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LETTERS TO THE EDITOR

Stereoselective Synthesis of Sarmentine

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Natural (2E,4E)-dienamides of *Piperaceae* and *Echinacea* species have a broad spectrum of therapeutic action, and in the past decade they attracted a wide attention [1–7]. Sarmentine, the 1-[(2E,4E)-deca-2,4-dienoyl]pyrrolidine, was first isolated from the *Piper sarmentosum* fruits [8]. It possesses sedative, analgesic, and antibacterial activities [9]. Recently it was found to manifest antitubercular and antiplazmo-dial activities [10].

The main task in a sarmentine synthesis is the stereoselective construction of conjugated (2E,4E)-diene system bonded with the amide function. Previously for this purpose were used the elimination [11] and homologenization of (2E,4E)-pentadienyl-1-carbonyl precursors [12, 13], isomerization of alkynyl-amides [14] and stereoselective iodosulfonylation of (2E,4E)-pentadienamide followed by the cross-coupling with *n*-pentylmagnesium bromide [15]. These methods are characterized by the low overall yield of the target product and the low stereoselectivity.

We investigated the possibility of stereoselective synthesis of 1-[(2E,4E)-deca-2,4-dienoyl]pyrrolidine I on the basis of direct cross-coupling of (1E)-1-iodohept-1-ene II with 1-acryloylpyrrolidine III (the Mizoroki–Heck reaction) [16–19]. The initial (1E)-1-iodohept-1-ene II was obtained via the hydroalumination-iodination of 1-heptyne IV in an yield of 88% by the optimized procedure [20]. The other building-block in the cross-coupling reaction, 1-acryloylpyrrolidine III, was synthesized through the acrylic acid chlorodehydroxylation followed by the amidation of acryloyl chloride V with pyrrolidine.

The reaction of (1E)-1-iodohept-1-ene with 1acrciloylpyrrolidine in the presence of Pd(OAc)₂, a base, and tetrabutylammonium chloride in DMF yields quantitatively 1-[(2E,4E)-deca-2,4-dienoyl]pyrrolidine with an insignificant content of isomeric products (1%). The total yield of the desired product is 83% relative to the starting 1-heptyne.



The structure and stereochemical purity of sarmentine I was confirmed by the ¹H and ¹³C NMR spectrometry, GLC analysis, and gas chromatography– mass spectrometry. In the ¹H NMR spectrum of the product I the coupling constants of vinyl hydrogen atom at C^2 atom is 7.14 Hz, which indicates clearly the double bond of transoid configuration.

The IR spectra were recorded from a thin layer on a Prestige-21 Shimadzu IR-Fourier spectrometer. The ¹H and ¹³C NMR spectra were recorded on a Bruker AM-300 instrument operating at 300 and 75.47 MHz, respectively) from solutions in CDCl₃. The chromatographic and mass spectral analysis was performed on a Chromatec-Crystal 5000 complex equipped with a Finnigan DSQ mass-selective detector (electron impact at 70 eV). A Restek Rtx-5ms capillary column (5% diphenylpolysiloxane, 95% dimethylpolysiloxane, length 30 m) was used, the evaporator temperature 250°C, the ionization cell temperature 250°C. The analysis was carried out with programming temperature from 50 to 250°C at a heating rate 10 deg min⁻¹, carrier gas helium (1.1 ml min⁻¹).

(1E)-1-Iodohept-1-ene (II) was obtained by the modified procedure [20]. To a solution of 0.96 g of 1heptyne IV in 10 ml of anhydrous hexane 15 ml of 1 M diisobutyl aluminum hydride solution in hexane was added. The mixture was stirred for 6 h at 55°C in an argon atmosphere and cooled to -50° C. Then to this mixture a solution of 2.79 g of iodine in 10 ml of anhydrous THF was added over 30 minutes. The reaction mixture was warmed within 1 h to room temperature and stirred at this temperature for another 12 h. To the mixture 25 ml of 10% sulfuric acid solution was added under ice-cooling. The organic layer was separated, and the aqueous layer was extracted with hexane (3×15 ml). The combined organic solutions were washed with brine, dried over Na₂SO₄, and concentrated. The product was isolated by the column chromatography (SiO₂, hexane-chloroform, 6:1). Yield 1.98 g (88%). IR spectrum, v, cm^{-1} : 2955, 2924, 2855, 1605, 1458, 1209, 1173, 939. ¹H NMR spectrum, δ, ppm: 0.87 t (3H, CH₃), 1.21–1.43 m (6H, 3CH₂), 2.03 q (2H, CH₂CH=, J 6.9 Hz), 5.96 d $(1H, C^{1}H=, J 14.4 Hz), 6.45-6.54 m (1H, C^{2}H=).$ ¹³C NMR spectrum, δ_{C} , ppm: 13.91 (C⁷), 22.33 (C⁶), 27.96 (C^4) , 31.02 (C^5) , 35.93 (C^3) , 74.24 (C^1) , 146.65 (C^2) . Mass spectrum, m/z (I_{rel} , %): 224 (34) $[M]^+$, 167 (24), 154 (55), 97 (22), 69 (13), 55 (100), 41 (23), 39 (13).

1-Acryloylpyrrolidine (III). To a solution of 1.81 g of acryloyl chloride V in 20 ml of anhydrous dichloroethane was slowly added 2.84 g of pyrrolidine in 15 ml of anhydrous dichloroethane at 0–5°C. The reaction mixture was stirred for 3 h at room temperature. The precipitate was filtered off and washed with dichloromethane $(2 \times 10 \text{ ml})$. The organic layer was successively washed with 10 ml of water, 2 ml of 5% HCl solution, 2 ml of NaHCO₃ saturated solution, and dried over Na₂SO₄. The solvent was removed, and the crude product was purified by the column chromatography (SiO₂, hexane–ethyl acetate, $5:1\rightarrow 1:1$). Yield 1.38 g (55%). IR spectrum, v, cm⁻¹: 2972, 2872, 1647, 1609, 1436, 1375, 982, 797. ¹H NMR spectrum, δ, ppm: 1.65-1.85 m (4H, 2CH₂), 3.31-3.40 m (4H, 2 CH₂N), 5.48 d (1H, CH₂=, J 10.1 Hz), 6.12–6.35 m (2H, CH₂=, CH=). ¹³C NMR spectrum, $\delta_{\rm C}$, ppm: 23.67 (CH₂), 25.53 (CH₂), 45.24 (CH₂N), 45.96 (CH₂N), 126.46 (CH₂=), 128.38 (CH=), 163.76 (C=O). Mass spectrum, m/z (I_{rel} , %): 125 (75) $[M]^+$, 124 (36), 97 (14), 96 (24), 70 (39), 69 (32), 68 (18), 56 (13), 55 (100), 43 (12), 42 (19), 41 (22), 39 (14).

1-[(2E,4E)-Deca-2,4-dienoyl]pyrrolidine (I). To a mixture of 0.46 g of K₂CO₃, 0.37 g of Bu₄NCl, 0.3 g of (1E)-1-iodohept-1-ene II, and 0.25 g of 1-acryloylpyrrolidine III in 0.9 ml of DMF was added a solution of 0.006 g of Pd(OAc)₂ in 0.1 ml of DMF. The reaction mixture was purged with argon and heated for 5 h at 70°C under stirring. After consumption of (1E)-1iodohept-1-ene II (TLC control), 3 ml of water and 3 ml of hexane was added to the mixture. The organic layer was separated, and the aqueous layer was extracted with hexane (2×3 ml). The combined organic solutions were washed with water (5 ml), dried over Na₂SO₄, and concentrated. The crude product was purified by the column chromatography $(SiO_2,$ hexane-ethyl acetate, $9:1\rightarrow 4:6$). Yield 0.28 g (94%). IR spectrum, v, cm⁻¹: 2955, 2926, 2870, 1653, 1624, 1600, 1425, 999. ¹H NMR spectrum, δ, ppm: 0.89 t (3H, CH₃), 1.26–1.47 m (6H, 3 CH₂), 1.81–2.03 m (4H, 2 CH₂CH₂N), 2.15 q (2H, CH₂CH=, J 7.0 Hz), 3.46–3.58 m (4H, 2 CH₂N), 6.02–6.23 m (2H, 2 CH=), 6.10 d (1H, C²H=, J 14.7 Hz), 7.22–7.35 m (1H, CH=). ¹³C NMR spectrum, δ_{C} , ppm: 13.76 (C¹⁰), 22.24 (C⁹), 24.10 (CH₂CH₂N), 25.86 (CH₂CH₂N), 28.23 (C^7), 31.11 (C⁸), 32.69 (C⁶), 45.57 (CH₂N), 46.17 (CH₂N), 119.66 (C²), 128.50 (C⁴), 141.86 (C⁵ or C³), 142.79 (C³ or C⁵), 164.89 (C¹). Mass spectrum, m/z (I_{rel} , %): 221 $(22) [M]^+$, 151 (29), 150 (100), 98 (27), 95 (26), 81 (76), 70 (46), 69 (30), 67 (27), 55 (36), 53 (23), 41 (28).

REFERENCES

- Silva, R.V., Navickiene, H.M.D., Kato, M.J., Bolzani, V.S., Meda, C.I., Young, M.C.M., and Furlan, M., *Phytochemistry*, 2002, vol. 59, p. 521.
- Tsukamoto, S., Tomise, K., Miyakawa, K., Cha, B.C., Abe, T., Hamada, T., Hirota, H., and Ohta, T., *Bioorg. Med. Chem.*, 2002, vol. 10, p. 2981.
- Stohr, J.R., Xiao, P.G., and Bauer, R., J. Ethnopharmacol., 2001, vol. 75, p. 133.
- Venkatasamy, R., Faas, L., Young, A.R., Raman, A., and Hider, R.C., *Bioorg. Med. Chem.*, 2004, vol. 12, p. 1905.
- Reddy, S.V., Srinivas, P.V., Praveen, B., Kishore, K.H., Raju, B.C., Murthy, U.S., and Rao, J.M., *Phytomedicine*, 2004, vol. 11, p. 697.
- 6. Hinz, B., Woelkart, K., and Bauer, R., Biochem. Biophys. Res. Commun., 2007, vol. 360, p. 441.
- Lee, S.W., Kim, Y.K., Kim, K., Lee, H.S., Choi, J.H., Lee, W.S., Jun, C.D., Park, J.H., Lee, J.M., and Rho, M.C., *Bioorg. Med. Chem. Lett.*, 2008, vol. 18, p. 4544.

- 8. Likhitwitayawuid, K., Ruangrungsi, N., Lange, G.L., and Decicco, C.P., *Tetrahedron*, 1987, vol. 43, p. 3689.
- Strunz, G.M., Stud. Nat. Prod. Chem., 2000, vol. 24, p. 683.
- Rukachaisirikul, T., Siriwattanakit, P., Sukcharoenphol, K., Wongvein, C., Ruttanaweang, P., Wongwattanavuch, P., and Suksamrarn, A., *J. Ethnopharmacol.*, 2004, vol. 93, p. 173.
- 11. Mandai, T., Moriyama, T., Tsujimoto, K., Kawada, M., and Otera, J., *Tetrahedron Lett.*, 1986, vol. 27, p. 603.
- 12. Lewis, N., McKen, P.W., and Taylor, R.J.K., *Synlett*, 1991, p. 898.
- 13. Babudri, F., Fiandanese, V., Naso, F., and Punzi, A., *Tetrahedron Lett.*, 1994, vol. 35, p. 2067.

- 14. Trost, B.M. and Kazmaier, U., J. Am. Chem. Soc., 1992, vol. 114, p. 7933.
- 15. Bernabeu, M.C., Chinchilla, R., and Najera, C., *Tetrahedron Lett.*, 1995, vol. 36, p. 3901.
- Handbook of Organopalladium Chemistry for Organic Synthesis, Negishi, E., Ed., New York: Wiley Interscience, 2002.
- 17. Heck, R.F., Org. React., 1982, vol. 27, p. 345.
- Metal-Catalyzed Cross-Coupling Reactions, de Meijere, A. and Diederich, F., Eds., New York: Wiley-VCH, 2004.
- 19. Ishbaeva, A.U., Shakhmaev, R.N., and Zorin, V.V., *Zh. Org. Khim.*, 2010, vol. 46, no. 2, p. 183.
- 20. Zweifel, G. and Whitney, C.C., J. Am. Chem. Soc., 1967, vol. 89, p. 2753.