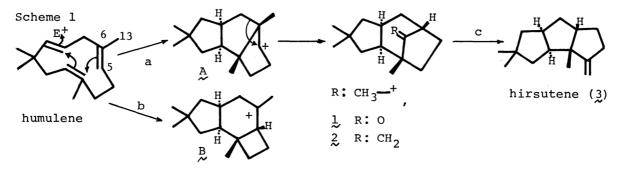
A CONVERSION OF HUMULENE TO HIRSUTENE

Shunjiro MISUMI, Hiroshi MATSUSHIMA, Haruhisa SHIRAHAMA, and Takeshi MATSUMOTO*

Department of Chemistry, Faculty of Science, Hokkaido University, Sapporo 060

Humulene has been transformed into 1,4,4-trimethyltricyclo- $[6.2.1.0^2,6]$ undecan-ll-one, a key intermediate for the synthesis of hirsutene, through a process which simulates the biosynthetic pathway assumed for the latter sesquiterpene.

In the previous paper 1) we reported chemical synthesis of tricyclic compound 2 and its rearrangement to hirsutene 3 through a biogenetic-like path (Scheme 1, c) 2). Compound 2 in turn was prepared from its norketone 1. In another paper we showed 3) that humulene can be converted to protoilludyl cation B under suitable conditions, but not to secondary cation A, presumably because of the presence of the 13-methyl group. It was therefore anticipated that a properly substituted 13-norhumulene would give an A type cation rather than a B type cation. We should like to describe herein transformation of humulene into the norketone 1 through 13-nor-cation of A. The transformation means the success of the first, even if not direct, synthesis of tricyclic hirsutene from its monocyclic biogenetic precursor.

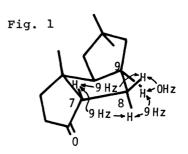


A 3,6-secoprotoilludene derivative 4, which was previously derived from humulene by us, 3) was first converted to 5^4) (0s0₄-NaIO₄/^tBuOH-H₂O/rt/2 hr, 95%). The ketone was then reduced to a 1:1 mixture of epimeric alcohols, 6 and 7 (DIBAH/THF/-78°/10 min, almost quantitative) which was separated by passing through a silica gel column [6^4 : mp 108-110 °C⁵; ir(CCl₄, 10⁻³ mol) 3610 cm⁻¹, 7^4 : oil; ir(CCl₄, 10⁻³ mol) 3515 cm⁻¹]. Configuration of the hydroxyl group of each

isomer was determined by the presence of an intramolecular H-bond in the ir spectrum of 7. Treatment of 6 with CBr_4/ϕ_3P gave β -bromide 8^{4} (ϕ H/reflux/5 min, 95%) (mp 91-2°). The β -configuration of the bromine atom was deduced from resembrance of the nmr coupling constants of the proton at C-6 (dt, 11, 4 Hz) to those of the C-6 proton of 6 (dt 10, 5 Hz). On the other hand, α -ol 7 gave exclusively olefin 9^{4} on the similar treatment (95%). Cleavage of the ether linkage of $\frac{8}{8}$ (Zn-EtOH/reflux/12 hr) afforded $\frac{10}{20}$ (90%). The nmr spectrum of $\frac{10}{20}$ in the presence of Eu $^{3+}$ [Eu(fod) $_3/10=0.51$, CCl $_4$] showed two double triplets at $^\delta$ 7.75(J=6, 10 Hz, 6-H) and at 7.25(J=7, 10 Hz, 7-H). Therefore the configuration of the double bond was assigned to be $\underline{\text{cis}}$ on the basis of the coupling constant 10 Hz. The cyclooctenol 10 was heated to reflux in HCO₂H-Ac₂O for 30 min under argon to give rise to a mixture of formates, 11 and 12 (80%, 3:1(glc)). The formates were treated as such with LiAlH, (THF/reflux/10 min, nearly quantitative) and the product was next oxidized with Jones reagent (0°/15 min, 95%) to yield a mixture of ketones (13^4) and $14^4)$, 3:1(glc)), which was separated into two isomers by silica gel chromatography (13:ir 1745 cm⁻¹; nmr δ 0.92(3H, s), 1.06(6H, s), 14: ir 1740 cm⁻¹, nmr δ 0.99(3H, d, 7 Hz), 1.02, 1.10(each 3H, s)].

Structure of 13 was revealed by nmr decoupling experiments in the presence of a shift reagent [Eu(fod) $_3/13=0.3$, CCl $_4$, δ 7.01(S=11.87, lH, d, J $_{4\alpha-5\alpha}=8$ Hz, $5\alpha H$), 6.97(S=11.17, lH, dd, $J_{4\beta-5\beta}=8$ Hz, $J_{4\alpha-5\beta}=4$ Hz, $5\beta H$), 6.55(S=11.17, lH, t, $J_{7\beta-8\alpha}=J_{7\beta-8\beta}=9$ Hz, $7_{\beta H}$), 4.85(S=7.85, lH, dt, $J_{7\beta-8\alpha}=J_{8\alpha-9\alpha}=9$ Hz, $J_{8\beta-8\alpha}=13$ Hz, $8\alpha H$), 3.95(S=2.61, 1H, m, $9\alpha H$), 3.64(S=5.48, 1H, dd, $J_{8\alpha-8\beta}=13$ Hz, $J_{7\beta-8\beta}=9$ Hz, .8 β H), 3.20(S=3.91, 1H, ddd, J_{4 α -4 β}=12 Hz, J_{4 β -5 α}=5 Hz, J_{4 β -5 β}=8 Hz, 4 β H), 2.28(S= 3.22, 3H, s, 12Me), 2.25(S=1.00, 1H, dd, $J_{9\alpha-10\alpha}$ =7 Hz, $J_{10\alpha-10\beta}$ =13 Hz, 10 α H), 2.05 (S=1.20, 2H, d, $J_{1\alpha,\beta-2\alpha} = 9$ Hz, $1\alpha,\beta$ H), 1.82(S=1.83, 1H, dd, $J_{9\alpha-10\beta} = 8$ Hz, J =13 Hz, 10 βH). Moreover, the large S value of 9 αH in camparison with that of 86H and the observed J values($J_{7\beta-8\alpha}=J_{7\beta-8\beta}=J_{8\alpha-9\alpha}=9$ Hz, $J_{8\beta-9\alpha}=0$ Hz) indicated a cis stereochemistry as depicted in Fig. 1 for this ketone. On the other hand, structure of 14 was determined by leading it to the known compound, pentalenene, 15^{7} [(1) MeMgI/ether/reflux/2 hr, 90%, (2) MsCl/py/rt/12 hr, 90%]. Reduction of 13 (NaBH₄/ether/0°/10 min) furnished a pair of epimeric alcohols 164 (quantitative) which in turn was treated with MsCl-py (100°/30 min) to give epimeric chlorides (34%) and two olefins 17 and 18 (42%, 9:1, glc). The chlorides were separated from the olefins by column chromatography(SiO2)(19: nmr

 δ 0.92(3H, s), 1.08(6H, s), 3.86 and 4.28(1H, each m), 40%) and were further dehalogenated (^tAmoNa/DMSO/80°/8 hr) to afford a similar mixture of the olefins, 17 and 18 (9:1(glc)). Since the olefins could not be separated by the usual methods, they



were treated as such with $T1(C1O_4)_3^{6}$ ($^tBuOH-H_2O(1:1)/rt/3$ hr) to give two ketones (53%, 20/1 =5/1, silica gel column, 20: ir 1745 cm⁻¹, nmr δ 0.91, 1.02, 1.07(each 3H, s), 1: ir 1750 cm⁻¹; nmr δ 0.90, 0.98, 1.10(each 3H, s)). Stereostructure of 20 was deduced from the decoupling experiments of the nmr spectrum in the presence

of Eu³⁺: (Eu(fod) $_3/20$ =0.35, (CCl $_4$) δ 8.22(S=15.40, 1H, dd, $J_{6\alpha-6\beta}$ =19 Hz, $J_{6\beta-7\beta}$ =6 Hz), 6βH), 7.82(S=15.00, 1H, dd, $J_{6\alpha-6\beta}$ =19 Hz, $J_{6\alpha-7\beta}$ =2.5 Hz, 6αH), 7.80 and 7.68 (S=15.20 and 14.60, 2H, ABq, $J_{4\alpha-4\beta}$ =19 Hz, 4α and 4βH), 4.40(S=4.52, 1H, m, 7βH), 4.10(S=3.12, 2H, m, 9α, 8αH), 4.00(S=5.04, 1H, dt, $J_{1\alpha-2\alpha}$ =14 Hz, $J_{1\beta-2\alpha}$ = $J_{2\alpha-9\alpha}$ =9 Hz, 2αH), 3.20(S=2.40, 1H, dd, $J_{8\alpha-8\beta}$ =12 Hz, $J_{7\beta-8\beta}$ =8 Hz, 8βH), 2.92(S=4.20, 3H, s, 12Me), 1.72(S=0.92, 3H, s, 14Me), 1.60(S=1.08, 3H, s, 14Me). Ketone 1 was identified with an authentic sample of 13-norprotohirsutan-7-one, which had already been transformed to hirsutene¹⁾.

References

- 1. Y. Ohfune, H. Shirahama, and T. Matsumoto, Tetrahedron Lett., 1976, 2795.
- Biosynthesis: T. C. Feline, G. Mellows, R. B. Jones, and L. Phillips, J. C. S. Chem. Comm., 1974, 63; M. Tanabe, and K. T. Suzuki, Tetrahedron Lett., 1974, 2271; D. E. Cane and R. B. Nachbar, J. Am. Chem. Soc., 100, 3208 (1978); H. Shirahama, E. Osawa, and T. Matsumoto, ibid, 102, 3208 (1980).
- S. Misumi, T. Ohtsuka, Y. Ohfune, K. Sugita, H. Shirahama, and T. Matsumoto, Tetrahedron Lett., <u>1979</u>, 31; S. Misumi, T. Ohtsuka, H. Hashimoto, Y. Ohfune, H. Shirahama, and T. Matsumoto, Tetrahedron Lett., <u>1979</u>, 35.
- 4. Satisfactory spectral data were obtained for this compound.
 - 5: ir 1710 cm⁻¹; nmr δ 1.01, 1.08, 1.20 (each 3H, s), 4.15 (1H, d, 8 Hz).
 - 6: nmr δ 0.97, 1.04, 1.16 (each 3H, s), 3.62 (1H, dt, 10, 5 Hz), 4.08 (1H, m).
 - 7: nmr δ 0.99, 1.04, 1.19 (each 3H, s), 3.40 (1H, dt, 5, 2.5 Hz), 4.20 (1H, m).
 - 8: ir 1070, 1055 cm⁻¹; nmr δ 0.95, 1.02, 1.13 (each 3H, s), 4.00 (1H, dt, 11, 4 Hz), 4.35 (1H, m).
 - 9: ir 1650, 1090, 1065 cm⁻¹; nmr δ 0.99, 1.05, 1.11 (each 3H, s), 4.50 (1H, m), 5.25 (1H, dd, 2, 12 Hz), 5.35 (1H, d, 12 Hz).
 - 10: ir 3400, 1100 cm⁻¹; nmr δ 1.01, 1.11, 1.13 (each 3H, s), 5.80 (2H, m).
 - 13: ir 1743 cm⁻¹; nmr δ 0.90 (3H, s), 1.04 (6H, s).
 - 14: ir 1740 cm⁻¹; nmr δ 0.97 (3H, d, J=6.5 Hz), 1.02, 1.10 (each 3H, s).
 - 16:(a) ir 3100, 1070 cm $^{-1}$; nmr δ 0.92, 0.97, 1.01 (each 3H, s), 4.24 (1H, m).
 - (b) nmr δ 0.92, 1.07, 1.12 (each 3H, s), 3.83 (IH, m). 16a/16b=5/4.
- 5. Mps were noted for all crystalline compounds.
- 6. Y. Ohfune, H. Shirahama, and T. Matsumoto, Tetrahedron Lett., 1975, 4377.
- 7. Y. Ohfune, H. Shirahama, and T. Matsumoto, Tetrahedron Lett., 1976, 2869.

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