			72.8	73.0
C-6'	69.6		66.7			67.4	67.4
C-1''	100.2		99.0			96.1	96.3
C-2''	72.4		73.0			69.1	68.9
C-3''	75.5		74.2			70.7	71.0
C-4''	71.6		70.7			66.5	67.2
C-5''	74.1		74.2			69.8	70.4
C-6''	62.6		61.7			61.9	62.2
<i>Amino acid residue</i>							
C-2		59.2	59.5	52.5	55.6	53.3	55.5
C-3		26.2	26.5	27.3	24.8	27.4	24.9
C-4		30.6	30.9	30.0	29.3	29.9	29.1

<sup>a</sup>For solutions of **1-3** in  $\text{D}_2\text{O}$ , and for solutions of **5**, **6**, **9**, and **10** in chloroform-*d*.

deethylated with triethylamine in 50% methanolic solution at room temperature to give *N*-(L-glutam-5-oyl)- $\alpha$ -D-glucopyranosylamine (2).

Similarly, condensation of *O*-(2,3,4,6-tetra-*O*-acetyl- $\alpha$ -D-glucopyranosyl)-(1 $\rightarrow$ 6)-*O*-(2,3,4-tri-*O*-acetyl- $\beta$ -D-glucopyranosyl)-(1 $\rightarrow$ 6)-2,3,4-tri-*O*-acetyl- $\alpha$ -D-glucopyranosylamine<sup>6</sup> (8) with 1-ethyl *N*-(benzyloxycarbonyl)-L-glutamate gave, in 47% yield, *O*-(2,3,4,6-tetra-*O*-acetyl- $\alpha$ -D-glucopyranosyl)-(1 $\rightarrow$ 6)-*O*-(2,3,4-tri-*O*-acetyl- $\beta$ -D-glucopyranosyl)-(1 $\rightarrow$ 6)-2,3,4-tri-*O*-acetyl-*N*-(1-ethyl *N*-benzyloxycarbonyl-L-glutam-5-oyl)- $\alpha$ -D-glucopyranosylamine (9), the <sup>1</sup>H-n.m.r. spectrum of which showed signals for a benzyloxy group, 10 acetyl groups, and an ethoxymethyl group. The <sup>13</sup>C-n.m.r. data showed signals for three anomeric carbon atoms, and three signals for C-2, -3, and -4 of the glutamic acid residue (see Table I). Removal of the benzyloxycarbonyl group of 9 by catalytic hydrogenation afforded *O*-(2,3,4,6-tetra-*O*-acetyl- $\alpha$ -D-glucopyranosyl)-(1 $\rightarrow$ 6)-*O*-(2,3,4-tri-*O*-acetyl- $\beta$ -D-glucopyranosyl)-(1 $\rightarrow$ 6)-2,3,4-tri-*O*-acetyl-*N*-(1-ethyl L-glutam-5-oyl)- $\alpha$ -D-glucopyranosylamine (10). Subsequently, 10 was *O*-deacetylated and deethylated with triethylamine in 50% methanolic solution at room temperature to give the target compound, *O*- $\alpha$ -D-glucopyranosyl-(1 $\rightarrow$ 6)-*O*- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 6)-*N*-(L-glutam-5-oyl)- $\alpha$ -D-glucopyranosylamine (3), the <sup>1</sup>H-n.m.r. data of which show signals for three anomeric protons.



All the compounds described gave elemental analyses in agreement with assigned structures, and no anomeric impurities were observed by <sup>1</sup>H- and <sup>13</sup>C-n.m.r. spectroscopy. The <sup>13</sup>C shifts of the glycosylamine derivatives and related compounds are listed in Table I. The data for 3, 9, and 10 are essentially the same as those for the corresponding aspartic acid derivatives<sup>6</sup>, except for the amino acid residue. The <sup>13</sup>C-n.m.r. spectra indicated that the chemical shifts of the sugar residues of 1 and 3 are very similar.

## EXPERIMENTAL

*General methods.* — Melting points were determined with a Yanagimoto microapparatus and are uncorrected.  $^1\text{H}$ -N.m.r. spectra were recorded with a JNM MH-100 spectrometer, and  $^{13}\text{C}$ -n.m.r. spectra with a FX-100 instrument, tetramethylsilane being the internal standard in both cases. Optical rotations were recorded with a Union Giken PM-201 automatic digital polarimeter. T.l.c. was conducted on precoated silica gel plates (Merck GF-254), and column chromatography on silica gel (Merck Kieselgel 60).

*2,3,4,6-Tetra-O-acetyl-N-(1-ethyl N-benzyloxycarbonyl-L-glutam-5-oyl)- $\alpha$ -D-glucopyranosylamine (5).* — To a solution of **4** (ref. 3, 532 mg) in oxolane (19 mL) were added 1-ethyl *N*-(benzyloxycarbonyl)-L-glutamate (455 mg), diethyl phosphorocyanidate (524 mg), and triethylamine (0.54 mL). The mixture was stirred for 12 h at  $0^\circ$ , and then diluted with ethyl acetate (100 mL), and successively washed with 5% hydrogen chloride, water, saturated sodium hydrogencarbonate, and saturated sodium chloride. Drying, followed by evaporation, gave a syrup, which was chromatographed on silica gel with 4:1 (v/v) chloroform-methanol. Evaporation of the eluate gave a pure syrup (800 mg, 82%),  $[\alpha]_{\text{D}}^{20} +42.4^\circ$  (*c* 5.6, chloroform);  $^1\text{H}$ -n.m.r. ( $\text{CDCl}_3$ ):  $\delta$  7.38 (m, 5 H, arom), 2.02–2.00 (12 H, 4 OAc), and 1.32 (t, 3 H,  $\text{CH}_2\text{CH}_3$ ).

*Anal.* Calc. for  $\text{C}_{29}\text{H}_{38}\text{N}_2\text{O}_{14}$ : C, 54.54; H, 6.00; N, 4.39. Found: C, 54.49; H, 5.98; N, 4.25.

*2,3,4,6-Tetra-O-acetyl-N-(1-ethyl L-glutam-5-oyl)- $\alpha$ -D-glucopyranosylamine (6).* — A solution of **5** (90 mg) in oxolane (8 mL) was hydrogenated in the presence of platinum dioxide (30 mg) for 12 h at room temperature. The catalyst was filtered off, and the filtrate evaporated to dryness to give **6** (66.1 mg, 93%) syrup  $[\alpha]_{\text{D}}^{22} +64.1^\circ$  (*c* 0.77, chloroform).

*Anal.* Calc. for  $\text{C}_{21}\text{H}_{32}\text{N}_2\text{O}_{12}$ : C, 50.00; H, 6.39; N, 5.55. Found: C, 50.12; H, 6.33; N, 5.50.

*N-(L-Glutam-5-oyl)- $\alpha$ -D-glucopyranosylamine (2).* — To a solution of **6** (25 mg) in 50% methanol (1 mL) was added triethylamine (0.24 mL). The mixture was kept overnight at room temperature, and then evaporated *in vacuo* to give a syrup (15 mg, 98%),  $[\alpha]_{\text{D}}^{22} +53.1^\circ$  (*c* 0.64, water);  $^1\text{H}$ -n.m.r. ( $\text{D}_2\text{O}$ ):  $\delta$  5.52 (d, 1 H, *J* 4 Hz, H-1).

*Anal.* Calc. for  $\text{C}_{11}\text{H}_{20}\text{N}_2\text{O}_8$ : C, 42.86; H, 6.54; N, 9.09. Found: C, 42.72; H, 6.49; N, 9.13.

*O-(2,3,4,6-Tetra-O-acetyl- $\alpha$ -D-glucopyranosyl)-(1 $\rightarrow$ 6)-O-(2,3,4-tri-O-acetyl- $\beta$ -D-glucopyranosyl)-(1 $\rightarrow$ 6)-2,3,4-tri-O-acetyl-N-(1-ethyl N-benzyloxycarbonyl-L-glutam-5-oyl)- $\alpha$ -D-glucopyranosylamine (9).* — To a solution of **8** (ref. 6, 204 mg) in oxolane (6 mL) were added 1-ethyl *N*-(benzyloxycarbonyl)-L-glutamate (90 mg), diethyl phosphorocyanidate (66 mg), and triethylamine (0.18 mL). The mixture was stirred for 40 h at  $0^\circ$ , and then diluted with ethyl acetate (100 mL), and successively washed with 5% hydrogen chloride, water, saturated sodium hydrogencarbonate, and saturated sodium chloride. Drying followed by evaporation gave a syrup, which was chromatographed on silica gel with 3:2 (v/v) benzene-acetone. The eluate was

evaporated to dryness to give **9** (126 mg, 47%),  $[\alpha]_D^{20} + 63.8^\circ$  (*c* 7.26, chloroform);  $^1\text{H-n.m.r.}$  ( $\text{CDCl}_3$ ):  $\delta$  7.36 (5 H, arom). 2.08–2.00 (30 H, 10 OAc), and 1.26 (t, 3 H,  $\text{CH}_2\text{CH}_3$ ).

*Anal.* Calc. for  $\text{C}_{53}\text{H}_{70}\text{N}_2\text{O}_{30}$ : C, 52.39; H, 5.81; N, 2.31. Found: C, 52.44; H, 5.89; N, 2.36.

O-(2,3,4,6-Tetra-O-acetyl- $\alpha$ -D-glucopyranosyl)-(1 $\rightarrow$ 6)-O-(2,3,4-tri-O-acetyl- $\beta$ -D-glucopyranosyl)-(1 $\rightarrow$ 6)-2,3,4-tri-O-acetyl-N-(1-ethyl L-glutam-5-oyl)- $\alpha$ -D-glucopyranosylamine (**10**). — A solution of **9** (50 mg) in oxolane (5 mL) was hydrogenated in the presence of platinum dioxide (50 mg) for 3 h at room temperature. The catalyst was filtered off, and the filtrate was evaporated to dryness to give **10** (42 mg, 94.5%),  $[\alpha]_D^{22} + 74.0^\circ$  (*c* 0.5, chloroform); t.l.c. (10:1, v/v, chloroform–methanol):  $R_F$  0.51.

O- $\alpha$ -D-Glucopyranosyl-(1 $\rightarrow$ 6)-O- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 6)-N-(L-glutam-5-oyl)- $\alpha$ -D-glucopyranosylamine (**3**). — To a solution of **10** (38 mg) in 50% methanolic solution (2 mL) was added triethylamine (0.18 mL). The mixture was kept overnight at room temperature, and then evaporated *in vacuo* to give a powder (20 mg, 89.9%),  $[\alpha]_D^{22} + 47.6^\circ$  (*c* 1.68, water); t.l.c. (2:4:1, v/v, 1-butanol–acetic acid–water):  $R_F$  0.32;  $^1\text{H-n.m.r.}$  ( $\text{D}_2\text{O}$ ):  $\delta$  5.56 (d, *J* 4 Hz, H-1), 4.95 (d, *J* 3.5 Hz, H-1''), and 4.50 (d, *J* 8 Hz, H-1').

*Anal.* Calc. for  $\text{C}_{23}\text{H}_{40}\text{N}_2\text{O}_{18}$ : C, 43.67; H, 6.37; N, 4.43. Found: C, 43.72; H, 6.42; N, 4.49.

#### ACKNOWLEDGMENT

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