Synthesis of Some 4-Oxo-1-phenyl-5,6-dihydro-1*H*,4*H*-pyrano[2,3-*c*]pyrazole Derivatives

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In recent years, a number of studies have been published on the *C*-alkylation of 2-pyrazolin-5-ones 1^{1-5} , but $\alpha.\beta$ -unsaturated acyl chlorides have not been used. The present work describes the synthesis of 4-oxo-1-phenyl-5,6-dihydro-1*H*,4*H*-pyrano[2,3-c]pyrazole derivatives 4 by *C*-acylation of compounds 1 with $\alpha.\beta$ -unsaturated acyl chlorides 2.

Treatment of 2-pyrazolin-5-ones 1 with magnesium ethoxide in tetrahydrofuran generates the corresponding anions which, on reaction with the alkenoyl chlorides 2, afford the 4-acylated derivatives 3. Compounds 3 are easily converted to 4 by treatment with concentrated sulfuric acid at room temperature without prior purification. The structures of products 4 are confirmed by microanalytical, 1.R., U.V., and ¹H-N.M.R. spectral data (Table).

1.
$$\frac{Mg(OC_2H_5)_2}{R^3}C = C \frac{CO - CI}{R^4}$$
2. $\frac{R^3}{R^2}C = C \frac{CO - CI}{R^4}$
2. $\frac{R^3}{R^2}C = C \frac{CO - CI}{R^4}$
3. $\frac{R^3}{R^2}C = C \frac{R^4}{R^4}$
3. $\frac{R^3}{R^2}C = C \frac{R^4}{R^4}$
3. $\frac{R^3}{R^2}C = C \frac{R^4}{R^4}$
4. $\frac{R^4}{R^3}C = C \frac{R^4}{R^4}$
4. $\frac{R^4}{R^3}C = C \frac{R^4}{R^4}$

3-Methyl- (1a)⁶, 3-n-propyl- (1b)⁷, and 3-phenyl- (1c)⁸ 1-phenyl-2-pyrazolin-5-ones are prepared according to the procedures described in the literature.

4-Oxo-1-phenyl-5,6-dihydro-1*H*,4*H*-pyrano[2,3-c]pyrazole 4, General Procedure:

A mixture of 1 (0.01 mol), magnesium ethoxide (1.14 g, 0.01 mol), and dry tetrahydrofuran (50 ml) is magnetically stirred and heated to reflux for 4 h. The mixture is then cooled to 0-5°C with an ice/water bath and a solution of the $\alpha.\beta$ -unsaturated acyl chloride 2 (0.01 mol) in tetrahydrofuran (25 ml) is added dropwise with efficient stirring. The cooling bath is removed and the mixture is allowed to stand at room temperature for 3 h, then poured on to cold 10% hydrochloric acid (200 ml), and extracted with chloroform (3×50 ml). The combined extracts are washed with water (2×50 ml), dried with sodium sulfate, and rotary-evaporated. The residue (crude compounds 3) is cooled and concentrated sulfuric acid is added (50 ml) with stirring. The mixture is allowed to stand at room temperature overnight, then poured on to ice/water (400 g), and extracted with chloroform (3×50

Table. 4-Oxo-1-phenyl-5,6-dihydro-1H,4H-pyrano[2,3-c]pyrazoles 4 prepared

Produ No.	ict R ¹	\mathbb{R}^2	\mathbb{R}^3	R ⁴	Yield [%]	m.p. [°C] (solvent)	Molecular formula ^a	I.R. (CHCl ₃) $v_{C==0}$ [cm ⁻¹]	U.V. (C_2H_5OH) λ_{max} $[nm]$ (ε)	1 H-N.M.R. b (CDCl ₃) δ [ppm]
4a	CH ₃	CH ₃	Н	Н	60	93-94° (hexane)	C ₁₄ H ₁₄ N ₂ O ₂ (242.3)	1680	256 (20600)	1.60 (d, 3 H, J =6 Hz); 2.50 (s, 3 H); 2.53 (s, 1 H); 2.62 (dd, 1 H, J_{AB} = 16 Hz) ^{ct} ; 4.9 (m, 1 H); 7.3-7.7 (m, 3 H); 7.7-7.9 (m, 2 H)
4b	<i>n</i> -C ₃ H ₇	CH ₃	Н	Н	65	97-98° (hexane)	$C_{16}H_{18}N_2O_2$ (270.3)	1680	256 (22600)	1.00 (t, 3 H, J =7 Hz); 1.60 (d, 3 H, J =6 Hz); 1.77 (sex, 2 H, J =7 Hz) $^{\circ}$; 2.52 (s, 1 H); 2.61 (dd, 1 H, J _{AB} =16 Hz) d : 2.83 (t, 2 H, J =7 Hz); 4.9 (m, 1 H); 7.25-7.6 (m, 3 H); 7.7-7.9 (m, 2 H)
4c	C ₆ H ₅	CH ₃	Н	Н	85	134-135° (2:1 C ₂ H ₅ OAc/ hexane)	C ₁₉ H ₁₆ N ₂ O ₂ (304.3)	1680	254 (31500)	1.55 (d, 3 H, J =6 Hz); 2.56 (s, 1 H); 2.65 (dd, 1 H, J _{AB} =16 Hz) ^d ; 4.9 (m, 1 H); 7.25-7.65 (m, 6 H); 7.75-8.0 (m, 2 H); 8.15-8.4 (m, 2 H)
4d	CH ₃	Н	Н	CH ₃	35°	110-111° (hexane)	$\begin{array}{c} C_{14}H_{14}N_2O_2\\ (242.3) \end{array}$	1675	256 (19 400)	(m, 21); 2.48 (d, 3 H, J = 7 Hz); 2.48 (s, 3 H); 2.80 (m, 1 H); 4.38 (dd, 1 H, J_{AB} = 11 Hz, J_{AX} = 10 Hz); 4.75 (dd, 1 H, J_{AB} = 11 Hz, J_{AX} = 5 Hz); 7.3–7.6 (m, 3 H); 7.7–7.9 (m, 2 H)
4e	C ₆ H ₅	Н	Н	CH ₃	50°	133-132° (C ₂ H ₅ OH)	C ₁₉ H ₁₆ N ₂ O ₂ (304.3)	1685	254 (29 900)	1.26 (d, 3 H, $J=7$ Hz); 2.85 (m, 1H); 4.41 (dd, 1 H, $J_{AB}=11$ Hz, $J_{AX}=10$ Hz); 4.77 (dd, 1 H, $J_{AB}=11$ Hz, $J_{BX}=5$ Hz) ¹ ; 7.2-7.7 (m, 6H); 7.8-8.0 (m, 2 H); 8.15-8.5 (m, 2 H)
4f	CH ₃	CH ₃	CH ₃	Н	69	122-123° (1:2 C ₂ H ₅ OAc/ hexane)	$C_{15}H_{16}N_2O_2$ (256.3)	1675	260 (21 200)	1.58 (s, 6 H); 2.50 (s, 3 H); 2.61 (s, 2 H); 7.25–7.6 (m, 3 H); 7.7–7.85 (m, 2 H)
4g	C ₆ H ₅	CH ₃	CH ₃	Н	40	143-144° (1:4 C ₂ H ₅ OAc/ hexane)	$C_{20}H_{18}N_2O_2$ (318.3)	1675	255 (32 500)	1.61 (s, 6H); 2.71 (s, 2H); 7.3-7.65 (m, 6H); 7.8-8.0 (m, 2H); 8.25-8.45 (m, 2H)

^a The microanalyses were in satisfactory agreement with the calculated values (C ± 0.21 ; H ± 0.19 ; N ± 0.13).

ml). The combined extracts are washed with 5% aqueous sodium carbonate $(2\times30 \text{ ml})$, water $(2\times30 \text{ ml})$, dried with sodium sulfate, and evaporated to give the crude compounds 4. Purified compounds are obtained in the case of 4a, b, c, f, and g by recrystallization from a suitable solvent (Table) and in the case of 4d, by column chromatography on silica gel with ether as eluent.

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^b 80-MHz Bruker Spectrometer.

^c Yield after column chromatography.

^d AB part of degenerated ABX system.

^e Partially masked by the CH₃ group at C-6.

AB part of ABX system, in first order treatment.

B. S. Jensen, Acta Chem. Scand. 13, 1668 (1959).

A. S. Sarenko, I. Ya. Kvitko, L. S. Efros, Khim. Geterotsikl. Soedin 6, 799 (1972); C. A. 77, 88384 (1972).

³ H. A. DeWald, S. Lobbestael, B. P. H. Poschel, J. Med. Chem. 24, 982 (1981) and references cited therein.

Japan Kokai, 81 63 964, Kumiai Chemical Industry Co., Ltd. (1981); C. A. 96, 6719 (1982).

⁵ A. Terebenina, N. Iordanov, B. Iordanov, G. Borisov, *Izv. Akad. Nauk SSSR Ser. Khim.* 14, 118 (1981); C. A. 96, 52225 (1982).

⁶ D. Biquard, P. Grammaticakis, Bull. Soc. Chim. Fr. 8, 246 (1941).

⁷ A. O. Zoss, G. F. Hennion, J. Am. Chem. Soc. 63, 1151 (1941).

⁸ V. H. Wallingford, A. H. Homeyer, U. S. Patent 2407 942 (1946); C. A. 41, 1699 (1947).