## INSIDE-OUTSIDE STEREOISOMERISM III1:

## THE SYNTHESIS OF TRANS-BICYCLO[4.3.1]DECAN-10-ONE+

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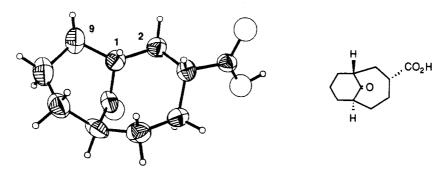
Summary: The synthesis of <a href="mailto:trans-bicyclo">trans-bicyclo</a>[4.3.1]decan-10-one, the smallest bicycloalkane to possess the inside-outside stereochemical relationship, is described, in which the <a href="mailto:trans">trans</a> intrabridgehead stereochemistry is established <a href="mailto:via">via</a> the intramolecular dioxolenone photocycloaddition.

We have recently reported that the intramolecular photoaddition of dioxolenones to alkenes, i.e, 1--3, 5 leads to the formation of six-, seven- and eight-membered rings in good yield, and that this reaction can be applied to the synthesis of bicycloalkanes with the inside-outside intrabridgehead stereochemical relationship. 6,7 In this Letter, we describe the synthesis and characterization of trans-bicyclo[4.3.1]decan-10-one, 4, a compound which is ca. 20 kcal/mole more strained than the corresponding cis-bicyclodecanone, 5.

The photosubstrate  $\underline{8}$  was prepared as outlined below. Bianion alkylation of t-butyl cyclopentanone 2-carboxylate with 4-pentenyl iodide (89% yield), followed by dioxolenone formation (acetic anhydride, acetone, trifluoroacetic acid, room temp., 58% yield) provided  $\underline{8}$ . In striking contrast to the irradiation of  $\underline{12}$ , which resulted in

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the exclusive formation of photoproduct, 13, with the inside-outside stereochemistry, irradiation of 8 (0.02M in 1:9 acetone/acetonitrile; pyrex immersion well; 0°C) for four hours resulted in the formation of two diastereomeric photoadducts, which, upon fragmentation (0.1 equivs. p-TsOH/methanol/reflux/72 hrs.), led to a 1:5 mixture of ketoesters  $\underline{10}$  and  $\underline{11}$  (as a mixture of epimers, vide infra) in 65% overall yield. To determine the intrabridgehead stereochemical relationships of  $\underline{10}$  and  $\underline{11}$ , the separated ketoesters were submitted to ester hydrolysis (3 equivs. 1 M aqueous lithium hydroxide/methanol/tetrahydrofuran, 25°C), acid chloride formation (10 equivs. oxalyl chloride, catalytic dimethylformamide, benzene, 25°C, 1 h) and Barton decarboxylation (1.2 equivs. of the sodium salt of 2-mercaptopyridine-1-oxide, 0.1 equivs. dimethylaminopyridine, 10 equivs. tert-butyl thiol, diethyl ether, 100W sun lamp, two hours)  $^{12}$  to provide the bicyclo[4.3.1] decanones,  $\underline{4}$  and  $\underline{5}$ , respectively. Inspection of the  $^{13}\text{C}$  NMR of these compounds led to the assignment of the major product, 5 [IR = 1692]  $cm^{-1}$  (CHCl<sub>3</sub>);  $^{13}C$ -NMR (CDCl<sub>3</sub>): 19.6, 27.6, 31.7, 33.1, 48.7, 218.0] as the outside-outside <u>cis</u>-bridged compound and the minor product,  $\underline{4}$  [IR = 1748 cm<sup>-1</sup> (CHCl<sub>3</sub>);  $^{13}$ C-NMR (CDCl<sub>2</sub>): 21.0, 25.5, 28.1, 29.3, 31.9, 34.9, 35.3, 49.5, 49.6, 219.1] as the trans-bridged stereoisomer, which contains neither a plane nor an axis of symmetry. Unambiguous proof for the inside-outside intrabridgehead stereochemical relationship followed from the single crystal X-ray analysis  $^{13}$  of 10 (R = H), the most striking feature of which is that, to accomodate the inside-outside stereochemical relationship, the C9-C1-C2 bond angle in 4 is  $130^{\circ}$ .



The dramatic increase in strain energy difference between <u>cis</u>- and <u>trans</u>-bicyclo[5.3.1]undecanones (ca. 10 kcal/mol) and <u>cis</u>- and <u>trans</u>-bicyclo[4.3.1]undecanones (ca. 20 kcal/
mol)<sup>14</sup> leads to a striking difference in the stereoselectivity of the photocycloadditions
of <u>8</u> and <u>12</u>. In the photocyclization of <u>12</u>, the exclusive formation of the inside-outside
photoadduct, <u>13</u>, was explained <u>via</u> a chair-like six-membered ring transition state in the
photocycloaddition.<sup>6</sup> It was not expected that this orientation should be particularly
sensitive to the ketoester ring size, i.e., <u>8</u> instead of <u>12</u>. However, photocycloaddition
of <u>8</u> results in the predominant formation of the "outside-outside" or <u>cis</u>-bridged bicyclic
products. A later transition state for the photocycloaddition of <u>8</u> than <u>12</u>, reflecting the
difference in stability between the <u>cis</u>- and <u>trans</u>-bridged products, would account for the
predominant formation of the more stable <u>cis</u>-bridged stereoisomer in the irradiation of <u>8</u>.

In conclusion, we note again that the intramolecular photocycloaddition of dioxolenones has important advantages over the more classical deMayo diketone sequence.  $^{15,16}$  Aside from the benefits of regiochemical control afforded by the use of the  $\beta$ -ketoesters, this new methodology makes possible the synthesis of <u>trans</u>-bicyclo[n.3.1]alkanones, which cannot otherwise be prepared.

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## References:

- + Dedicated to Professors Gerhard Closs and N. C. Yang on the occasion of their sixtieth birthdays.
- For the previous paper in this Inside-Outside Stereoisomerism series, see J. Winkler, K. Henegar, P. Williard, <u>J. Am. Chem. Soc.</u>, <u>109</u> 2850 (1987).
- Recipient of a Merck Grant for Faculty Development (1985-1986) and a National Institutes of Health Research Career Development Award (CA01337) (1988-1993). Fellow of the Alfred P. Sloan Foundation (1987-1989).
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- 5. J. Winkler, J. Hey, F. Hannon, P. Williard, Heterocycles 25 55 (1987).
- 6. J. Winkler, J. Hey, P. Williard, J. Am. Chem. Soc. 108 6425 (1986).
- 7. For a recent review, see R. Alder, <u>Acc. Chem. Res. 16</u> 321 (1983). For previous syntheses of inside-outside bicycloalkanes, see a) P. Gassman, S. Korn, T. Bailey, T. Johnson, J. Finer, J. Clardy, <u>Tet. Lett.</u> 3401 (1979); b) A. Haines, P. Karntiang, <u>J. Chem. Soc., Perkin I</u> 2577 (1979); c) P. Gassman, R. Thummel, <u>J. Am. Chem. Soc. 94</u> 7183 (1972); d) C. Park, H. Simmons, <u>J. Am. Chem. Soc. 94</u> 7184 (1972); e) J. McMurry, C. Hodge, C. <u>J. Am. Chem. Soc. 106</u> 6450 (1984).
- 8. All new compounds were characterized by full spectroscopic (NMR, IR, MS) data. Yields refer to spectroscopically and chromatographically homogeneous (>95%) materials.
- 9. S. Huckin, L. Weiler, L. J. Am. Chem. Soc. 96 1082 (1974).
- 10. D. Banerjee, S. Mahapatra, Tetrahedron 11 234 (1960).
- 11. M. Sato, H. Ogasawara, K. Oi, T. Kato, Chem. Pharm. Bull. 31 1896 (1983).
- 12. D. Barton, D. Crich, W. Motherwell, J. Chem. Soc., Chem. Comm. 939 (1983).
- 13. Data were collected on a Nicolet R3m X-ray diffractometer. All crystallographic computations were carried out using the SHELXTL programs. Lattice constants a = 7.077(2) A, b = 6.361(2) A, c = 23.050(4) and  $\beta$ = 97.12(2)°, monoclinic (P 2<sub>1</sub>/a). R = 0.0615, R<sub>w</sub>=0.0873.
- 14. Calculated using the Gajewski/Gilbert modification of the Allinger MM2 program (#395, Quantum Chemistry Program Exchange, Indiana University), which is commercially available through Serena Software, Bloomington, IN.
- 15. a)P. de Mayo, <u>Pure and Appl. Chem. 9</u> 597 (1964); b) P. de Mayo, <u>Acc. Chem. Res. 4</u> 41 (1971). For a recent review on the intramolecular version of this reaction, see W. Oppolzer, <u>Accts. Chem. Res. 15</u> 135 (1982).
- 16. a) M. Begley, M. Mellor, G. Pattenden, <u>J. Chem. Soc.</u>, <u>Perkin Trans</u>. <u>I</u> 1905 (1983), and references cited therein.

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