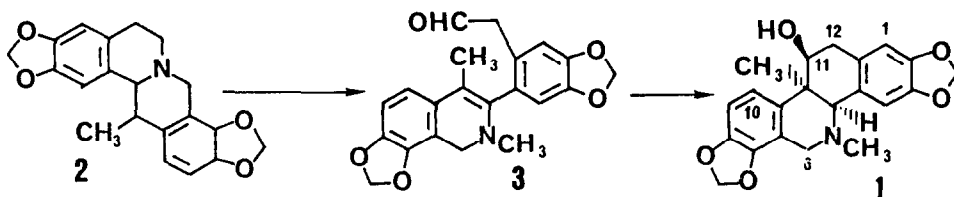


**A BIOMIMETIC SYNTHESIS OF (+)-CORYNOLINE, (+)-11-EPICORYNOLINE,
(+)-ISOCORYNOLINE, AND (+)-11-EPIISOCORYNOLINE FROM CORYSAMINE**

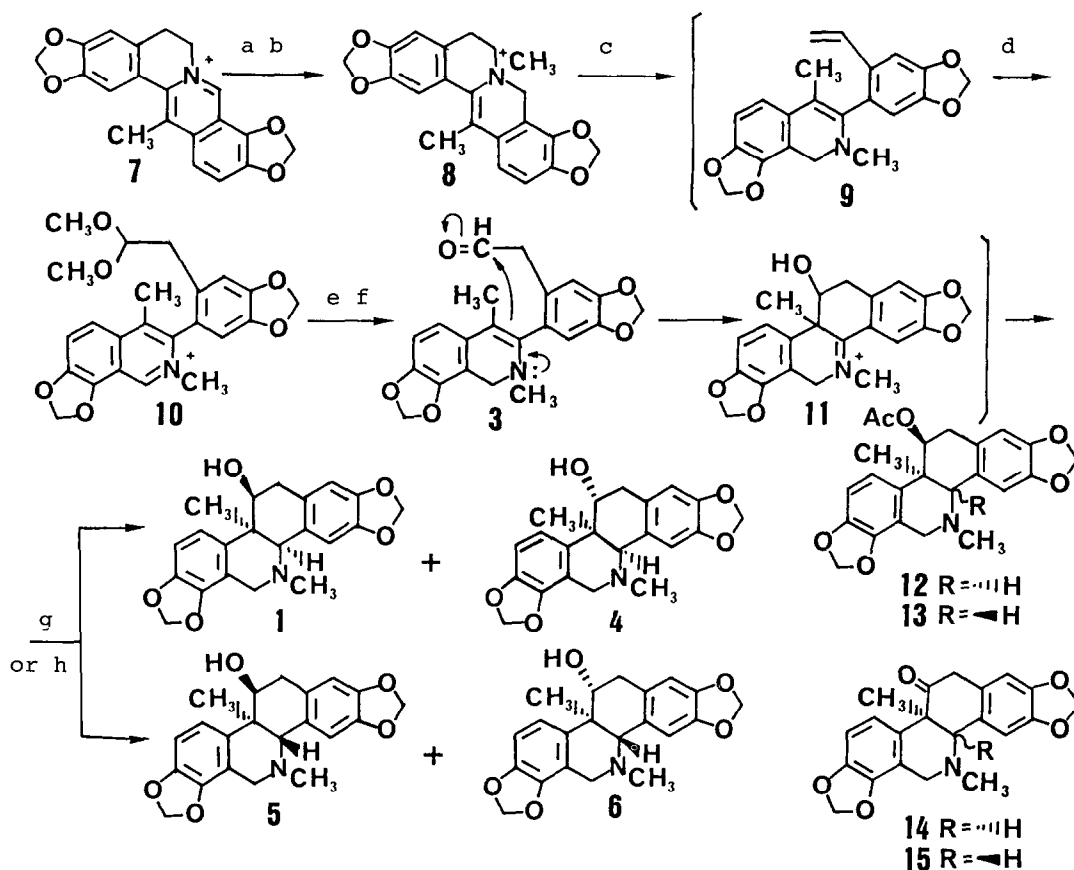
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Summary: A novel and efficient synthesis of hexahydrobenzo[c]phenanthridine alkaloids, (+)-corynoline, (+)-11-epicorynoline, (+)-isocorynoline, and (+)-11-epiisocorynoline was accomplished from protoberberine alkaloid, corysamine through a biogenetic route.

Corynoline (1),¹⁾ a representative 10b-methylhexahydrobenzo[c]phenanthridine alkaloid,²⁾ has been isolated both as a racemate and (+)-form and shown to be biosynthesized³⁾ from the corresponding protoberberine alkaloid, tetrahydrocorysamine (2) via a hypothetical intermediate enamine-aldehyde (3). Three other diastereoisomeric alkaloids, 11-epicorynoline (4),⁴⁾ isocorynoline (5),^{1e,4-6)} and 11-epiisocorynoline (6),^{7,8)} have also been isolated. Several total synthesis of corynoline⁹⁻¹¹⁾ and its related alkaloids such as 12-hydroxycorynoline,⁹⁾ 11-epicorynoline,⁹⁾ 6-oxocorynoline,¹⁰⁾ isocorynoline,^{10,12)} and 11-epiisocorynoline¹⁰⁾ have been reported. However, no report has so far appeared on a synthesis of corynoline and its diastereoisomers according to the above biogenetic process. This communication deals with a first biomimetic transformation of a protoberberine alkaloid, corysamine into corynoline, 11-epicorynoline, isocorynoline, and 11-epiisocorynoline via a proposed biogenetic intermediate, enamine-aldehyde (3).



Oxygenation of the enamine (9),¹³⁾ the Hofmann degradation product of 8 derived from corysamine (7), with thallium (III) nitrate¹⁴⁾ in methanol at



room temperature for 30 min afforded the acetal (10). Successive treatments of the crude product (10) with sodium borohydride at 0°C for a few min, 15% hydrochloric acid in methanol at room temperature overnight, and then sodium cyanoborohydride at 0°C for 1 hr effected reduction of a C-N double bond in 10, deacetalization, cyclization, and further reduction of a C-N double bond in 11 to provide (+)-corynoline (1) [45% from 8; mp 219-220°C (lit.⁹) 218-220°C]; m/z 367 (M^+); 6.92, 6.81 (1H each, AB-q, $J=8.3$ Hz), 6.66 (2H, s), 3.98 (1H, s), 3.38 (1H, d, $J=1.2$ Hz), 2.26 (3H, s), 1.15 (3H, s)] along with (+)-11-epicorynoline (4) [13% from 8; mp 191-192°C (lit.⁴) 195.5-196.5°C]; m/z 367 (M^+); 4.56 (1H, dd, $J=9.5$; 7.1 Hz), 3.22 (1H, s), 2.19 (3H, s), 1.11 (3H, s)] via the enamine-aldehyde (3)³ and the iminium (11)³. Synthetic (+)-corynoline and (+)-11-epicorynoline were identical with natural (+)-corynoline and the authentic (+)-11-epicorynoline, respectively. Acetylation of 1 with acetic anhydride in pyridine afforded (+)-acetylcorynoline (12) [mp 157-158°C

(lit.^{5a}) 158-159°C), m/z 409 (M^+), 5.18 (1H, dd, $J=8.1, 6.7$ Hz), 1.86 (3H, s)].

Reduction of **11** with zinc-acetic acid instead of sodium cyanoborohydride in the last step of the above reaction gave (+)-isocorynoline (**5**) [28% from **8**; mp 204-205°C (lit.¹⁰) 174-176°C); m/z 367 (M^+); 4.51 (1H, s), 4.33 (1H, d, $J=4.9$), 3.22 (1H, dd, $J=17.8, 4.9$), 2.84 (1H, d, $J=17.8$) 2.49 (3H, s), 1.11 (3H, s)] accompanied with (+)-11-epiisocorynoline (**6**) [7% from **8**; mp 185-186°C (lit.^{10,15}) 238-240°C); m/z 367 (M^+); 3.97 (1H, s), 3.10 (1H, dd, $J=17.1, 7.1$), 2.81 (1H, dd, $J=17.1, 10.0$), 2.43 (3H, s), 1.19 (3H, s)] as well as **1** (4%) and **4** (7%). Acetylation of **5** gave (+)-acetylisocorynoline (**13**) [mp 178-180°C, m/z 409 (M^+), 5.52 (1H, dd, $J=5.1, 1.7$ Hz), 1.77 (3H, s)]. Synthetic (+)-isocorynoline was identical with natural (+)-isocorynoline. ¹H-NMR spectral data of synthetic (+)-11-epicorynoline were in good agreement with those reported for natural alkaloid.⁷⁾

The Swern oxidation of **1** or **4** with trifluoroacetic anhydride in dimethyl sulfoxide afforded the ketone (**14**)^{4,5b)} in 77% or 72% yield, respectively. The product was reduced either with lithium aluminum hydride¹⁶⁾ in tetrahydrofuran or with sodium cyanoborohydride in t-butanol in the presence of 15% hydrochloric acid to furnish stereoselectively (+)-corynoline (**1**) in 94% or 91% yield, respectively.¹⁷⁾ On the other hand, similar reduction of the diastereoisomeric ketone (**15**), derived from **5** (84%) or **6** (75%), with lithium aluminum hydride afforded (+)-isocorynoline (**5**) (81%) along with (+)-11-epiisocorynoline (**6**) (10%), whereas that with sodium cyanoborohydride provided **5** (19%) and **6** (54%).

Thus, we developed a novel and biomimetic synthesis of corynoline, 11-epicorynoline, isocorynoline, and 11-epiisocorynoline. As the starting protoberberine alkaloid is readily accessible and the reaction procedure is very simple, the present synthesis provides a general method for a synthesis of hexahydrobenzo[c]phenanthridine alkaloids.

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References and Note

- 1) a) R. H. F. Manske, J. Am. Chem. Soc., **72**, 3207 (1950); b) C. Tani and N. Takao, Yakugaku Zasshi, **82**, 594 (1962); c) N. Takao, Chem. Pharm. Bull.,

- 11, 1306 (1963); d) S. Naruto, S. Arakawa, and H. Kaneko, *Tetrahedron Lett.*, **1968**, 1705; e) T. Kametani, M. Ihara, and T. Honda, *Pytochemistry*, **10**, 1881 (1971); f) N. Takao, K. Iwasa, M. Kamigauchi, and M. Sugiura, *Chem. Pharm. Bull.*, **24**, 2859 (1976); g) N. Takao, M. Kamigauchi, and K. Iwasa, *Tetrahedron*, **35**, 1977 (1979).
- 2) V. Šimánek, "The Alkaloids," Vol. 26, ed. by A. Brossi, Academic Press, New York, 1985, p185.
- 3) A. Yagi, G. Nonaka, S. Nakayama, and I. Nishioka, *Phytochemistry*, **16**, 1197 (1977); E. Leete and S. J. B. Burrill, *ibid.*, **6**, 231 (1967); A. R. Battersby, J. Staunton, M. C. Summers, and R. Southgate, *J. Chem. Soc., Perkin Trans. 1*, **1979**, 45; N. Takao, M. Kamigauchi, and M. Okada, *Helv. Chim. Acta*, **66**, 473 (1983). and references cited therein.
- 4) G. Nonaka and I. Nishioka, *Chem. Pharm. Bull.*, **23**, 521 (1975).
- 5) a) G. Nonaka, H. Okabe, I. Nishioka, and N. Takao, *Yakugaku Zasshi*, **93**, 87 (1973); b) N. Takao, H. W. Bersch, and S. Takao, *Chem. Pharm. Bull.*, **21**, 1096 (1973).
- 6) This alkaloid is also named as 14-epicorynoline.
- 7) W. Zeng, W. Liang, C. He, Q. Zheng, and G. Tu, *Phytochemistry*, **27**, 599 (1988).
- 8) This alkaloid was originally named as 13-epicorynoline, however, it is better to designate it as 11-epiisocorynoline.
- 9) I. Ninomiya, O. Yamamoto, and T. Naito, *J. Chem. Soc., Perkin Trans. 1*, **1980**, 212.
- 10) M. Cushman, A. Abbaspour, and Y. P. Gupta, *J. Am. Chem. Soc.*, **105**, 2873 (1983).
- 11) M. Cushman and J.-K. Chen, *J. Org. Chem.*, **52**, 1517 (1987).
- 12) J. R. Falck, S. Manna, and C. Mioskowski, *J. Am. Chem. Soc.*, **105**, 631 (1983).
- 13) M. Hanaoka, S. Yoshida, and C. Mukai, *J. Chem. Soc., Chem. Commun.*, **1984**, 1703.
- 14) A. Mckillop, J. D. Hunt, F. Kienzle, E. Bigham, and E. C. Taylor, *J. Am. Chem. Soc.*, **95**, 3635 (1973).
- 15) The spectral data of the previous synthetic product¹⁰⁾ are not coincident with those of natural product.⁷⁾
- 16) cf. M. Onda, K. Yuasa, and J. Okada, *Chem. Pharm. Bull.*, **22**, 2365 (1974).
- 17) The Meerwein-Ponndorf Reduction of the ketone (**14**) derived from natural corynoline afforded **1** and **4** in a ratio of 1:5.^{5b)}

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