

STEREOSELECTIVE TOTAL SYNTHESIS OF (±)-EREMOFORTIN B,
A SESQUITERPENOID MYCOTOXIN OF *Penicillium roqueforti*¹

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The stereoselective total synthesis of (±)-eremofortin B (1), a sesquiterpenoid mycotoxin of *Penicillium roqueforti*, from 5β,6β-dimethyl-2,7-dioxo-Δ¹⁽¹⁰⁾-octalin (2) is described. Stereochemistry of eremofortin B was confirmed as shown in 1.

Eremofortin B, a sesquiterpenoid mycotoxin of *Penicillium roqueforti*, was found by Moreau *et al.*² The structure of eremofortin B was proposed to be an eremophilane type sesquiterpenoid as shown in 1', however the stereochemistry was not clear.³

We previously reported the total synthesis of (±)-isopetasol,^{4a} (±)-warburgiadione,^{4a} (±)-petasitol,^{4a} and (±)-phomenone^{4b} from 5β,6β-dimethyl-2,7-dioxo-Δ¹⁽¹⁰⁾-octalin (2). In this communication, we wish to report the first total synthesis of (±)-eremofortin B (1) from the key intermediate (2), the synthesis confirming the stereochemistry of 1.

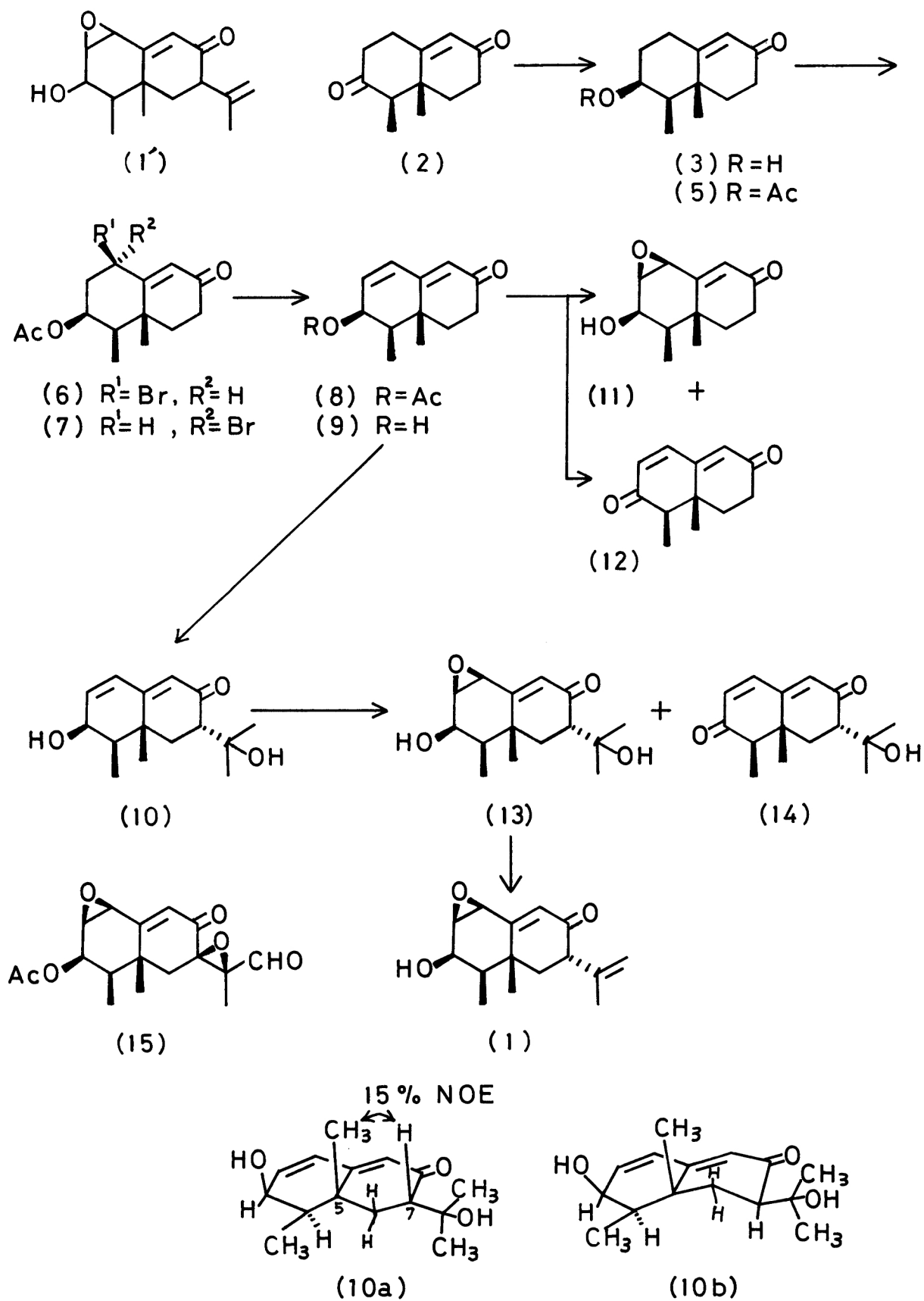
Reduction of the enone (2) with NaBH₄ in MeOH at 0°C for 30 min gave 3β-OH (3) (90%) together with a small amount of 3α-epimer (4) (10%). The configuration of C-3 hydroxyl group of 3 and 4 were determined by their NMR spectra [3: δ 3.94, W_{1/2}=7.5 Hz, 3α-H and 4: δ 3.55, W_{1/2}=25 Hz, 3β-H]. Acetylation of 3 with Ac₂O-pyridine-DMAP gave the 3β-acetate (5), mp 71-2°C [NMR δ: 5.08 (W_{1/2}=7.5 Hz, 3α-H)]. Bromination of 5 with NBS in refluxing CCl₄ for 1.5 hr afforded bromides (70-80%), which were separated by preparative TLC to give 6, mp 96-8°C (28.5%), and 7, mp 102-104°C (44.5%), while the reaction under 30 min reflux gave 6 (44%) and 7 (21%). The bromide (6) was initially formed and then epimerized to the bromide (7) on prolonged heating. The stereochemistry of the bromides (6) and (7) were confirmed to be 1β(axial)- and 1α(equatorial)-bromide, respectively, by their physical

properties [6: MS m/z : 314, 316, M^+ ; UV $\lambda_{\max}^{\text{EtOH}}$ 243.5 nm (ϵ 14000); NMR δ : 4.89 (m, $W_{1/2}=9$ Hz, 1-H), 5.95 (s, 9-H); 7: MS m/z : 314, 316, M^+ ; UV $\lambda_{\max}^{\text{EtOH}}$ 233.5 nm (ϵ 14000); NMR δ : 4.98 (m, $W_{1/2}=21$ Hz, 1-H), 6.41 (d, $J=2$ Hz, 9-H)]. Dehydrobromination of the bromides (6) and (7) with LiBr, Li_2CO_3 -DMF gave the same dienone (8), mp 99-100°C [MS m/z : 234, M^+ ; UV $\lambda_{\max}^{\text{EtOH}}$ 274 nm (ϵ 22000); IR cm^{-1} : 1728, 1655, 1628] in an 81% and 67% yield, respectively. Hydrolysis of 8 with methanolic K_2CO_3 gave the dienone (9), mp 129-130°C, in a 50% overall yield from 5. An alternative preparation of 9 was carried out as follows: bromination of 3 with NBS gave a bromide, mp 112-3°C, and then dehydrobromination under the same conditions as described above gave 9, however the overall yield from 3 was unfavorable (30%).

The dienone (9) was treated with LDA and condensed with acetone in the presence of ZnCl_2^{4a} afforded 10, mp 181-2°C (66%) [IR cm^{-1} : 3360; UV $\lambda_{\max}^{\text{EtOH}}$ 281.5 nm (ϵ 16000)]. Dreiding models revealed that two isomers 10a and 10b were possible to be assigned to 10. In ^1H -NMR of 10, the 7-H was observed as a double doublet at δ 2.61, $J=15$ and 4.5 Hz, and the coupling constants were attributed to trans. When 5- CH_3 protons (δ 1.29) was irradiated, the integration of 7-H showed a 15% NOE enhancement. This fact indicated a 1,3-diaxial relationship between 5- CH_3 and 7-H, and consequently demonstrated that 10 should be shown as the 7α -side chain isomer (10a).

Stereoselective epoxidation of allyl alcohols have been reported,⁵ whereas stereoselective epoxidation of linear dienone allyl alcohols has not yet been reported. Epoxidation of 9 with *tert*-butyl hydroperoxide in the presence of $\text{VO}(\text{acac})_2$ in CH_2Cl_2 , according to the Sharpless' procedure,^{5a} gave $1\beta,2\beta$ -epoxide (11) (17%), mp 117-8°C, together with dienone (12) (54%), mp 103-4.5°C. Treatment of 10 under similar conditions (room temperature, 2 days) gave $1\beta,2\beta$ -epoxide (13) (20%), mp 180.5-2°C, and dienone (14) (54%), mp 125.5-8°C. The epoxidation in benzene containing a small amount of THF gave 13 (24%), 14 (9%), and unchanged 10 (28% recovery), but α -epoxy compound was not detected. Attempted preparation of the β -epoxide (13) from 10 with *m*-CPBA or basic H_2O_2 was unsuccessful.

For the purpose of conversion of 13 into eremofortin B (1), mild cis-elimination conditions were applied to the compound 13 without dehydration of the C-3 hydroxyl group by the trans-elimination. A solution of 13 in benzene containing a small amount of THF with (methylcarboxysulfamoyl)triethylammonium hydroxide inner salt⁶ was warmed at 50°C for 3 hr. The product was separated by prepara-



tive TLC to give (\pm) -1, mp 114-7°C (from hexane-EtOAc), as colorless plates (20%) and unchanged 13 (27%). Compound (\pm) -1: MS m/z : 248, M^+ ; UV $\lambda_{\max}^{\text{EtOH}}$ 241 nm (ϵ 15600); IR cm^{-1} : 3460, 1660, 1610; NMR δ : 1.09 (3H, d, $J=7.5$ Hz, 4-CH₃), 1.35 (3H, s, 5-CH₃), 1.73 (3H, bs, 11-CH₃), 3.26 (1H, dd, $J=13.5, 6.0$ Hz, 7-H), 3.67 (1H, d, $J=3.5$ Hz, 1-H), 3.84 (1H, dd, $J=4.5, 3.5$ Hz, 2-H), 4.09 (1H, m, $W_{1/2}=15$ Hz, 3-H, changed to a triplet with $J=4.5$ Hz on addition of D₂O), 4.82 and 4.97 (each 1H, bs, <H), 6.17 (1H, s, 9-H). The IR and NMR spectra of (\pm) -1 were identical with those of (+)-eremofortin B.^{2,3}

The total synthesis of (+)-PR-toxin (15) isolated from *P. roqueforti* by Wei *et al.*⁷ and the other eremofortins is now in progress.

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