SYNTHESIS OF TERTIARY ALKOXY- AND ALKYLTHIOENEDIYNIC ALCOHOLS

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The discovery of new highly effective plant growth stimulators among the unsaturated cycloaliphatic ketols and ketones [1, 2] gives a basis for broad investigations of these compounds. The study of the relationship between their chemical structures and physiological action, and also the location of the active center of the molecule which is characteristic for this structure, is acquiring particular importance. In recent years a large number of polyacetylenic compounds has been isolated from various plants [3] from which it follows that the typical bond apparently takes part in physiological processes occurring in the plant organism. Since some of the alkoxy- and alkylthioenynic alcohols that we have synthesized previously exhibit growth-regulating activity, it appeared desirable to study the change in this activity in dependence on the number of conjugated triple bonds in the molecule. For this purpose we have undertaken the synthesis of alkoxy- and alkylthioenediynic alcohols (I) from ethynyl and vinyl ethers and sulfides (II) by their unsymmetrical condensation with bromo-substituted acetylenic alcohols (III) by Chodkiewiez' method

$$\begin{array}{c} R_1 \\ CC \equiv CBr + HC \equiv CCH = CH - XR_3 \xrightarrow{CuCl} \\ R_2 \xrightarrow{OH} \\ (III) \end{array} \qquad \begin{array}{c} R_1 \\ CC \equiv CC \equiv CCH = CH - XR_3 \\ R_2 \xrightarrow{OH} \\ (I) \end{array}$$

where $R_1 = R_2 = CH_3$, $XR_3 = OCH_3$ (a); $R_1 + R_2 = (CH_2)_5$, $XR_3 = OCH_3$ (b); $R_1 + R_2 = (CH_2)_5$, $XR_3 = OC_4H_9$ (c); $R_1 = R_2 = CH_3$, $XR_3 = SC_2H_5$ (d); $R_1 + R_2 = (CH_2)_5$, $XR_3 = SC_2H_5$ (e).

It was found that this reaction readily takes place in the presence of cuprous chloride, and the tertiary enediynic alcohols (I) formed were isolated with good yields. However, the analogous reaction with 1-diethylaminobut-1-ene-3-yne did not lead to the desired result, and only the corresponding acetylenic glycols, formed as the result of the oxidative condensation of the bromo-substituted acetylenic alcohols (III), were isolated from the reaction mixture.

The alkoxyenediynic alcohols (Ia-c) are viscous liquids which are unstable in the pure state but can be stored satisfactorily in solution. The alkylthioenediynic alcohols (Id, e) are crystalline and completely stable substances. To confirm their structure, the alkoxyenediynic alcohols (Ia-c) were subjected to exhaustive hydrogenation with 5% Pd/CaCO₃ to form the saturated alcohols (IVa-c) (Table 1). The structure of all the compounds isolated was also confirmed by their IR and UV spectra.

EXPERIMENTAL

The bromine-substituted acetylenic alcohols (III) were obtained by treating the alcohols with a solution of potassium hypobromite [7].

Synthesis of the Alkoxy- and Alkylthioenediynic Alcohols (I). With stirring, to 25 ml of a 40% aqueous solution of ethylamine containing 1 g of CuCl and 0.7 g of hydroxylamine hydrochloride was added a solution of 0.1 mole of the appropriate ethynyl vinyl ether or sulfide (II) in 20 ml of methanol and then, in drops, a solution of 0.1 mole of the bromine-substituted acetylenic alcohol (III) in 25 ml of methanol in such a way that the temperature of the reaction mixture did not exceed 35°. The mixture was stirred for 2 hr at 20° and was then treated with 10% HCl and extracted with ether; the extract was twice washed with dilute HCl (1:5) and with water and was dried over MgSO₄, and then the ether was driven

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TABLE 1. Characteristics of the Compounds Synthesized

Calculated, % IR spectrum, c H v, cm ⁻¹		7,37 1280, 1628, 2136, 2220, 3330—3420	7,89 1270, 1620, 2120, 2200, 3320—3380	9,00 1278, 1625, 2138, 2225, 3350—3430	7,27 1270 1550, 1615, 2140, 2225, 3360—3430	7,74 1276, 1560, 1615, 2150, 2220, 3320—3360	12,72 1125, 3380—3460	12,23 1125, 34203480	12,58 1125, 3400—3480
Calc	υ	73,14	76,44	78,01	68,02	71,17	68,91	72,84	74,94
Empirica l formula		C10H12O2	C ₁₃ H ₁₆ O ₂	C16 H22 O2	C11H14OS	C14 H18 OS	$C_{10}H_{22}O_{2}$	C ₁₃ H ₂₆ O ₂	C16 H 82 O2
Found, %	# 	7,44	7,93	8,86 9,02	7,57	7,67	12,50 12,40	12,30 12,10	12,62
	O	73,27 73,09	76,63 76,65	78,10 78,02	68,08 68,08	72,02	69,00 68,96	73,02 73,27	75,02
Viold 0	of theo- retical	80	80	82	87	94	86	98	96
	02u	1,5690	1,5680	1,5652	1		1,4380	1,4688	1,4650
	B.p., °C (p, mm Hg)	117(1)	130 (0,05)	172 (1,5)	M.p.36—37 (n-Hexane)	M.p.70-72 (n-Hexane)	84 (1,5)	121,5(2)	130—132
Formula		CH, C—C=CC=CCH~CH—OCH, CH, OH	C=CC=CCH-CH-OCH,	C≡CC≡CCH=CH−OC,H,	CH-SC,H,SC,H,SC,H,SC,H,SC,H,SC,H,SC,H,SC	$\left(\begin{array}{c} \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\$	C-(CH ₂),-OCH ₂ CH ₃ OH	(CH ₃) ₄ -OCH ₃	(CH ₂),—OC,H,
Com- pound		Ia	el el	* oI	Idī	Te‡	IVa	IVb	Ivc

* UV spectrum: λ_{max} 223, 262, 276, 291 m μ (ϵ 16,600, 7000, 9600, 8000) (alcohol). † Found %: S 16.51, 16.48. Calculated %: S 16.47. UV spectrum: λ_{max} 222, 233, 295, 307 m μ (ϵ 8200, 7000, 15,000, 18,000) (alcohol). † Found %: S 13.48, 13.75. Calculated %: S 13.66. UV spectrum: λ_{max} 224, 234, 298, 307 m μ (ϵ 7800, 6800, 15,000, 17,400) (alcohol).

off and the enynic alcohols (I) were isolated by distillation or crystallization. The constants of the compounds synthesized and also the results of elementary analysis and the IR and UV spectra are given in Table 1.

CONCLUSIONS

A number of alkoxy- and alkylthioenediynic alcohols have been synthesized to study the growth-regulating activity of polyacetylenic compounds.

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