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pathway in Rhodococcus jostii RHA1

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In vitro reconstitution of the catabolic reactions catalyzed by PcaHG, PcaB, and PcaL: the protocatechuate branch of the β -ketoadipate pathway in *Rhodococcus jostii* RHA1

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The β -ketoadipate pathway is a major pathway involved in the catabolism of the aromatic compounds in microbes. The recent progress in genome sequencing has led to a rapid accumulation of genes from the *B*-ketoadipate pathway in the available genetic database, yet the functions of these genes remain uncharacterized. In this study, the protocatechuate branch of the *β*-ketoadipate pathway of Rhodococcus jostii was reconstituted in vitro. Analysis of the reaction products of PcaHG, PcaB, and PcaL was achieved by high-performance liquid chromatography. These reaction products, β-ketoadipate enol-lactone, 3-carboxy-cis, cis-muconate, γ -carboxymuconolactone, muconolactone, and β-ketoadipate, were further characterized using LC-MS and nuclear magnetic resonance. In addition, the in vitro reaction of PcaL, a bidomain protein consisting of y-carboxy-muconolactone decarboxylase and β-ketoadipate enol-lactone hydrolase activities, was demonstrated for the first time. This work provides a basis for analyzing the catalytic properties of enzymes involved in the growing number of β-ketoadipate pathways deposited in the genetic database.

Key words: β-ketoadipate pathway; *Rhodococcus jostii* RHA1; protocatechuate

Aromatic compounds, one of the most abundant classes of naturally occurring compounds, are primarily found in microbial cells as degradation products of plant-derived molecules such as lignins.¹⁾ Microorganisms have developed the ability to use an impressive variety of aromatic compounds as carbon and energy sources.^{2,3)} The β -ketoadipate pathway is one of the main biodegradation pathways for aromatic compounds in aerobic soil bacteria and fungi.²⁾ This pathway includes the protocatechuate and catechol branches, which catabolize protocatechuate (1) and catechol, respectively. Both branch converges at β -ketoadipate enol-lactone (2) (Fig. 1). 2 is then further catabolized to the tricarboxylic acid cycle intermediates, acetyl-CoA and succinyl-CoA, by downstream enzymes of the β-ketoadipate pathway.²⁾ Since the 1940s, the enzymology of the B-ketoadipate pathway of Pseudomonas and Acinetobacter has been extensively studied,⁴⁾ leading to the unraveling of this complex reaction pathway and the chemical structures of the intermediates involved, including 2, 3-carboxy-cis, cis-muconate (3), γ -carboxymuconolactone (4), and muconolactone (5) (Fig. 1). 51 The protocatechuate branch of the β-ketoadipate pathway begins with an ortho-cleavage of 1 by PcaHG, a protocatechuate 3,4-dioxygenase (Fig. 1),6,7) resulting in the formation of 3. PcaB, a 3-carboxy-cis, cis-muconate cycloisomerase, catalyzes the lactonization of 3 to form **4**.^{8,9)} Decarboxylation of **4** is catalyzed by PcaC, a γ -carboxy-muconolactone decarboxylase,¹⁰⁾ but this reaction can also occur spontaneously.¹¹) The resulting decarboxylated product, 2, can be isomerized to form 5, and the two exist in equilibrium.¹¹⁾ PcaD, β-ketoadipate enol-lactone hydrolase, also known as 3-oxoadipate-enol lactonase, hydrolyzes **2** to form β -ketoadipate ($\hat{\mathbf{6}}$).^{12,13}

Rhodococcus jostii RHA1 was first isolated from y-hexachlorocyclohexane-contaminated soil and initially characterized for its polychlorinated biphenyl-degrading properties.¹⁴⁾ Genome sequence analyses of *R. jostii* RHA1 further indicated the potential ability of R. jostii RHA1 to degrade a broad range of aromatic compounds.¹⁵⁾ For example, Hara et al. demonstrated the growth of R. jostii RHA1 on both phthalate and terephthalate through the use of pad and tpa genes, which encode the enzymes for the initial steps of phthalate and terephthalate degradation, respectively. These gene clusters were identified on two large linear plasmids of R. jostii RHA1, namely, pRHL1 and pRHL2.¹⁶⁾ By proteomic analysis, Patrauchan et al. confirmed the translation of the *pad* genes, *pcaB*, *pcaL*, and *pcaF* of *R*. *jostii* RHA1 grown on phthalate.¹⁷⁾ In addition, they detected the protocatechuate 3,4-dioxygenase activity from the cell-free extracts of R. jostii RHA1 grown on phthalate.17)

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COOH spontaneously COOH COOH or PcaHG PcaB PcaL ĊOOH 0 Ò ĊOOH ĊOOH ĊООН PcaC PcaD OН COOH COOH ÔН Ô Ο $\begin{array}{ccc} 3\text{-carboxy-}\\ cis, cis-muconate (\textbf{3}) & muconolactone (\textbf{4}) \end{array}$ β -ketoadipate enol-lactone (2) β -ketoadipate (6) protocatechuate (1)muconolactone (5)

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Fig. 1. The protocatechuate branch of the β-ketoadipate pathway from *R. jostii* RHA1.

Note: PcaL, instead of PcaC and PcaD, is responsible of transfer from 4 to 6 via 2 in R. jostii RHA1.

In *R. jostii* RHA1, the chromosomal *pca* cluster consists of genes predicted to encode the enzymes Pca-JIHGBLF,^{15,17)} which are required to convert **1** to acetyl-CoA and succinyl-CoA. The genes *pcaH* and *pcaG* encode the two subunits of protocatechuate 3,4-dioxygenase,¹⁵⁾ and *pcaB* encodes a 3-carboxy-*cis*,*cis*-muconate lactonizing enzyme.¹⁵⁾ *pcaL* is a fused gene encoding for an enzyme with both γ -carboxy-mucono-lactone decarboxylase and β -ketoadipate enol-lactone hydrolase activities (Fig. 1).^{15,17,18)}

In this study, the protocatechuate branch of the β -ketoadipate pathway from *R. jostii* RHA1 was reconstituted *in vitro* using recombinant PcaHG, PcaB, and PcaL. The bifunctional nature of PcaL was characterized biochemically for the first time. All compounds of the protocatechuate branch, including the highly unstable compound, **3**, were characterized by high-performance liquid chromatography (HPLC), nuclear magnetic resonance (NMR), and liquid chromatography-electrospray ionization/high-resolution mass spectrometry (LC-ESI/HRMS) analyses.

Materials and methods

Chemicals and bacterial strains. Protocatechuate was purchased from Wako Pure Chemical Industries, Ltd. (Osaka, Japan). Luria-Bertani (LB) Lennox medium was purchased from Sigma-Aldrich (St. Louis, MO, USA). *Escherichia coli* HST08, the pColdI plasmid, restriction enzymes, and other DNA-modifying enzymes used for DNA manipulation were purchased from Takara Bio Inc (Shiga, Japan). *E. coli* BL21 was purchased from the National BioResource Project (National Institute of Genetics of Japan). *R. jostii* RHA1 was obtained from Prof. Masao Fukuda. All other reagents were purchased from Wako Pure Chemicals.

Construction of pColdI-pcaHG, pColdI-pcaB, and pColdI-pcaL. Media, growth conditions, and general recombinant DNA techniques of *E. coli* were described by Sambrook et al.¹⁹ The chromosomal DNA of *R. jostii* RHA1 was used as a template for PCR experiments. The

accession numbers of the amino acid sequences of PcaG, PcaH, PcaB, and PcaL are ABG93160, ABG93159, ABG93161, and ABG93162, respectively. An NdeI site was introduced at the start codon of *pcaHG* by PCR with primer I: 5'-CGGAATTCCATATGCTGCATCTGCCA-GCCC-3' (the EcoRI site is underlined; the NdeI site is italicized), and primer II: 5'-CGCAAGCTTCGA-CACCTTTCTGCGTTGTG-3' (the HindIII site is underlined). The amplified fragments were cloned between the EcoRI and HindIII sites of pUC19, resulting in pUC19pcaHG. An NdeI site was introduced at the start codon of pcaB by PCR with primer III: 5'-CGGAATTCCA-TATGAACTCTCCGGAGCCTT-3' (the EcoRI site is underlined; the NdeI site is italicized), and primer IV: 5'-CGCAAGCTTGTGTGCGAGTGCGACTG-3 (the HindIII site is underlined). The amplified fragments were cloned between the EcoRI and HindIII sites of pUC19, resulting in pUC19-pcaB. An NdeI site was introduced at the start codon of pcaL by PCR with primer V: 5'-CGTCTAGACATATGACAGTCGCACTCGCA-3' (the XbaI site is underlined; the NdeI site is italicized), and primer VI: 5'-CGCAAGCTTCGTCGGTTCGGTAG-CGG-3' (the HindIII site is underlined). The amplified fragments were cloned between the XbaI and HindIII sites of pUC19, resulting in pUC19-pcaL. DNA sequencing of pUC19-pcaHG, pUC19-pcaB, and pUC19-pcaL confirmed the correct sequence. Each NdeI-HindIII fragment excised from pUC19-pcaHG, pUC19-pcaB, and pUC19-pcaL was cloned between the NdeI and HindIII sites of pColdI, resulting in pColdI*pcaHG*, pColdI-*pcaB*, and pColdI-*pcaL*, respectively.

Expression and purification of recombinant PcaHG, PcaB, and PcaL. E. coli BL21 harboring pColdIpcaHG, pColdI-pcaB, or pColdI-pcaL was grown at 37 °C in 50 mL of LB broth containing 100 μ g mL⁻¹ of ampicillin. When an optical density at 600 nm reached 0.5, cells were incubated for 30 min at 15 °C and then protein expression was induced by the addition of 0.1 mM isopropyl β-D-thiogalactoside. Cells were then cultured for an additional 24 h at 15 °C. Cells were harvested by centrifugation and resuspended in 50 mM NaH₂PO₄ (pH 8.0), 300 mM NaCl, 10 mM

3

imidazole, and 10% glycerol. The resuspended cells were collected by centrifugation, and a crude cell lysate was prepared by sonication and cell debris was removed by centrifugation at $20,000 \times g$ for 20 min. The cleared lysate was applied to Ni-nitrilotriacetic acid columns (Qiagen, Hilden, Germany), washed twice with 50 mM NaH₂PO₄ (pH 8.0), 300 mM NaCl, 20 mM imidazole, and 10% glycerol. The purified histidine-tagged protein was eluted with a buffer containing 50 mM NaH₂PO₄ (pH 8.0), 300 mM NaCl, 250 mM imidazole, and 10% glycerol and was dialyzed two times against 2 L of 10 mM Tris-HCl (pH 7.5) containing 10% glycerol. The protein concentration was measured by Bradford assay with a Bio-Rad Protein Assay Kit (Bio-Rad Laboratories, Hercules, CA, USA) using bovine serum albumin as a standard. The purity of N-terminal His-tagged (MNHKVHHH-HHHIEGRH) PcaHG, PcaB, and PcaL was verified by SDS analysis (Supplemental Fig. S1). The proteins were stored at -80 °C before use.

In vitro reactions with protocatechuate. The enzymatic reactions, containing 1 µg each of PcaHG, PcaB, and/or PcaL, 1 mM of protocatechuate, and 50 mM Tris-HCl (pH 7.5), were performed in a total volume of 300 µL. The boiled enzymes were used as negative controls. The reaction mixtures were incubated at 30 °C for 5 min and were stopped by filtration with Amicon Ultra-0.5 Centrifugal Filters (Merck Millipore, Darmstadt, Germany). Aliquots (5 µL) of the filtrate were directly analyzed via HPLC (JASCO International Co., Tokyo, Japan). Reverse-phase HPLC analysis was carried out using a COSMOSIL 5C18-PAQ column (4.6 × 150 mm; Nacalai Tesque Inc., Kyoto, Japan), and the analytes were subjected to an isocratic elution with 3% acetonitrile in water (both containing 0.1% trifluoroacetic acid) at a flow rate of 1 mLmin^{-1} at 40 °C. UV absorbance was detected at 200 nm.

NMR and MS analyses. The compounds, 2, 3, 5, and 6, used for NMR analyses were prepared from the in vitro reaction, which was scaled up to 10 mL. For preparation of 2, 5, and 6, the reaction mixtures were acidified with 1 mL of 1 M HCl and extracted with ethyl acetate. The organic layers were evaporated until dry. The crude materials were dissolved in a small amount of methanol and purified using a reverse-phase preparative HPLC equipped with a COSMOSIL 5C₁₈-AR-II column (20 × 250 mm; Nacalai Tesque Inc., Kyoto, Japan). The compounds were chromatographed using 10% methanol and 0.1% trifluoroacetic acid in water as a mobile phase, at a flow rate of 3 mL/min at ambient temperature. Because 3 was easily transformed into a geometical isomer under acidic conditions, the reaction containing 3 was filtrated with Amicon Ultra-0.5 Centrifugal Filters and the combined filtrate was lyophilized. The lyophilizate was used directly for NMR analyses. NMR data were collected on a Bruker AVANCE III 400 FT-NMR spectrometer (Bruker Corporation, Billerica, MA, USA). LC-ESI/HRMS analysis was carried out using a Q Exactive™ (Thermo Fisher Scientific Inc., Waltham, MA, US). For LC-ESI/HRMS analysis, a COSMOSIL 5C₁₈-PAQ column (2.0 × 150 mm; Nacalai Tesque Inc., Kyoto, Japan) was used, and analytes were eluted under the isocratic condition of 3% acetonitrile in water (both containing 0.1% formic acid) at a flow rate of 0.2 mL min⁻¹ at 40 °C.

NMR and MS data. β *-ketoadipate enol-lactone (2).*

¹H NMR (400 MHz, D₂O): δ 3.36 (d, 2H, J = 2.0 Hz, C2H₂), 3.49 (s, 2H, C3H), 5.58 (s, 1H, C5H₂). Negative mode LC-ESI/HRMS, [M – H]⁻ = 141.01835 (calculated for C₆H₅O₄⁻, 141.01933).

3-carboxy-cis,cis-muconate (3). ¹H NMR (400 MHz, CD₃OD): δ 6.01(dd, 1H, J = 0.9, 12.0 Hz, C5H), 6.50 (m, 1H, C2H), 6.77 (dd, 1H, J = 1.7, 12.0 Hz, C4H). Negative mode LC-ESI/HRMS, $[M - H]^-$ = 185.00800 (calculated for C₇H₅O₆⁻, 185.00916).

 γ -carboxymuconolactone (4). Negative mode LC-ESI/HRMS, $[M - H]^- = 185.00841$ (calculated for $C_7H_5O_6^-$, 185.00916).

muconolactone (5). ¹H NMR (400 MHz, D₂O): δ 2.71 (dd, 1H, J = 8.2, 16.6 Hz, C1'Ha), 2.95 (dd, 1H, J = 4.8, 16.6 Hz, C1'Hb), 5.57 (dddd, 1H, J = 1.4, 1.9, 4.8, 8.2 Hz, C5H), 6.24 (dd, 1H, J = 1.9, 5.8 Hz, C3H), 7.81 (dd, 1H, J = 1.4, 5.8 Hz, C4H). Negative mode LC-ESI/HRMS, $[M - H]^- = 141.01831$ (calculated for C₆H₅O₄⁻, 141.01933).

β-ketoadipate (6). ¹H NMR (400 MHz, CD₃OD): δ 2.55 (t, 2H, J = 6.5 Hz, C5H), 2.86 (t, 2H, J = 6.5 Hz, C4H), 3.39 (s, 2H, C2H₂). Negative mode LC-ESI/ HRMS, $[M - H]^- = 159.02898$ (calculated for C₆H₇O₅⁻, 159.02990).

Results

In vitro catabolism of protocatechuate by PcaHG

The first step of the catabolism of protocatechuate (1) in the β -ketoadipate pathway is a ring-opening reaction catalyzed by PcaHG.6,7) An SDS PAGE-verified pure heterodimer of PcaHG (Supplemental Fig. S1) was incubated with 1 as described in the Methods section. HPLC analysis confirmed that 1 was readily converted to the product, 3, which migrated at a retention time (RT) of 5.0 min. In control experiments with boiled, inactive PcaHG, 1 was not converted to product as evidenced by an RT of 10.2 min (Fig. 2(A)). The structure of the reaction product was confirmed by the negative mode LC-ESI/HRMS and ¹H NMR analyses (Fig. 2(B) and Supplemental Fig. S2). The unstable nature of the product hindered us from conducting ¹³C NMR and two-dimensional NMR analyses. The m/z of the product, 185.00800, was in good agreement with the calculated theoretical value, 185.00916, of the $[M - H]^{-}$ ion of 3. The geometrical isomerism of the product was characterized from the coupling constants calculated from the ¹H NMR spectrum (Supplemental Fig. S2). A coupling constant of 12.0 Hz characteristic of a cis-coupling across a double bond was observed for C4- and C5-protons (Supplemental Fig. S3), indicating that the geometrical isomerism of the olefin at C4–C5 position is in a *cis*-configuration. In addition, a long-range coupling was observed between the C2- and C5-protons, since the C5 proton appeared to be a



Fig. 2. HPLC and MS analyses of the product from PcaHG reaction.

Note: (A) HPLC chromatograms of *in vitro* reactions of PcaHG. Boiled PcaHG (*i*) and active PcaHG (*ii*) were used as enzymes. UV absorption was detected at 200 nm. (B) LC-ESI/HRMS spectrum of 3-carboxy-*cis*,*cis*-muconate (3). $[M - H]^-$, $[M - CO_2 - H]^-$, and $[M - 2CO_2 - H]^-$ ions were observed.

doublet of doublets (Supplemental Fig. S2). The geometrical isomerism of the olefin at the C2–C3 position was deduced to be in a *cis*-configuration, since the coupling constant of the dienylic coupling was calculated to be 1.7 Hz (Supplemental Fig. S3). Taken together, molecule **3** was determined to be the product of the reaction of PcaHG with **1** via an enzyme catalyzed oxidation.

In vitro catabolism of protocatechuate by PcaHG and PcaB

The second step of the β -ketoadipate pathway is a cycloisomerization of 3 catalyzed by PcaB, forming molecule 4.8) When active PcaB and PcaHG were simultaneously incubated with 1, a product was observed at an RT of 3.0 min via HPLC analysis (Fig. 3(A)). This product was not generated in the reactions when boiled PcaB or PcaHG were used instead of active enzymes (Fig. 3(A)). The m/z of the product, 185.00841, was in good agreement with the calculated theoretical value, 185.00916, of the $[M - H]^-$ ion of 4 (Fig. 3(B)). A filtrate of the reaction was incubated at 30 °C to examine whether facile decarboxylation of the product occurred (Supplemental Fig. S4). As a result, a rapid decrease in the product with a concomitant increase in 5 (see below for a structural characterization) was observed. This suggests that the product of PcaHG and PcaB reaction is 4, which then undergoes spontaneous decarboxylation to form 2. It is known that 2 and 5 exist in equilibrium,¹¹⁾ even though the



Fig. 3. HPLC and MS analyses of the product from PcaHG and PcaB reaction.

Note: (A) HPLC chromatograms of *in vitro* reactions of PcaHG and PcaB. All enzymes were active unless inactivated by boiling prior to use in assay. Boiled PcaHG and PcaB (*i*), PcaHG and boiled PcaB (*ii*), and PcaHG and PcaB (*iii*) were used as enzymes. UV absorption was detected at 200 nm. (B) LC-ESI/HRMS spectrum of γ -carboxymuconolactone (4). $[M - H]^-$, $[M - CO_2 - H]^-$, and $[M - 2CO_2 - H]^-$ ions were observed. Decarboxylation of 4 occurred during ionization resulting in the observation of both $[M - CO_2 - H]^-$ and $[M - 2CO_2 - H]^-$ as major ions.

form of **5** is more energetically favorable than **2** (Supplemental Fig. S4). Unfortunately, no NMR data of **4** were successfully obtained, because **4** proved to be extremely unstable.

The molecular formula for 5 was determined to be C₆H₆O₄ since the LC-ESI/HRMS analysis showed that the m/z of the product of decarboxylation of 4 was 141.01831-a value that was in agreement with the theoretical value, 141.01933, of the $[M-H]^-$ ion of 5 (Supplemental Fig. S5). Similarly, the molecular formula of 2 was also determined as $C_6H_6O_4$ (Supplemental Fig. S6). The ¹H NMR data of 5 were assigned as follows: the doublets at 6.24 and 7.81 ppm were assigned to the C4 and C3 protons of the dihydrofuran ring; the dddd at 5.57 ppm was assigned to the C5 proton; the doublet of doublets at 2.71 and 2.95 ppm were assigned to the germinal protons at C1', since the coupling constant between these protons was 16.6 Hz (Supplemental Fig. S7). In addition, the ¹H NMR data of 2 were consistent with the structure of β -ketoadipate enol-lactone (Supplemental Fig. S8).

In vitro catabolism of protocatechuate by PcaHG, PcaB, and PcaL

PcaL is a fusion protein composed of PcaC and PcaD that catalyzes the decarboxylation of 4 and the hydrolysis of 2, by their respective activities. When PcaL, PcaB, and PcaHG were incubated with 1, a product was observed at an RT of 3.3 min via HPLC trace (Fig. 4(A)). This product was not generated in reactions with boiled, inactive PcaL, PcaB, or PcaHG (Fig. 4(A)). Similar results were obtained from the two-step reaction in which PcaL was added to the filtrate of the reaction of PcaHG and PcaB, as seen with HPLC and LC-MS analyses (Supplemental Fig. S9). The m/z of the product, 159.02892, was in good agreement with the calculated theoretical value, 159.02990, of the $[M - H]^-$ ion of 6 (Fig. 4(B)). In addition, the ¹H NMR spectrum of **6** was identical to β -ketoadipate (Supplemental Fig. S10).²⁰⁾ Furthermore, PcaL exhibited a β-ketoadipate enol-lactone hydrolase activity with 2 in vitro, resulting in the formation of 6. HPLC analysis revealed that 5 was not utilized as a substrate in this reaction (Fig. 5). These results indicate that PcaL is a bifunctional enzyme, functioning as both γ -carboxy-muconolactone decarboxylase and β -ketoadipate enol-lactone hydrolase.

Discussion

(A)

0

(B)

4

Time (min)

Although a complete picture of the β -ketoadipate pathway was depicted by the Ornston group in the 1970s,⁴⁾ chromatographic, spectroscopic, and mass spectrometric studies of the intermediates in this

1 iv

8

[M-H]

159.02892

iii

ii

i

12

170



m/z

145

Note: (A) HPLC chromatograms of *in vitro* reactions of PcaHG, PcaB, and PcaL. All enzymes were active unless inactivated by boiling prior to use in assay. Boiled PcaHG, PcaB, and PcaL (*i*), PcaHG, boiled PcaB, and PcaL (*ii*), PcaHG, PcaB, and boiled PcaL (*iii*), PcaHG, PcaB, and PcaL (*iv*) were used as enzymes. UV absorption was detected at 200 nm. (B) LC-ESI/HRMS spectrum of β -ketoadipate (6). [M – H]⁻ ion was observed.



Fig. 5. HPLC chromatograms of *in vitro* reaction of PcaL. Note: Reactions of boiled PcaL and 5 (*i*), active PcaL and 5 (*ii*), boiled PcaL and 2 (*iii*), and active PcaL and 2 (*iv*).

pathway have not been thoroughly investigated. Although Ravi et al. reported the proton NMR spectrum of **3**, they did not observe the long-range couplings of **3**.⁵⁾ On the other hand, Ainsworth et al. reported the coupling constants of the dienylic (J = 0.9 Hz) and allylic (J = 2.1 Hz) couplings of **3**.²¹⁾ They explained that these "abnormal" coupling constants can be attributed to the nonplanarity of **3**. However, we have observed "normal" coupling constants for the dienylic and allylic couplings as 1.7 and 0.9 Hz, respectively (Supplemental Fig. S4). We suppose that the rapid sample handling allowed us to analyze the intact **3**.

In this study, enzymes of the protocatechuate branch of the β-ketoadipate pathway from R. jostii, whose functions were predicted based on sequence similarity, were reconstituted in vitro. We accomplished the structural determination of the reaction products of PcaHG, PcaB, and PcaL using HPLC, LC-MS, and NMR, leading to a biochemical characterization of these enzymes. PcaHG catalyzes the oxidative cleavage of the aromatic ring of protocatechuate to yield 3. Following cycloisomerization by PcaB, the formation of the lactone ring gives 4. PcaL, a bidomain protein consisting of PcaC and PcaD activities, catalyzes the sequential decarboxylation and hydrolysis reactions of 4 that result in the formation 6. The Pca enzymes of R. jostii share moderate sequence similarity with the enzymes whose catalytic properties were characterized biochemically. For example, PcaG is 42% identical in amino acid sequence to the alpha chain of protocatechuate 3,4-dioxygenase⁷⁾ of *P. aeruginosa*. Similarly, PcaH is 50% identical to the beta chain of protocatechuate 3,4-dioxygenase⁷⁾ of *P. aeruginosa*. In addition, PcaB is 41% identical to 3-carboxy-cis,cismuconate lactonizing enzyme⁸⁾ of *P. putida*. While the N-terminus (residues 1-261) of PcaL is 45% identical to β -ketoadipate enol-lactone hydrolase¹¹⁾ of *P. putida*, the C-terminus (residues 272-400) of PcaL is 49% identical to γ -carboxy-muconolactone decarboxylase¹⁰⁾ of *P*. putida. Therefore, it is not surprising that the catalytic reactions of the Pca enzymes of R. jostii were identical to those of the homologous enzymes of Pseudomonas species. To our knowledge, this is the first report

demonstrating the *in vitro* reactions of the Pca enzymes from actinomycetes.

In the course of our study, 4 was found to be an extremely unstable compound that was readily transformed to 5 via spontaneous decarboxylation and isomerization. It is possible that the bifunctional PcaL is more advantageous than the separate enzymes of PcaC and PcaD in carbon and energy acquisition due to the necessity of PcaC to release intermediate 2, allowing it to equilibrate with 5, and resulting in the loss of the carbon source from the pathway. We also found that 3 was an unstable compound and was readily transformed into 3-carboxy-trans, trans-muconate (results not shown). There may be unidentified enzymes that scavenge these side products and bring them back to the β -ketoadipate pathway to recover the invested energy. Recent progress in genome sequencing has revealed that the β -ketoadipate pathway genes are distributed widely among microbial taxa. This work provides a basis for analyzing the catalytic properties of enzymes involved in the growing number of the β -ketoadipate pathway genes deposited in the seemingly continuously updated genetic databases.

Supplemental material

The supplementary material for this paper is available at http://dx.doi.org/10.1080/09168451.2014.993915.

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