

## CLAISEN REARRANGEMENTS—XV

### STRUCTURE REVISION OF THE COUMARIN, CELERIN, BY SYNTHESIS

R. D. H. MURRAY\* and Z. D. JORGE

Department of Chemistry, University of Glasgow, Glasgow, G12 8QQ, Scotland

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**Abstract**—The four possible structures, 9, 12, 17 and 26 for celerin have been synthesised and the structure of the natural product revised to 8-hydroxy-7-methoxy-5-(1,1-dimethylallyl)coumarin 26. Efficient alternative synthetic routes to sibiricol 5, coumurrayin 6 and pinnarin 13 have been established.

Studies by Garg, Gupta and Sharma of the seed constituents of *Apium graveolens* (Umbelliferae) have revealed the presence of thirteen coumarins.<sup>2-6</sup> Apart from celerin,<sup>2</sup> m.p. 154–156°, three of these are 7-oxygenated coumarins—umbelliferone,<sup>3</sup> osthenol<sup>3</sup> and seselin;<sup>3</sup> three are 5,7-dioxygenated-bergapten,<sup>3</sup> isoimperatorin<sup>3</sup> and celereoside;<sup>4</sup> four are 7,8-dioxygenated-apigravin<sup>3</sup> (7-hydroxy-8-methoxy-6-prenylcoumarin), apiumetin,<sup>3</sup> rutaretin<sup>5</sup> and apiumoside;<sup>6</sup> while two have a 5,7,8-trioxygenated nucleus—8-hydroxybergapten<sup>3</sup> and isopimpinellin.<sup>3</sup> Celerin was deduced to be 6-hydroxy-7-methoxy-8-(1,1-dimethylallyl)coumarin 1<sup>2</sup> but recent synthetic studies in our laboratory revealed<sup>7</sup> that celerin is neither of the monomethyl ethers of obliquetol [6,7-dihydroxy-8-(1,1-dimethylallyl)-coumarin].

Biogenetic considerations<sup>8</sup> require that celerin possesses an oxygen substituent at C-7. The possibility of this being OH was precluded from the absence of a bathochromic shift in the UV maxima of celerin on addition of sodium acetate.<sup>24</sup> However, the Indian workers report<sup>2</sup> of a bathochromic shift in the UV maxima of demethylated celerin, on addition of sodium acetate, is consistent with the presence of a 7-OMe group in the natural product.

The <sup>1</sup>H NMR signals reported<sup>2</sup> for celerin reveal the H-4 doublet to be centred at  $\delta$  8.28. The normal range for H-4 (in CDCl<sub>3</sub>) for coumarins is  $\delta$  7.5–7.9<sup>8,9</sup> and only when C-5 carries an alkyl, OH or alkoxy substituent does the H-4 resonance shift downfield to  $\delta$  7.9–8.2.<sup>8-10</sup> Consequently celerin must have either the OH or the 1,1-dimethylallyl unit at C-5 and thus four structures must be considered for the natural product, two with the OH at C-5 and the 1,1-dimethylallyl group at C-6 or C-8, 9 and 12, and conversely two with the alkenyl residue at C-5 and the OH at C-6 or C-8, 17 and 26. The synthesis of each of these compounds is reported herein.

The two 5,7-dioxygenated coumarins 9 and 12 were obtained as follows. In 1975, we discovered<sup>11</sup> that 5-prenyloxy-7-methoxycoumarin 2 underwent exclusive *para*-Claisen rearrangement on heating in butyric anhydride/diethylaniline giving the butyrate 3 despite the availability of an *ortho* position for rearrangement. Recently we found<sup>12</sup> that the problem of freeing products from traces of reagents, attendant on such reactions, could be circumvented by using acetic anhydride and omitting the amine. Thus 5-prenyloxy-7-acetoxycoumarin in refluxing anhydride gave the *para*-Claisen rearrangement product,

5,7-diacetoxy-8-prenylcoumarin.<sup>13</sup> When 5-prenyloxy-7-methoxycoumarin 2 was treated in the same way, the anticipated *para* rearrangement to the acetate 4 occurred, again in quantitative yield. Deacetylation of 4 using the mild conditions of Zn dust in MeOH<sup>14</sup> gave the corresponding phenol, sibiricol 5,<sup>15</sup> the penultimate product in our previous synthesis of coumurrayin 6.<sup>16</sup> However, overnight exposure of 4 to ten equivalents of 1% methanolic NaOH afforded an equilibrium mixture (85 : 14) of sibiricol and the 5-hydroxycoumarin 7 from lactone-ring isomerisation.<sup>12</sup> Although the yield of the latter phenol is low, the two products can readily be separated by chromatography thereby providing an alternative synthetic route to toddaculin 8.<sup>11</sup>

We envisaged that 9 could be obtained by a Claisen rearrangement of 2 if the intermediate *ortho*-dienone<sup>11</sup> could be induced to enolise instead of undergoing the further sigmatropic shift to the *para* position. In our recent synthesis of hortiolone,<sup>12</sup> the combination of acetic anhydride and sodium acetate was utilised to prevent incursion of a rapid abnormal Claisen rearrangement. Under these conditions, 2 afforded the acetate 10 of the desired *ortho*-rearrangement product in 86% yield although *para* rearrangement to 4 (11%) could not be totally excluded. The structure of 10 followed from its <sup>1</sup>H NMR spectrum which disclosed signals for a 1,1-dimethylallyl unit and long-range coupling of H-4 with H-8, requiring the alkenyl substituent to be located at C-6.<sup>17</sup> The acetate 10 was inert to attempted deacetylation with Zn dust in MeOH but after treatment with one equivalent of 0.5% methanolic NaOH for 4 hr, a mixture was obtained of the desired phenol 9 (75%), m.p. 105–107°, the isomer 12 (6%), m.p. 175–177° resulting from lactone-ring isomerisation<sup>11</sup> and residual starting material (9%). An equilibrium mixture of 12 (91%) having the 1,1-dimethylallyl group at C-8 and 9 (9%), where this bulky group is at C-6, was obtained after 12 hr treatment of 10 with ten equivalents 3% methanolic NaOH. The melting points of these two 5-hydroxycoumarins indicate that neither can be celerin, as do their <sup>1</sup>H NMR spectra. Their singlet aromatic protons resonate at  $\delta$  6.52 and 6.47, respectively, compared with  $\delta$  7.03 for celerin, while the H-4 doublets are centred at  $\delta$  8.05 and 7.98, respectively. Confirmation of the structure 12 of the phenol obtained after lactone-ring isomerisation came from methylation, which gave pinnarin 13.<sup>18</sup> It should be noted that in both syntheses of this natural

coumarin, the nuclear 1,1-dimethylallyl group is introduced by a prenyl ether *ortho*-Claisen rearrangement, in our first synthesis<sup>18</sup> directly to C-8 from 5-methoxy-7-prenyloxycoumarin and now indirectly from the positional isomer 2.

The two OMe resonances of pinnarin at  $\delta$  3.84 and 3.93 are close to those of 5,7-dimethoxycoumarin ( $\delta$  3.83 and 3.88)<sup>19</sup> whereas the OMe signals for 11 resonate at  $\delta$  3.83 and 3.65. The last value is at higher field than normally found<sup>19</sup> for methoxycoumarins ( $\delta$  3.8–4.4) and implies that one of the OMe groups, presumably that at C-5, adopts an out-of-plane conformation<sup>20</sup> due to the proximity of the bulky 1,1-dimethylallyl group at C-6 and the *peri* hydrogen at C-4. In contrast the OMe signals for toddaculin 8<sup>11</sup> the isomer of 11 having a 3,3-dimethylallyl, instead of a 1,1-dimethylallyl, group at C-6, resonate at  $\delta$  3.82 and 3.86.

The third possibility 17 for celerin, which again contains a 1,1-dimethylallyl group *ortho* to a phenol, was obtained by the first example of an *ortho*-Claisen rearrangement of a 6-allyloxycoumarin. The major product 18 (80%) from heating 6-prenyloxy-7-methoxycoumarin 15 in acetic anhydride containing sodium acetate was accompanied however by the acetate 16 (14%) from prenyl ether cleavage, a known alternative reaction pathway.<sup>21</sup>

The acetate 18 was inert to Zn dust in MeOH, a deacetylation process which we have found to be ineffective for hindered acetates. Complete hydrolysis of 18 required eight equivalents of 2% methanolic NaOH for 8 hr but the desired phenol 17 (65%) m.p. 92–94° was accompanied by the 2,3,3-trimethyl-dihydrofuranocoumarin 20 (31%). Treatment of 18 with a vast excess (150 equivalents) of 10% methanolic NaOH for 10 min gave the phenol 17 (97%) uncontaminated with the cyclic isomer. The <sup>1</sup>H NMR spectrum of 17 is close to that reported<sup>2</sup> for celerin with H-4 centred at  $\delta$  8.30 and the benzenoid proton at  $\delta$  6.73, but the difference of 62° in m.p.s precludes this possibility for the natural coumarin. Moreover, there is no indication<sup>2</sup> that solutions of celerin are unstable whereas 17 was readily converted to the dihydrofuran 20 on standing in CHCl<sub>3</sub> or ethyl acetate at room temperature.

We envisaged that the final possibility, 26, for celerin could be obtained by the *para*-Claisen rearrangement<sup>1</sup> of the 1,1-dimethylallyl ether 24. The requisite phenol, 8-hydroxy-7-methoxycoumarin 21 was conveniently obtained by the known<sup>22</sup> selective demethylation of 7,8-dimethoxycoumarin 22 in sulphuric acid at 70°. Extending the reaction time from the recommended 14 hr to 6 days however has resulted in a yield improvement from 29 to 77%.

The phenol 21 was completely inert to 3-chloro-3-methylbut-1-yne using the normal etherification conditions<sup>11,23</sup> and even prolonged exposure to potassium carbonate and potassium iodide in moist acetone in a closed system, used in the synthesis of benahorin,<sup>1</sup> gave the desired ether 23 in only 15% yield, the *C,O*-dialkylated product 29 (4%) but mainly unreacted starting material. However, in the presence of cesium carbonate and dry acetone in a closed system, 21 reacted completely after 36 hr to give the ether 23 (46%) and a number of by-products, though not 29, which are currently being investigated.

Partial hydrogenation of 23 over 5% Pd-BaSO<sub>4</sub>

followed by chromatography gave the tetrahydro compound 25 (17%) and the required 1,1-dimethylallyl ether 24 (71%) which quantitatively underwent a *para*-Claisen rearrangement in acetic anhydride and sodium acetate giving 27. Hydrolysis of 27 with 1% methanolic NaOH gave the phenol 26, m.p. 158–159.5°. This melting point is only 4° higher than that reported<sup>2</sup> for celerin while the <sup>1</sup>H NMR signals of 26 are similar to the published values for celerin except that those for the synthetic product are upfield by ~0.1 ppm. In particular, H-4 is centred at  $\delta$  8.22 and the aromatic H-6 proton at  $\delta$  6.93. The mass spectra of celerin and 26 each show a parent molecular ion and important fragmentation ions at *m/z* 245 and 213 with two consecutive losses of CO from the latter. The loss of 42 mass units (MeOH) from the first fragment ion in both spectra is quite unique for methoxycoumarins.<sup>8</sup> Although no sample of the natural product is available, direct comparison of the <sup>1</sup>H NMR and mass spectra of celerin (kindly provided by Prof. Gupta) and 26 has revealed their probable identity. In support of structure 26, it should be noted that the reported<sup>2</sup> demethylation of celerin with pyridinium bromide to a diphenol followed by methylation to celerin methyl ether precludes the 1,1-dimethylallyl group being *ortho* to either oxygen function since cyclisation to a dihydrofuran would otherwise have undoubtedly occurred in the strongly acidic demethylation conditions.<sup>8</sup>

Apart from 5-methylcoumarins, only eight C-5 alkylated coumarins are known<sup>8</sup> all of which possess a 7,8-dioxygenated coumarin nucleus as significantly do four of the *A. graveolens* coumarins. Our results suggest that celerin should be reformulated as 8-hydroxy-7-methoxy-5-(1,1-dimethylallyl)coumarin 26, the second example of a coumarin with a 1,1-dimethylallyl group at C-5, benahorin being the first.<sup>1</sup>

## EXPERIMENTAL

For general experimental see preceding paper.

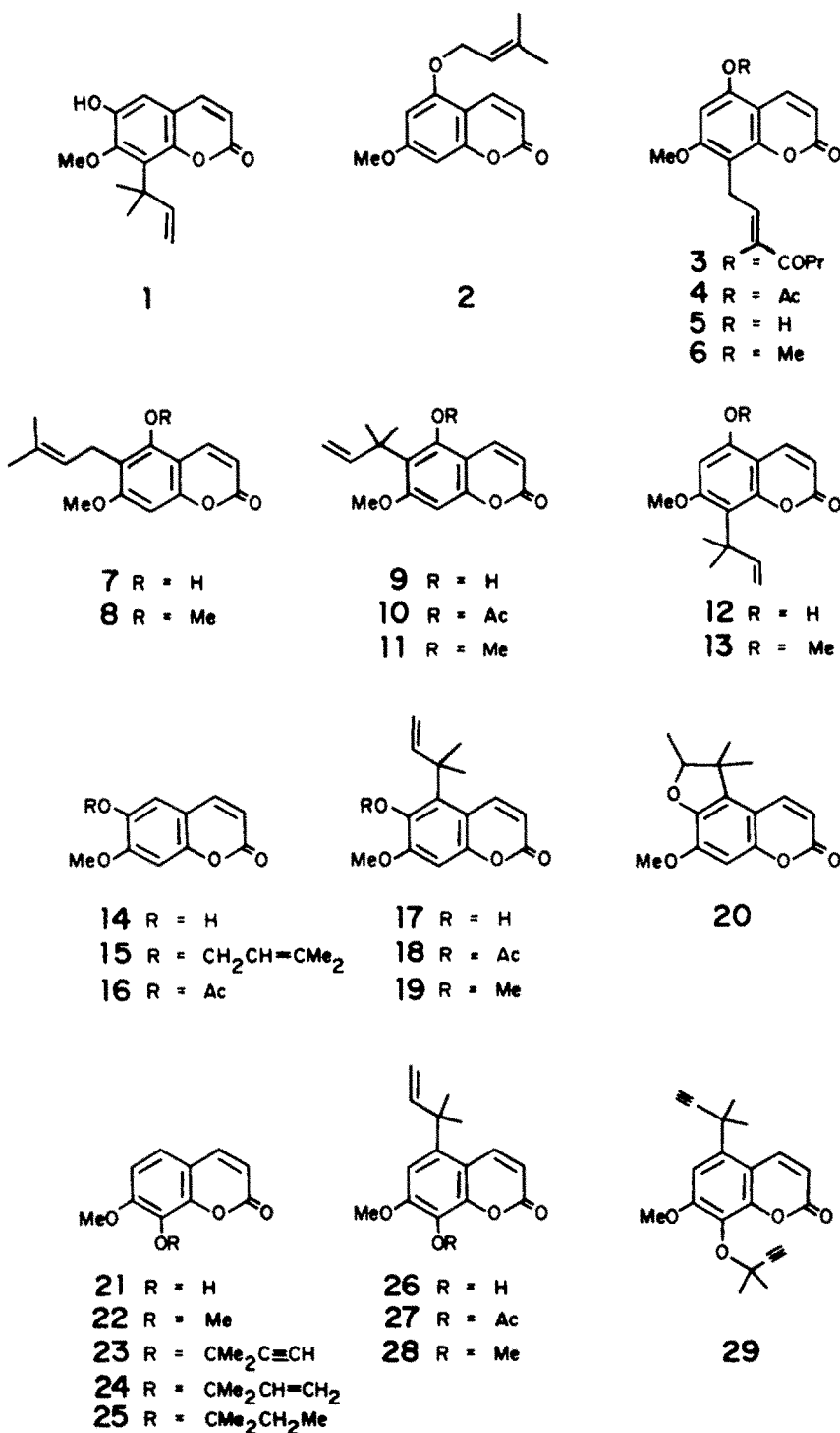
### Claisen rearrangements of 2

(i) A soln of 2 (80 mg) in Ac<sub>2</sub>O (10 ml) was refluxed for 20 hr. Evaporation of the solvent under reduced pressure gave 5-acetoxy-7-methoxy-8-(3-methylbut-2-enyl)coumarin 4 (92 mg, 100%) colourless needles, m.p. 120–121° (lit.<sup>13</sup> 124–125°) (from ether). (Found: C, 67.65; H, 6.0. C<sub>17</sub>H<sub>18</sub>O<sub>5</sub> requires: C, 67.55; H, 6.0%;  $\nu_{\max}$  1770, 1730, 1620, 1610 and 1560 cm<sup>-1</sup>; NMR signals at  $\delta$  1.63 (3H, bs), 1.81 (3H, bs), 2.35 (3H, s), 3.50 (2H, bd, *J* 7.5 Hz), 3.87 (3H, s), 5.2 (1H, bt, *J* 7.5 Hz), 6.22 (1H, d, *J* 9.5 Hz), 6.65 (1H, s) and 7.59 (1H, d, *J* 9.5 Hz).

(ii) A mixture of 2 (228 mg), NaOAc (1.53 g) and Ac<sub>2</sub>O (15 ml) was refluxed with stirring for 20 hr. The residue from filtration and evaporation was chromatographed on silica gel G. Elution with EtOAc–light petroleum (1:4) gave 5-acetoxy-7-methoxy-8-(1,1-dimethylallyl)coumarin 10 (227 mg, 86%) colourless plates, m.p. 144–145° (from ether). (Found: C, 67.3; H, 6.2. C<sub>17</sub>H<sub>18</sub>O<sub>5</sub> requires: C, 67.55; H, 6.0%;  $\nu_{\max}$  1770, 1730, 1620, 1610 and 1560 cm<sup>-1</sup>; NMR signals at  $\delta$  1.48 (6H, s), 2.29 (3H, s), 3.89 (3H, s), 4.88 (1H, dd, *J* 10 and 2 Hz), 4.92 (1H, dd, *J* 17.5 and 2 Hz), 6.23 (1H, d, *J* 9.5 Hz), 6.27 (1H, dd, *J* 17.5 and 10 Hz), 6.74 (1H, s) and 7.43 (1H, d, *J* 9.5 Hz); and 4 (30 mg, 11%).

### Hydrolysis of 4

(i) A soln of 4 (13.5 mg) in MeOH (5 ml) was stirred with activated Zn<sup>14</sup> (from Zn dust, 150 mg) for 12 hr. After filtration through celite and washing with MeOH, the combined filtrates were evaporated under reduced pressure. The residue was partitioned between EtOAc and dil HCl,



the organic layer washed with dil HCl, brine, dried and evaporated to give sibiricol 5 (11.5 mg 100%) colourless needles, m.p. 195–197° (lit.<sup>15,16</sup> 193–194°) (from EtOAc–light petroleum) identical (mmp, NMR and TLC) with an authentic sample.

(ii) A soln of 4 (56 mg, 0.185 mmole) in MeOH (5 ml) and 1% NaOH/MeOH (7.5 ml, 1.87 mmole) was stirred for 24 hr. After careful neutralisation with dil HCl, the solvent was evaporated and the residue partitioned between EtOAc and brine, the organic layer washed with brine, dried and

evaporated. The residue was chromatographed on silica gel G. Elution with EtOAc–light petroleum afforded 7 (7 mg, 14%) colourless needles, m.p. 151–153° (lit.<sup>11</sup> 150–153°) (from EtOAc–light petroleum) identical (mmp, NMR and TLC) with an authentic sample; and 5 (41 mg, 85%).

#### Hydrolysis of 10

(i) A soln of 10 (55 mg, 0.182 mmole) in MeOH (10 ml) and 0.5% NaOH/MeOH (1.42 ml, 0.182 mmole) was stirred for 4 hr until TLC disclosed maximal concentration of a less

polar product. Work up as for 7 followed by chromatography on silica gel G and elution with EtOAc–light petroleum (1:4) gave 5 - hydroxy - 7 - methoxy - 6 - (1,1 - dimethylallyl)coumarin 9 (35 mg, 75%) tan-yellow plates, m.p. 105–107° (from MeOH). (Found: C, 69.25; H, 6.1.  $C_{15}H_{16}O_4$  requires: C, 69.2; H, 6.2%;  $\nu_{\max}$  3410, 1775, 1620, 1610 and 1570  $\text{cm}^{-1}$ ; NMR signals at  $\delta$  1.55 (6H, s), 3.90 (3H, s), 5.45 (1H, dd,  $J$  10 and 2 Hz), 5.53 (1H, dd,  $J$  17.5 and 2 Hz), 6.17 (1H, d,  $J$  9.5 Hz), 6.47 (1H, s), 6.53 (1H, dd,  $J$  17.5 and 10 Hz), 7.40 (1H, s, OH) and 7.98 (1H, d,  $J$  9.5 Hz); 10 (5 mg, 9%) and 12 (3 mg, 6%).

(ii) A soln of 10 (51 mg, 0.169 mmole) in MeOH (10 ml) and 3% NaOH/MeOH (2 ml, 1.5 mmole) was stirred overnight. Work up and chromatography as above gave 9 (4 mg, 9%) and 5 - hydroxy - 7 - methoxy - 8 - (1,1 - dimethylallyl)coumarin 12 (40 mg, 91%) colourless plates, m.p. 175–177° (from EtOAc–light petroleum). (Found: C, 69.5; H, 6.1.  $C_{15}H_{16}O_4$  requires: C, 69.2; H, 6.2%;  $\nu_{\max}$  (KBr) 3300 (b), 1690, 1640 and 1575  $\text{cm}^{-1}$ ; NMR signals (acetone- $d_6$ ) at  $\delta$  1.61 (6H, s), 3.80 (3H, s), 4.79 (1H, dd,  $J$  10 and 2 Hz), 4.83 (1H, dd,  $J$  17.5 and 2 Hz), 6.06 (1H, d,  $J$  9.5 Hz), 6.27 (1H, dd,  $J$  17.5 and 2 Hz), 6.52 (1H, s) and 8.05 (1H, d,  $J$  9.5 Hz).

#### Methylation of 9

A mixture of 9 (16 mg), MeI (0.5 ml),  $K_2CO_3$  (100 mg) and acetone (10 ml) was refluxed with stirring for 1 hr. The cooled mixture was filtered, the solvent evaporated and the residue partitioned between EtOAc and brine. The organic layer was washed with satd. aq.  $NaHCO_3$ , brine, dried and evaporated to give 5,7 - dimethoxy - 6 - (1,1 - dimethylallyl)coumarin 11 (17 mg) colourless needles, m.p. 88–90° (from ether–light petroleum). (Found: C, 69.45; H, 6.40.  $C_{16}H_{18}O_4$  requires: C, 70.05; H, 6.6%;  $\nu_{\max}$  1725, 1640, 1612, 1600 and 1565  $\text{cm}^{-1}$ ; NMR signals at  $\delta$  1.51 (6H, s), 3.65 (2H, s), 3.85 (3H, s), 4.82 (1H, dd,  $J$  10 and 2 Hz), 4.87 (1H, dd,  $J$  17.5 and 2 Hz), 6.24 (1H, d,  $J$  9.5 Hz), 6.35 (1H, dd,  $J$  17.5 and 10 Hz), 6.63 (1H, s) and 7.85 (1H, d,  $J$  9.5 Hz).

#### Pinnarin 13

Methylation of 12 (24 mg) as above gave pinnarin 13 (25.5 mg) colourless plates, m.p. 170–172° (lit.<sup>24</sup> 162–163°; lit.<sup>18</sup> 166–167°) (from ether). (Found for natural sample<sup>24</sup> 170–172°—found for synthetic sample<sup>18</sup> 169–171°) identical (mmp, NMR, IR and TLC) with authentic samples.

#### Synthesis and Claisen rearrangement of 15

A mixture of 14<sup>25</sup> (514 mg),  $K_2CO_3$  (1.45 g), prenyl bromide (540 mg) and acetone (50 ml) was refluxed for 1 hr. Work up as for 11 gave 6 - (3 - methylbut - 2 - enyloxy) - 7 - methoxycoumarin 15 (680 mg, 98%) colourless plates, m.p. 116–118° (from EtOAc–hexane). (Found: C, 69.5; H, 6.35.  $C_{15}H_{16}O_4$  requires: C, 69.2; H, 6.2%;  $\nu_{\max}$  1720, 1615, 1560 and 1515  $\text{cm}^{-1}$ ; NMR signals at  $\delta$  1.73 (3H, bs), 1.76 (3H, bs), 3.92 (3H, s), 4.58 (2H, bd,  $J$  7.5 Hz), 5.51 (1H, bt,  $J$  7.5 Hz), 6.25 (1H, d,  $J$  9.5 Hz), 6.80 (1H, s), 6.88 (1H, s) and 7.59 (1H, d,  $J$  9.5 Hz). A mixture of 15 (154 mg), NaOAc (1.1 g) and  $Ac_2O$  (15 ml) was refluxed with stirring for 36 hr. Work up as for 4 and chromatography on silica gel 60 with elution with EtOAc–light petroleum (3:17) gave 6 - acetoxy - 7 - methoxy - 5 - (1,1 - dimethylallyl)coumarin 18 (151 mg, 80%) colourless plates, m.p. 124–125° (from EtOAc–light petroleum). (Found: C, 67.45; H, 5.85.  $C_{17}H_{18}O_5$  requires: C, 67.55; H, 6.0%;  $\nu_{\max}$  1760, 1728, 1600 and 1560  $\text{cm}^{-1}$ ; NMR signals at  $\delta$  1.55 (6H, s), 2.28 (3H, s), 3.87 (3H, s), 4.97 (1H, dd,  $J$  17.5 and 2 Hz), 5.07 (1H, dd,  $J$  10 and 2 Hz), 6.15 (1H, d,  $J$  9.5 Hz), 6.23 (1H, dd,  $J$  17.5 and 10 Hz), 6.80 (1H, s) and 8.23 (1H, d,  $J$  9.5 Hz); and 16 (20 mg, 14%) colourless needles, m.p. 166–167° (lit.<sup>26</sup> 167–168°) (from EtOAc–light petroleum) identical (mmp, NMR and TLC) with a sample prepared by acetylation of 14.

#### Hydrolysis of 18

(i) A soln of 18 (56 mg, 0.185 mmole) in MeOH (10 ml) and 2% NaOH/MeOH (3 ml, 1.5 mmole) was stirred for 8 hr

until TLC indicated complete disappearance of starting material. Work up as for 7 gave a mixture of two products as an oil. Preparative TLC on silica (Fluka GF 254) and elution with MeOH– $CHCl_3$ –light petroleum (1:4:20)  $\times$  3 afforded the dihydrofuran 20 (15 mg, 31%) colourless needles, m.p. 154–156° (from EtOAc–light petroleum). (Found: C, 69.0; H, 6.25.  $C_{15}H_{16}O_4$  requires: C, 69.2; H, 6.2%;  $\nu_{\max}$  1725, 1620, 1612 and 1565  $\text{cm}^{-1}$ ; NMR signals at  $\delta$  1.20 (3H, s), 1.42 (3H, d,  $J$  6.5 Hz), 1.47 (3H, s), 3.93 (3H, s), 4.46 (1H, q,  $J$  6.5 Hz), 6.28 (1H, d,  $J$  9.5 Hz), 6.71 (1H, s) and 7.86 (1H, d,  $J$  9.5 Hz); and 6 - hydroxy - 7 - methoxy - 5 - (1,1 - dimethylallyl)coumarin 17 (31 mg, 65%) tan-yellow plates, m.p. 92–94° (from ether–light petroleum). (Found: C, 69.2; H, 5.95.  $C_{15}H_{16}O_4$  requires: C, 69.2; H, 6.2%;  $\nu_{\max}$  3530, 1720, 1610 and 1570  $\text{cm}^{-1}$ ;  $\lambda_{\max}$  (MeOH) 352, 302, 260 and 238 nm ( $\log \epsilon$  3.87, 3.81, 3.48 and 3.91);  $\lambda_{\max}$  (MeOH + base) 352, 302, 260 and 238 nm ( $\log \epsilon$  3.85, 3.80, 3.58 and 3.91); NMR signals at  $\delta$  1.62 (6H, s), 3.96 (3H, s), 4.95 (1H, dd,  $J$  17.5 and 2 Hz), 5.05 (1H, dd,  $J$  10 and 2 Hz), 6.00 (1H, s, OH), 6.12 (1H, d,  $J$  9.5 Hz), 6.25 (1H, dd,  $J$  17.5 and 10 Hz), 6.73 (1H, s) and 8.30 (1H, d,  $J$  9.5 Hz).

(ii) A soln of 18 (62 mg, 0.203 mmole) in MeOH (5 ml) and 10% NaOH/MeOH (12 ml, 30 mmole) was stirred for 10 min. Work up as above gave 17 (52 mg, 97%); no trace of 20 could be detected by NMR or TLC.

#### Methylation of 17

A mixture of 17 (21 mg), MeI (1.0 ml),  $K_2CO_3$  (100 mg) and acetone (10 ml) was refluxed with stirring for 3 hr. Work up as for 11 gave 6,7 - dimethoxy - 5 - (1,1 - dimethylallyl)coumarin 19 (21 mg) colourless plates, m.p. 75–76° (from light petroleum); mass spectral peaks at  $m/z$  274 ( $M^+$ , 100%), 259(50), 243(30), 231(27), 128(28) and 115(45);  $\nu_{\max}$  1720, 1593, 1559 and 1463  $\text{cm}^{-1}$ ; NMR signals at  $\delta$  1.62 (6H, s), 3.75 (3H, s), 3.93 (3H, s), 4.90 (1H, dd,  $J$  17.5 and 2 Hz), 5.05 (1H, dd,  $J$  10 and 2 Hz), 6.13 (1H, d,  $J$  9.5 Hz), 6.25 (1H, dd,  $J$  17.5 and 10 Hz), 6.78 (1H, s) and 8.24 (1H, d,  $J$  9.5 Hz).

#### Partial demethylation of 7,8-dimethoxycoumarin 22

A cooled soln of conc  $H_2SO_4$  (27 ml) and water (9 ml) was added to 22 (4.55 g) and the yellow soln stirred in an oil bath kept at 70°. Monitoring by TLC revealed maximal concentration of a new product after 6 days compared with starting material and the more polar 7,8-dihydroxycoumarin. The cooled soln was poured on to crushed ice, extracted with EtOAc, the organic layer washed with brine to neutrality, dried and evaporated. The residue was crystallised from EtOAc–light petroleum to give 21 (3.29 g, 77%), m.p. 173–175° (lit.<sup>22</sup> 173–174°; 29%).

#### Dimethylpropargylation of 21

(i) A mixture of 21 (440 mg) in 2% aqueous acetone (30 ml),  $K_2CO_3$  (1.0 g), KI (150 mg) and 3-chloro-3-methylbut-1-yne (1.5 ml) was refluxed with stirring for 12 hr in a closed system. More  $K_2CO_3$  (500 mg), KI (50 mg) and 3-chloro-3-methylbut-1-yne (1.5 ml) were added and refluxing continued for a further 12 hr. Two further additions of 3-chloro-3-methylbut-1-yne (1.5 ml each) were made after 36 and 48 hr. After 60 hr, the mixture was filtered through celite and washed with acetone. The combined filtrates were evaporated and the residue partitioned between EtOAc and water, the organic layer washed with 5%  $Na_2CO_3$  soln, brine, dried and evaporated. The residual oil was chromatographed on silica gel 60. Gradient elution with EtOAc–light petroleum (1:9 to 1:4) afforded 7 - methoxy - 8 - (1,1 - dimethylpropargyloxy) - 5 - (1,1 - dimethylpropargyl)coumarin 29 (14 mg, 4%) colourless plates, m.p. 150–152° (from EtOAc–light petroleum). (Found: C, 74.0; H, 6.15.  $C_{20}H_{20}O_4$  requires: C, 74.05; H, 6.2%; mass spectral peaks at  $m/z$  258 ( $M^+ - C_5H_6$ , 100%) and 243(79);  $\nu_{\max}$  3305, 1723, 1590 and 1495  $\text{cm}^{-1}$ ; NMR signals at  $\delta$  1.72 (6H, s), 1.77 (6H, s), 2.30 (1H, s), 2.38 (1H, s), 3.92 (3H, s), 6.25 (1H, d,  $J$  9.5 Hz), 6.90 (1H, s) and 8.72 (1H, d,  $J$  9.5 Hz); 7 - methoxy - 8 - (1,1 - dimethylpropargyloxy)coumarin 23 (89 mg, 15%) as a pale yellow oil. (Found: C, 69.75; H, 5.85.

$C_{15}H_{14}O_4$  requires: C, 69.75; H, 5.45%;  $\nu_{\max}$  3300, 1725, 1608 and  $1560\text{ cm}^{-1}$ ; NMR signals at  $\delta$  1.79 (6H, s), 2.30 (1H, s), 3.92 (3H, s), 6.24 (1H, d,  $J$  9.5 Hz), 6.86 (1H, d,  $J$  8 Hz), 7.21 (1H, d,  $J$  8 Hz) and 7.60 (1H, d,  $J$  9.5 Hz); and recovered 21 (270 mg).

(ii) A mixture of 21 (298 mg),  $Ca_2CO_3$  (900 mg), 3-chloro-3-methylbut-1-yne (2 ml) and acetone (25 ml) was refluxed with stirring for 36 hr in a closed system. Work up and chromatography as above gave 23 (130 mg, 46%).

#### Synthesis and Claisen rearrangement of 24

A soln of 23 (69 mg) in EtOAc (10 ml) was hydrogenated over 5% Pd-BaSO<sub>4</sub> (22 mg) for 2.5 hr. Removal of catalyst and solvent followed by flash column chromatography on silica gel (Fluka GF 254) and elution with EtOAc-light petroleum (3:17) gave 7-methoxy-8-(1,1-dimethylpropoxy)coumarin 25 (13 mg, 17%) colourless needles, m.p. 39–40° (from EtOAc-light petroleum); NMR signals at  $\delta$  1.15 (3H, t,  $J$  Hz), 1.30 (3H, s), 1.85 (2H, q,  $J$  7 Hz), 3.88 (3H, s), 6.21 (1H, d,  $J$  9.5 Hz), 6.84 (1H, d,  $J$  8 Hz), 7.14 (1H, d,  $J$  8 Hz) and 7.57 (1H, d,  $J$  9.5 Hz); 7-methoxy-8-(1,1-dimethylallyloxy)coumarin 24 (49 mg, 71%) colourless needles, m.p. 58–59.5° (from EtOAc-light petroleum). (Found: C, 69.15; H, 6.25.  $C_{15}H_{16}O_4$  requires: C, 69.2; H, 6.2%; mass spectral peaks at  $m/z$  192 ( $M^+ - C_2H_5$ , 100%), 164(21), 163(29) and 149(36);  $\nu_{\max}$  1725, 1605, 1560 and  $1505\text{ cm}^{-1}$ ; NMR signals at  $\delta$  1.51 (6H, s), 3.88 (3H, s), 5.00 (1H, dd,  $J$  10 and 2 Hz), 5.12 (1H, dd,  $J$  17.5 and 2 Hz), 6.20 (1H, d,  $J$  9.5 Hz), 6.25 (1H, dd,  $J$  17.5 and 10 Hz), 6.83 (1H, d,  $J$  8 Hz), 7.15 (1H, d,  $J$  8 Hz) and 7.57 (1H, d,  $J$  9.5 Hz); and 23 (5 mg, 7%).

A mixture of 24 (27 mg), NaOAc (50 mg) and  $Ac_2O$  (1.5 ml) was refluxed with stirring for 90 min. Work up as for 10 gave 7-methoxy-8-acetoxy-5-(1,1-dimethylallyl)coumarin 27 (32 mg, 100%) colourless needles, m.p. 159–160° (from EtOAc-light petroleum). (Found: C, 67.55; H, 5.5.  $C_{15}H_{18}O_5$  requires: C, 67.55; H, 6.0%;  $\nu_{\max}$  1760, 1730, 1605 and  $1495\text{ cm}^{-1}$ ; NMR signals at  $\delta$  1.49 (6H, s), 2.38 (3H, s), 3.93 (3H, s), 5.04 (1H, dd,  $J$  17.5 and 2 Hz), 5.13 (1H, dd,  $J$  10 and 2 Hz), 6.14 (1H, d,  $J$  9.5 Hz), 6.15 (1H, dd,  $J$  17.5 and 10 Hz), 6.97 (1H, s) and 8.20 (1H, d,  $J$  9.5 Hz).

#### Celerin 26 and its methyl ether

A soln of 27 (21 mg, 0.07 mmole) in MeOH (5 ml) and 1% NaOH/MeOH (0.5 ml, 0.125 mmole) was stirred for 15 min. Work up as for 7 gave 7-methoxy-8-hydroxy-5-(1,1-dimethylallyl)coumarin (celerin) 26 (18 mg, 100%) colourless plates, m.p. 158–159.5° (lit.<sup>2</sup> 154–156°) (from EtOAc-light petroleum). (Found: C, 69.45; H, 6.15.  $C_{15}H_{16}O_4$  requires: C, 69.2; H, 6.2%; mass spectral peaks at  $m/z$  260 ( $M^+$ , 100%), 245(24), 213(66) and 185(23);  $\nu_{\max}$  3550, 1730, 1612, 1580 and  $1500\text{ cm}^{-1}$ ;  $\lambda_{\max}$  (MeOH) 325, 264 and 228 nm ( $\log \epsilon$  4.10, 4.06 and 3.70);  $\lambda_{\max}$  (MeOH + base) 332, 284 and 232 nm ( $\log \epsilon$  3.99, 4.23 and 3.79); NMR signals at  $\delta$  1.49 (6H, s), 4.01 (3H, s), 5.01 (1H, dd,  $J$  17.5 and 2 Hz), 5.12 (1H, dd,  $J$  10 and 2 Hz), 6.14 (1H, dd,  $J$  17.5 and 10 Hz), 6.15 (1H, d,  $J$  9.5 Hz), 6.93 (1H, s) and 8.22 (1H, d,  $J$  9.5 Hz); extremely similar (NMR and mass spectra) to those of a natural sample.

A mixture of 26 (14 mg),  $K_2CO_3$  (150 mg), MeI (0.5 ml) and acetone (10 ml) was refluxed with stirring for 15 min. Work up gave 7,8-dimethoxy-5-(1,1-dimethylallyl)coumarin 28 (15 mg, 100%) colourless plates, m.p. 99–100° (from light petroleum). (Found: C, 70.2; H, 6.65.

$C_{16}H_{18}O_4$  requires: C, 70.05; H, 6.6%; mass spectral peaks at  $m/z$  274 ( $M^+$ , 100%), 259(64), 231(21), 228(63) and 200(23);  $\nu_{\max}$  1725, 1590 and  $1492\text{ cm}^{-1}$ ; NMR signals at  $\delta$  1.49 (6H, s), 3.98 (6H, s), 5.02 (1H, dd,  $J$  17.5 and 2 Hz), 5.12 (1H, dd,  $J$  10 and 2 Hz), 6.14 (1H, dd,  $J$  17.5 and 2 Hz), 6.14 (1H, d,  $J$  9.5 Hz), 6.93 (1H, s) and 8.18 (1H, d,  $J$  9.5 Hz).

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#### REFERENCES

- Part XIV, R. D. H. Murray, Z. D. Jorge and D. M. Boag, *Tetrahedron* **40**, 5225 (1984).
- S. K. Garg, S. R. Gupta and N. D. Sharma, *Planta Med.* **38**, 186 (1980).
- S. K. Garg, S. R. Gupta and N. D. Sharma, *Phytochemistry* **18**, 1580 (1979).
- S. K. Garg, S. R. Gupta and N. D. Sharma, *Planta Med.* **38**, 363 (1980).
- S. K. Garg, S. R. Gupta and N. D. Sharma, *Phytochemistry* **17**, 2135 (1978).
- S. K. Garg, S. R. Gupta and N. D. Sharma, *Ibid.* **18**, 1764 (1979).
- R. D. H. Murray and Z. D. Jorge, *Tetrahedron* **39**, 3163 (1983).
- R. D. H. Murray, J. Méndez and S. A. Brown, *The Natural Coumarins*. Wiley, Chichester (1982).
- W. Steck, *Lloydia* **35**, 418 (1972).
- J. Reisch, I. Novák, K. Szendrei and E. Minker, *Pharmazie* **22**, 205 (1967); A. G. González, J. L. Bretón, H. López Dorta, M. A. Martínez Iñiguez and F. Rodríguez Luis, *An. Quím.* **69**, 1013 (1973).
- R. D. H. Murray, M. M. Ballantyne, T. C. Hogg and P. H. McCabe, *Tetrahedron* **31**, 2960 (1975).
- R. D. H. Murray, Z. D. Jorge and K. W. M. Lawrie, *Ibid.* **39**, 3159 (1983).
- R. D. H. Murray and Z. D. Jorge, *Tetrahedron Letters* **24**, 3403 (1983).
- A. G. González, Z. D. Jorge, H. López Dorta and F. Rodríguez Luis, *Ibid.* **353** (1981).
- R. Kumar, B. D. Gupta, S. K. Banerjee and C. K. Atal, *Phytochemistry* **17**, 2111 (1978).
- R. D. H. Murray, M. M. Ballantyne and K. P. Mathai, *Tetrahedron* **27**, 1247 (1971).
- E. V. Lassak and J. T. Pinhey, *J. Chem. Soc. (C)* (1967).
- R. D. H. Murray and M. M. Ballantyne, *Tetrahedron* **26**, 4667 (1970).
- W. Steck and M. Mazurek, *Lloydia* **35**, 418 (1972).
- A. I. Gray, R. D. Waigh and P. G. Waterman, *J. Chem. Soc. Perkin II* 391 (1978).
- M. M. Ballantyne, P. H. McCabe and R. D. H. Murray, *Tetrahedron* **27**, 871 (1971).
- F. M. Dean, A. M. B. S. R. C. S. Costa, J. B. Harborne and D. M. Smith, *Phytochemistry* **17**, 505 (1978).
- J. Hlubucek, E. Ritchie and W. C. Taylor, *Aust. J. Chem.* **24**, 2347 (1971).
- R. Estévez Reyes and A. G. González, *Phytochemistry* **9**, 833 (1970).
- F. S. Head and A. Robertson, *J. Chem. Soc.* 2434 (1930).
- T. Nakano, S. Terao, Y. Sacki and K. D. Jin, *Ibid.* (C) 1805 (1966).