Total Synthesis of Liverwort Sesquiterpene Ketone (\pm) -Chiloscyphone <u>via</u> Intramolecular Aldol Condensation

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 (\pm) -Chiloscyphone, a sesquiterpene ketone isolated from the liverwort, has been synthesized starting from 3,4-dimethyl-2-cyclohexen-1-one <u>via</u> intramolecular aldol condensation.

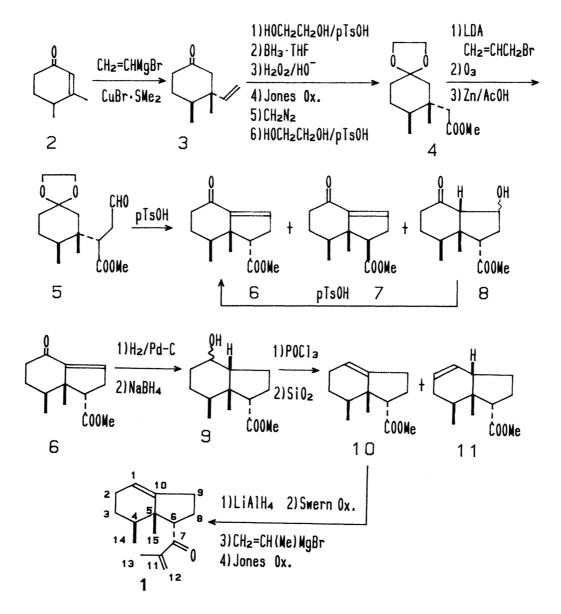
Chiloscyphone $(\underline{1})$ is a sesquiterpene ketone isolated from the liverwort <u>Chiloscyphus polyanthus</u> $(L.)^{1}$ and the structure was revised recently as depicted in $\underline{1}$ by Connolly and his coworkers.²) We have developed the new synthetic route³) to $\underline{1}$ through intramolecular aldol condensation. One of the intermediates could be used for optical resolution and determination of the absolute configuration. We now report the successful synthesis of $(\underline{+})$ -chiloscyphone $(\underline{1})$.

The cis dimethyl part was obtained by 1,4-addition⁴⁾ of the vinyl group to 3,4-dimethyl-2-cyclohexen-1-one (2) in the presence of $CuBr/SMe_2$. Ketal ester 4^{5} was derived from 3 in six steps conversion. Alkylation (LDA/allyl bromide) of 4and ozonolysis (CH₂Cl₂/-78°C) followed by Zn/AcOH treatment afforded ketal aldehyde 5 in a good yield. When 5 was subjected to acidic conditions (pTsOH/acetone-H₂O), three products $(\underline{6}, 5), \underline{7}, 5)$ and $\underline{8}^{5}$) were isolated in 35, 23, and 22% yield, respectively. Compound 6 showed the presence of the β proton of the α , β -unsaturated ketone system [δ 6.54 (dd, J=3.3 and 2.1 Hz)], two methyl groups [δ 1.02 (d, J=3.4 Hz) and 1.07 (s)] and methoxycarbonyl group [δ 3.67 (s)] in its ¹H NMR spectrum. The spectral data of compound $\frac{7}{2}$ also suggested the similar structure to $\frac{6}{2}$, indicating that these two are isomers in each other concerning the 6-position. When the methyl group at either C-4 or C-5 of 6 was irradiated, NOE's into H-6 were observed. While irradiation at H-6 of 7 caused NOE into H-4. These experiments indicated the orientation of the methoxycarbonyl group as depicted in the formulae. The third compound <u>8</u> [IR 3450 cm⁻¹, δ 9.83 (s)] was obviously hydrogen bonded ketol. Since $\underline{8}$ was converted to $\underline{6}$ under the same reaction conditions, the orientation of methoxycarbonyl group was determined to be trans to both methyl groups.

The conversion of enone <u>6</u> to olefin <u>10</u> was accomplished in three steps. Hydrogenation of enone <u>6</u> (H₂/Pd-C) followed by NaBH₄ reduction afforded two isomeric alcohols 9, which were dehydrated (POCl₃/Py) to give a mixture of trisubstituted olefin <u>10</u> [δ 5.39 (1H, m)] and disubstituted olefin <u>11</u> [δ 5.61 (2H, m)]. This mixture was separated by AgNO₃ impregnated silica-gel column chromatography. The olefin <u>10</u> was reduced (LiAlH₄), oxidized (Swern oxidation), alkylated [CH₂=C(Me)MgBr], and oxidized (Jones oxidation) to afford <u>1</u>.⁶)

Optical resolution of $\underline{9}$ should be possible and the absolute configuration

could be determined by conversion to the corresponding ketone. The work along these lines is under way.



References

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- J. D. Connolly, L. J. Harrison, and D. S. Rycroft, J. Chem. Soc., Chem. Commun., <u>1982</u>, 1236.
- 3) K. -G. Gerling and H. Wolf, Tetrahedron Lett., <u>26</u>, 1293 (1985).
- 4) F. E. Ziegler and P. A. Wender, Tetrahedron Lett., 1974, 449.
- 5) All compounds show satisfactory spectral and analytical data.
- 6) The spectral data of the synthetic compound were identical with those reported in Ref. 1.

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